

Interaction of Nanoparticles with Model Cell Membranes: Implications for Nanoparticle Toxicity

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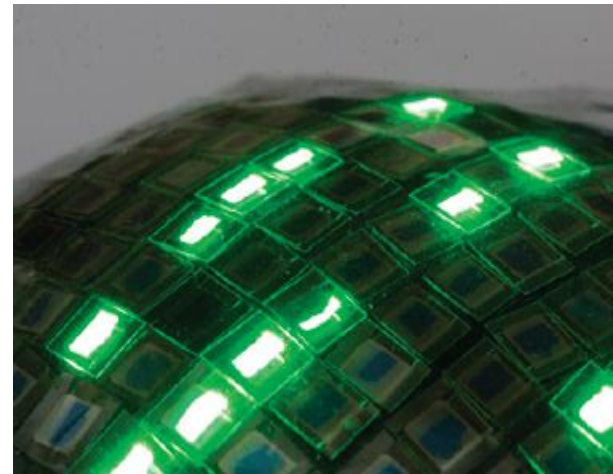
Nanoparticle Applications in Semiconductor Industry

- CeO_2 , Al_2O_3 , and SiO_2 nanoparticles are employed in chemical mechanical planarization



<http://www.levitronix.com/cmp-slurry.html>

- Carbon nanotubes (CNTs) are employed in organic LED



Sekitani et al., *Nature Materials*, **2009**, 494-499

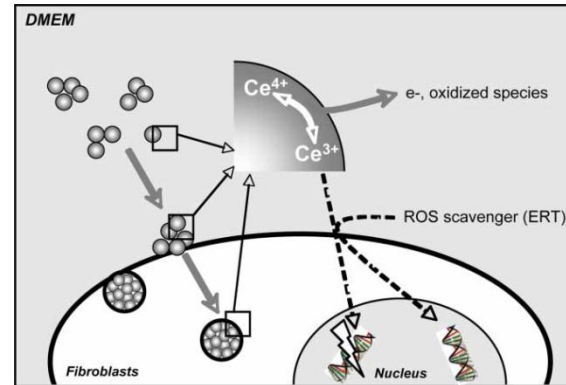
Toxicity of Nanoparticles

- CeO_2 and SiO_2 nanoparticles damage the membranes and DNA of human cells

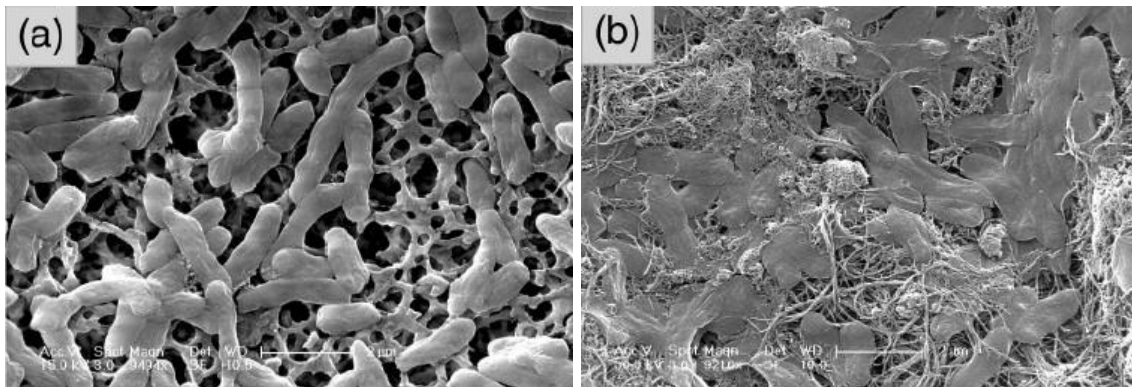
Lin et al., *International Journal of Toxicology*, **2006**, 451-457

Lin et al., *Toxicology and Applied Pharmacology*, **2006**, 252-259

Auffan et al., *Nanotoxicology*, **2009**, 161-171



- CNTs damage bacterial membranes and inhibit bacterial growth



Kang et al.,
Langmuir
2007, 8670-
8673

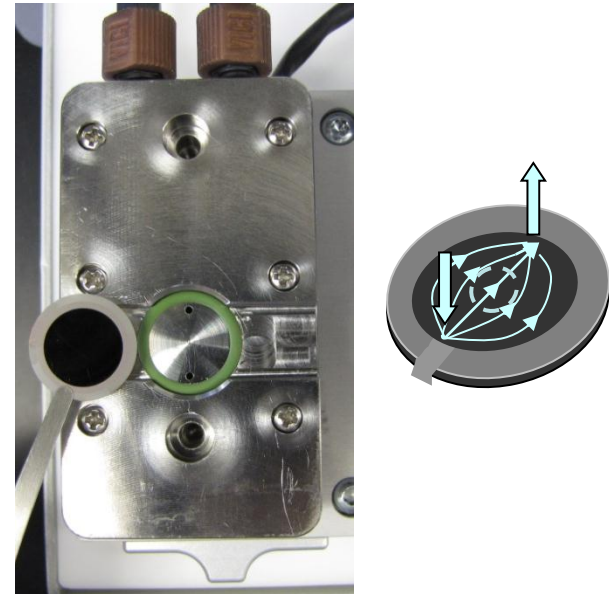
Objective

To investigate the adsorption and desorption behavior of nanoparticles on model cell membranes

Quartz Crystal Microbalance with Dissipation Monitoring (QCM-D)



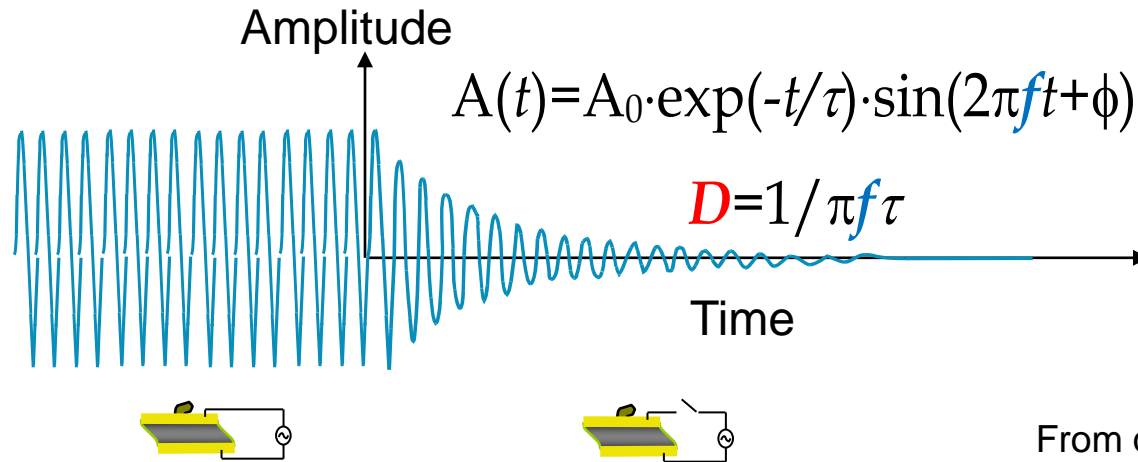
www.qsense.com



- Widely used in cell and molecular biology (e.g., interaction of biomacromolecules with model cell membranes)
- An emerging technique for studying the interaction of nanoparticles with environmental surfaces

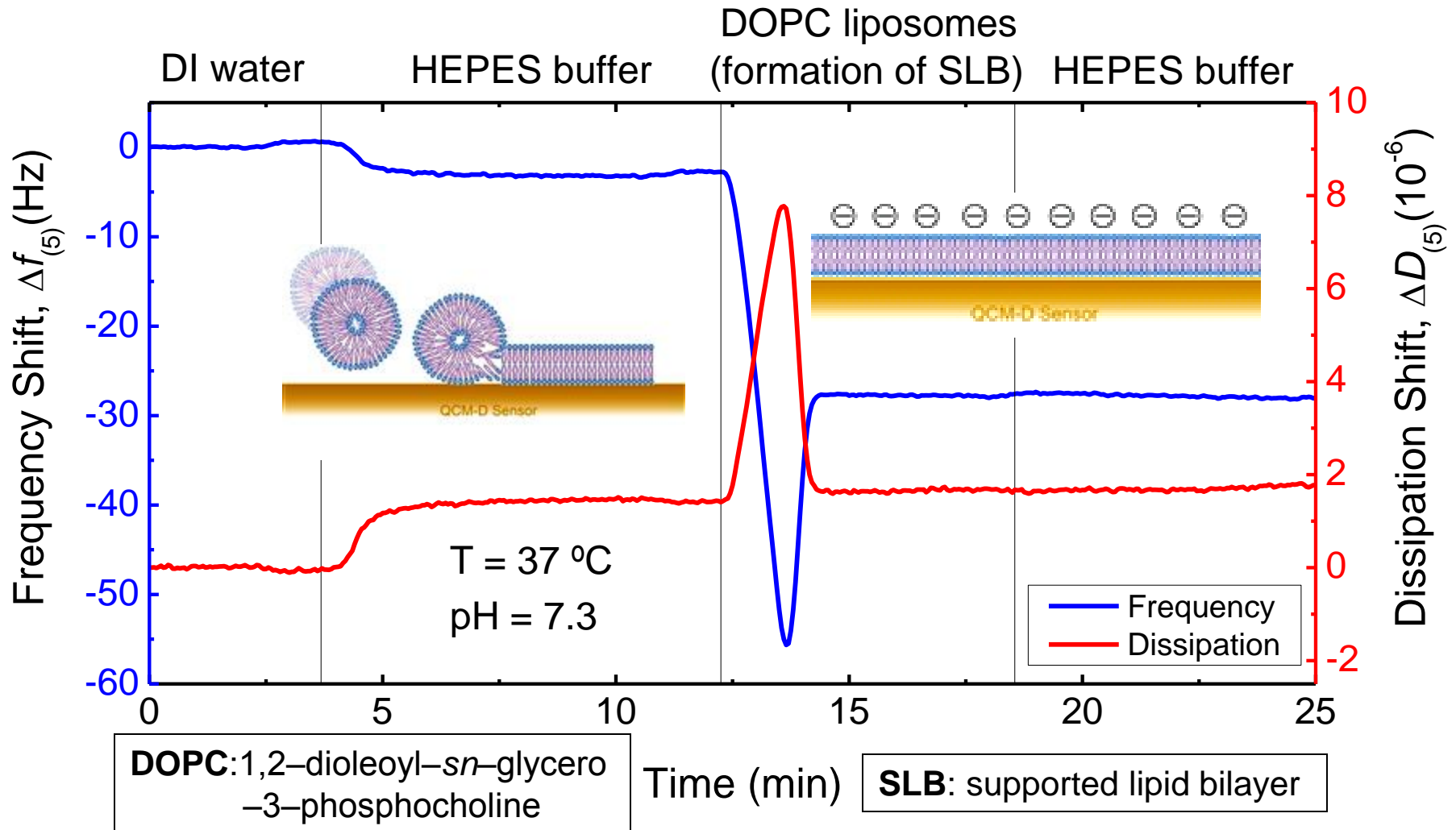
Principle of QCM-D

Frequency and Dissipation



- Generally, the frequency decreases as the deposited mass on the crystal increases; the dissipation increases as the softness of the deposited layer increases
- The mass of the deposited layer can be derived from Voigt-based modeling

Formation of Model Cell Membranes on QCM-D Crystals

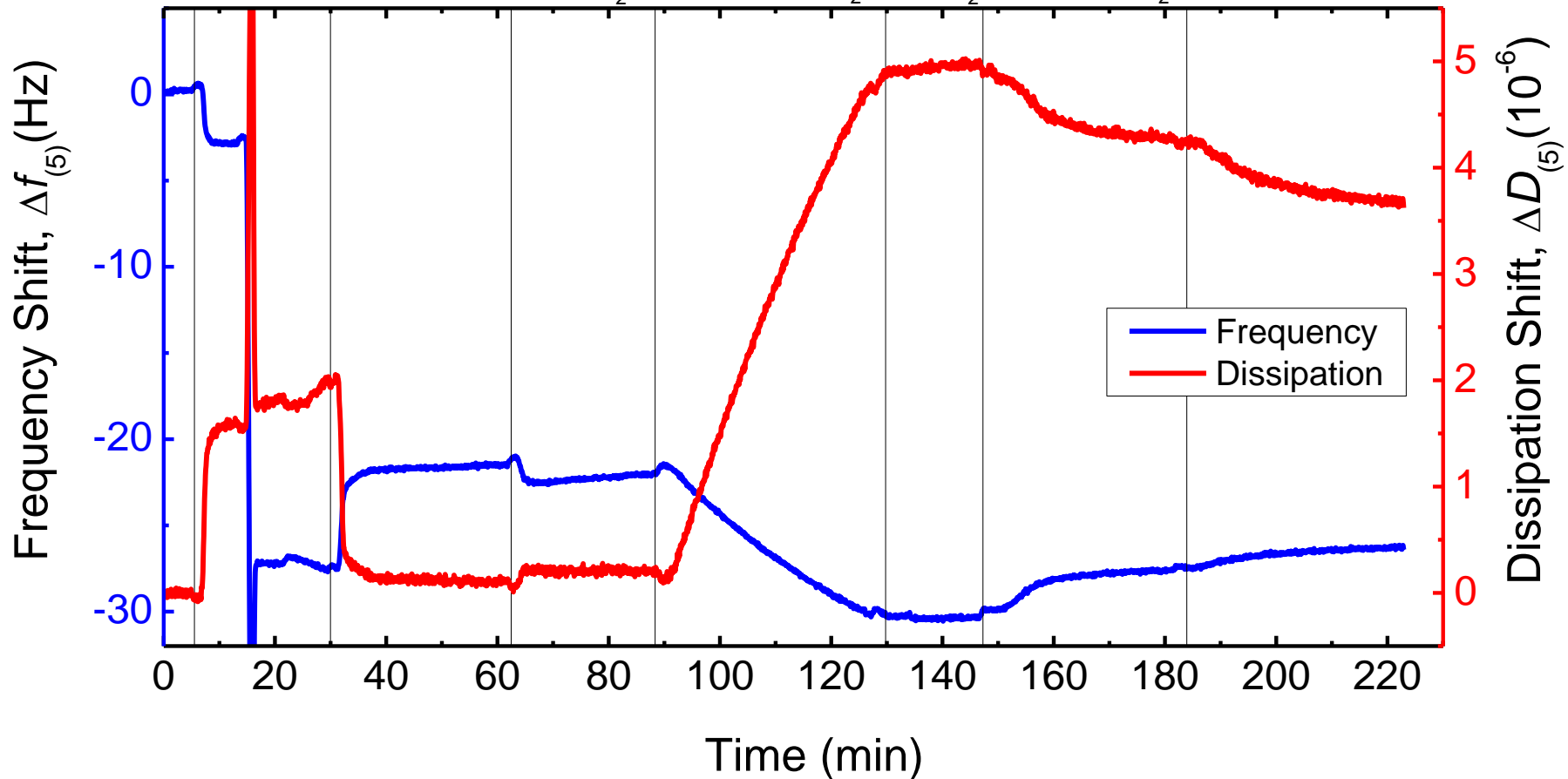


SRC Engineering Research Center for Environmentally Benign Semiconductor Manufacturing

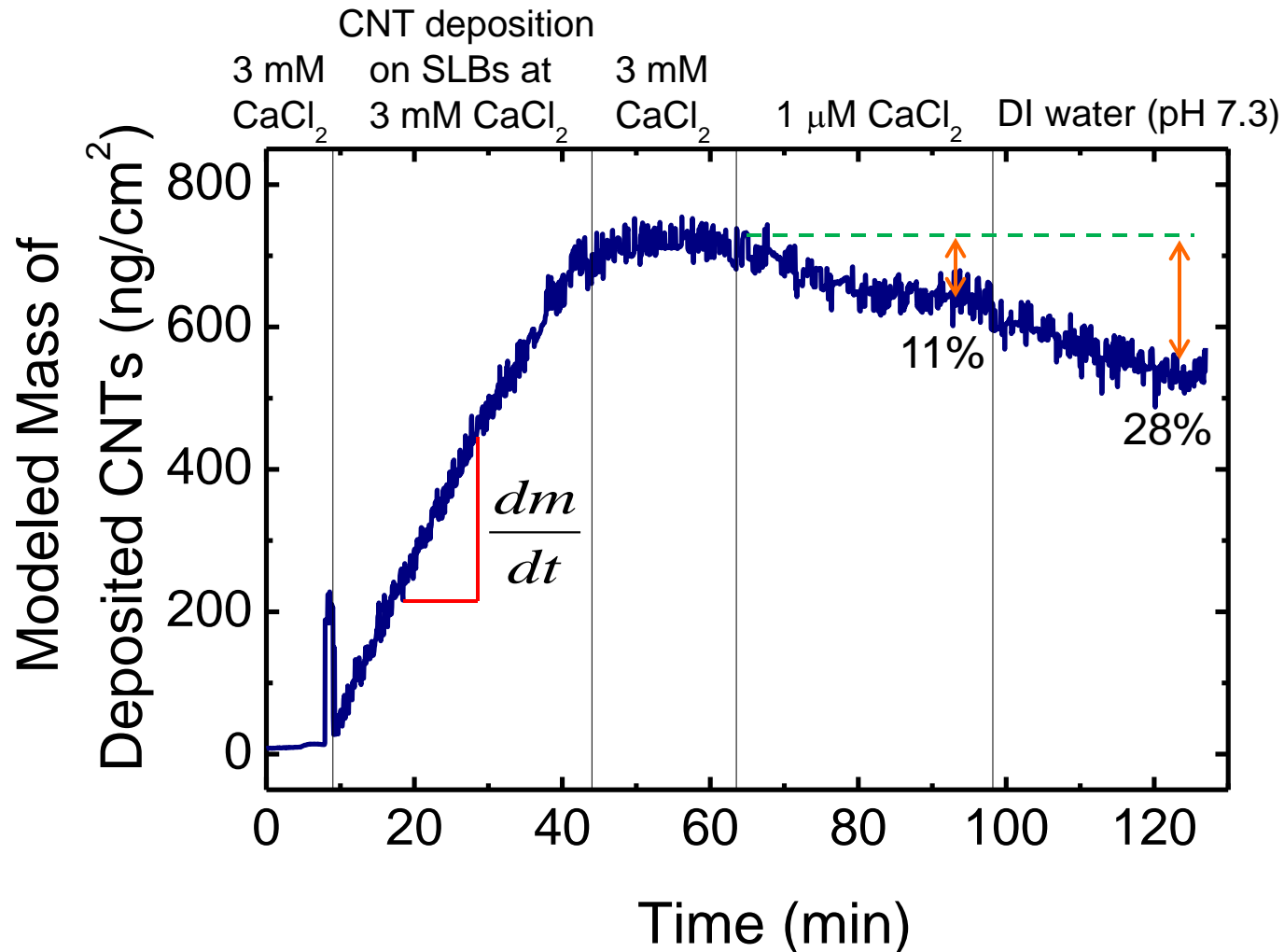
Adsorption and Desorption of CNTs on Model Cell Membranes

CNT deposition

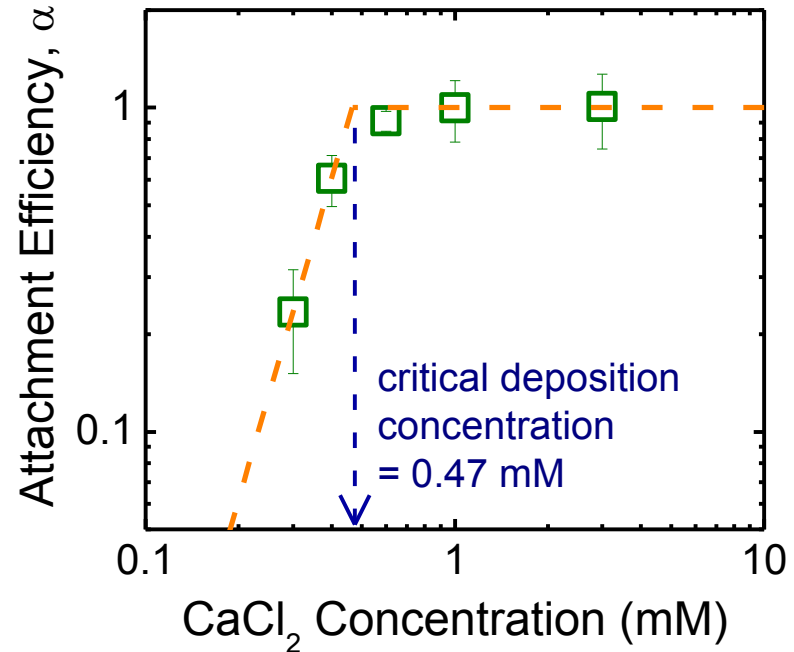
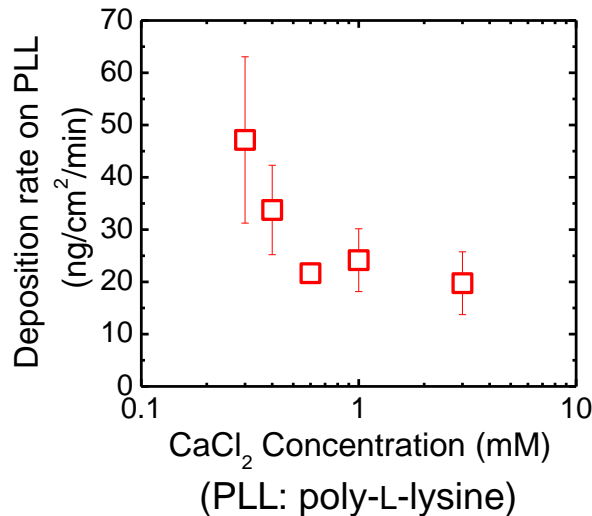
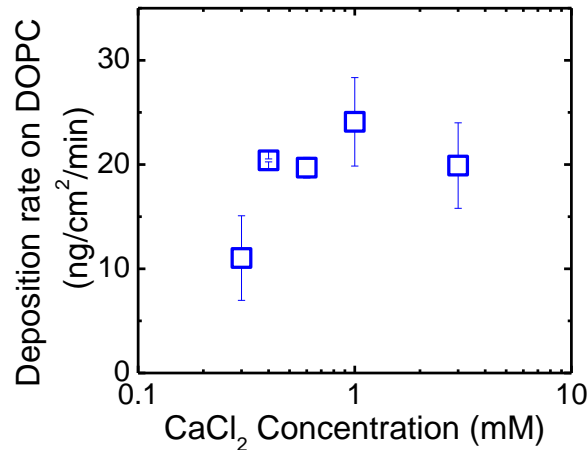
Formation of DOPC SLBs DI water (pH 7.3) 3 mM CaCl_2 on SLBs at 3 mM CaCl_2 3 mM CaCl_2 1 μM CaCl_2 DI water (pH 7.3)



Adsorption and Desorption of CNTs on Model Cell Membranes



Influence of Solution Chemistry on the Adsorption of CNTs on DOPC SLBs



$$\alpha = \frac{dm / dt}{(dm / dt)_{fav}}$$