

# Copper Recovery & Nanoparticle Removal from CMP Waste Water

## PIs:

*(Based on Previous Custom Project)*

- James Farrell, Chemical and Environmental Engineering, UA
- James C. Baygents, Chemical and Environmental Engineering, UA

## Graduate Students:

- Jake Davis, PhD candidate, Chemical and Environmental Engineering, UA
- David Hubler, PhD candidate, Chemical and Environmental Engineering, UA

## Undergraduate Students:

- Melody Hendricks, Chemical and Environmental Engineering, UA
- Dave Niselson, Chemical and Environmental Engineering, UA
- George Chac, Chemical and Environmental Engineering, UA

## Cost Share (other than core ERC funding):

- Intel Funding (\$60k)
- University of Arizona Funding (\$37.5k)
- Science Foundation Arizona Fellowship (Jake Davis, \$46k)
- GEP Smith Fellowship & the Triffet Prize (David Hubler, \$44k)

# Objectives

- **Validate a novel integrated CMP waste water treatment & reclamation process, building on results from a previous custom project, for:**
  - **removal of silica nanoparticle from CMP waste water**
  - **recovery of copper from CMP waste water**
  - **reclamation of CMP waste water for reuse as fab feed supply**
- **Develop best operating conditions for integrated process units**
- **Pilot test integrated process for CMP waste water treatment & reclamation**
- **Develop clean & cost effective technology for point of use acid (pH~1.5) & base (pH~12.5) generation using electrochemistry**

# ITRS ESH Metrics

## ITRS ESH Chemicals and Materials Management Technology

### Requirements—Near-term Years

- **Nanomaterials (Critical):** Conduct risk assessment by 2012-2017
- **Copper (Important):** 95% of copper recovered by 2014-2017

## ITRS ESH Process and Equipment Management Technology

### Requirements—Near-term Years

- **Chemicals (Important):** Maintain or improve chemical utilization\*; characterize process emissions & byproducts; improve PCU by 10% by 2012-2017
- **Water and other utilities (Important):** Optimize consumption. Reduce water and utilities requirements

## ITRS ESH Facilities Technology Requirements—Near-term Years

- **UPW recycled/reclaimed (% of 2007 use) (Important):** 60% by 2016-2017

# ITRS ESH Impact

## ITRS ESH Chemicals and Materials Management Technology Requirements—

### Near-term Years

- **Nanomaterials:** Eliminate the emission of silica nanoparticles in CMP waste water
- **Copper:** Recover copper from CMP waste water

## ITRS ESH Process & Equipment Management Technology Requirements—

### Near-term Years

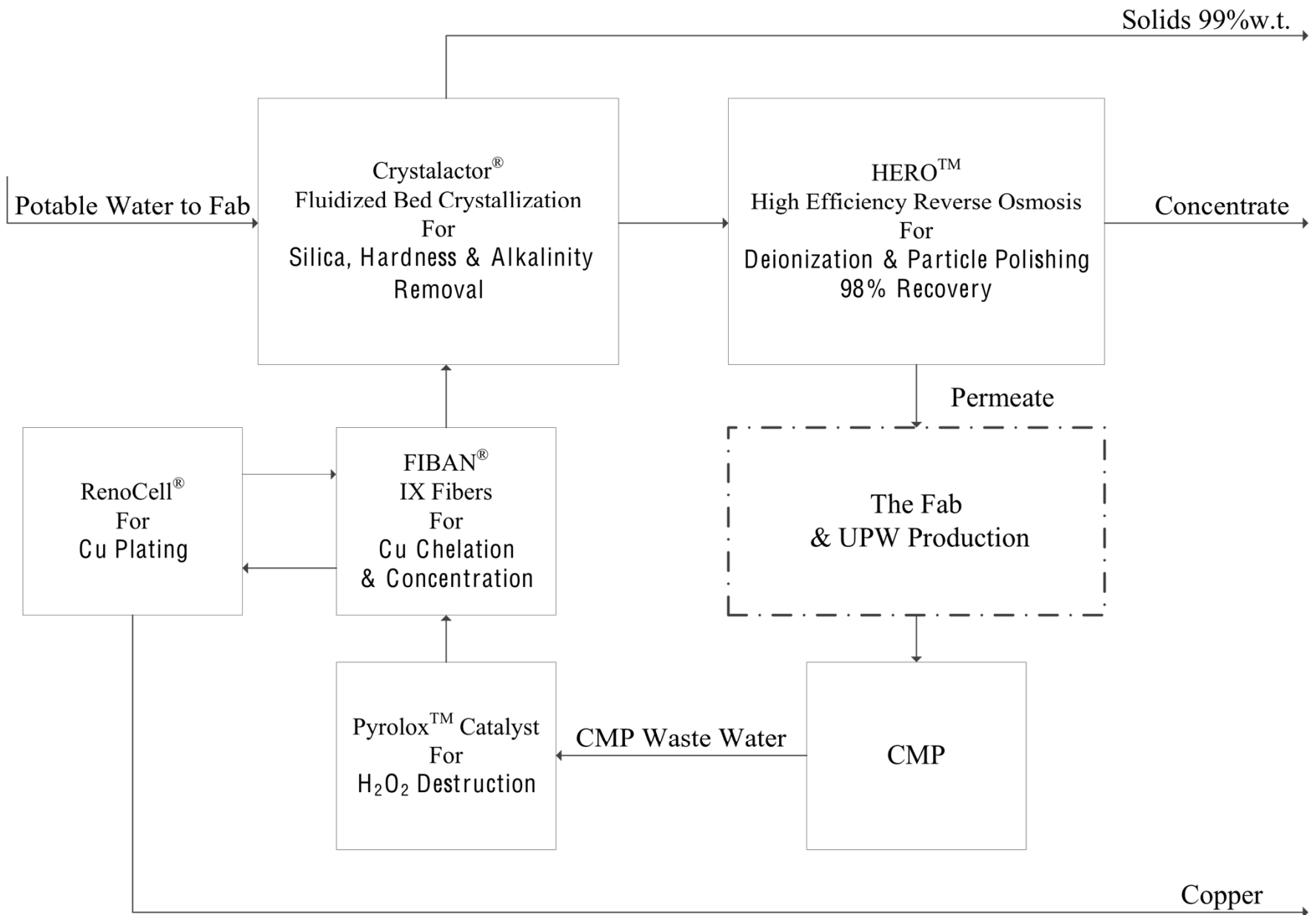
- **Chemicals:** Reduce strong acid use for primary IX regen. by 79%  
[Savings of \$0.213 per kgal treated (acid only, \$0.15 per liter of 38% HCl, assume 50ppm CaCO<sub>3</sub> in FCB effluent)]
- **Chemicals:** Reduction in emission of primary ion exchange regeneration brines by 79%
- **Water and other utilities:** Reduction in the amount of water used for regeneration of primary ion exchange resin by 79%

## ITRS ESH Facilities Technology Requirements—Near-term Years

- **UPW recycled/reclaimed:** Reduction in the use of potable water by reclaiming 98% of CMP waste water (CMP accounts for ~30% of the water used during IC fabrication)

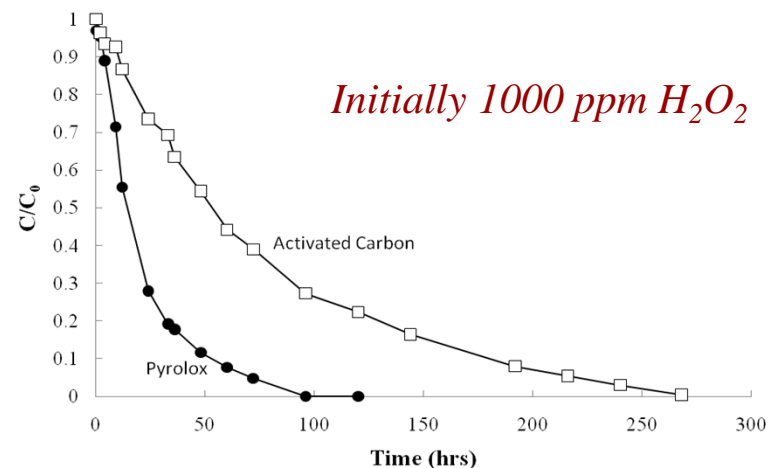
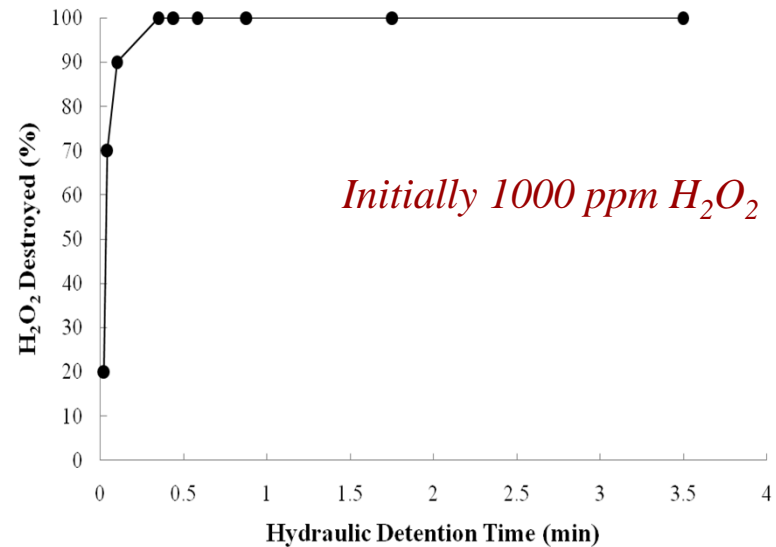
# Key & Distinguishing Features

- 1) *Cost effective copper recovery using RenoCell<sup>®</sup> technology*
- 2) *Novel use of Pyrolox<sup>™</sup> technology as a catalyst for H<sub>2</sub>O<sub>2</sub> destruction*
- 3) *Exploitation of new FIBAN<sup>®</sup> fibrous ion exchange technology to eliminate the need for expensive filtration or off-site regeneration of IX resin*
- 1) *Application of Crystalactor<sup>®</sup> & HERO<sup>™</sup> technology to CMP waste water to:*
  - *Eliminate silica nanoparticles*
  - *Eliminate chelating agents & corrosion inhibitors*
  - *Eliminate UPW primary ion exchange for control of:*
    - *Hardness*
    - *Alkalinity*
  - *Reclamation of CMP waste water for fab feed water*



# Catalytic Degradation of $H_2O_2$ with Pyrolox<sup>TM</sup>

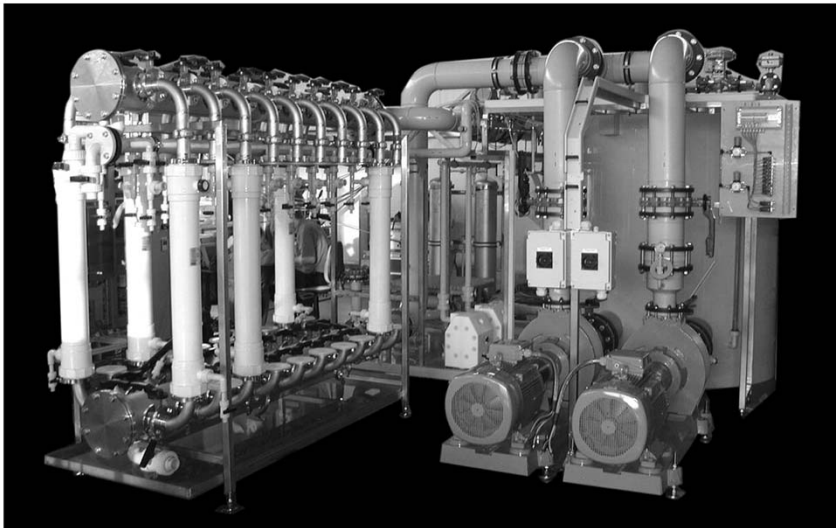
- *Pyrolox<sup>TM</sup> ( $MnO_2$ ) is granular media commonly used in water treatment*
- *Pyrolox<sup>TM</sup> has proven an effective catalyst for hydrogen peroxide destruction*
- *Pyrolox<sup>TM</sup> is better than GAC*
  - ✓ *Pyrolox<sup>TM</sup> is not fouled by chelators*
  - ✓ *Pyrolox<sup>TM</sup> has faster kinetics*
  - ✓ *Does not degrade*



# Current Copper Recovery Technologies

## Pall Microza<sup>1</sup> Filtration

*Actual Capital Cost Data: \$1.5 – \$2.5 Million*  
*Actual CoO Data: \$1.90-\$3.10 per kgal treated*  
*(excl. Capital & Labor Costs)*



## Siemens Copper Select™

*Requires transport of copper laden ion exchange resin off site for regeneration*



<sup>1</sup> *Microza is a trademark of Asahi Kasei Corporation*

SRC/SEMATECH Engineering Research Center for Environmentally Benign Semiconductor Manufacturing

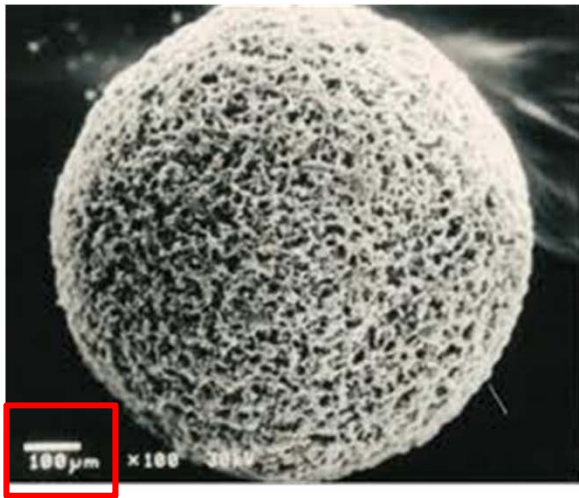


# FIBAN<sup>®</sup> Fibrous Ion Exchange Technology

about 0.65 mm



Porous IX Bead



○ Most IX sites in IX bead interior

○ Silica nanoparticles dissolve and foul bead during normal regeneration

○ IX beads require *off-site* regeneration

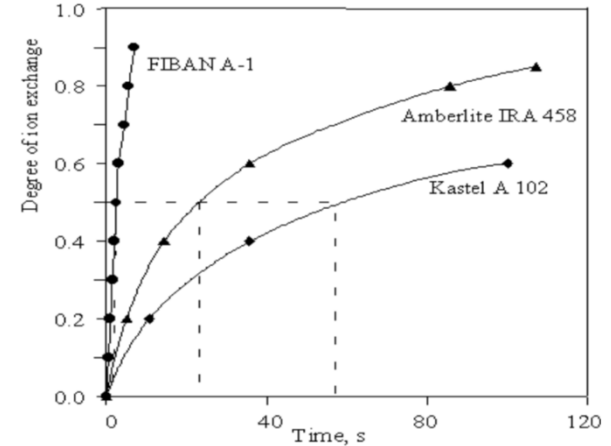
FIBAN<sup>®</sup> Filaments



✓ Most IX sites on FIBAN<sup>®</sup> surface

✓ FIBAN<sup>®</sup> can be regenerated at pH~1.5

✓ FIBAN<sup>®</sup> can be regenerated *on-site*



# Copper Recovery with RenoCell® Technology

*Regeneration of 1 m<sup>3</sup> of 50 mg-Cu per g-resin  
Recover \$320 of copper (\$4/lb)  
Costs \$8.40 (\$0.10/kWh)*



**RenoCell®**

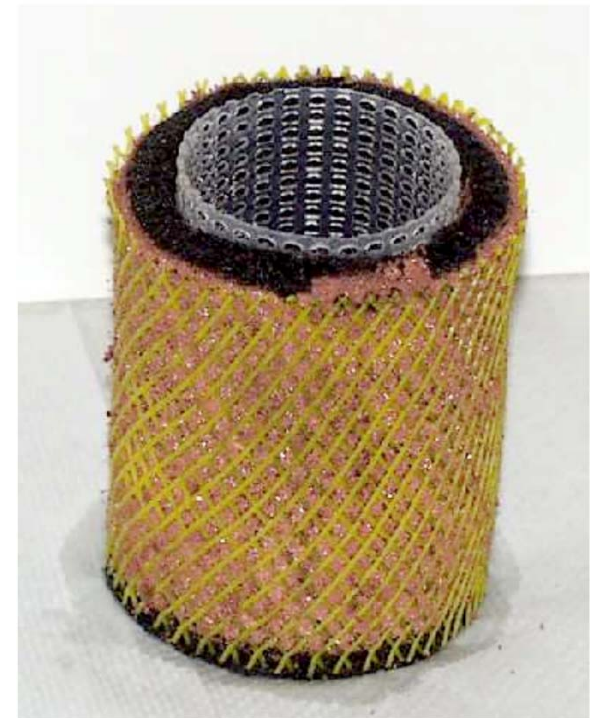


*Electroplating copper  
with  
zero acid consumption*

*Acid generation at anode  
Cu deposition on cathode*

*Throwaway carbon felt  
cathode (left)*

*Copper laden  
carbon felt cathode  
from bench reactor (right)*



# Fluidized Bed Crystallization *with Proven* DHV Crystalactor® Technology

*FCB is common practice in the Netherlands WTP  
(Waternet) since the late 1980s*

*The fluidized bed reactor is partly filled with seeding  
material (dia. 0.2-0.4 mm, garnet)*

*The seeding material in a fluidized condition*

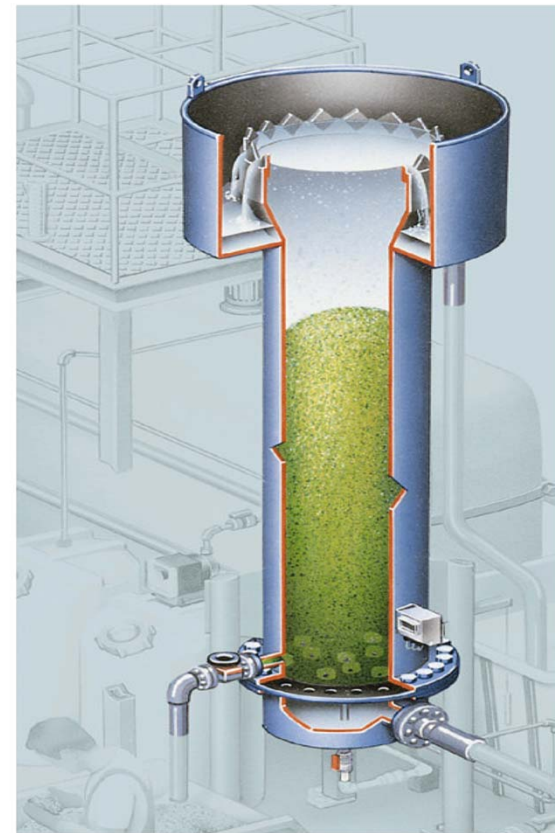
*In the bottom of the reactor, chemicals are dosed (caustic  
soda, soda ash or lime)*

*Carbonate & hydroxide minerals rapidly crystallize on the  
seeding material forming pellets*

*Pellets are removed from reactor bottom & can be re-  
used in industry*

*Dry bottom product, solids content of around 99%w.t.*

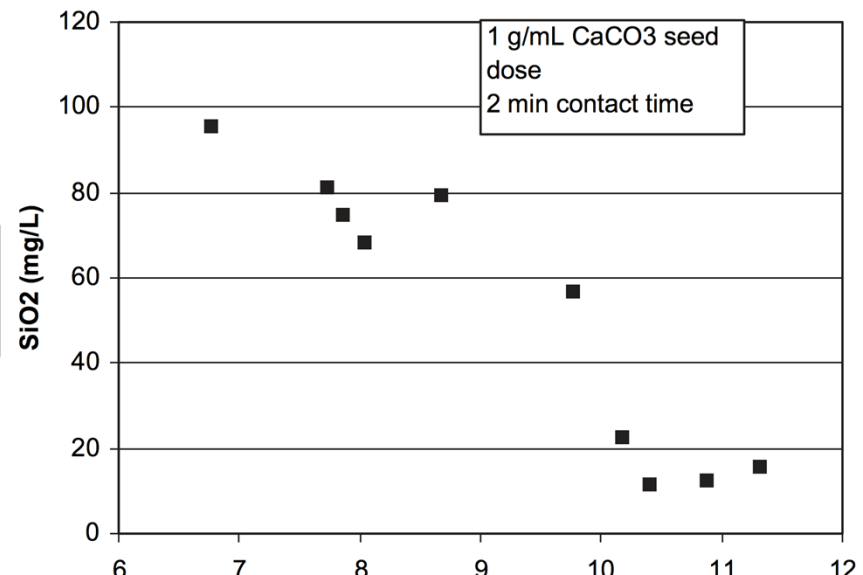
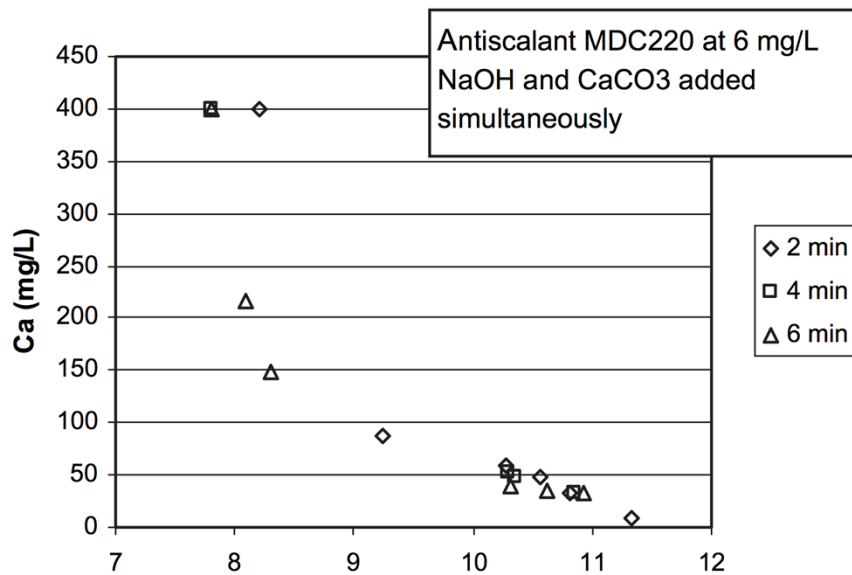
## DHV Crystalactor®



Dijk, J.C. and Wilms, D.A., Water treatment without waste material-fundamentals and state of the art of pellet softening, [J Water SRT-Aqua](#), 1991, 40, 263-280

[SRC/SEMATECH Engineering Research Center for Environmentally Benign Semiconductor Manufacturing](#)

Fluidized Bed Crystallization Investigation Funded By:  
City of Phoenix Water Services Department;  
California Energy Commission;  
AWWA Research Foundation;



***\$0.049 per kgal treated<sup>3)</sup> – FCB cost (including seed, base, & base neutralization)***

***\$0.213 per kgal treated – IX regeneration costs avoided with FCB***

***(acid only, \$0.15 per liter of 38%w.t. HCl, assuming 50ppm CaCO<sub>3</sub> in FCB effluent )***

1) Snyder, S.A., Inland Desalination: Current Practices, Environmental Implications, & Case Studies in Las Vegas, NV, *Sustain. Sci. and Eng.*, 2010, 2, 327-350

2) Bond, R. and Veerapaneni, S.V., Zero Liquid Discharge for Inland Desalination, *AWWA*, 2007

3) van Schagen, K., Rietveld, L., Babuska, R., Baars, E., Control of the fluidised bed in the pellet softening process, *Chem. Eng. Science*, 2008, 63, 1390-1400

# HERO<sup>®</sup> High Efficiency Reverse Osmosis

HERO<sup>®</sup> is a technology employed in semiconductor manufacture

*HERO<sup>®</sup> units are fed pH ~ 9 water*

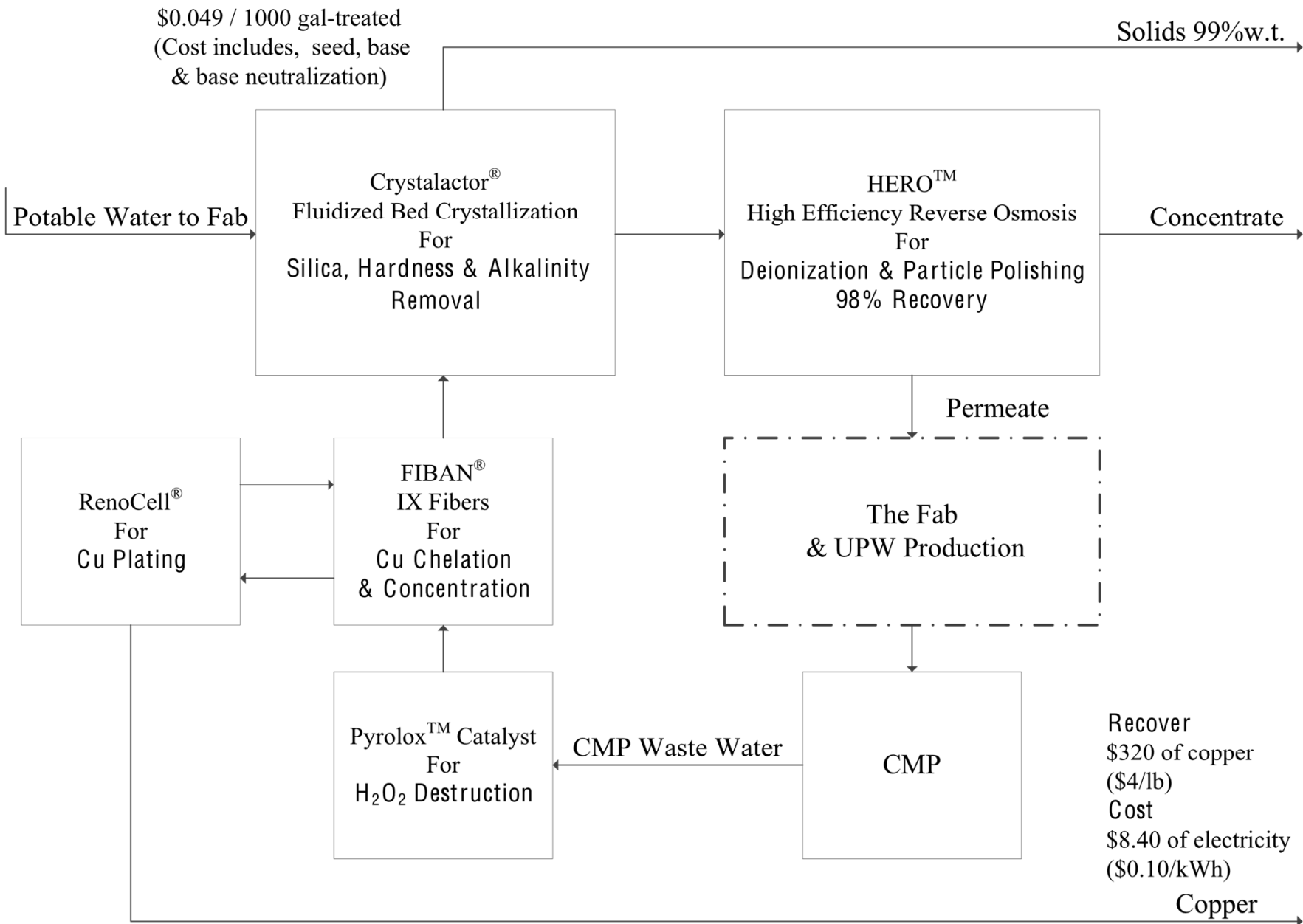
*Base accumulates on the concentrate side of the RO membrane*

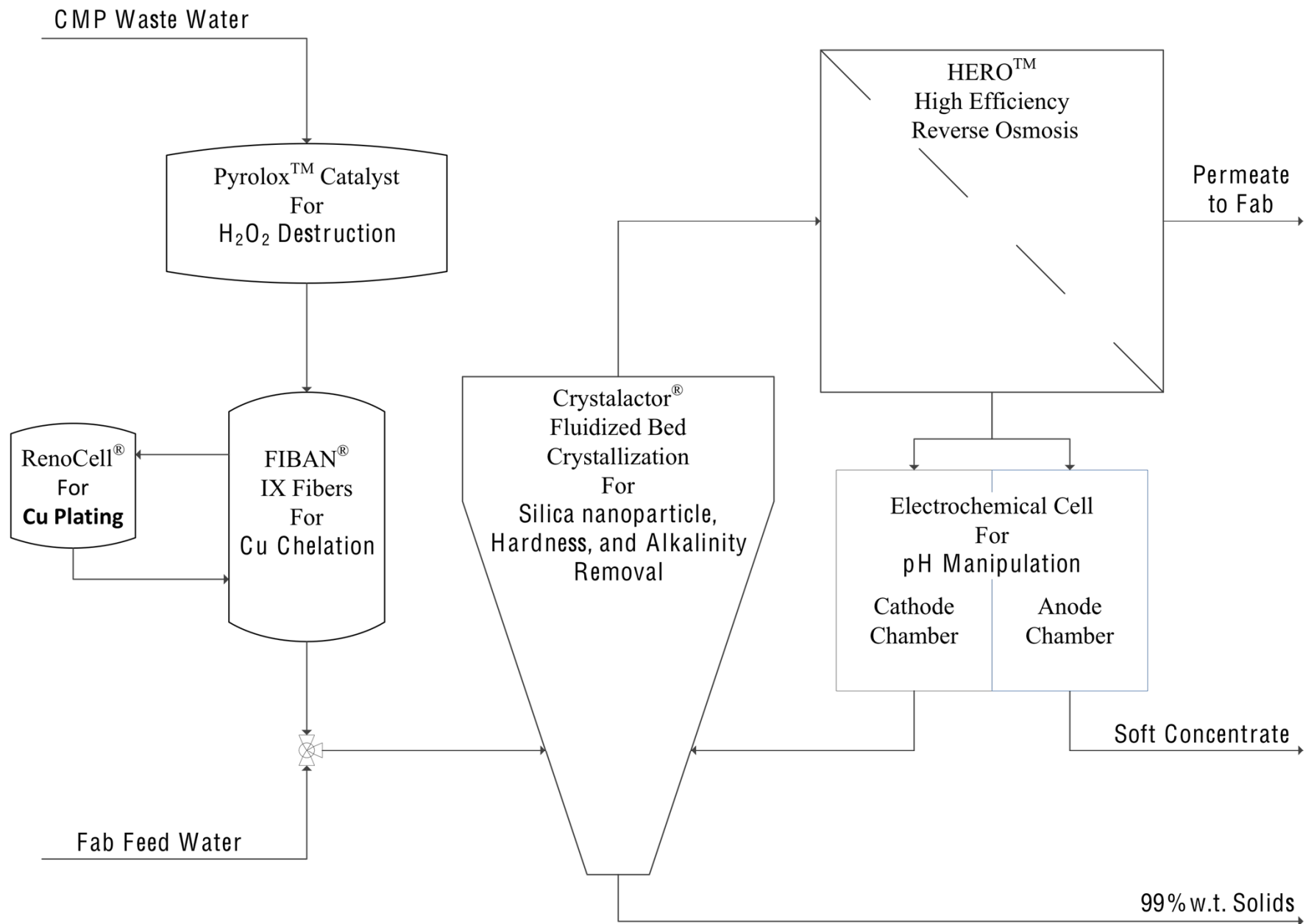
*Concentrate effluent pH ~ 11*

*The membrane and membrane foulants are  
negatively charged at pH > 9*

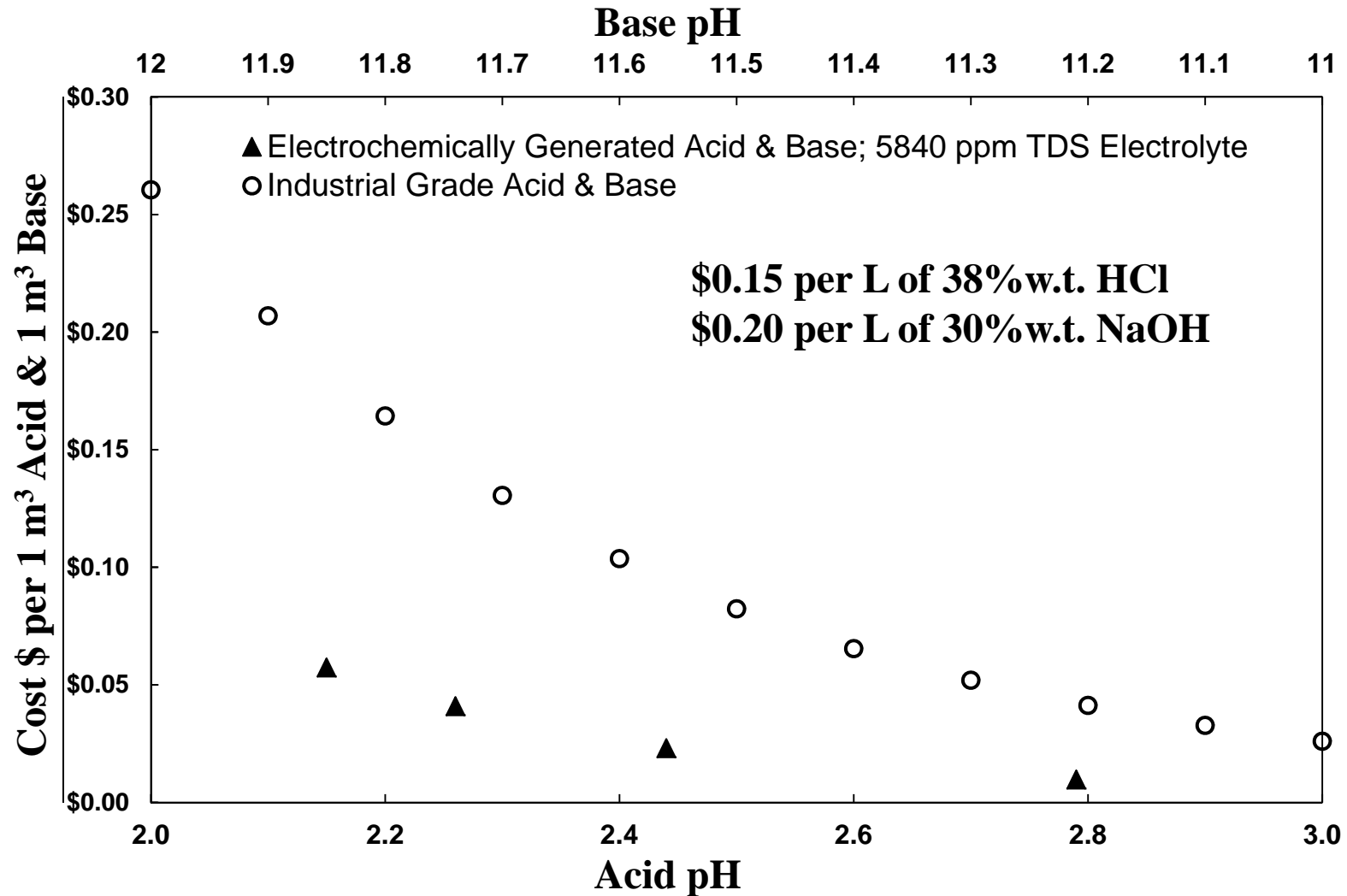
*Waters with silica concentrations in excess of 1000 ppm are effectively treated  
with HERO<sup>®</sup>*

*HERO<sup>®</sup> water recoveries in excess of 98%*





# Electrochemical Acid & Base Generation





# Intel Contacts & Collaborators

- **Dan Dodges**                      *dan.hodges@intel.com*
- **Don Hooper**                      *don.fab11.hooper@intel.com*
- **Allen Boyce**                      *allen.r.boyce@intel.com*

## Future Plans

### Next Year Plans

- Funding ended 4/1/10

### Long-Term Plans

- Seek partners for proposal

# Publications, Presentations, and Recognitions/Awards

- **David Hubler: Triffet Prize and GEP Smith Fellowship; First Place Award, University of Arizona 2009 Student Showcase, B.P.A. Division, November 6-7, 2009, Tucson, AZ**
- **Jake Davis: Science Foundation Arizona Fellowship, NASA Space Grant**
- **“Economic Benefit of Commercial and Industrial Water Uses in a Semi-arid Municipality,” presented at the Arizona Hydrological Society/American Institute of Hydrology 2009 Hydrological Symposium, August 30-September 2, 2009, Scottsdale, AZ**
- **“Electrochemical Methods for Water Reclaim in Semiconductor Manufacturing,” presented at the International Conference on Microelectronics Pure Water, November 11-12, 2008, Mesa, AZ**
- **“Electrochemical Water Treatment using Diamond Film Electrodes,” presented at the University of Illinois at Urbana-Champaign, November 7, 2008**
- **“Evaluating Economic Impacts of Semiconductor Manufacturing in Water-Limited Regions,” submitted to *Journal of the American Water Works Association***
- **“Reclamation of Copper-CMP Wastewater,” presented at the International Conference on Microelectronics Pure Water, November 16-17, 2010, Mesa, AZ**