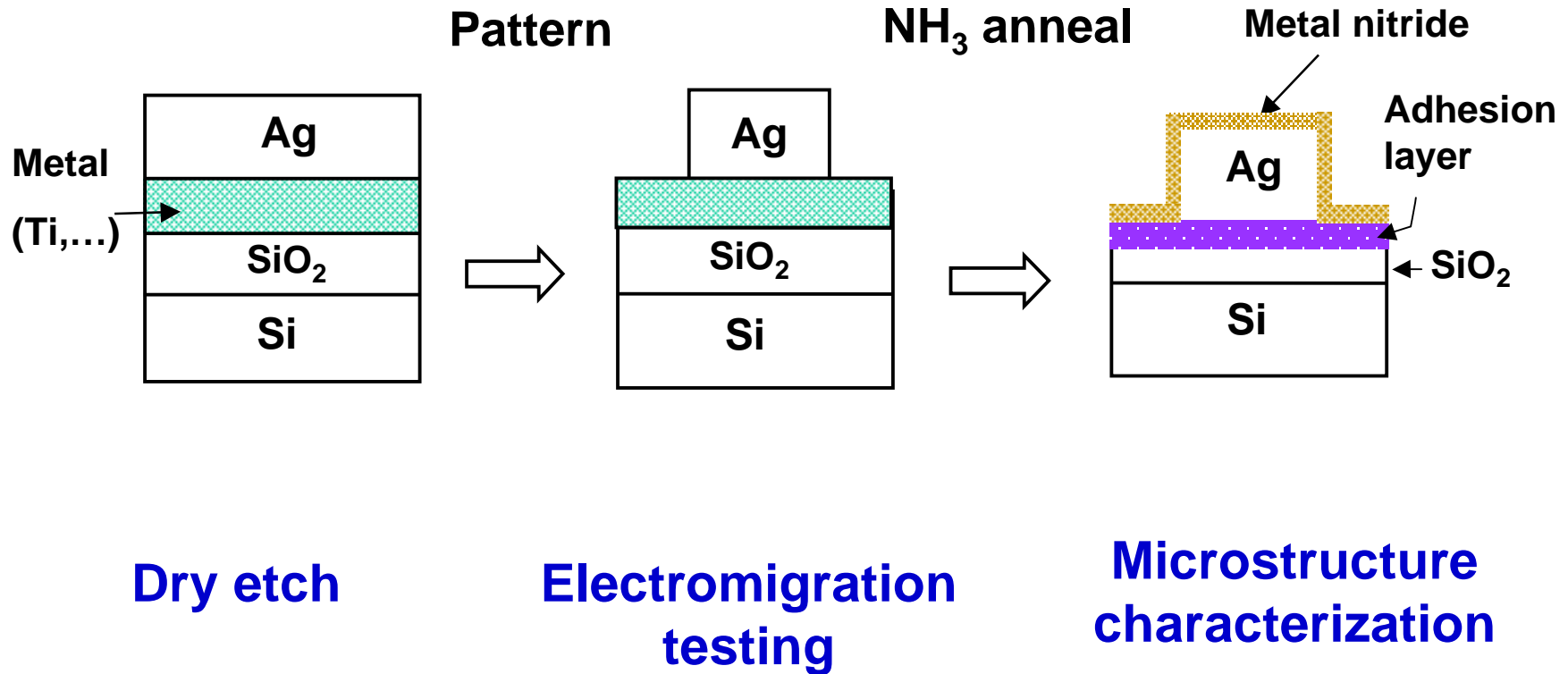
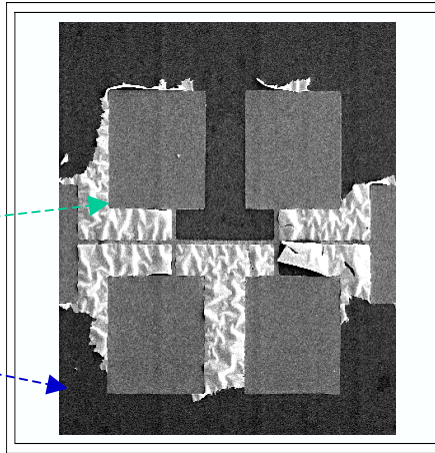
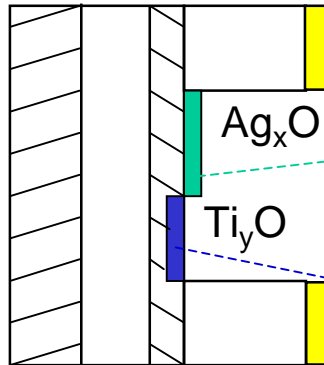


PHASE II: Patterned Ag Structures



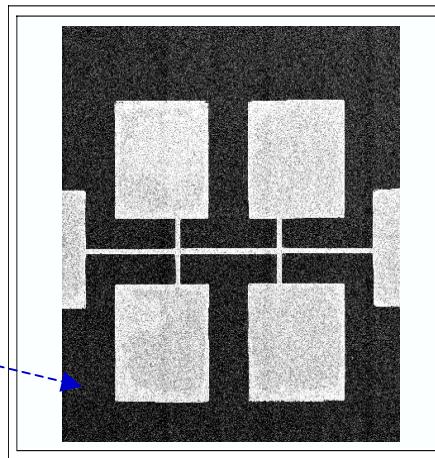
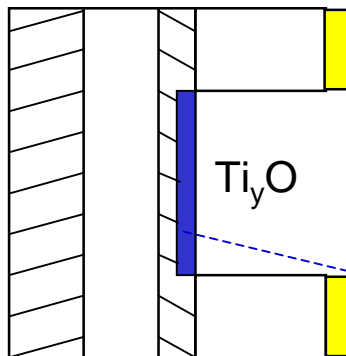
Ag Etch in O₂

Si SiO₂Ti AgResist



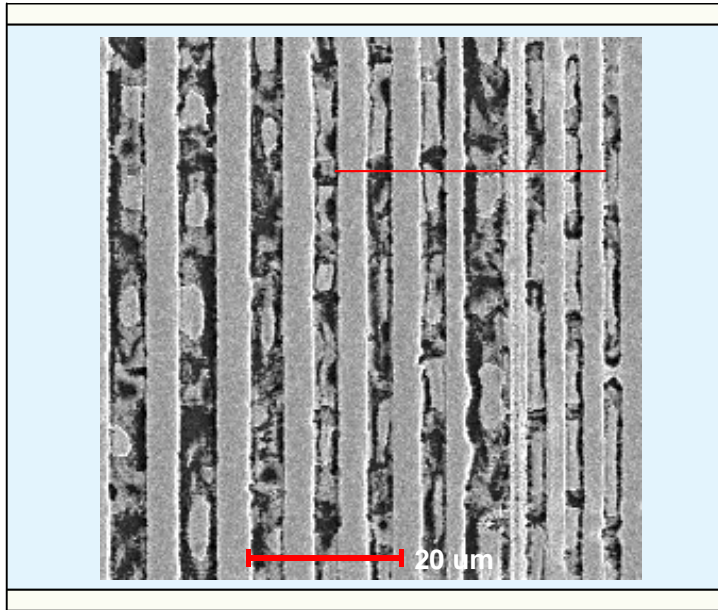
**50 watt, 100 mTorr,
4 min**

Si SiO₂Ti Ag

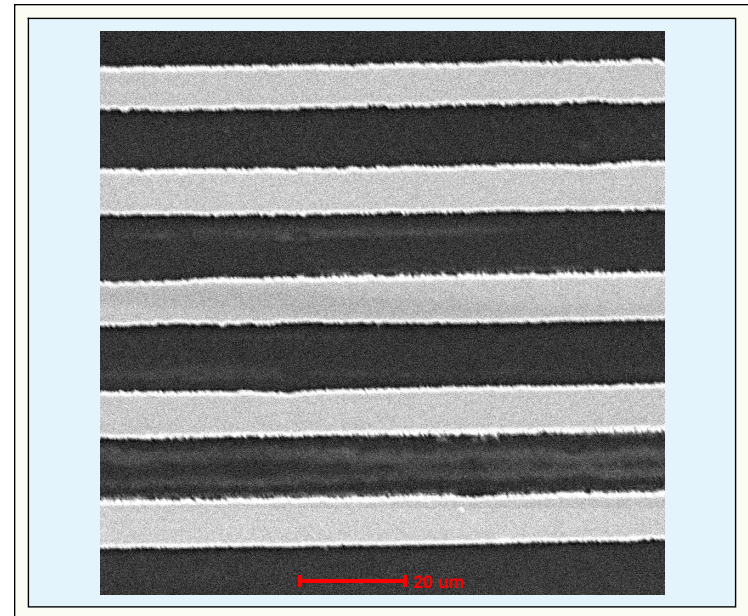


**50 watt, 100 mTorr,
5 min**

Comparison of O₂ and O₂/Cl₂ Chemistries

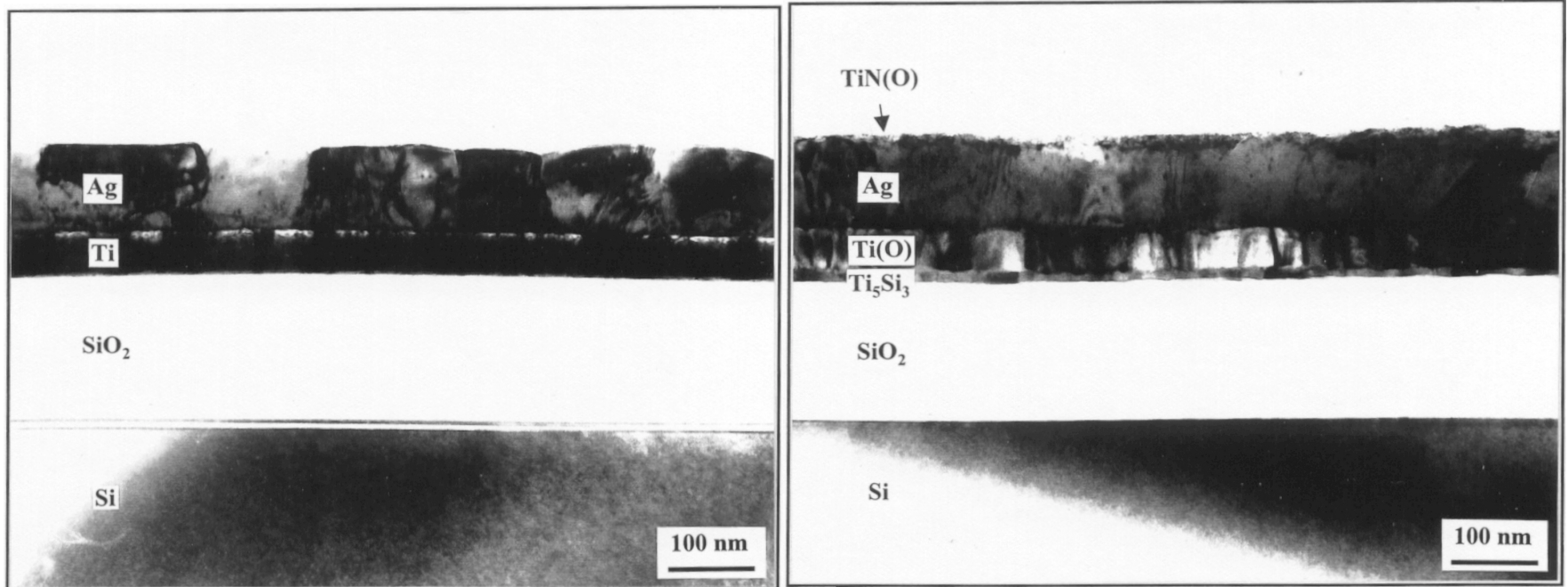


RF Power: 100 W
Pressure: 100 mtorr
O₂: 25 sccm



RF Power: 100 W
Pressure: 100 mtorr
O₂: 17 sccm
Cl₂: 8 sccm

Encapsulation of Ag Patterns

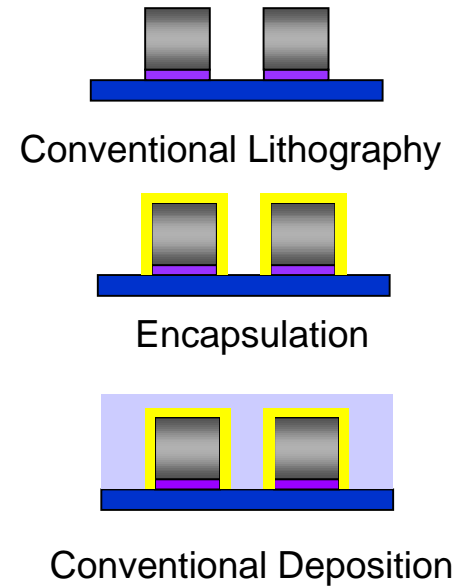
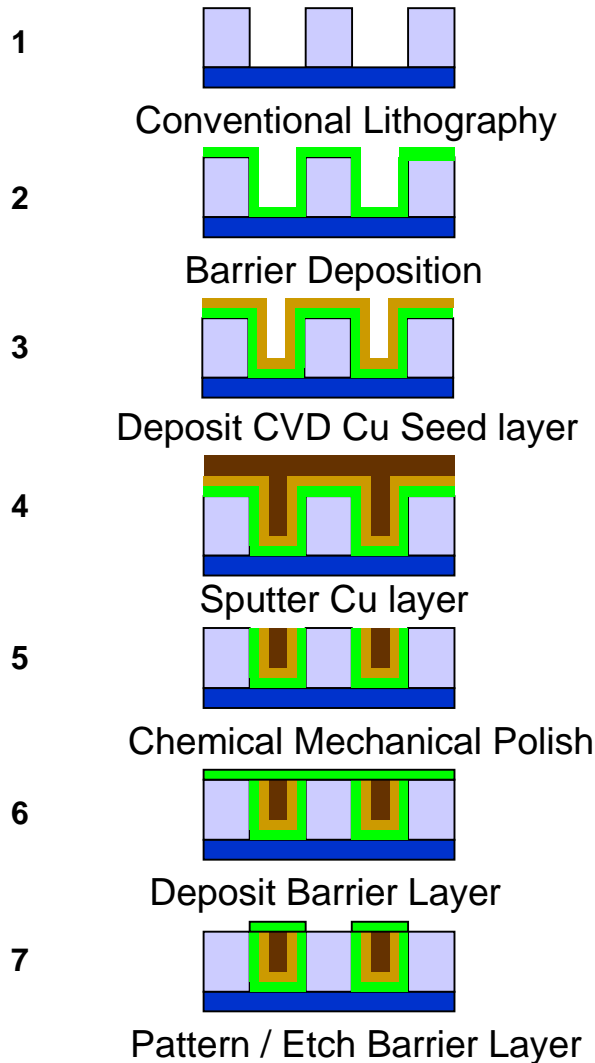










Before encapsulation

After encapsulation

- Cross-sectional TEM micrographs of 10- μ m-wide Ag lines
- Ag lines are encapsulated by TiN(O) layer

Comparison of Cu and Ag Metallization Processes



	Dielectric		Cu
	Barrier Oxide		Ag
	Barrier		Ti
	CVD Cu		TiNO

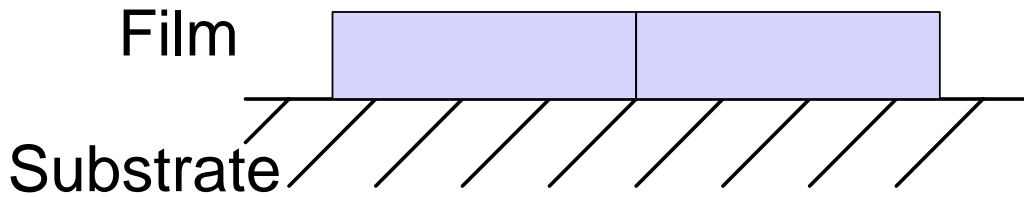
The Ag process is more manufacturable !

Thermal Stability

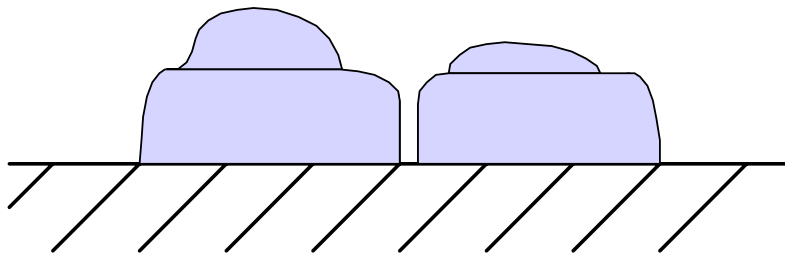
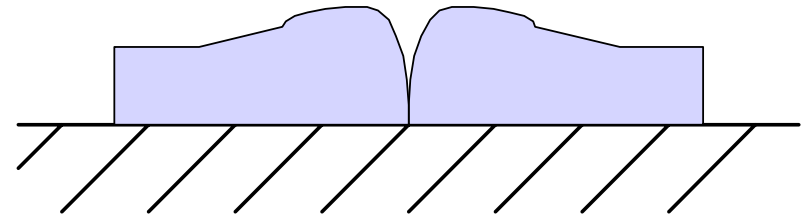
- **Thermal Stability Study**
 - **In-situ four-point probe to detect the onset temperature during annealing process**
 - **Onset temperature dependent on ramp rate**
 - **Calculate the activation energy for each film thickness, ambient, and its corresponding substrate - based on sheet resistance differences between measured values and linearity**
 - **Ramp rate only shifts the Arrhenius plot but does not change the slope**

Schematic of Films Evolution During Annealing

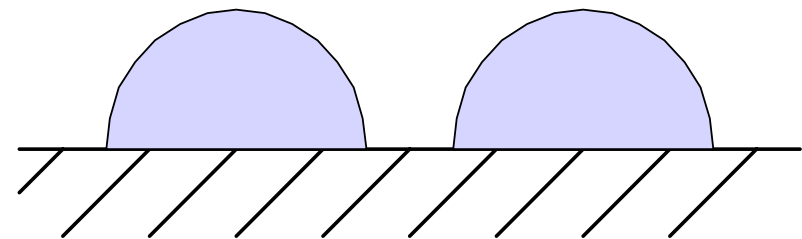
Initial Film



Thermal Grooving

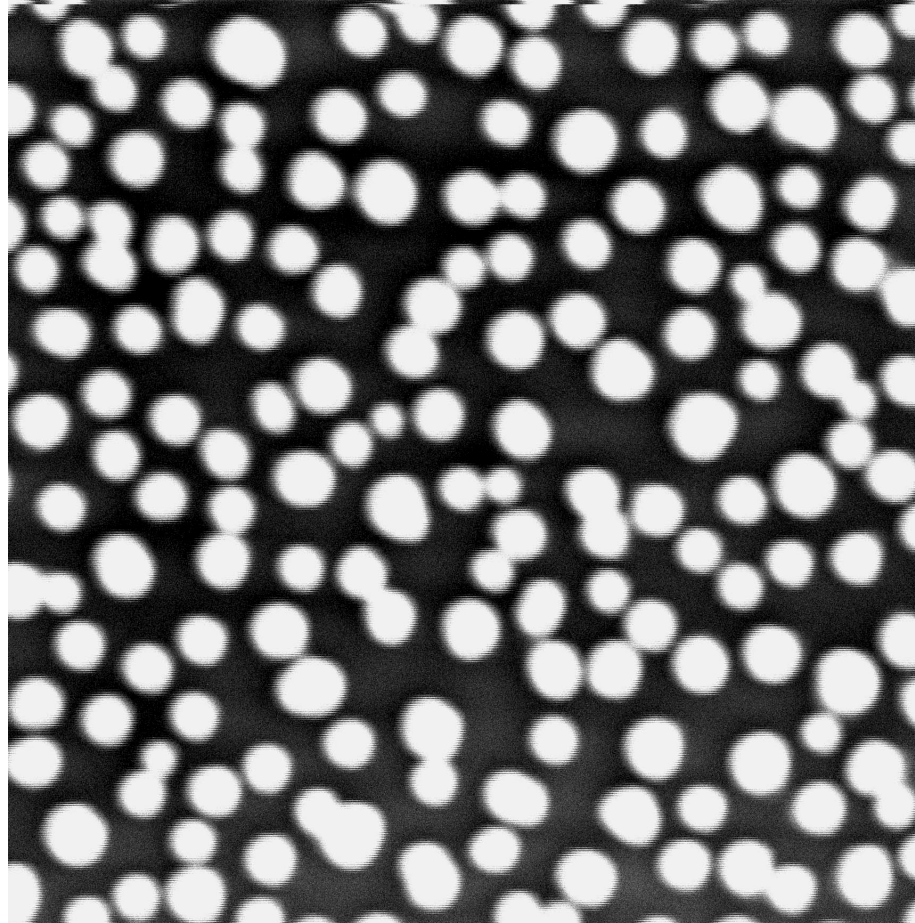


Hillocks and Voids Formation



Agglomeration

Demonstration of a Complete Agglomerated Ag/SiO₂ Annealed in Air



Arrhenius Relation

- Activation energies based on film thickness and onset temperature as follow:

$$\lambda^3 = C \exp\left(-\frac{E_a}{kT_i}\right)$$

λ - Film thickness

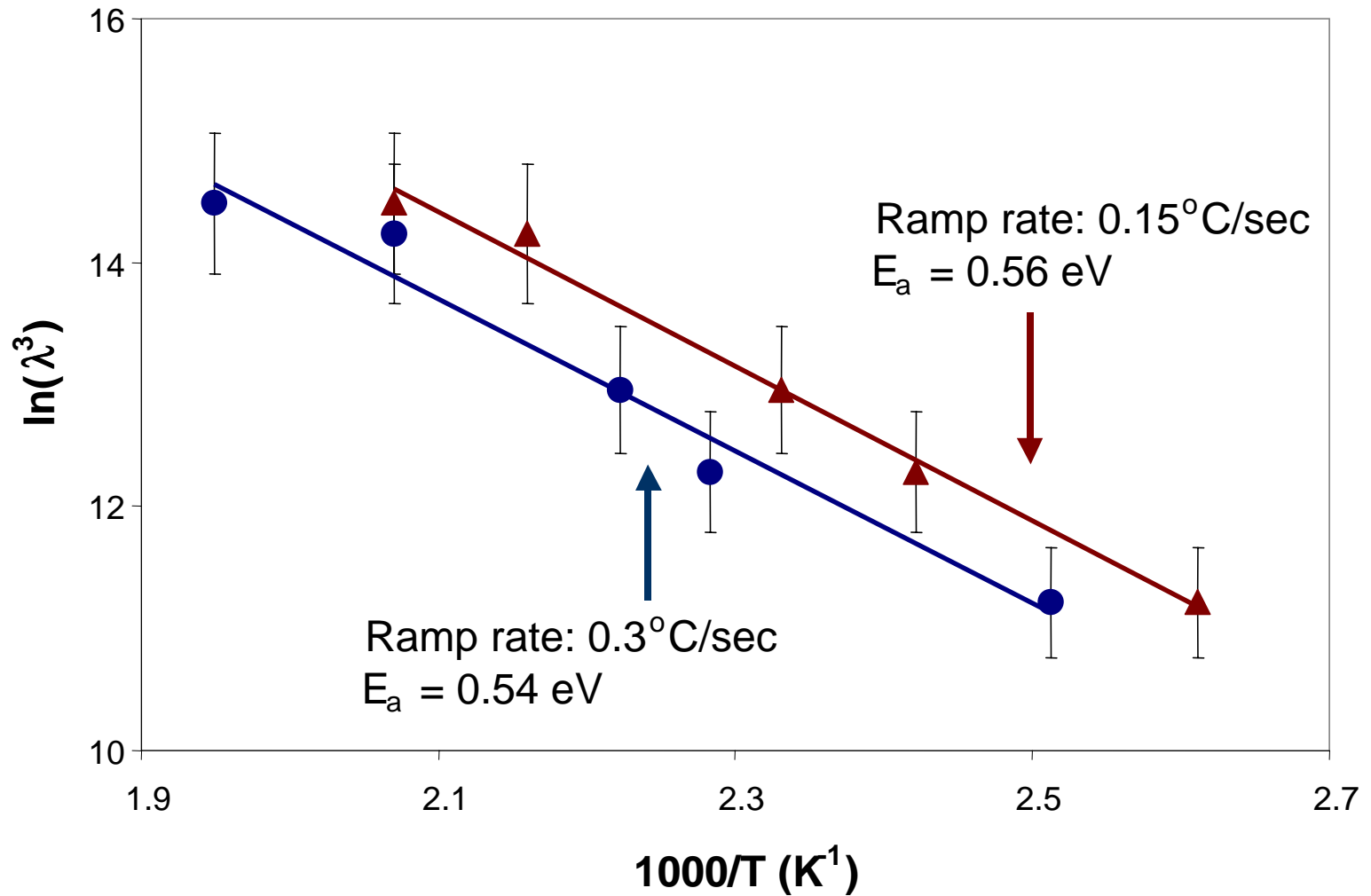
E_a - Activation energy

T_i - Onset temperature for void nucleation

k - Boltzman constant

C - Pre-exponent factor

Arrhenius plot of various Ag thicknesses (on SiO₂ annealed in air, at 0.15 and 0.3 °C/sec) vs.onset temperature



Calculation Formulation

- The Arrhenius dependence of each film thickness, its associated substrate, and ambient - based on the sheet resistance differences

$$\Delta\rho_s = \rho_o \exp\left(-\frac{E_a}{kT}\right) \quad \rho_{sm} = \rho_c + \alpha T + \rho_o \exp\left(-\frac{E_a}{kT}\right)$$

$$\Delta\rho_s = \rho_{sm} - (\rho_c + \alpha T)$$

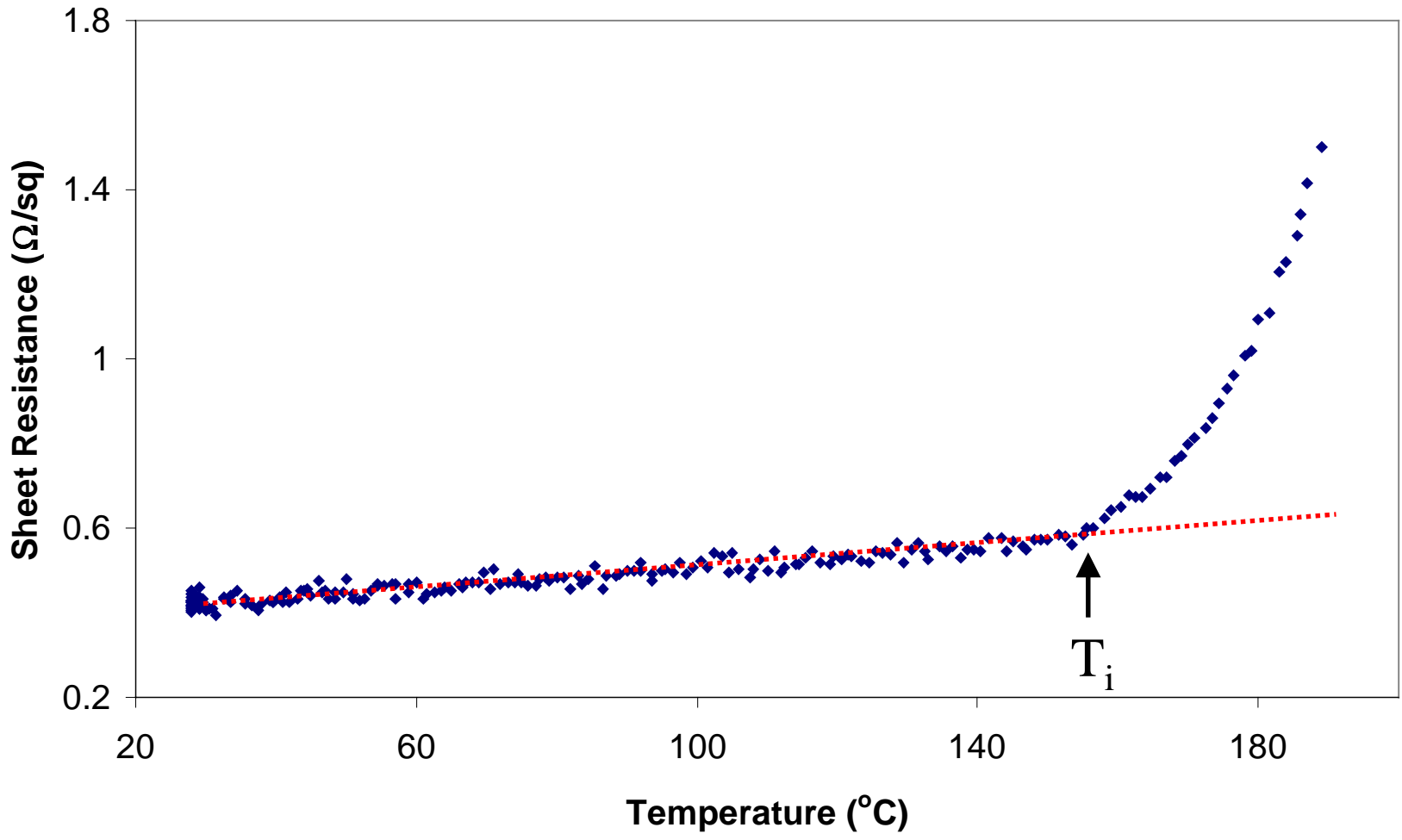
T - Temperature (K)

ρ_o - pre-exponential factor constant

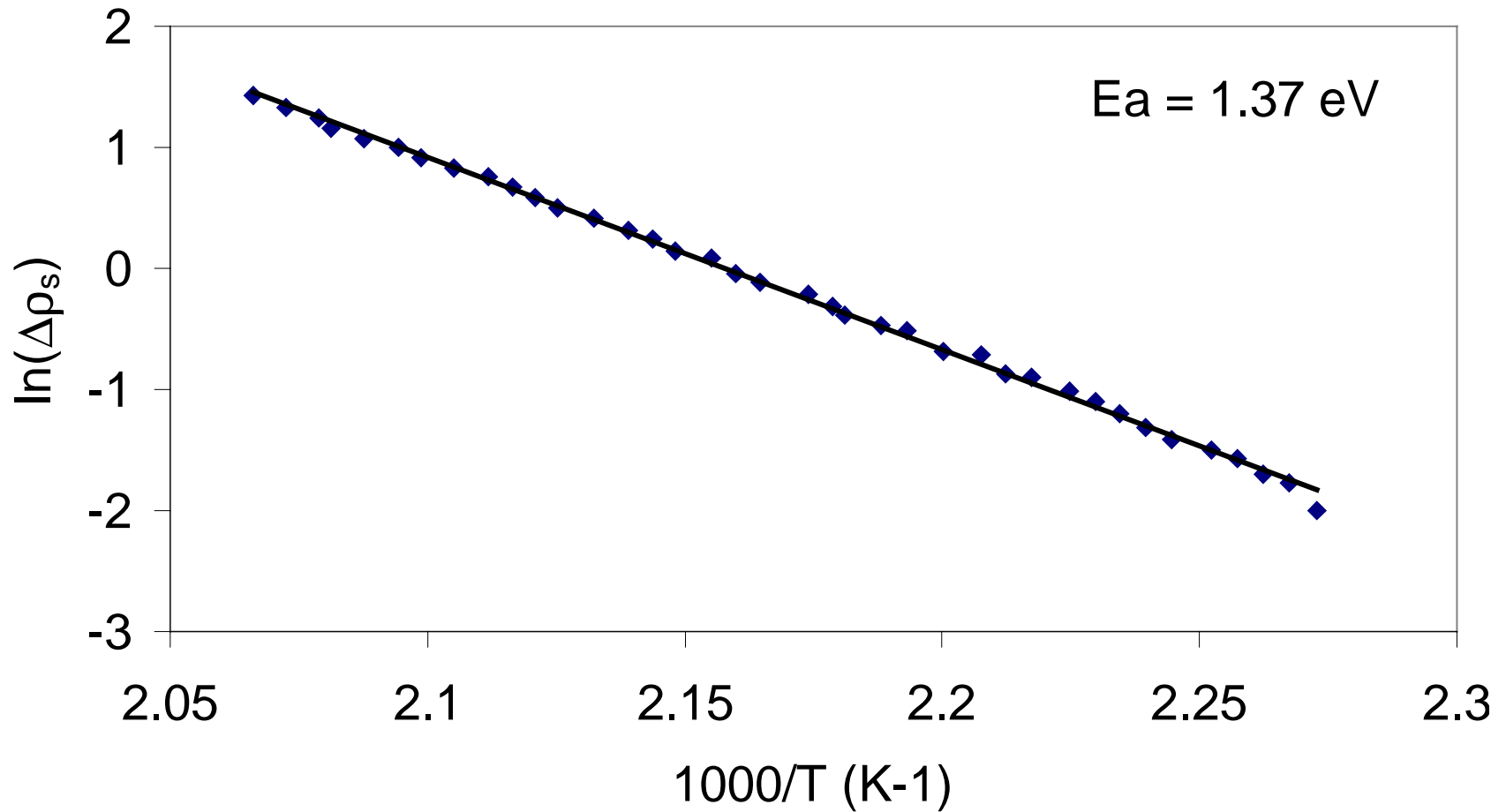
ρ_{sm} - measured sheet resistance

$\rho_c + \alpha T$ - sheet resistance change due to phonon scattering

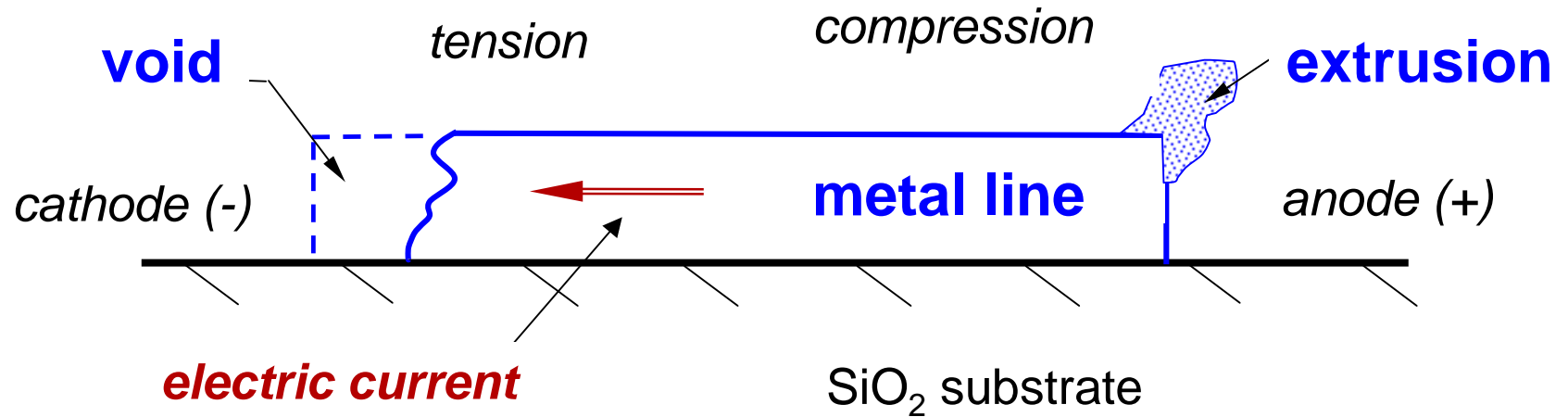
Sheet Resistance (SiO_2/Ag 60 nm, air, and $0.15\text{ }^\circ\text{C}/\text{sec}$) vs. Temperature.



Arrhenius plot of the sample shown in the previous slide

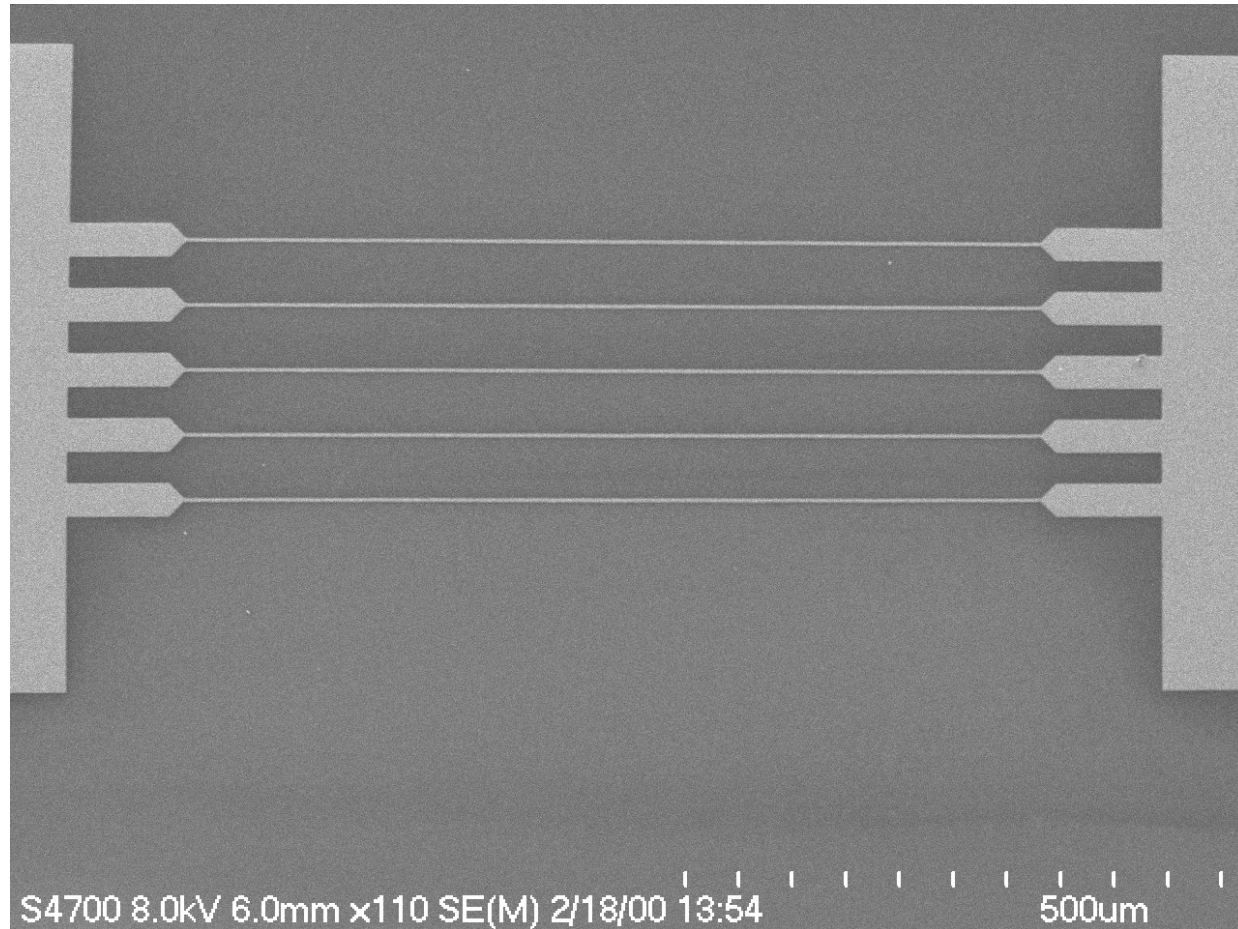


electromigration

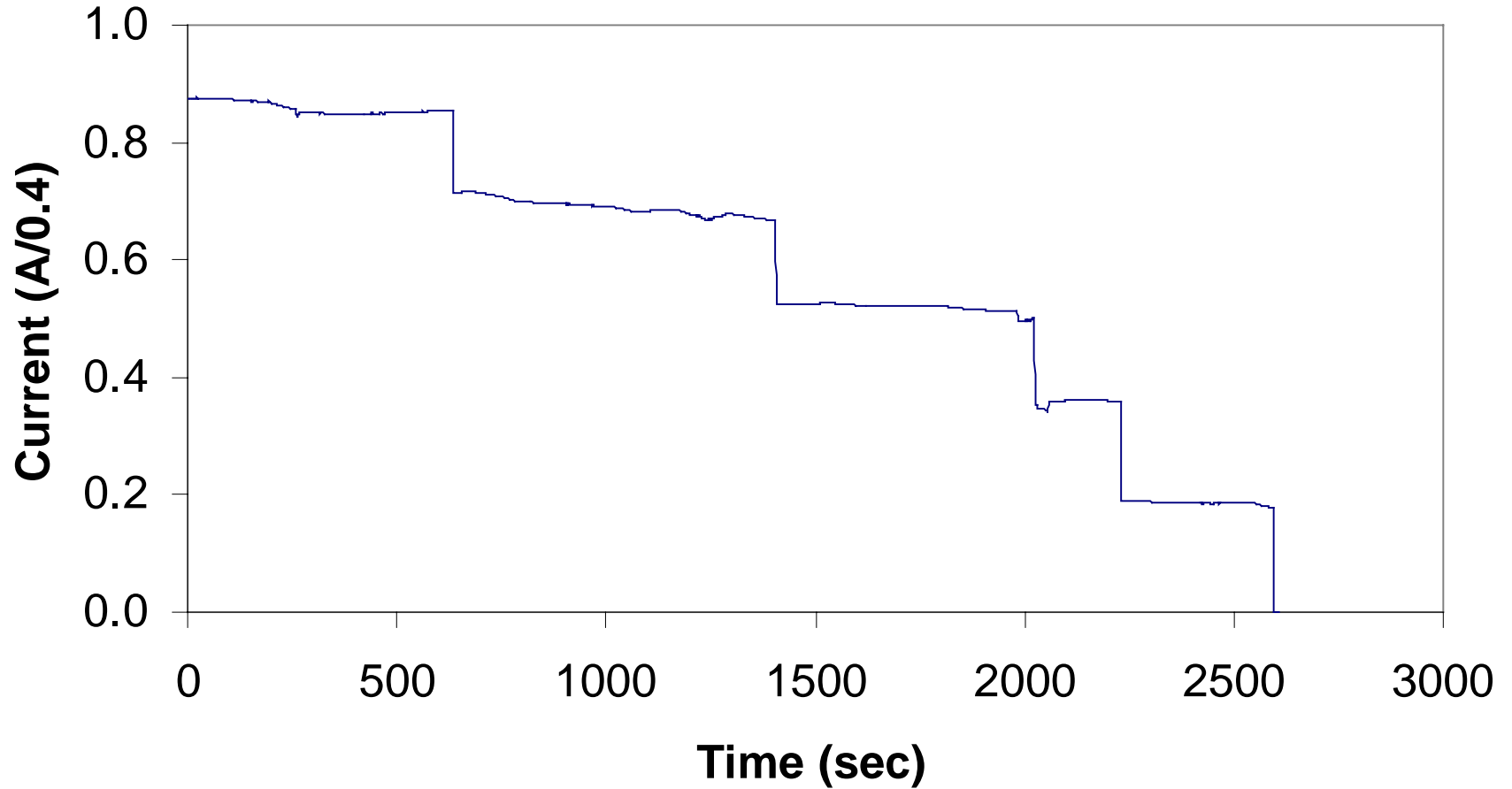


- *mass transport at high current levels*
- *critical failure mechanism*

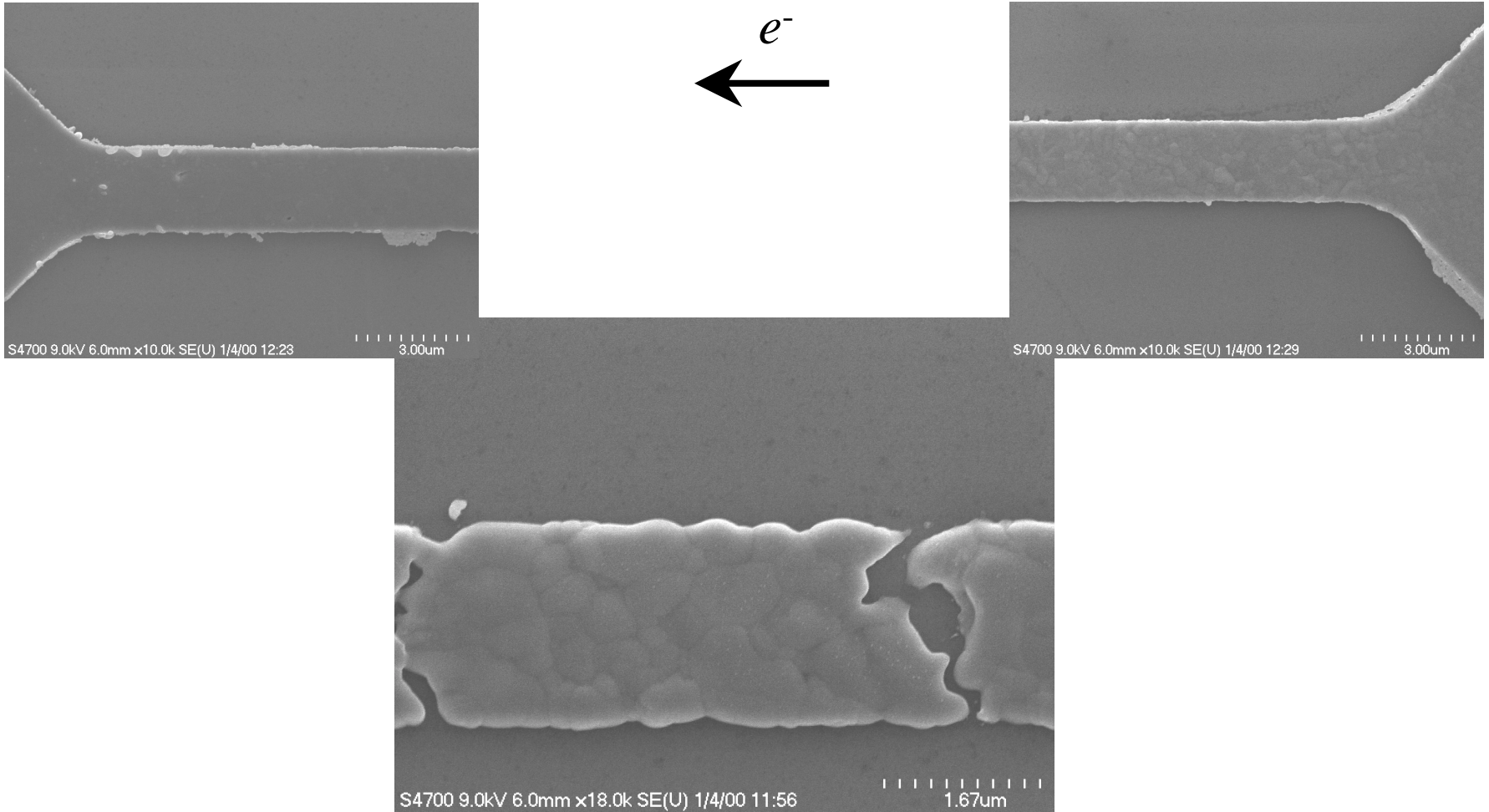
An EM Test Structure



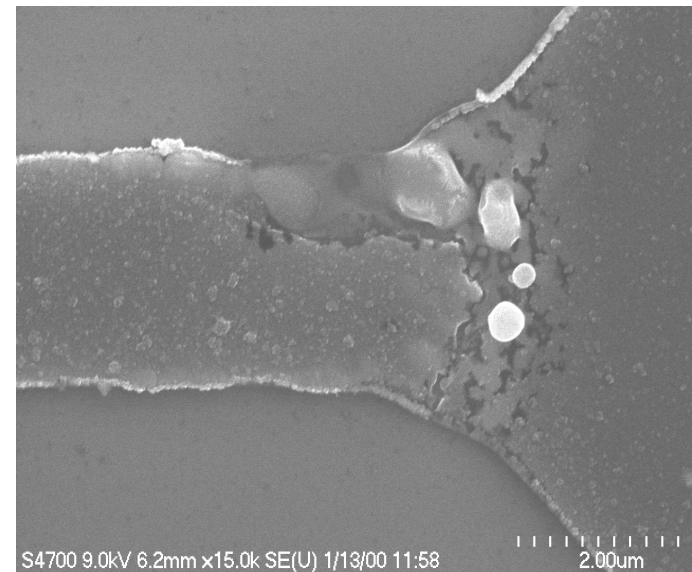
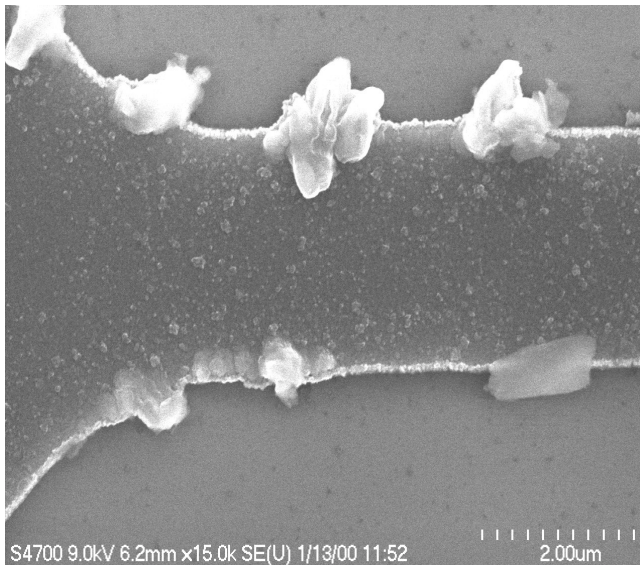
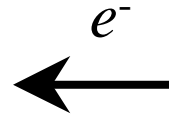
Actual Current Measurements during EM test
200 nm Ag/SiO₂, J ~ 17.5 MA/cm², T ~ 160 °C



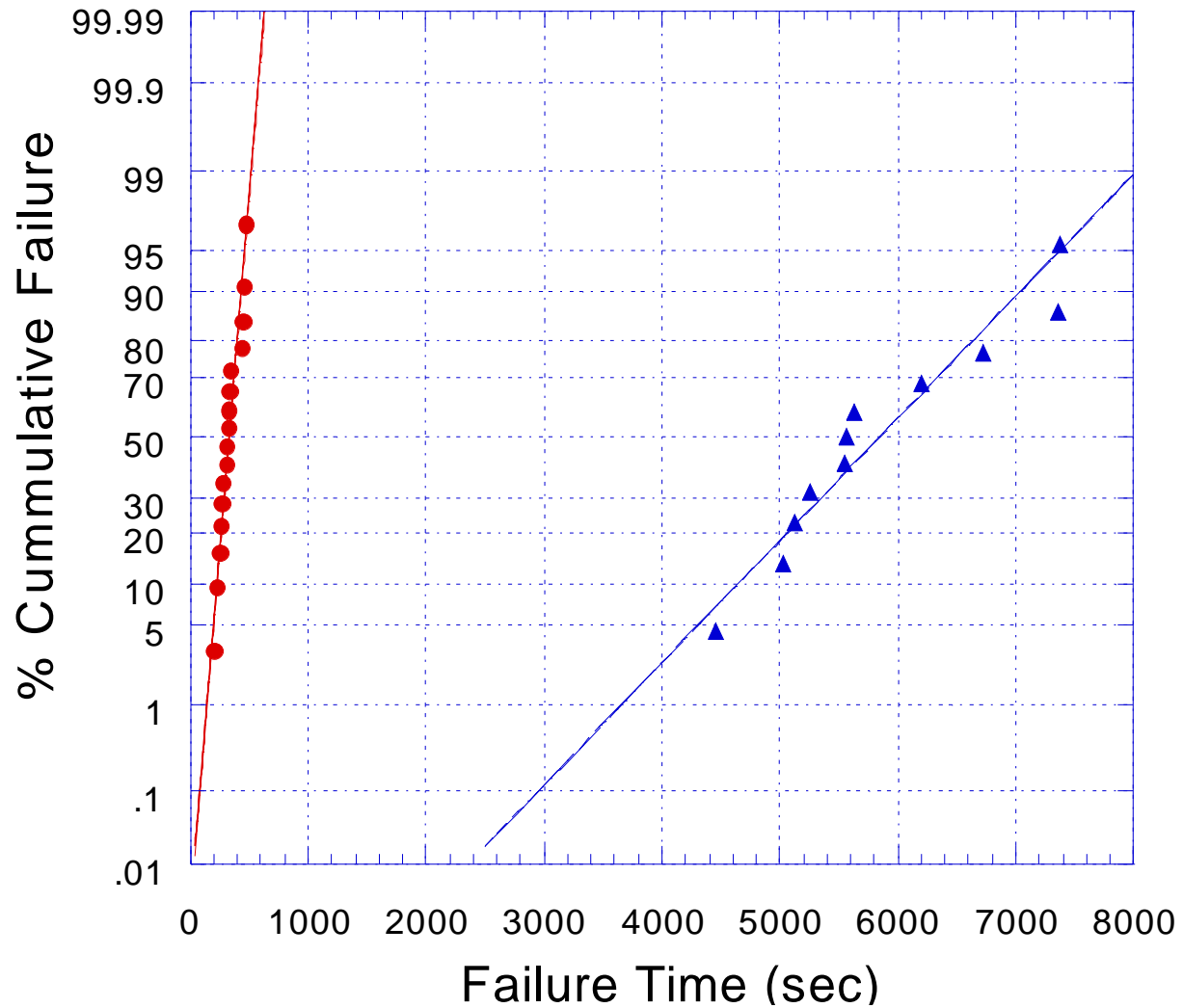
Ag/SiO₂, J ~ 22.5 MA/cm², T = 160 °C



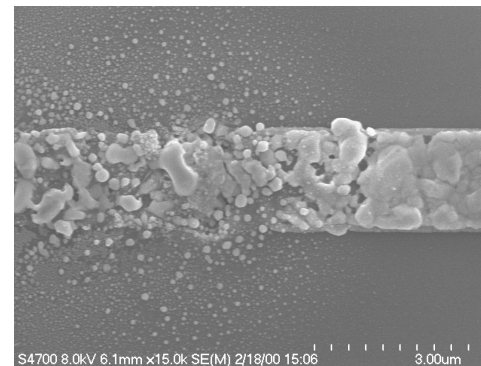
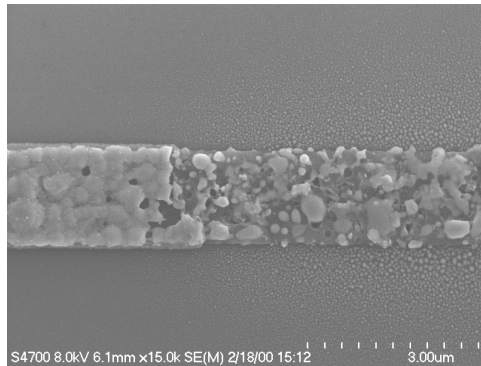
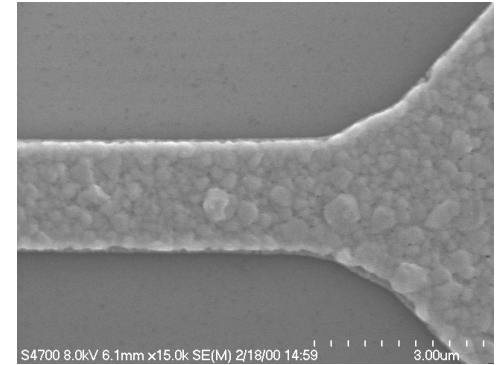
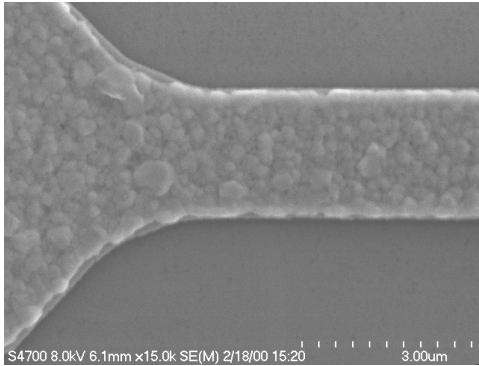
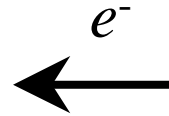
Al/SiO₂, J ~ 9 MA/cm², T = 160 °C



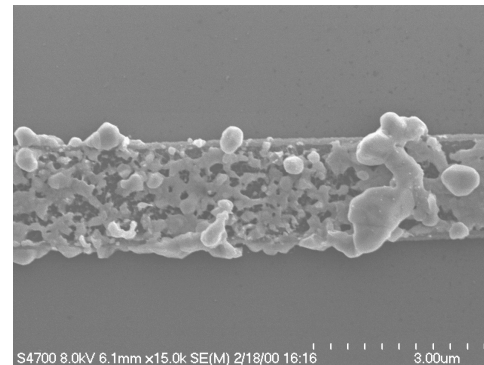
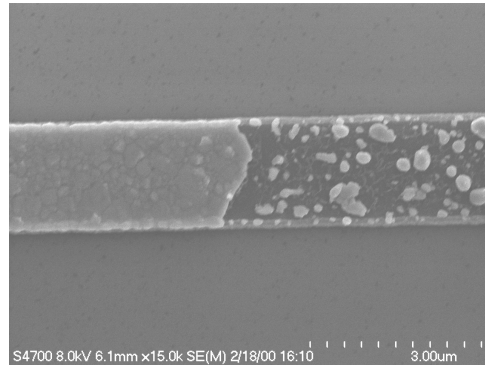
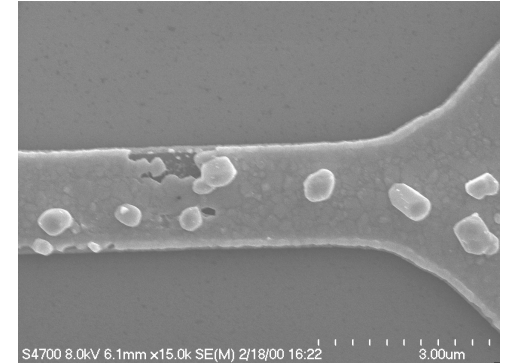
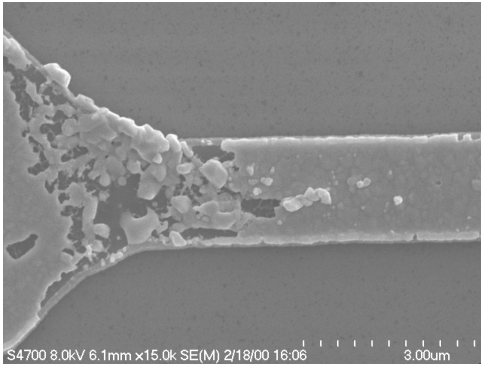
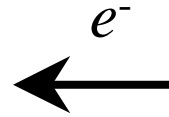
Ag/SiO₂ and Al/SiO₂, J ~ 9 MA/cm² T ~ 160°C



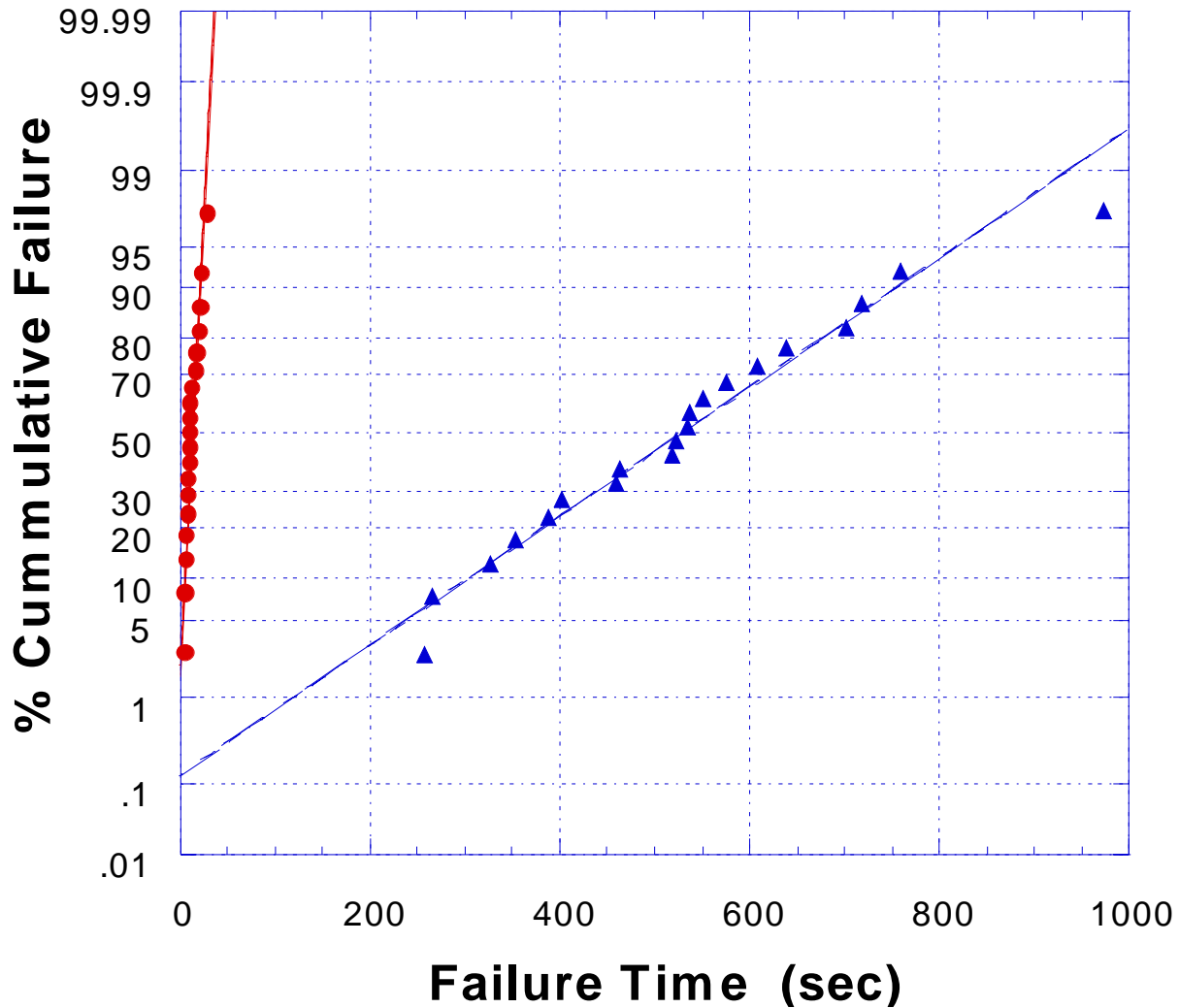
Ag/Ti/SiO₂, J ~ 18 MA/cm², T = 290 °C



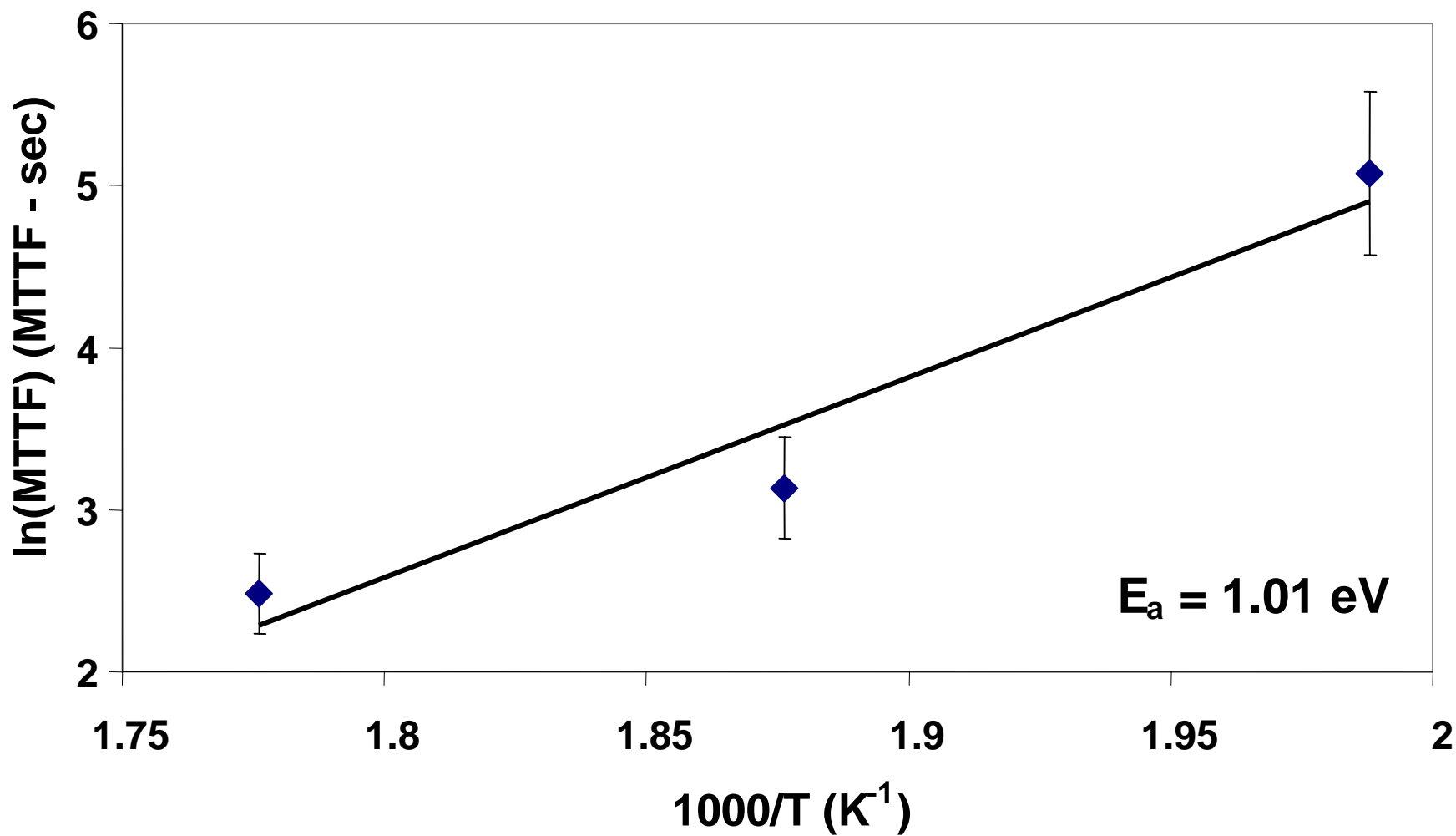
$\text{TiON}_x/\text{Ag}/\text{SiO}_2$, $J \sim 18 \text{ MA/cm}^2$, $T = 290 \text{ }^\circ\text{C}$

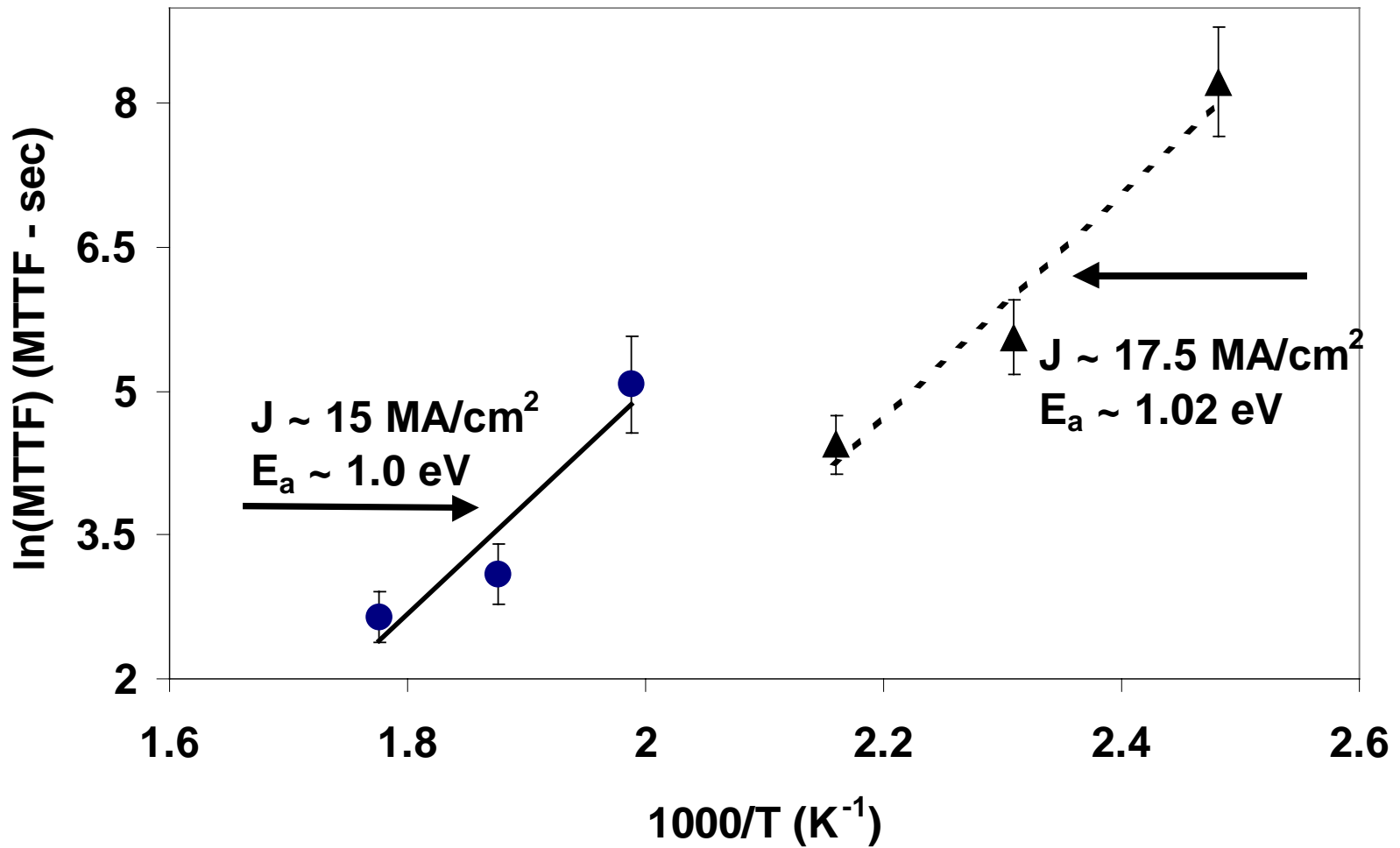


Ag/Ti and TiON/Ag/Ti, $J \sim 18 \text{ MA/cm}^2$, $T \sim 290^\circ\text{C}$



100 nm Ag/Ti/SiO₂ , J ~ 18 MA/cm² , 230 - 290°C





EM Summary

- EM test at temperatures $< T_i$ and at low current density:
 - Direction of transport is the same as that of electron flow
 - Voids form near to the cathode
 - More hillocks and less void from near the anode
- EM failure when tested at high current density show that open line is caused by slit-like formation
- EM activation energies (E_a) depend on current densities
 - $E_a \sim 0.6$ eV for $J < 5$ MA/cm²
 - $E_a \sim 1.0$ eV for $J > 15$ MA/cm²

Silver Metallization

- Lower ρ than Al and Cu → *Reduced RC Time Delay*
- Higher Z and melting point
- Encapsulation process → *Improved Reliability*
- Textured microstructure
- Dry etch capability → *Lower Fabrication Cost*