

SUPERCRITICAL CO₂ CLEANING

An Overview

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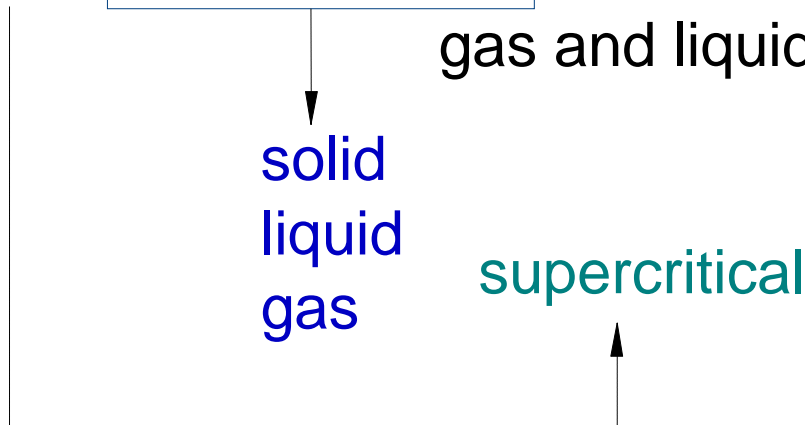
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- ◆ Karl Tiefert Agilent Technologies

What is a Supercritical Fluid?

A supercritical fluid can be considered as a **fourth state of matter** with combined properties of both, gas and liquid.



"supersolvent"

Solubility power can be modified and adapted.

Can dissolve nonpolar solids.

Is superior to conventional solvents for certain classes of chemicals.

Candidates for Supercritical Fluids



	Critical Temperature [°C]	Critical Pressure [psi]
Methane (CH ₄)	-82.4	667.2
Propane (C ₂ H ₆)	96.8	616.4
Methanol (CH ₃ -OH)	240.1	1173.4
Isopropanol (CH ₃ -CHOH-CH ₃)	235.8	690.4
Carbon Dioxide (CO ₂)	31.1	1070.4
Ammonia (NH ₃)	132.4	1646.2
Water (H ₂ O)	374.4	3208.2

Why Supercritical CO₂?



Acetal
Acetaldehyd
Acetic acid
Acetic anhydride
Acetone
Acetonitrile
Acetophenone
Acetyl chloride
Acrolein

Cyclohexane
Cyclohexanone
Diacetone alcohol
Di-sec-butylbenzene
p-Dichlorobenzene
2,4-Dichlorophenol
N,N-Dethylaniline
N,N-Diethylformamide
p-Dimethoxybenzene

1-Heptaldehyde
n-Heptane
2,5-Hexanedione
Hexyl Alcohol
Hydrogen sulfide
4-Hydroxy-4-methyl-2-pentanone
Beta-Hydroxypropionitrile
Isopropyl alcohol

Propylene
Pyridine
Salicylaldehyde
Stannic chloride
Sulfur dioxide
Sulfuryl chloride

CO₂ has a low critical point.

CO₂ dissolves readily many substances.

CO₂'s solubility power can be enhanced.

CO₂ is environmentally friendly and not rated.

Acrylonitrile
t-Amyl alcohol
Benzaldehyde
Benzene
Benzonitrile
Benzoyl chloride

N,N-Diethylacetamide
N,N-Dimethylacetamide
N,N-Dimethylformamide
2,2-Dimethylpentane
p-Dioxane
n-Dodecane
Ethyl alcohol
Ethyl acetoacetate
Ethyl benzoate
Ethyl carbonate
Ethyl chloroformate
Ethylene bromide
Ethylene diformate

2-Octanone
p-Oxathiane (thioxane)
Paraldehyde
Phenyl isocyanide
Phosphorus trichloride
2-Picoline
Pinene
Propionaldehyde
Thiophene
Thioxane
Toluene
Tolunitriles
Triacetin
11-sec-Butylbenzene
2,2,3-Trimethylbutane
Valeraldehyde

2-Butanone
n-Butyl ether
Butyl oxalate
n-Butylaldehyde
Caproic acid
Caprylic acid
Carbitol

Carbon disulfide
Cellulosolve
Chlorex
Chloroacetone
Chlorobenzene
Beta-Chloroethyl acetate
Chloroform
Chloromaleic anhydride
o-Chlorophenol
Crotonaldehyde

Ethyl ether
Ethyl formate
Ethyl lactate
Ethyl maleate
Ethyl oxalate
Ethyl phenylacetate
Ethyl salicylate
Ethyl succinate
Ethyl sulfate
Formic acid
Gasoline

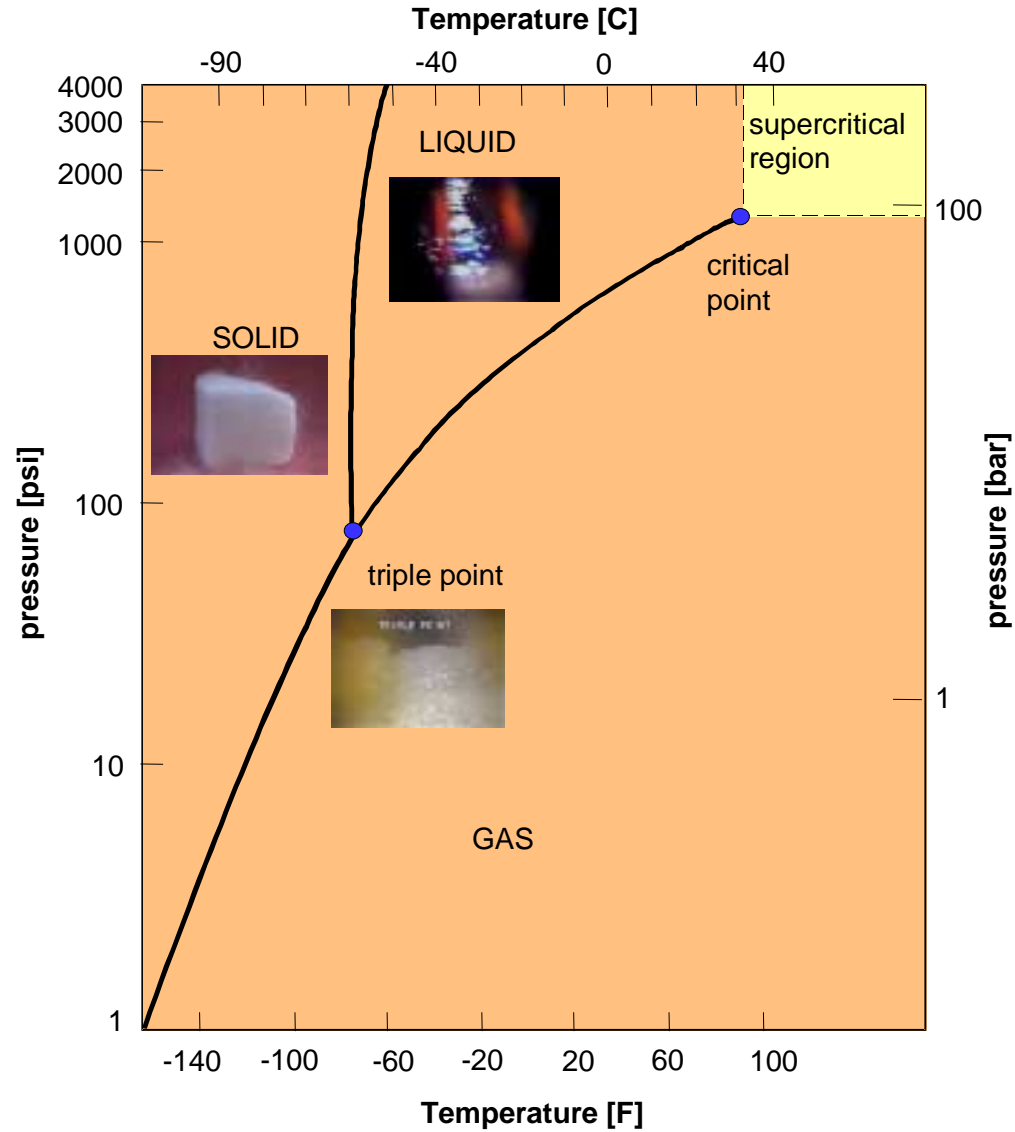
Methanol
Methyl acetate
Methylal
Methyl benzoate
Methylcyclohexane

Methyl salicylate
Methyl sulfate
Nitrobenzene
Nitroethane
Nitromethane
o-Nitrophenol
1-Nitropropane
o-Nitrotoluene

Advantages of SCCO₂

- ✓ Approved by EPA as an ODC alternative (SNAP list, 1995)
- ✓ Zero ODP
- ✓ No hazardous waste generated
- ✓ Low toxicity
- ✓ Non-flammable
- ✓ Non-corrosive
- ✓ Very inexpensive
- ✓ Low viscosity
- ✓ **Low surface tension**

Phase Diagram of CO₂



- ◆ **Critical Values**

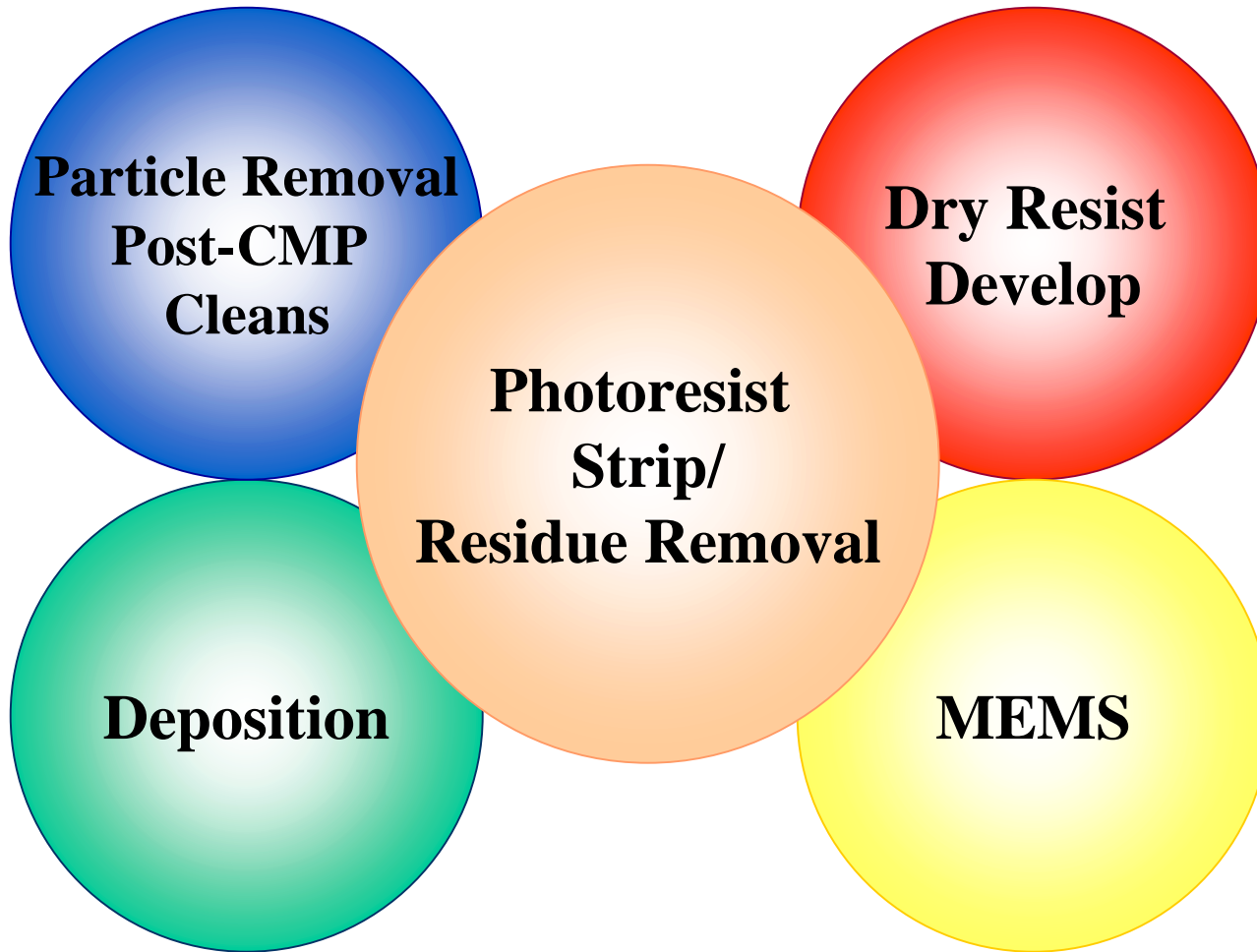
- **Critical temperature = 31°C (85°F)**
- **Critical Pressure = 73.8 atm (1070psi)**

- ◆ **No dipole moment**

- ◆ **Large Quadrupole moment**

- ◆ **CO₂ has both acidic and basic sites**

Supercritical Fluid Technology Applications



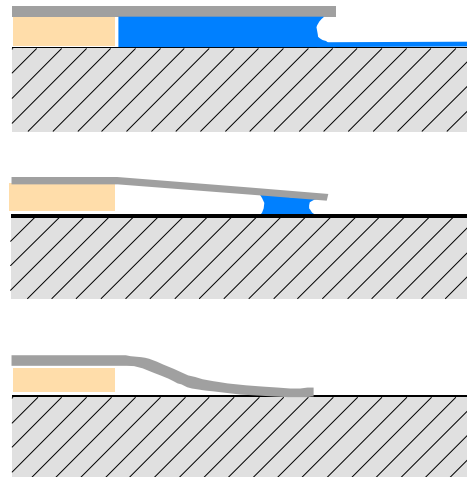
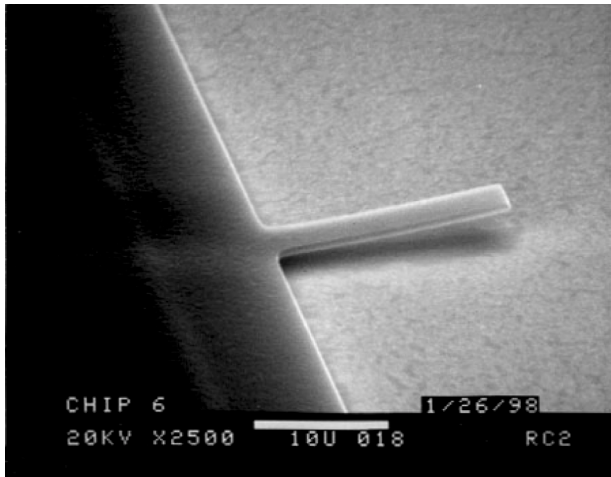
Current Application: MEMS Drying

MEMS = Micro Electro Mechanical Structures

Problem: **Liquid Bridging** or **STICTION**

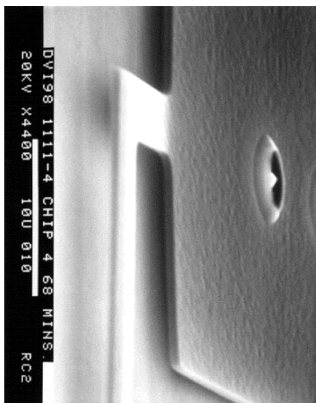
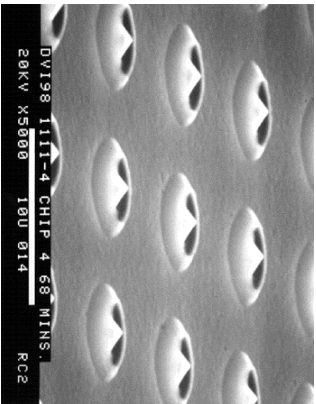
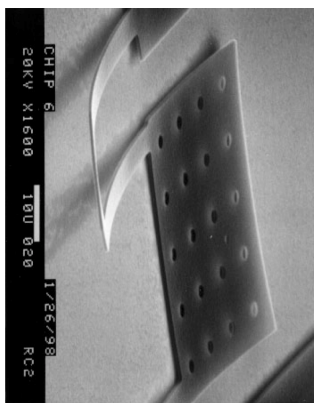
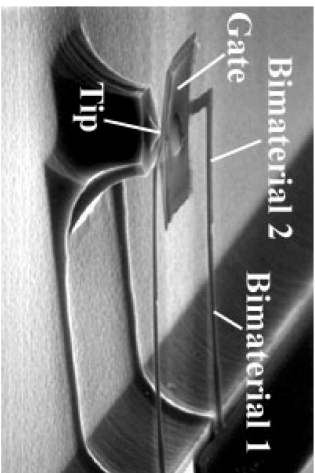
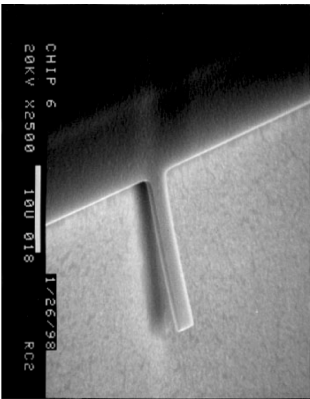
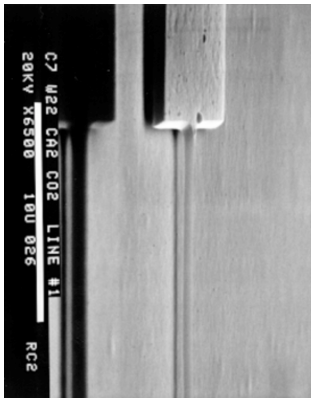
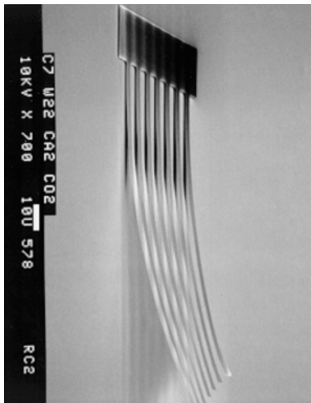
Stiction is due to the surface tension effects of trapped capillary liquids upon drying.

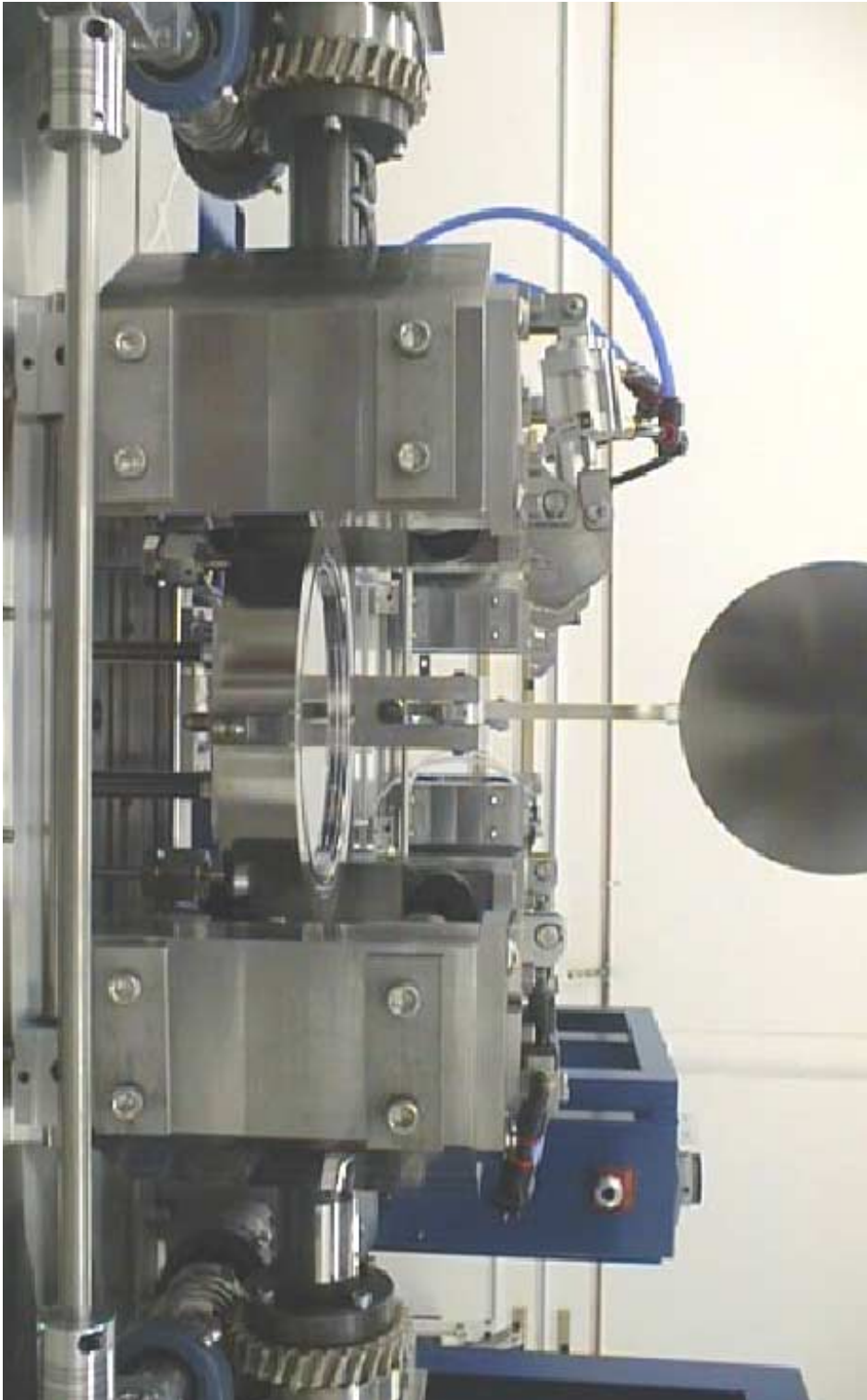
$$F_m = \frac{\gamma A}{h} (\cos \Theta_1 + \cos \Theta_2), \quad \gamma(H_2O) = 72 \text{ dynes / cm}, \quad \gamma(SCCO_2) = 0$$



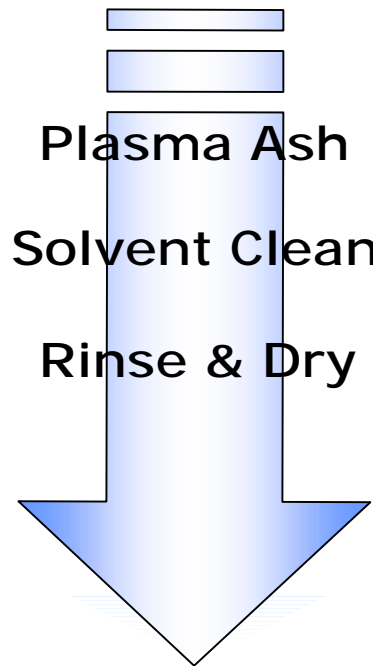
Development of stiction upon evaporation of the rinsing liquid.

MEMS: Process Results



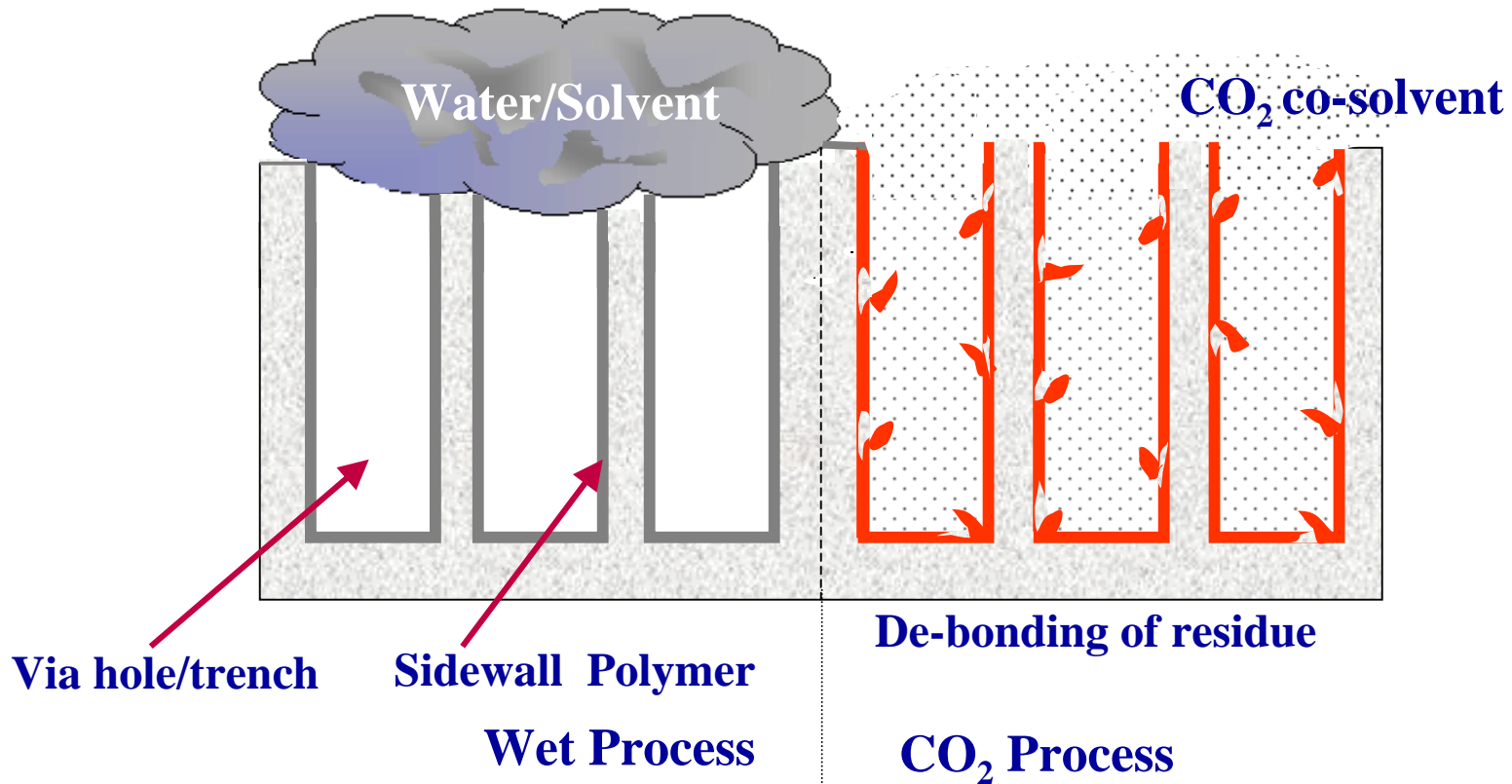


State-of-the-art Stripping Technology and Residue Removal



Enabling Technology ~ Why? Part 1

Photoresist Removal in a One-Step Process for 150nm Technology



<0.18 Features with Liquids

- ◆ Surface Tension
- ◆ Capillary Force

(Compositions)

◆ **Fluorinated Hydrocarbons**

◆ **Siloxanes**



◆ **Elemental Al or Refractory Metal or Cu**

Side-Wall Polymer/Residue Removal



(Chemistries)

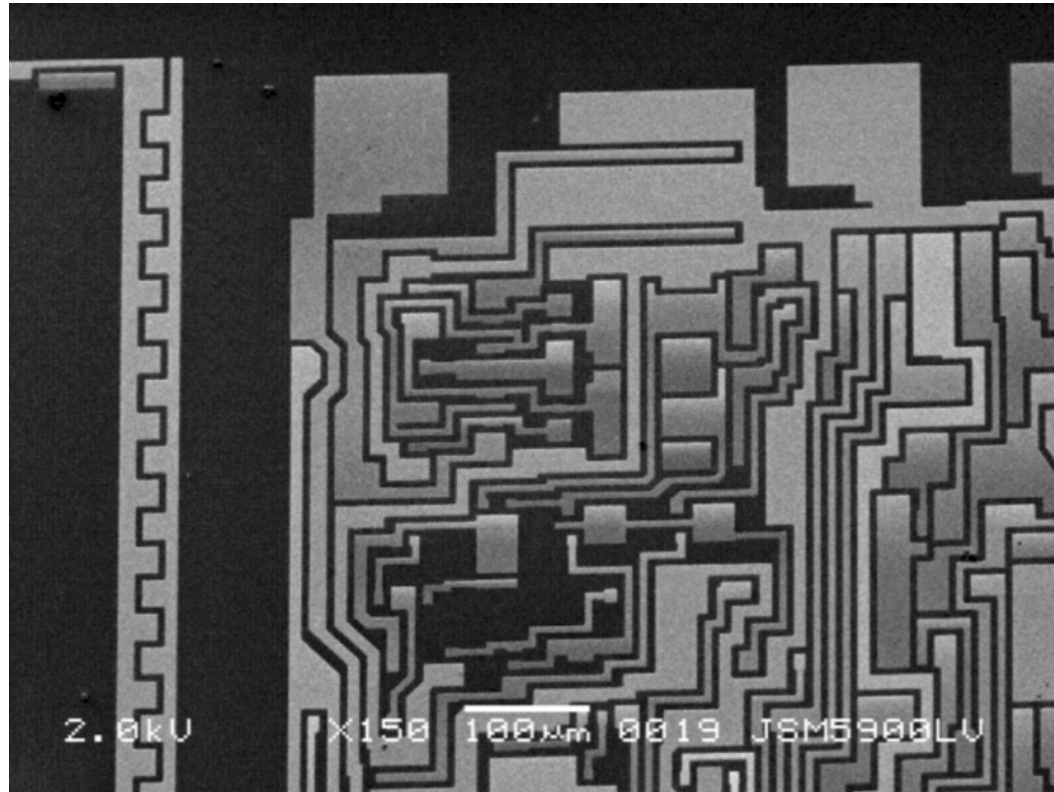
- ◆ **SCCO₂ + Co-Solvent**
- ◆ **SCCO₂ + Co-Solvent + Chelating Agent(s)**

Decompression Techniques

Process Results: Metal Pattern (TiW)

Prestrip: 5000 Å of resist

Result: Photoresist and sidewalls removed, no loading effect

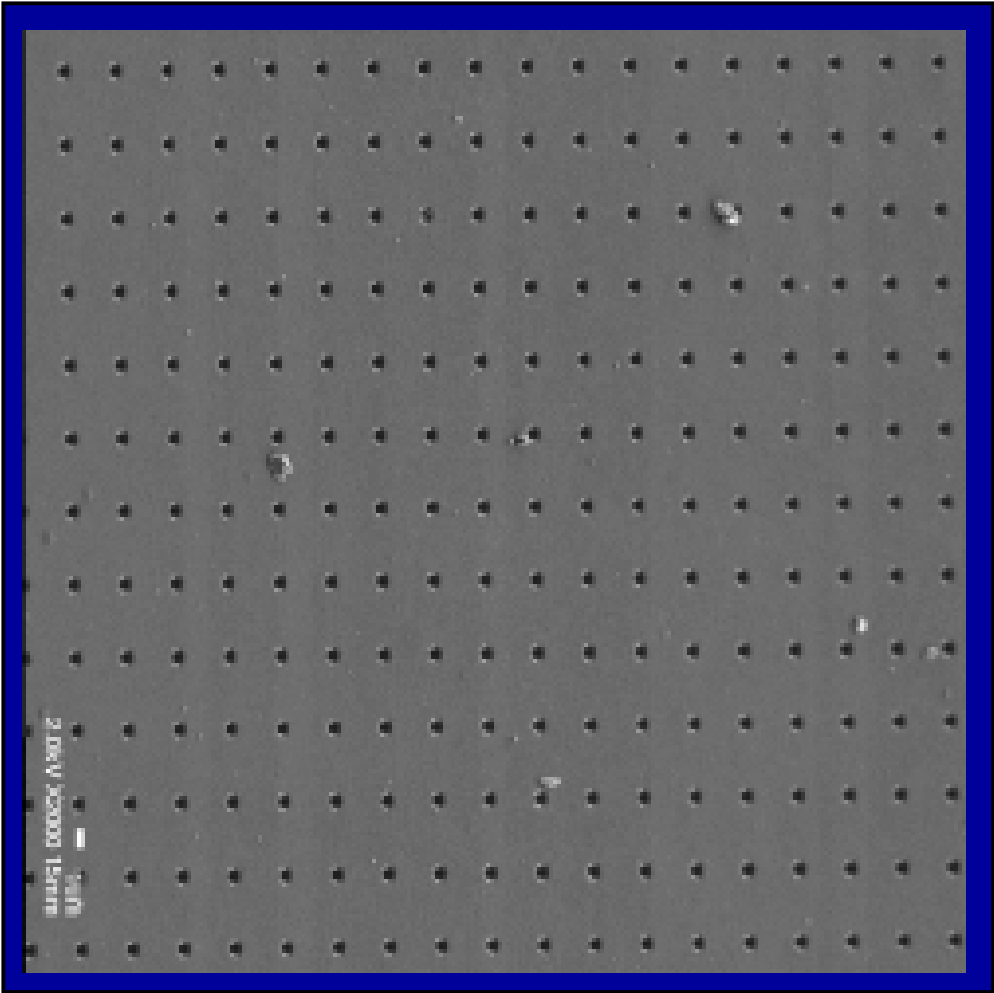


processed with CO₂

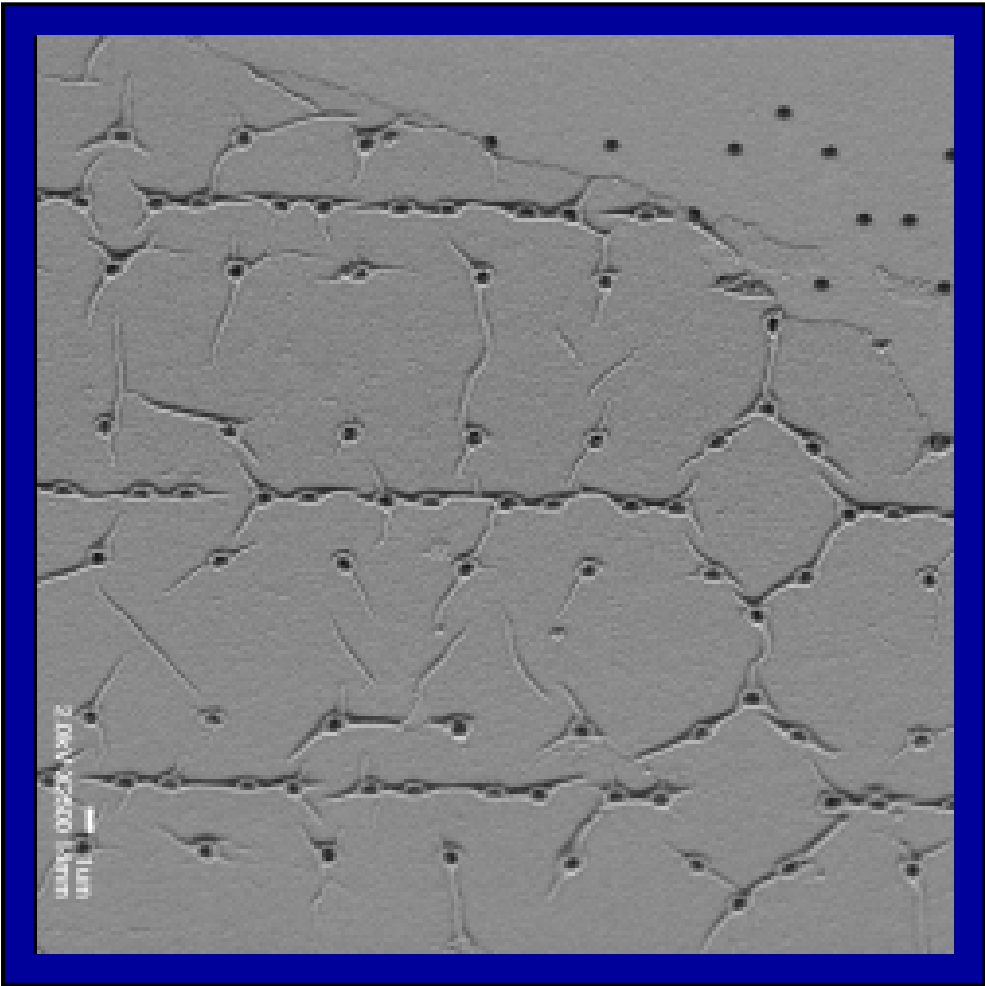
Oxide Via Field

4K Å Photo Resist

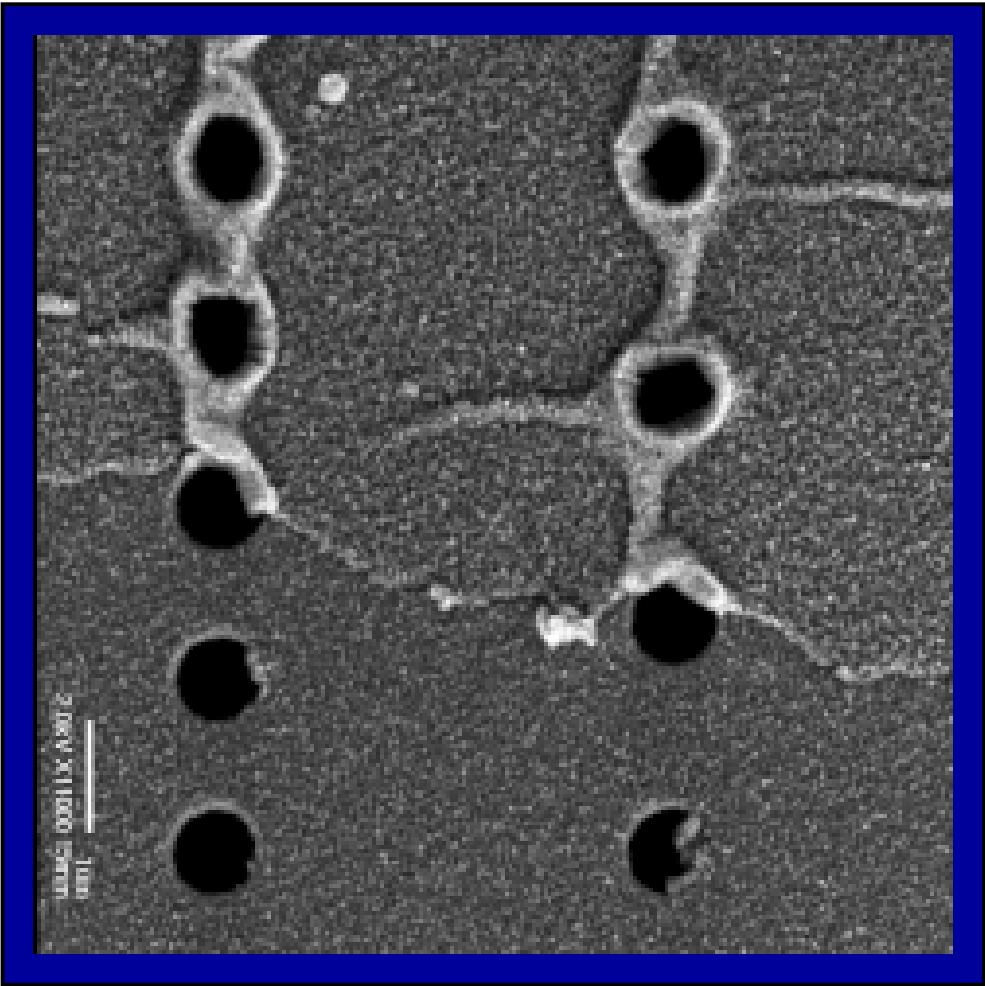
Pre-Strip



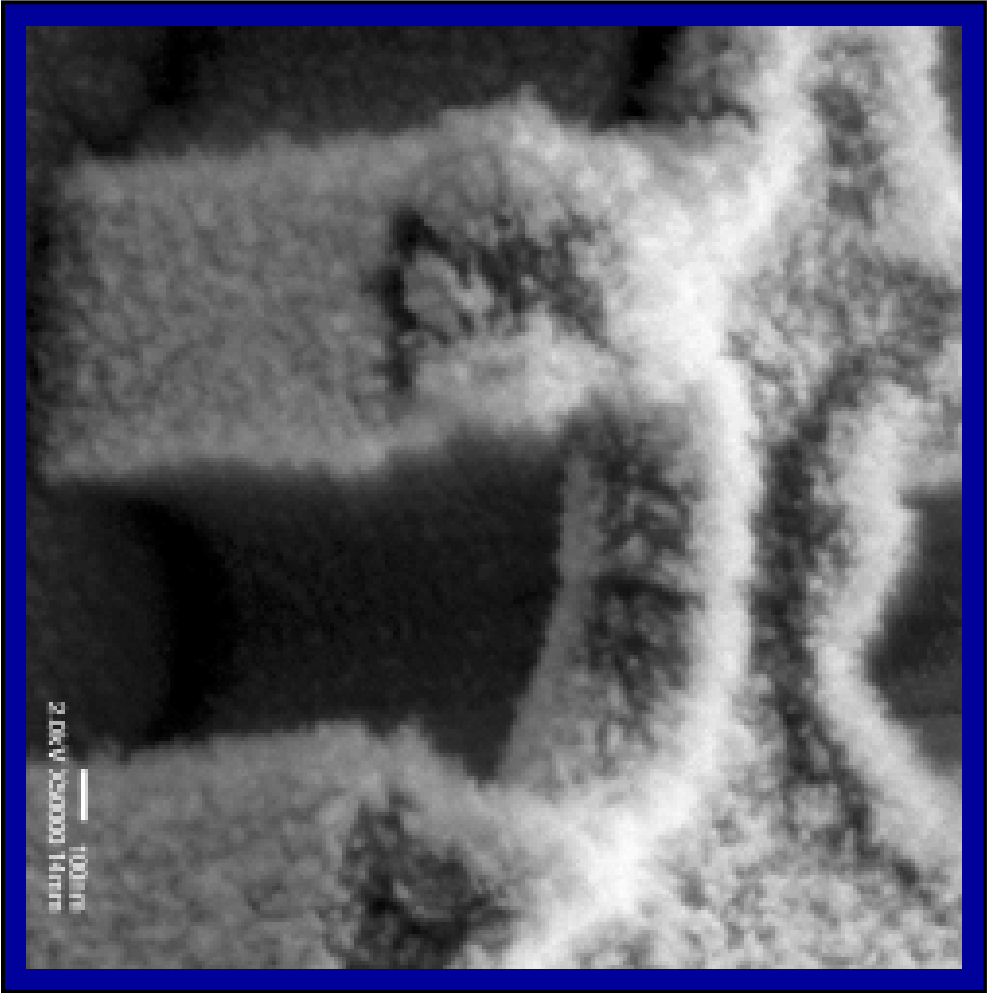
Oxide Via Field Partial Strip



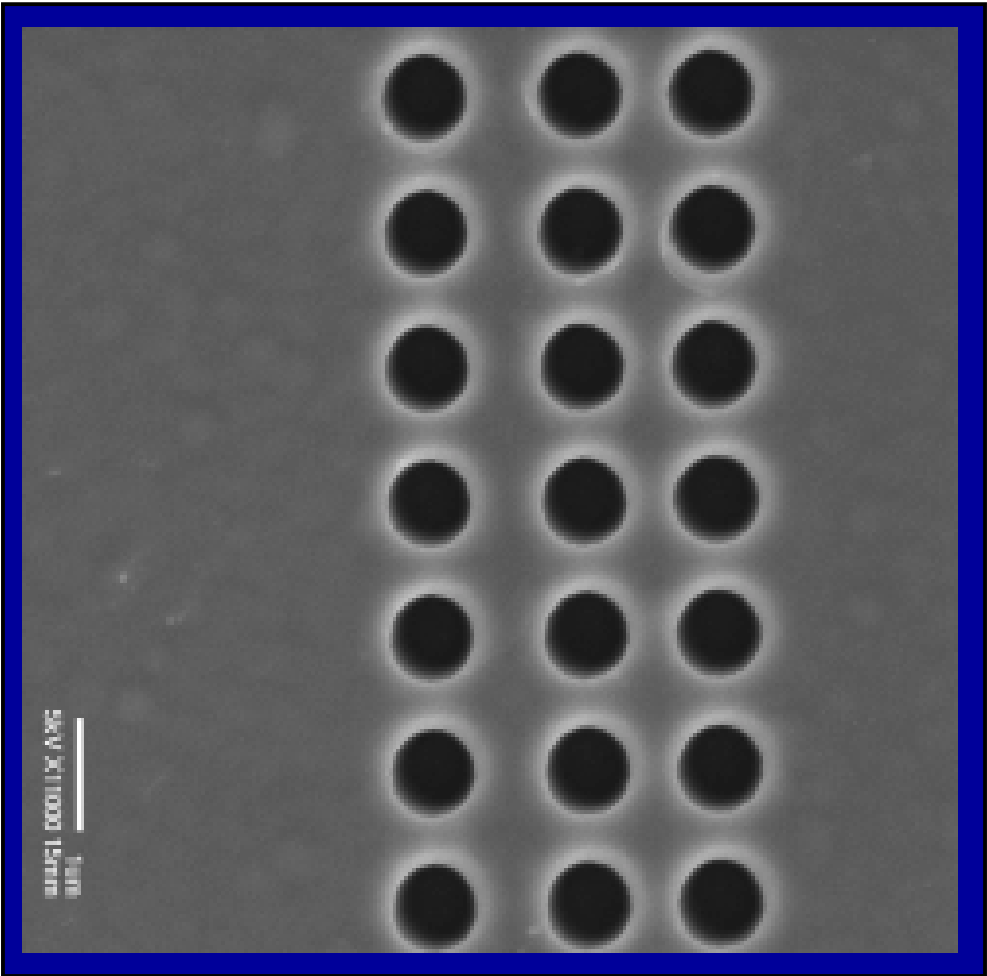
Oxide Via Field Partial Strip



Oxide Via Partial Strip

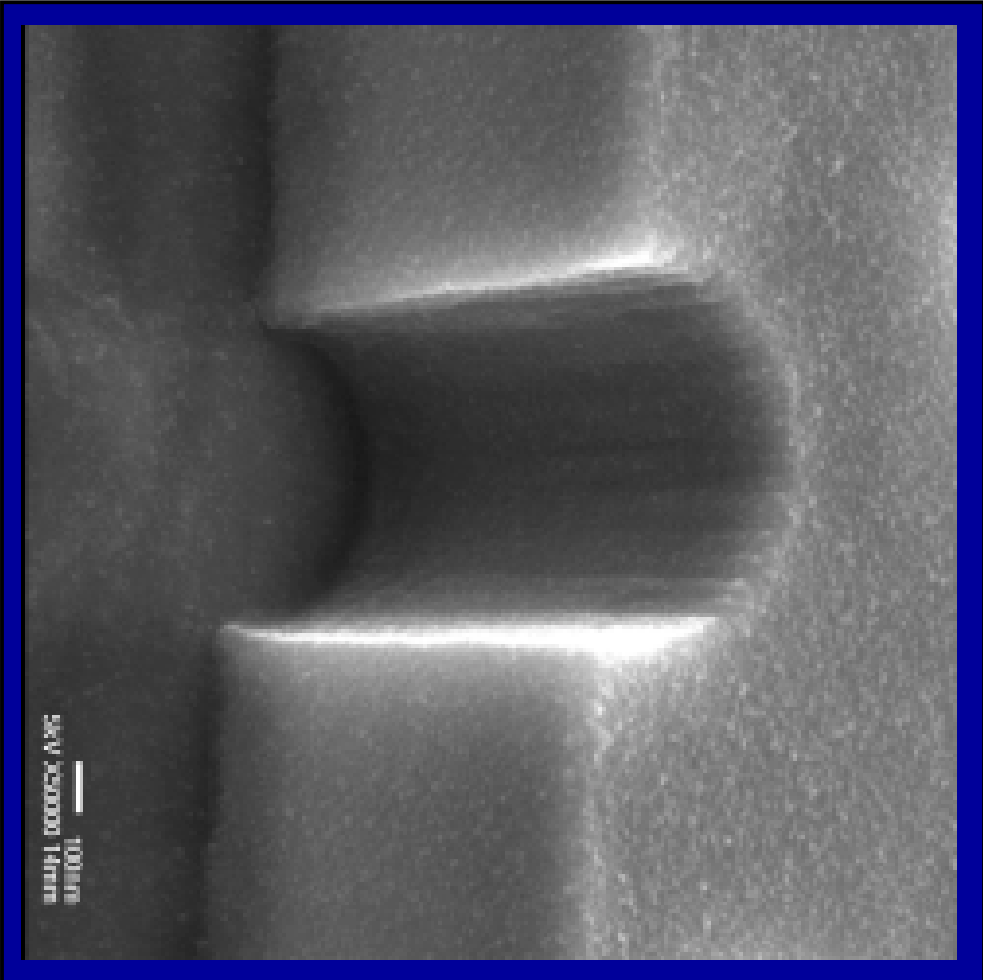


Oxide Via Field Complete Strip

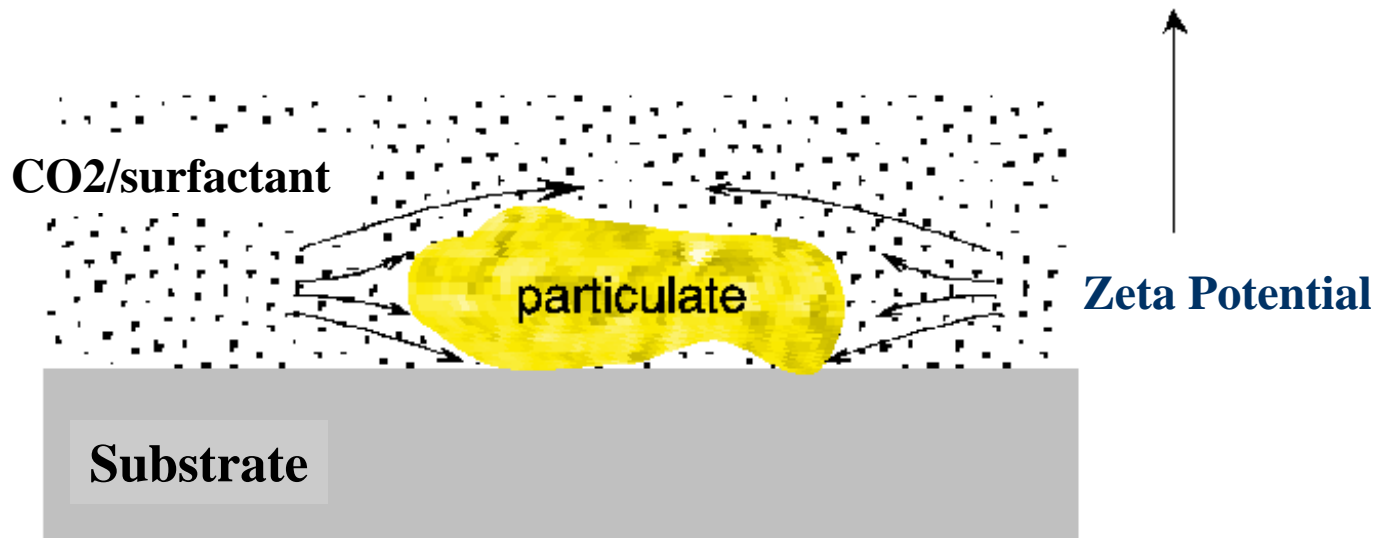


Oxide Via

- Resist Strip 4K Å
- Side-Wall Polymer Removed
- No Residue



Enabling Technology - Why? Part II



Changing Zeta Potential

Examples of Particle Morphology



Wafer Slot #25

Wafer Analysis Slot #	Particle #	Post X	Post Y	Tencor Size	SEM Size (um)	Gain	Content (SEM/EDX)
Box D Slot25	1	77753	189956	268.6	7.26	normal	Si, O
	2	35639	134713	382.4	41.4	expand	Si, C
	3	39299	71064	268.6	7.14	normal	Si
	4	32584	33351	258.6	9.46	expand	Si, C
	5	131021	54587	280.7	6.21	normal	Si, C
	6	134082	142740	261.1	12.43	expand	Si, Fe, Cr, O, Ni
	7	124651	158035	277.1	5.13	normal	Si, Fe, Cr
	8	125697	171080	324.4	9.87	expand	Si, O
	9	115564	155890	273.5		expand	Si, C
							Not Found

Slot#25 Particle #1



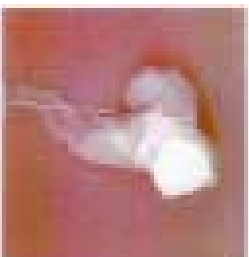
Slot#25 Particle #2



Slot#25 Particle #3



Slot#25 Particle #4



Slot#25 Particle #5



Slot#25 Particle #6



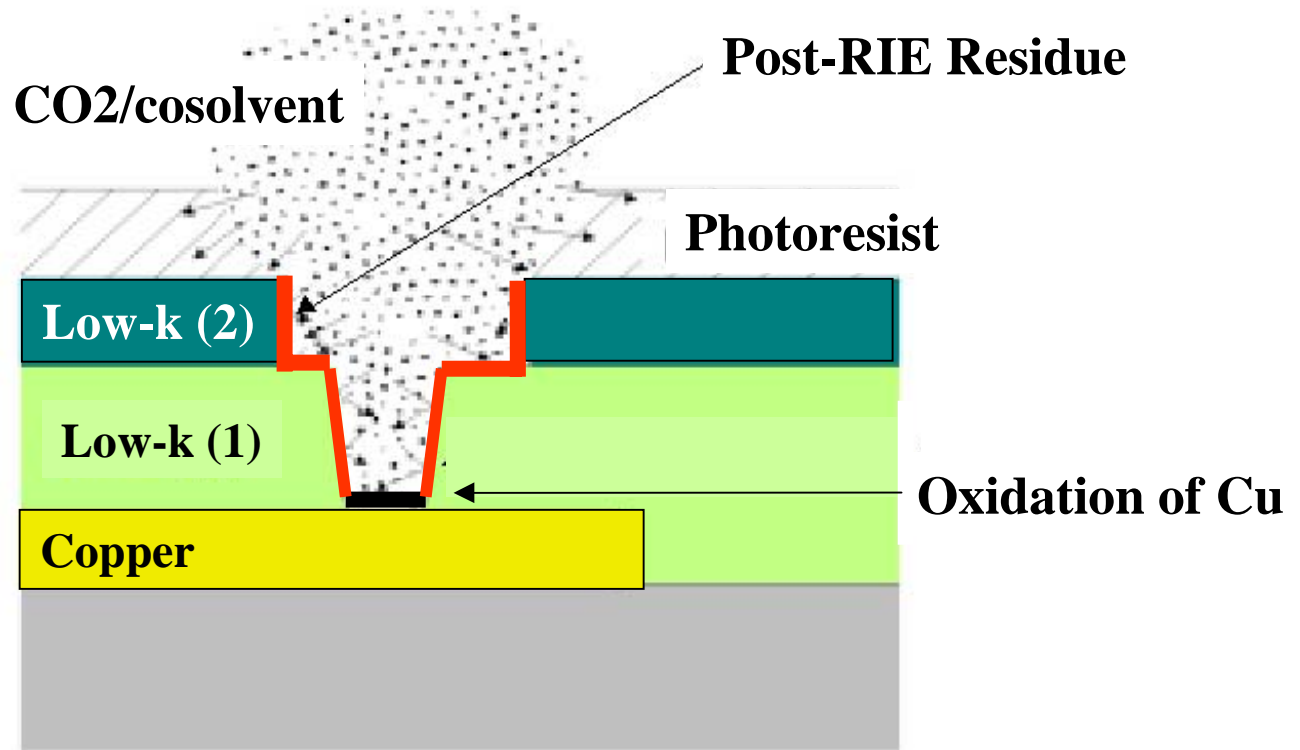
◆ “Engineered” Surfactants to Change Zeta Potential

R & D Required

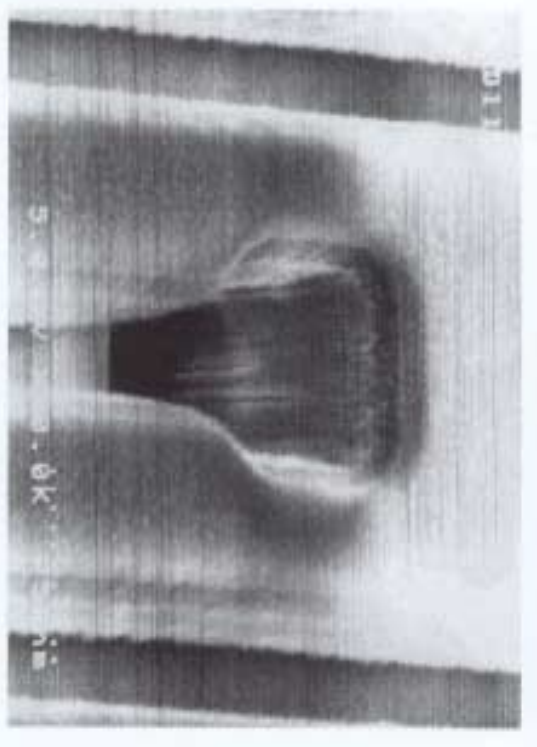
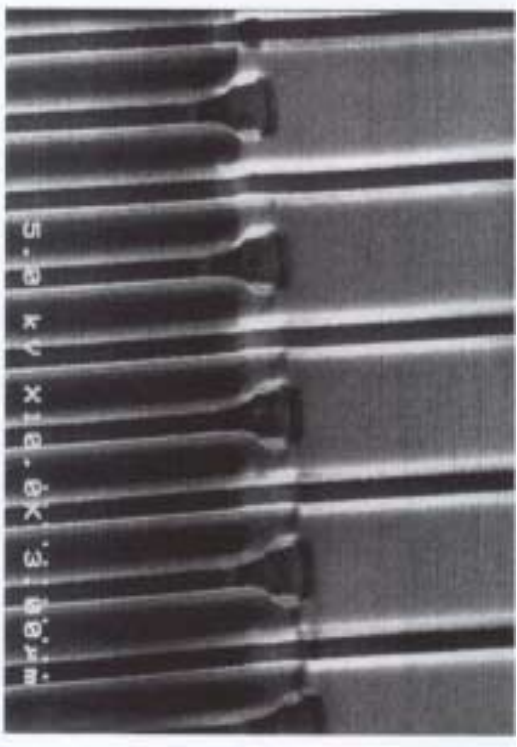
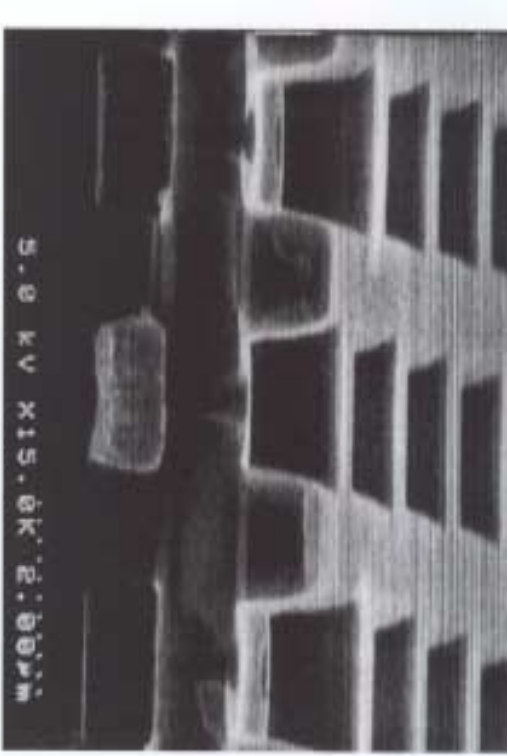
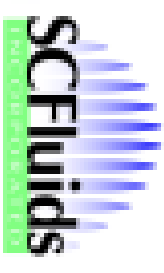
- ◆ Screen and Select Cationic and Amphoteric Surfactants
- ◆ Study Surface Forces
- ◆ Electro Kinetics and Colloid Behaviors of Surfactants

Enabling Technology - Why? Part III

Compatibility/Selectivity With Low-k Technology



Side Wall Removal - Process Matrix

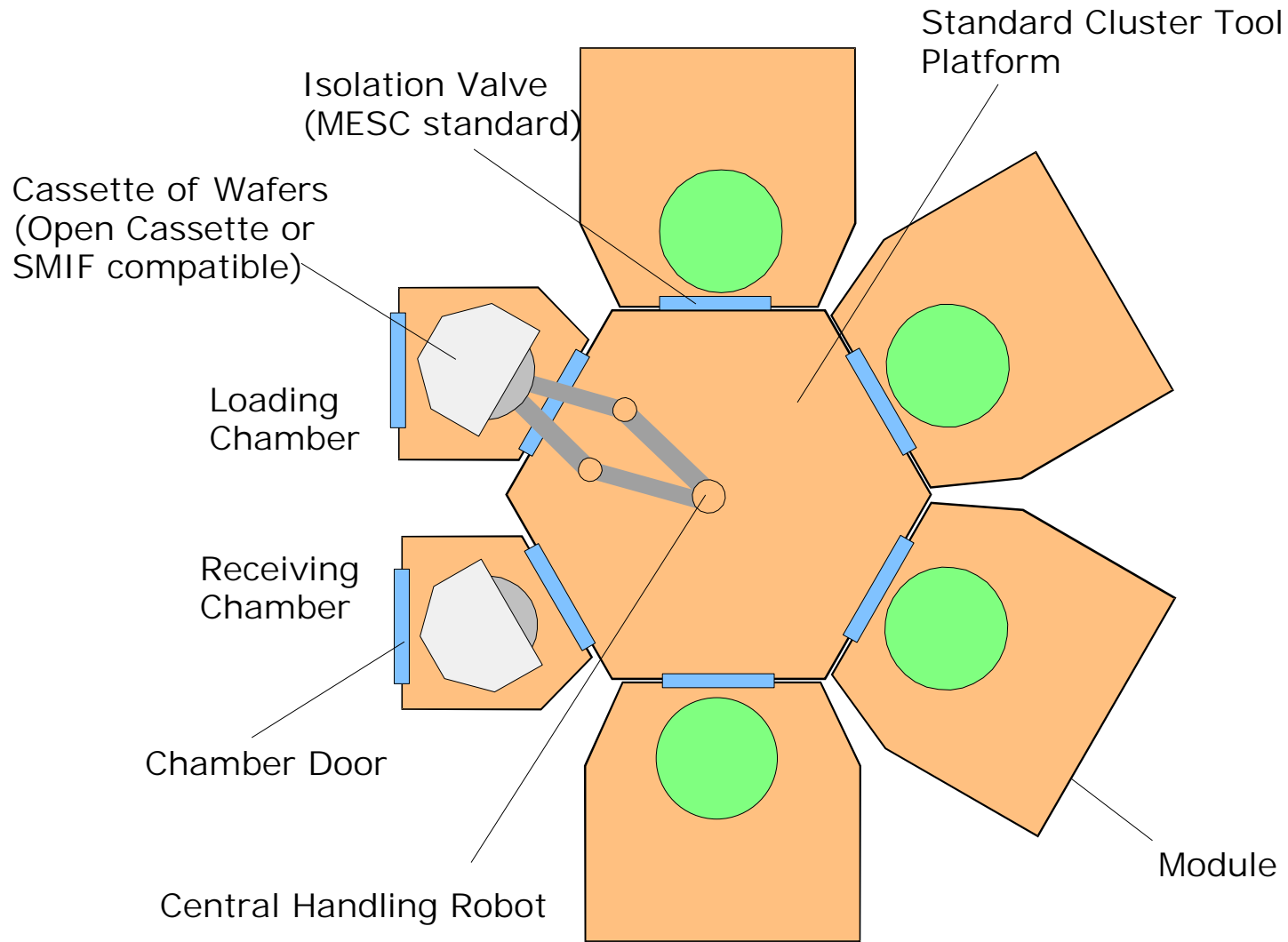


processed with CO₂

General Comments:

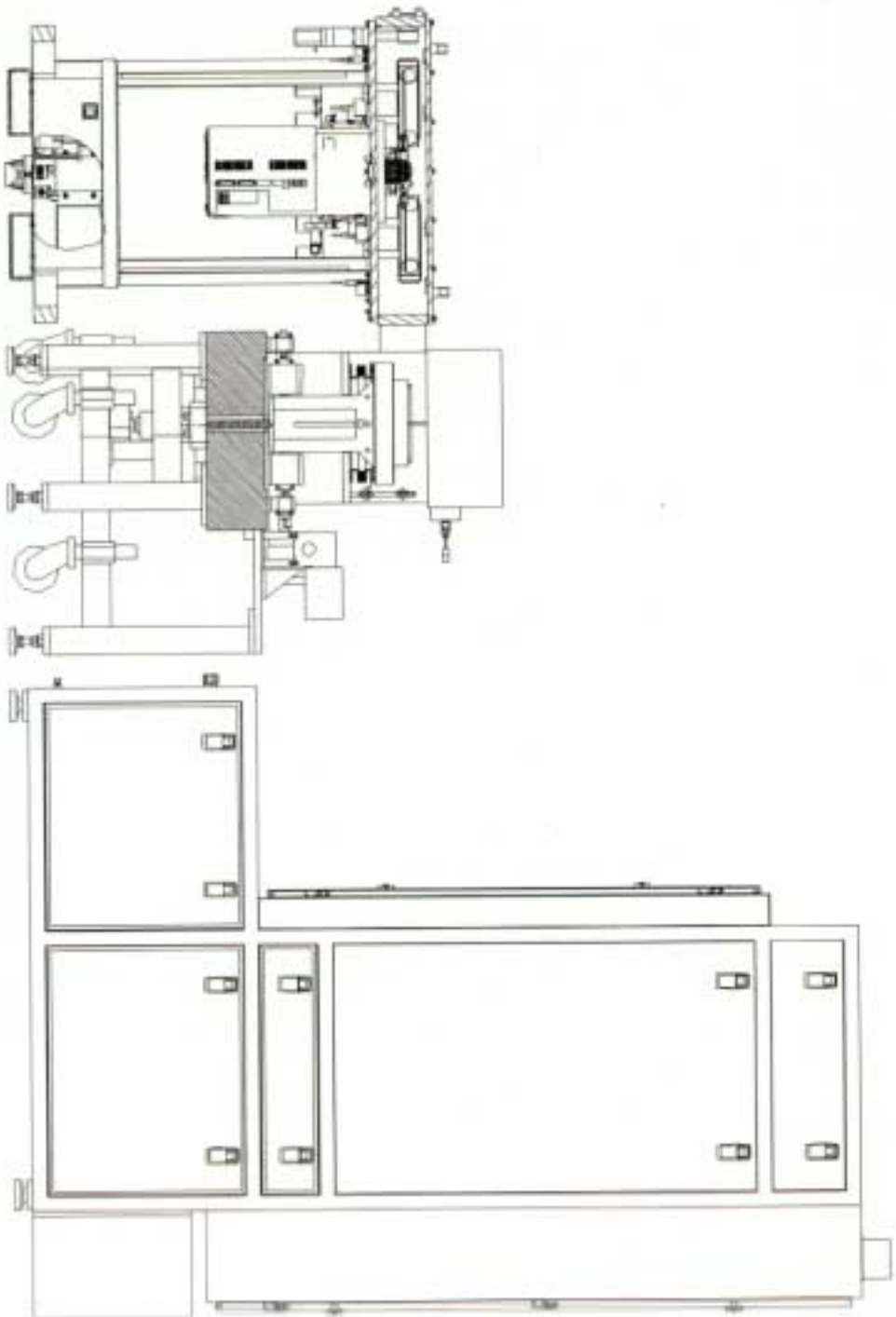
- ◆ Many substances dissolve in SCCO₂ (principally organic)
- ◆ Both organic acid and bases are soluble in SCCO₂
- ◆ Mixtures can be made with “other” supercritical fluids and organic and metal-organic compounds, in combination(s)
- ◆ Solubility of these compounds (parameters) are generally 1 - 8%
- ◆ Mixtures move the “critical values”, or the critical point of CO₂, (T crit and P crit), to higher values.

Manufacturing Tool Concept





Side View of ARROYO™ System

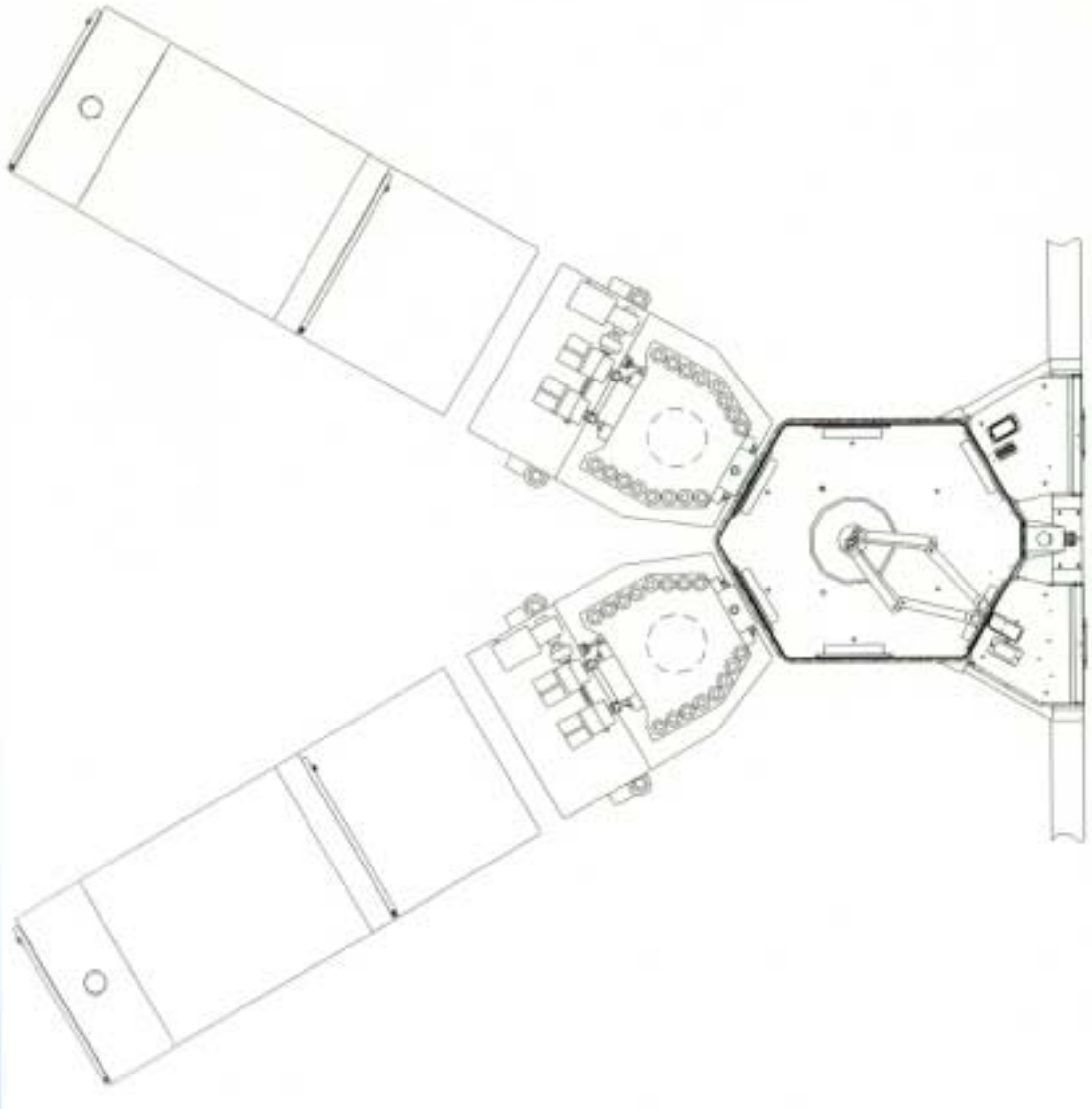


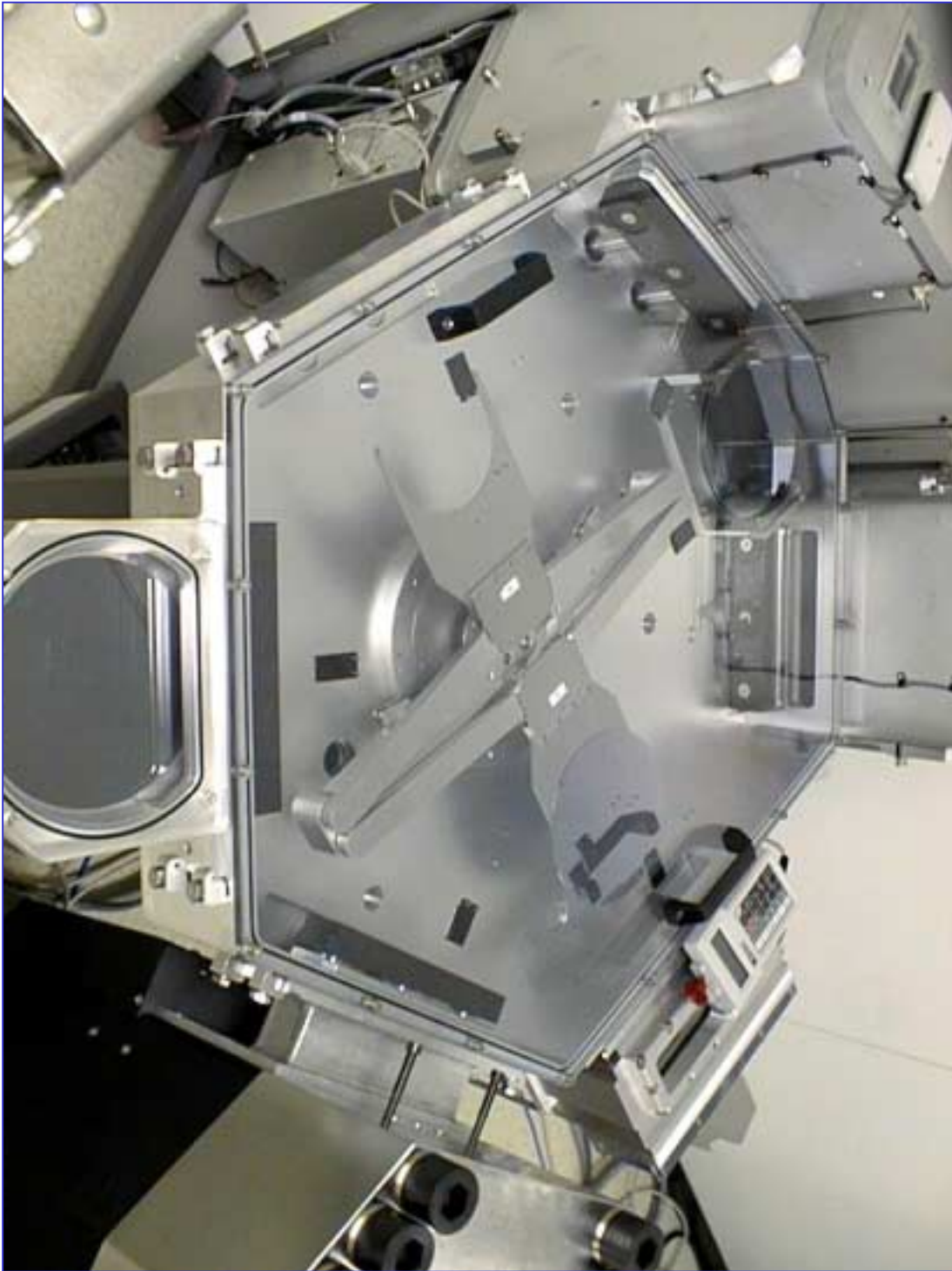
TRANSPORT MODULE

PROCESS MODULE

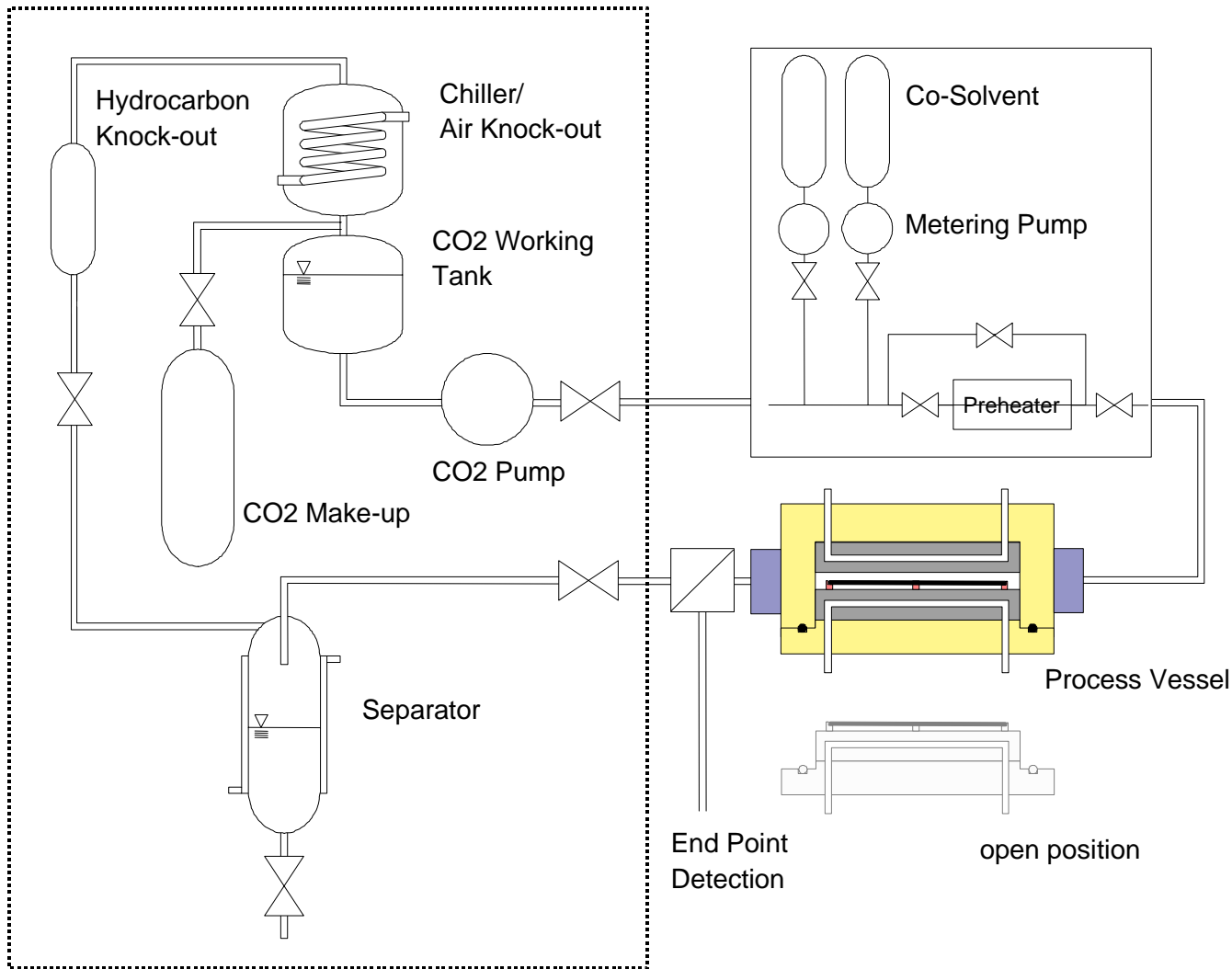
DELIVERY SYSTEM

Top View of ARROYO™ System





Process & Instrumentation Diagram



APPLICABILITY OF SC FLUIDS FOR 300mm WAFER:

Since SCCO_2 Technology is NOT Plasma or Vacuum-Based:

- ◆ Isostatic Pressure (No ΔP on Wafer)
- ◆ No RIE Lag
- ◆ No Loading Effects
- ◆ No Uniformity Gradients
- ◆ No Mass Transport Issues

Benefits of SCCO₂ Technology



- ▶ Benign Process Temperature (50-150°C)
- ▶ Non-toxic/non-flammable process media
- ▶ No environmental or employee health hazard
- ▶ No hazardous waste disposal
- ▶ No water consumption
- ▶ Small footprint
- ▶ Extendable to 300 mm wafers
- ▶ Dry-to-dry process
- ▶ Applicable to photoresist stripping, side wall polymer, particle removal, and deposition (future work)
- ▶ Viable for low-k dielectrics (no change in "k" value)
- ▶ Enabling technology for <math><0.18\mu\text{m}</math> structures

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