

A Surface Chemistry Approach to the Development of Gas Phase Wafer Cleaning Processes

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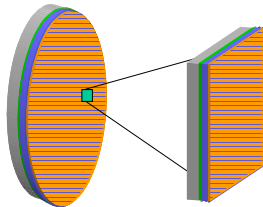
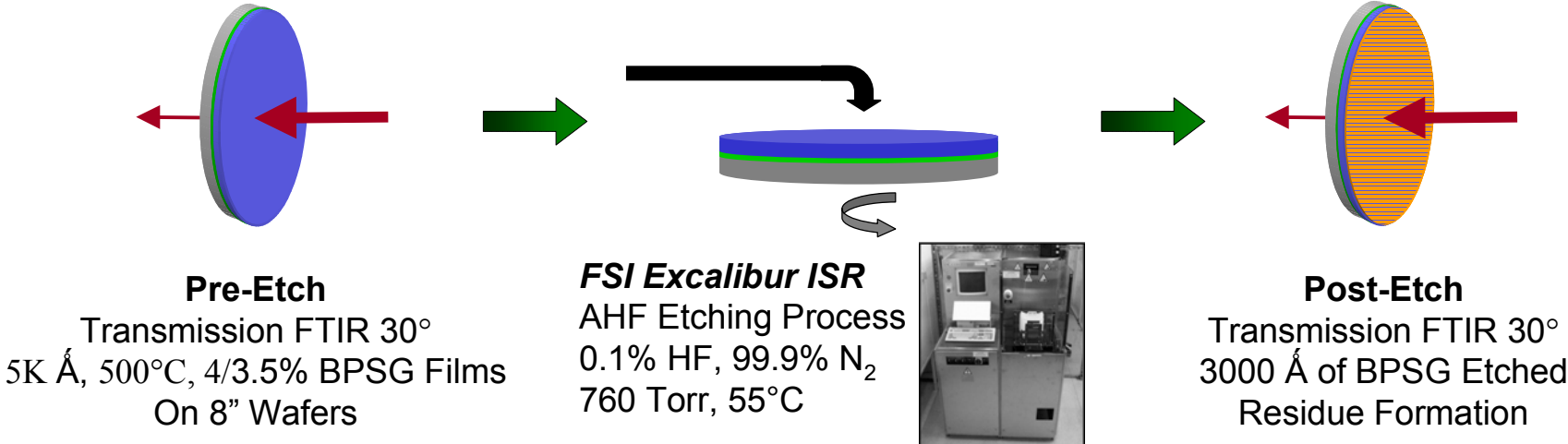


*NSF/SRC Engineering Research Center for Environmental Benign
Semiconductor Manufacturing, Thrust B Teleconference September 6, 2001*

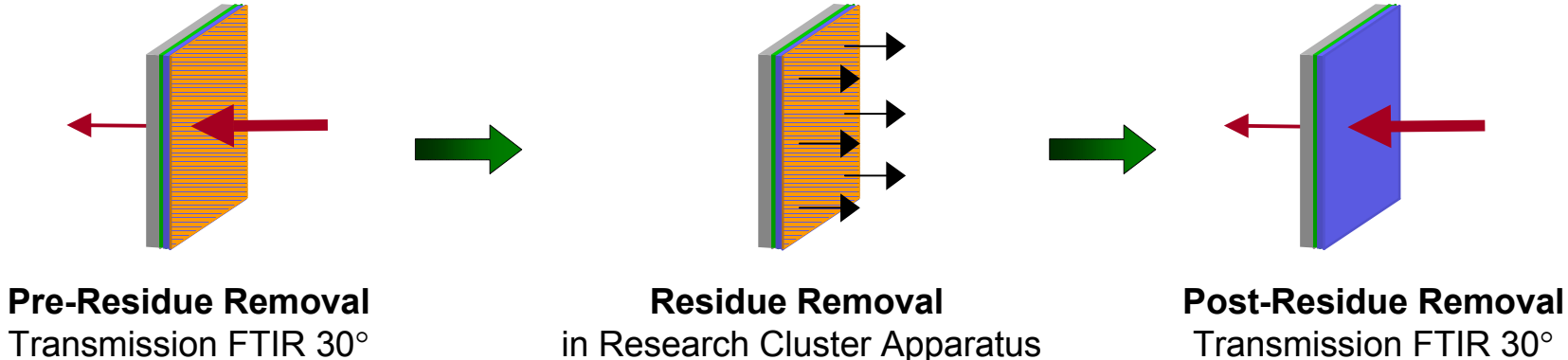
Outline

- Gas phase removal of doped oxide etching residues using UV-Cl₂ processes
- Mechanism of thermal oxide removal using low pressure gas phase HF/H₂O mixtures
- Future work

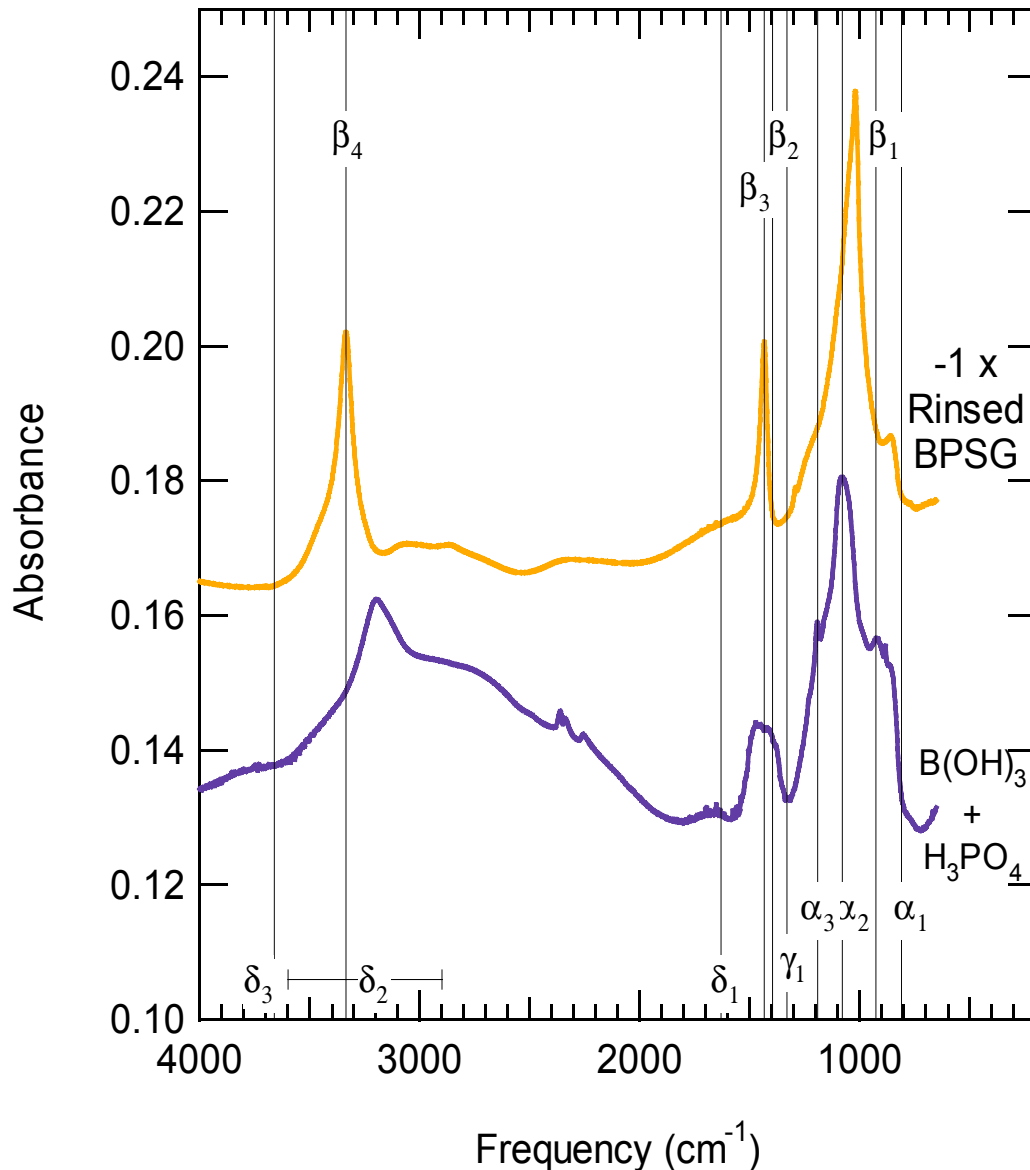
Residue Removal Process Flow



Dice 8" wafer into 1 x 1 cm pieces



Mixed Acid on Etched BPSG

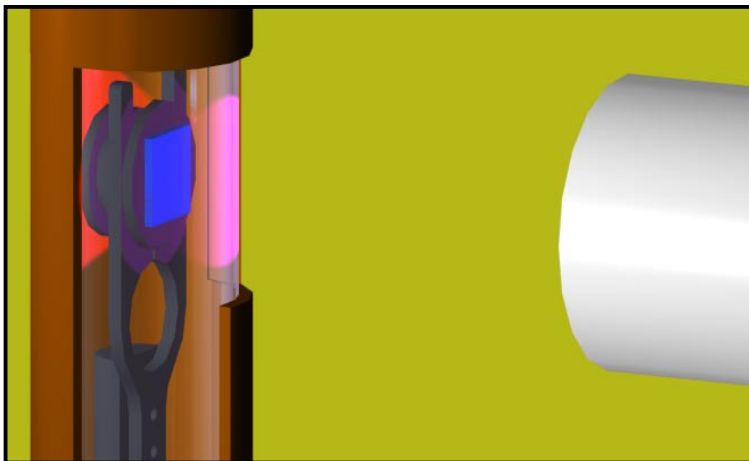
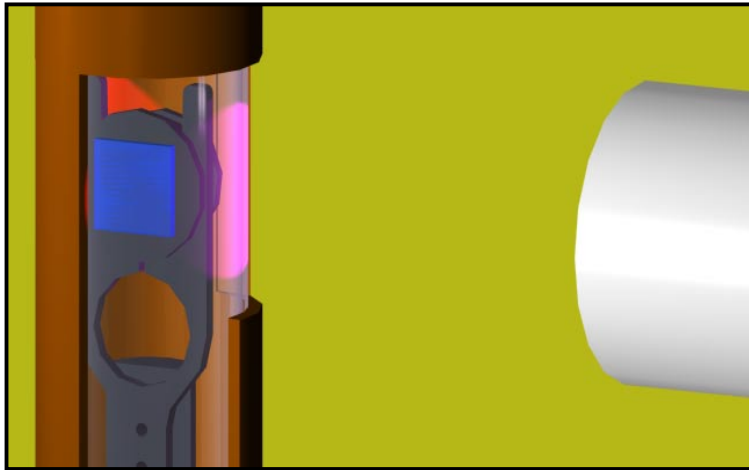


- Residue after etching and rinsing BPSG film compares closely with $B(OH)_3 + H_3PO_4$ mixture
- Peak Shift β_4
- Chemistry
 - $P_2O_5 + 3H_2O(l) = 2H_3PO_4(l)$
 - $B_2O_3 + 3H_2O(l) = 2B(OH)_3(a)$



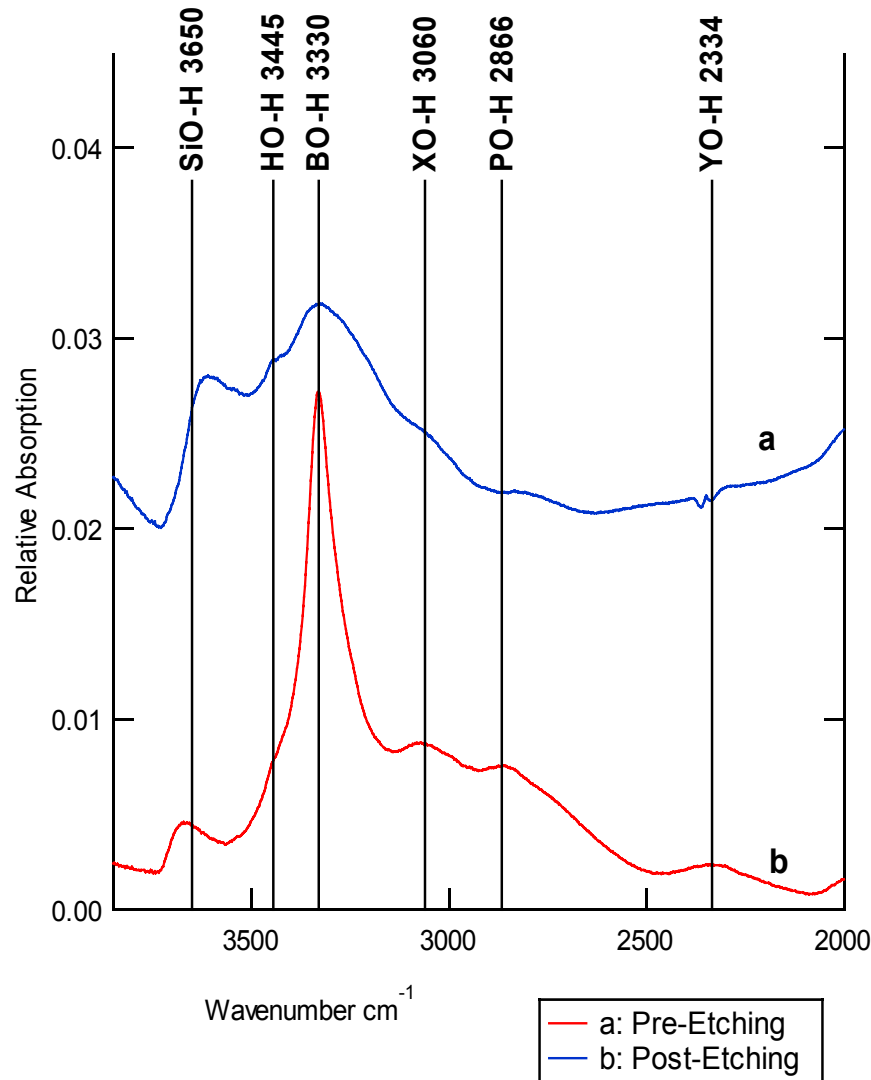
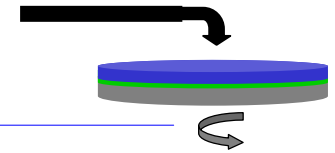
Muscat et al., JVSTA 19(4), 1854 (2001)

UV / Cl₂ Reactor



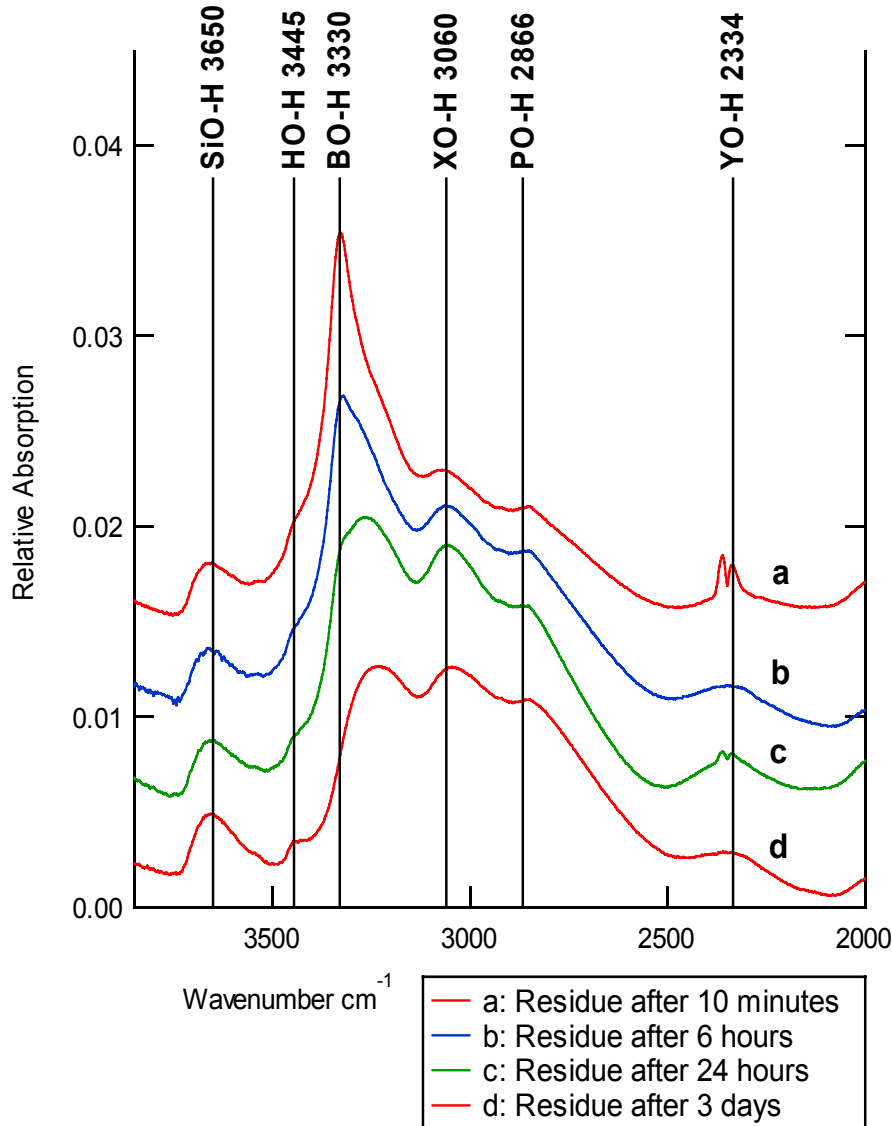
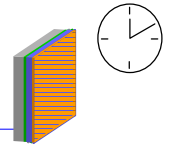
- Quartz Tube
 - ID = 33 mm
- Thermally Insulated
- Processing Parameters
 - 750 mTorr
 - 100 sccm
 - 10 – 35 mol% Cl₂
 - $\tau = 1$ s
 - 20 – 350°C
- 1000 watt high-pressure Xe Arc Lamp

Residue Formation



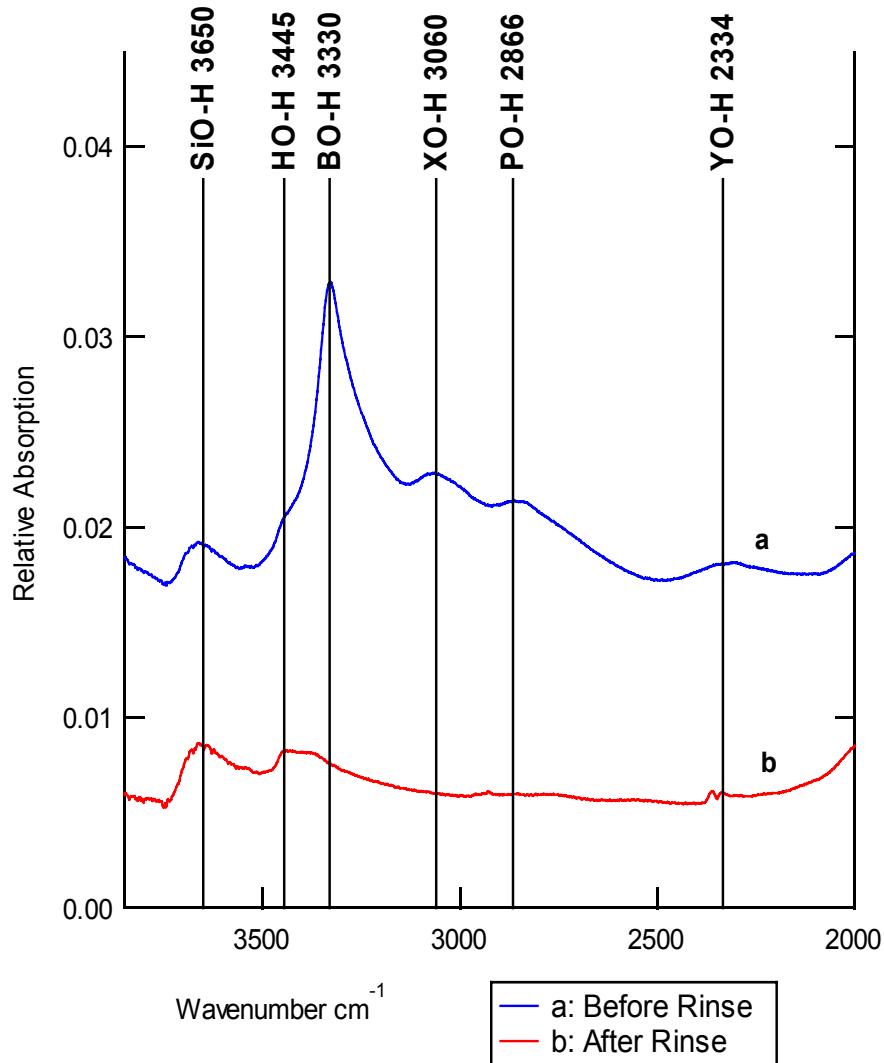
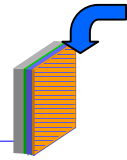
- 500°C Annealed BPSG
- FSI Excalibur ISR
 - AHF Etching Process
 - 760 Torr
 - 0.1% HF
 - 99.9% N₂
 - 55°C
- ~ 3000 Å oxide etched
- ~ 1000 Å thick mixed acid residue

Residue Change Over Time



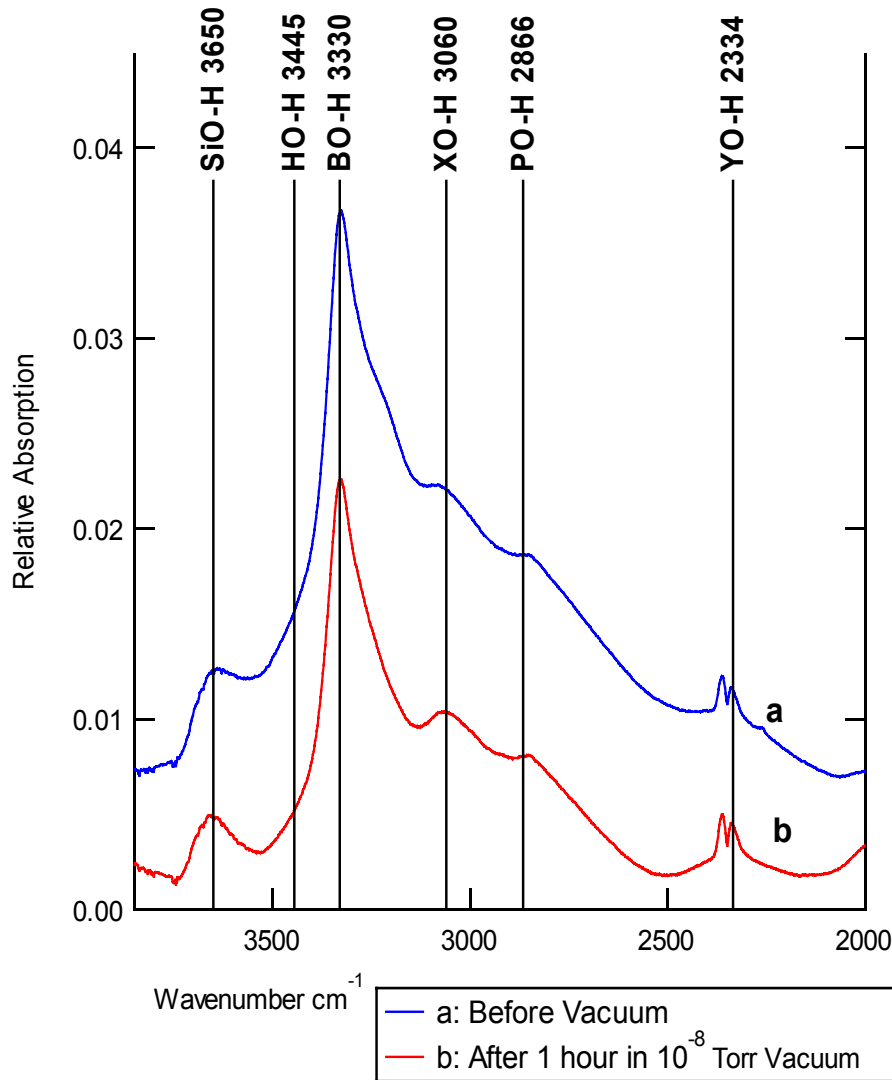
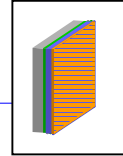
- Time
 - 10 minutes
 - 6 hours
 - 24 hours
 - 3 days
- Cleanroom
 - 760 Torr
 - 23°C

Rinsing The Residue



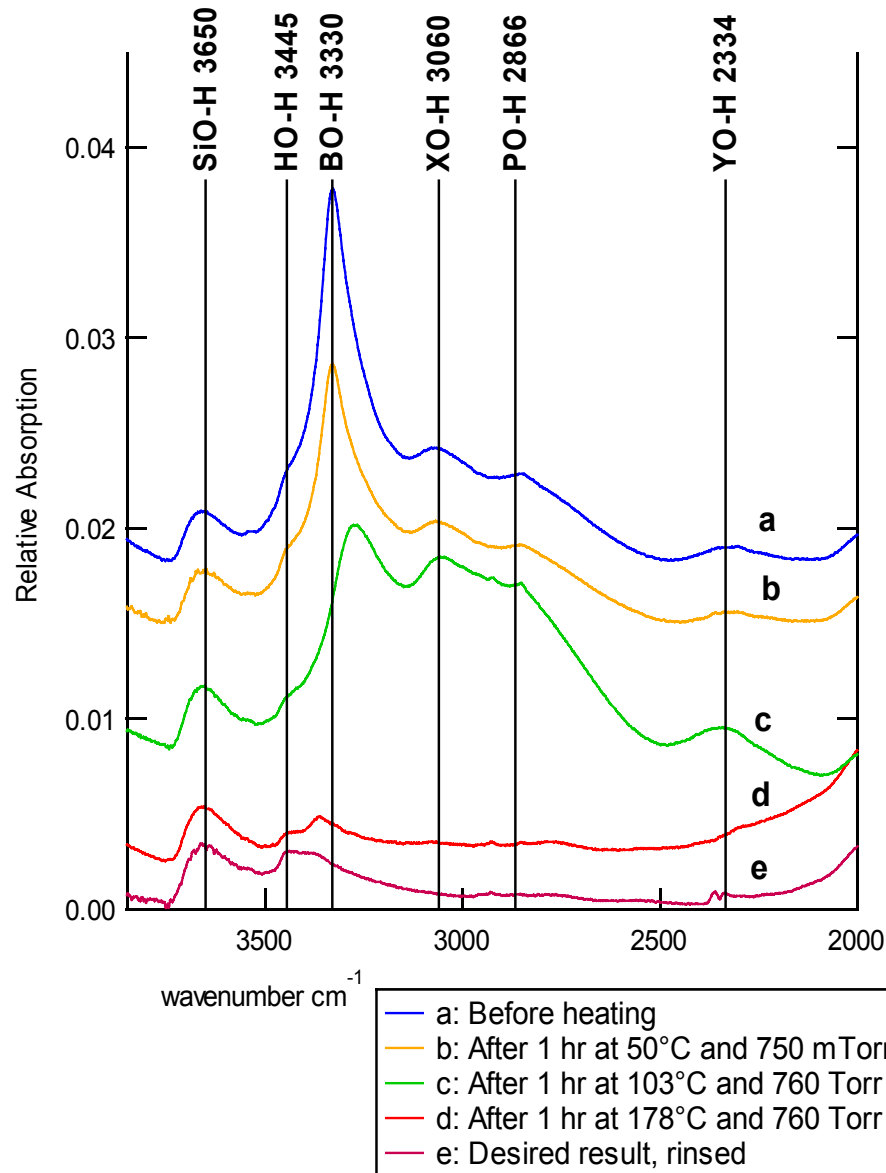
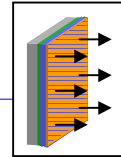
- Current industrial residue removal process
 - UPW rinse
 - UPW displacement with alcohol and super critical CO₂
- Rinsed with UPW
- No cloudy residue remains after process

Effect of Vacuum Alone

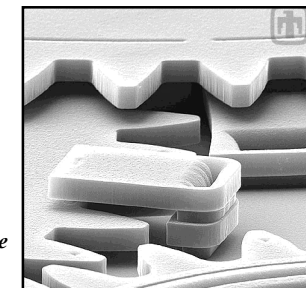


- Vacuum
 - 1 hour
 - 10^{-8} Torr
 - 22°C
- Cloudy residue remains after vacuum exposure

Removal By Heating

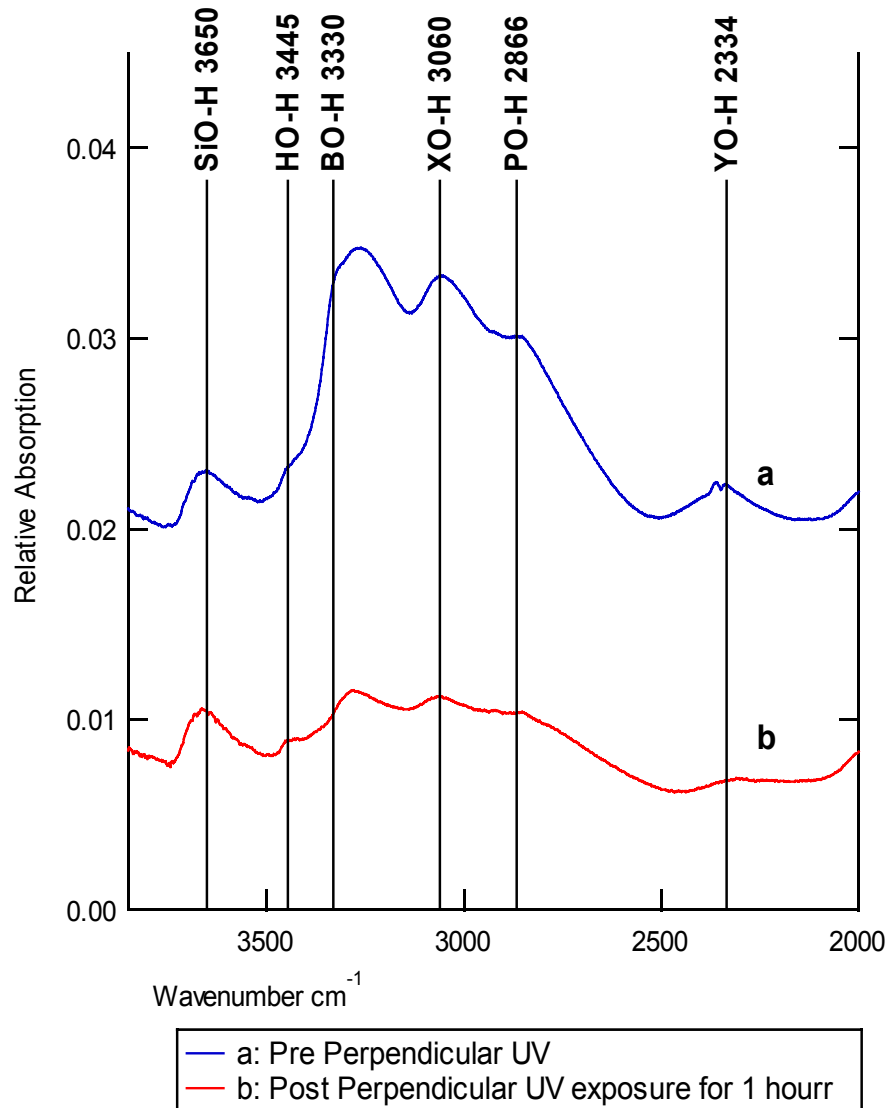
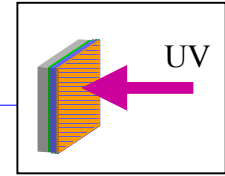


- Process
 - 1 hour
 - a \rightarrow b : 50°C, 750 mTorr N₂
 - a \rightarrow c : 103°C, 760 Torr
 - a \rightarrow d : 178°C, 760 Torr
- Physical properties
 - Phosphoric acid
 - 85% b135°C, Pure d70°C
 - Boric acid
 - m165°C, d185°C
- Cloudy residue remains after the heating process
- Evaporation of residue creates stiction
 - (A. Witvrouw *et al.*, 2001)



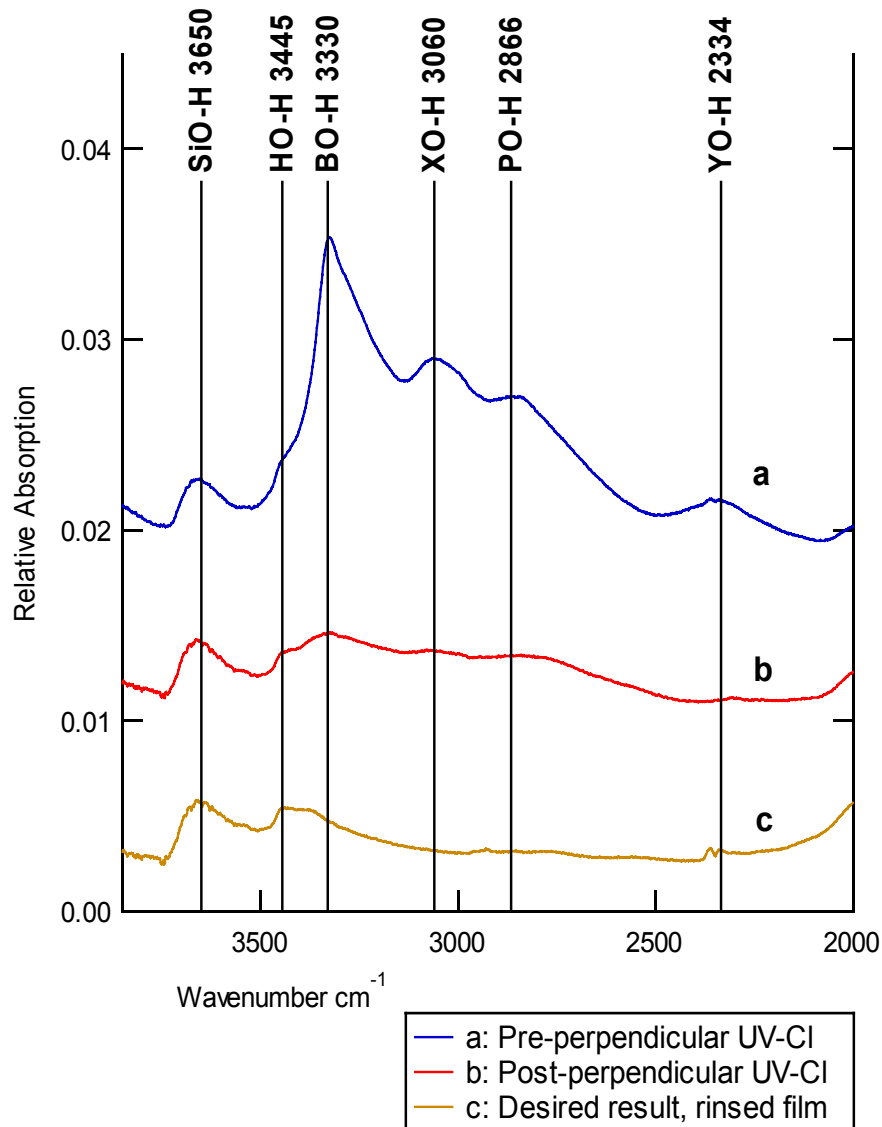
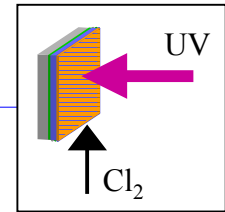
Courtesy of:
Sandia National Laboratories
Intelligent Micromachine initiative
www.mdl.sandia.gov

Perpendicular UV

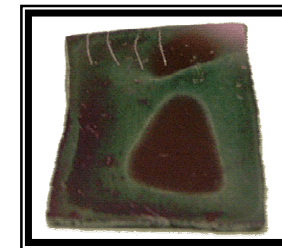


- UV Only Process
 - 1 hour
 - 750 mTorr
 - 100% N_2
 - 100°C
- Cloudy residue remains after UV only process

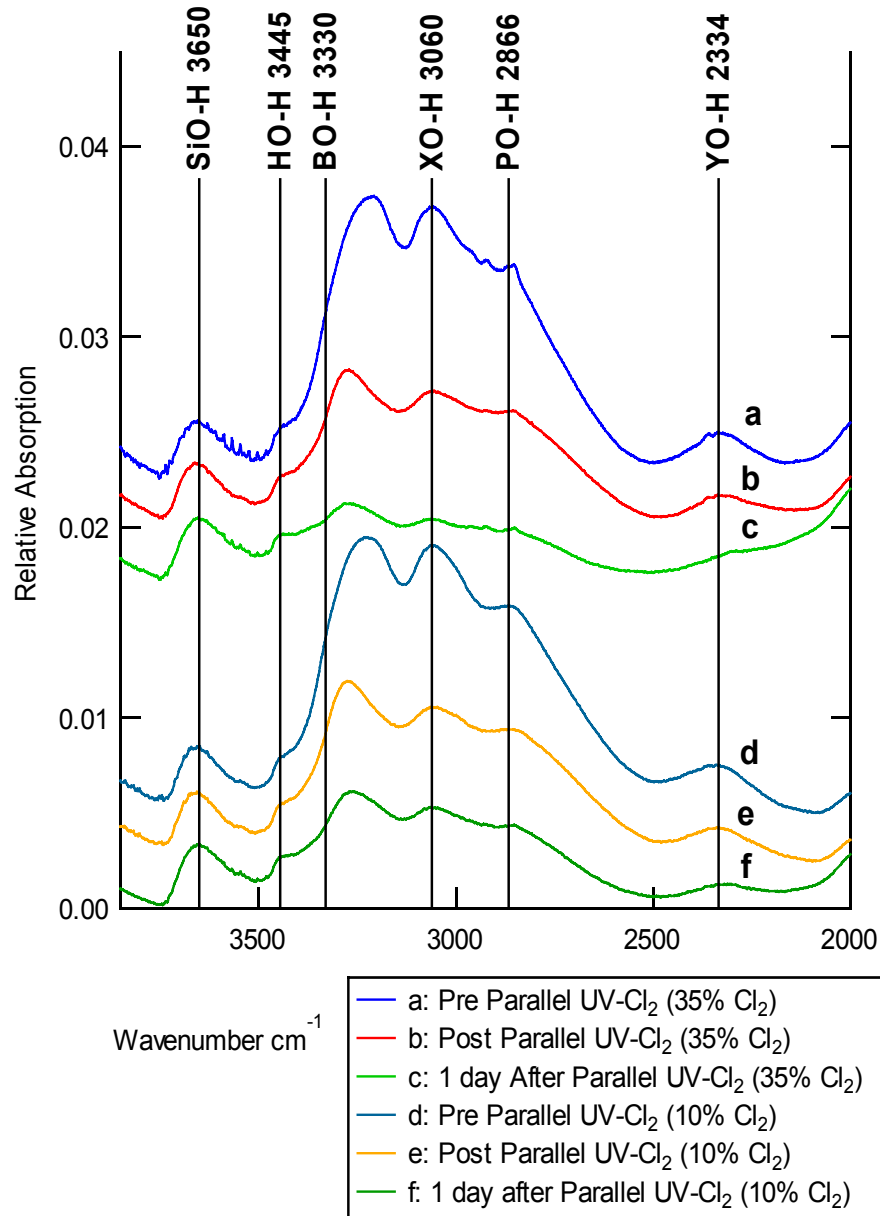
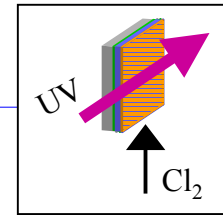
Perpendicular UV-Cl₂ Process



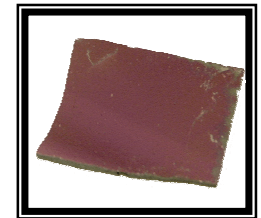
- UV-Cl₂ Process
 - 1 hour
 - 750 mTorr
 - 90% N₂, 10% Cl₂
 - 150°C
- Unrinsable opaque residue remains after process



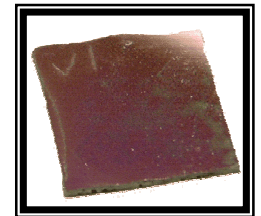
Parallel UV-Cl₂ Processes



- Process
 - 1 hour, 50°C, 750 mTorr
- a → b (35% Cl₂)
 - 47.5% reduction in peak envelope
- b → c (35% Cl₂)
 - 1 day in cleanroom
 - 80% reduction in peak envelope
- d → e (10% Cl₂)
 - 38.4% reduction in peak envelope
- e → f (10% Cl₂)
 - 1 day in cleanroom
 - 63.5% reduction in peak envelope

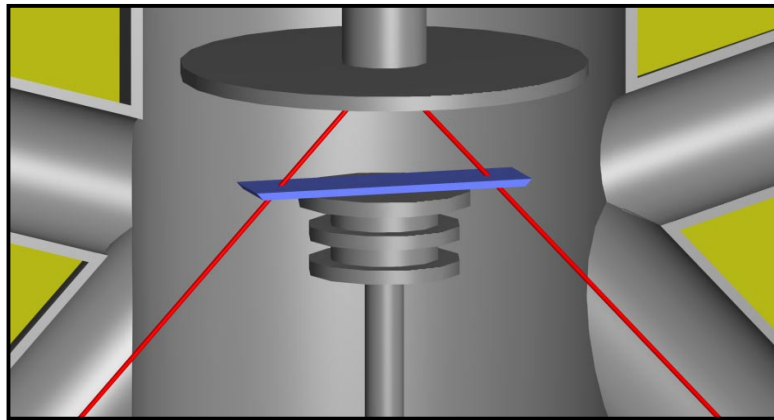


After b

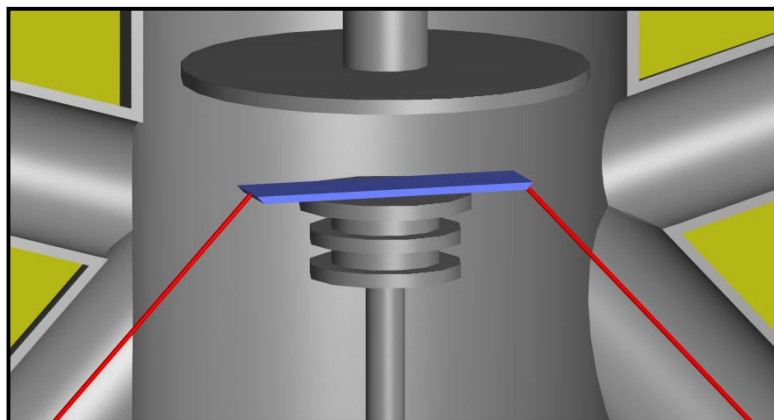


After e

HF/Vapor Etching Reactor



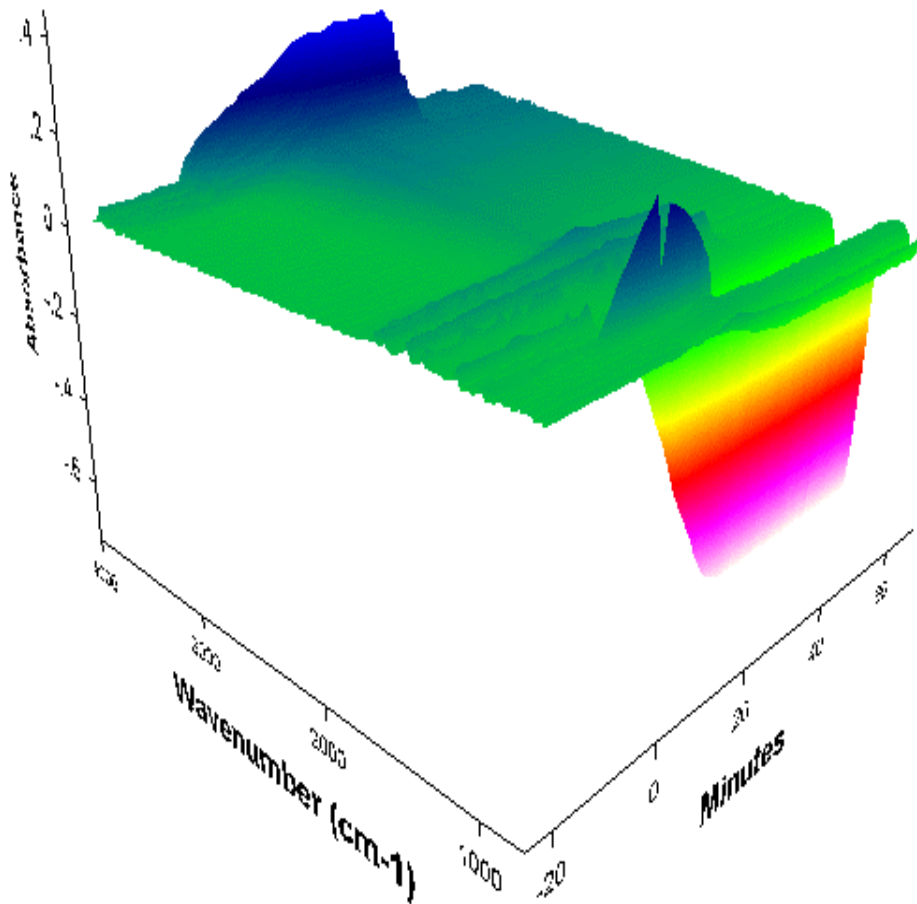
45° Transmission Mode



ATR Mode

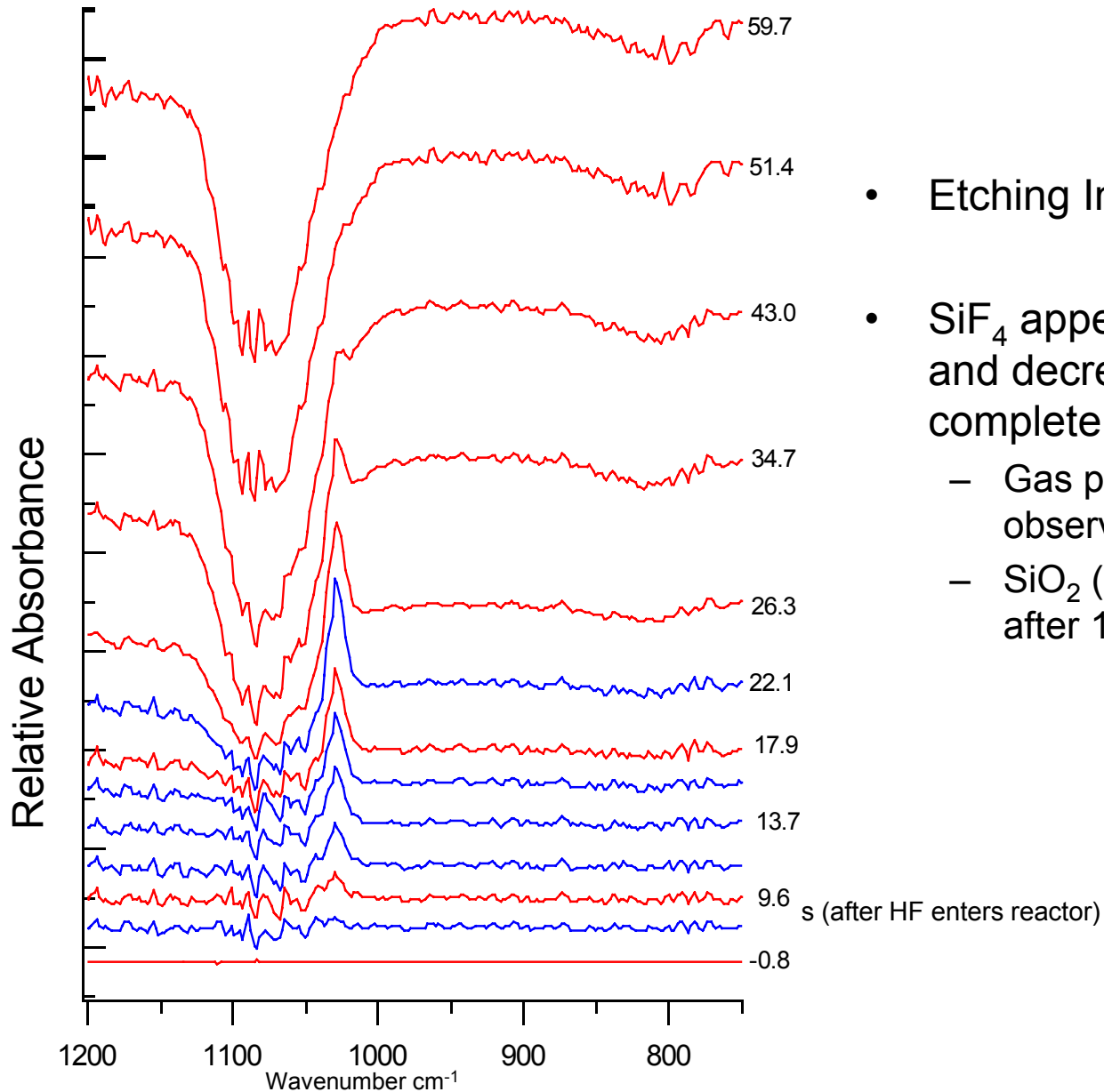
- Process Parameters
 - 50 - 800 Torr
 - 20 - 100°C
 - Broad range of HF and vapor partial pressures
- Direct introduction of vapor
 - Water and alcohols
- *In situ* Ellipsometry
- *In situ* FTIR
 - ATR and Transmission
- $\Delta P < 0.2$ Torr, transient < 1 s when HF or vapor introduced
- LabView computer control

Time Resolved FTIR of HF Vapor Process



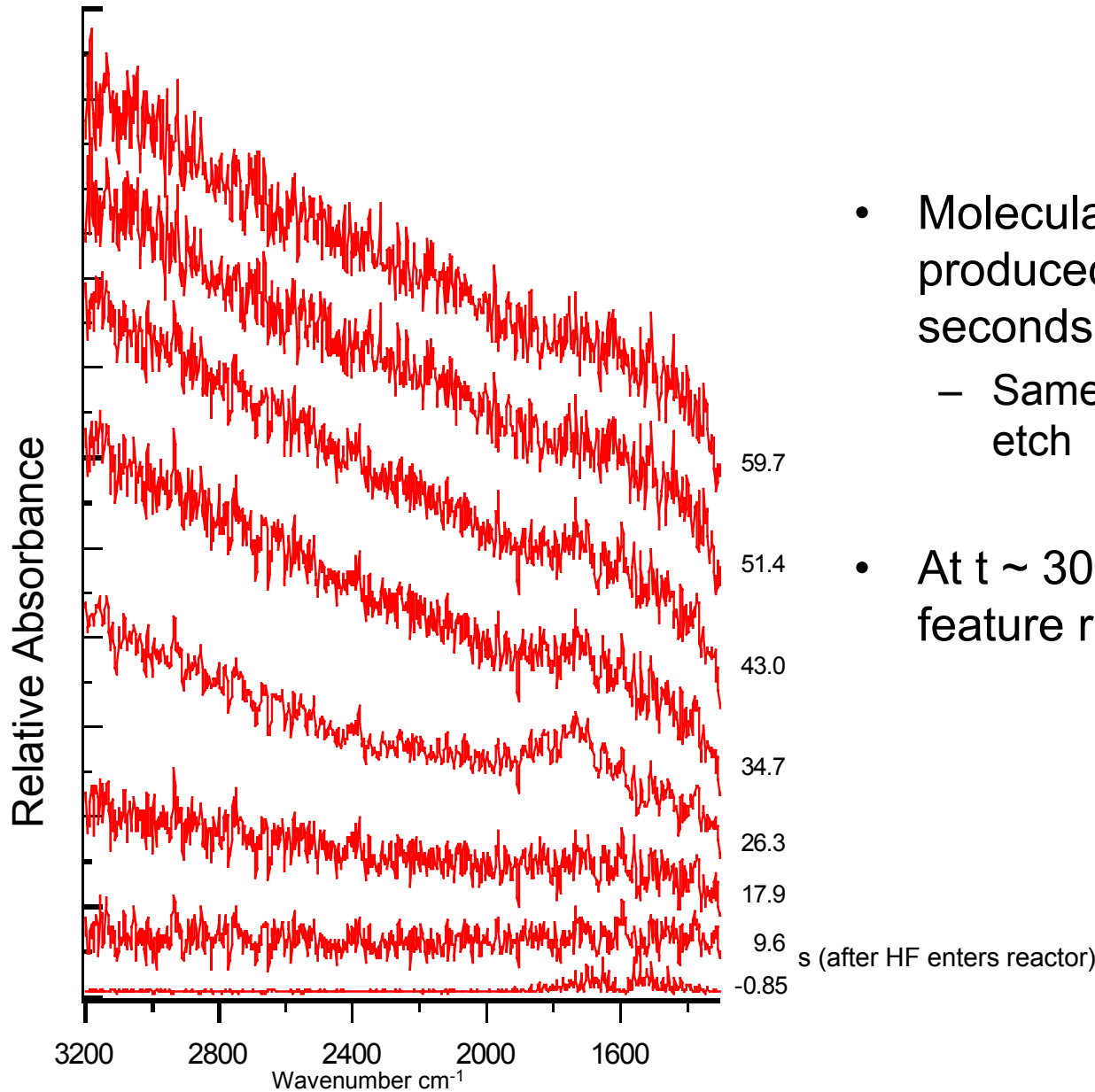
- Process parameters
 - 40°C
 - 250 Torr
 - Gas Flow
 - 400 sccm of N₂, 34 sccm of H₂O, 73 sccm of HF
- 2000 Å film was stripped in 36 s (56 Å/s)
- HF introduced into the reactor at time $\equiv 0$
- Reaction quenched with N₂ after 63 s

Low Frequency IR Vibrations



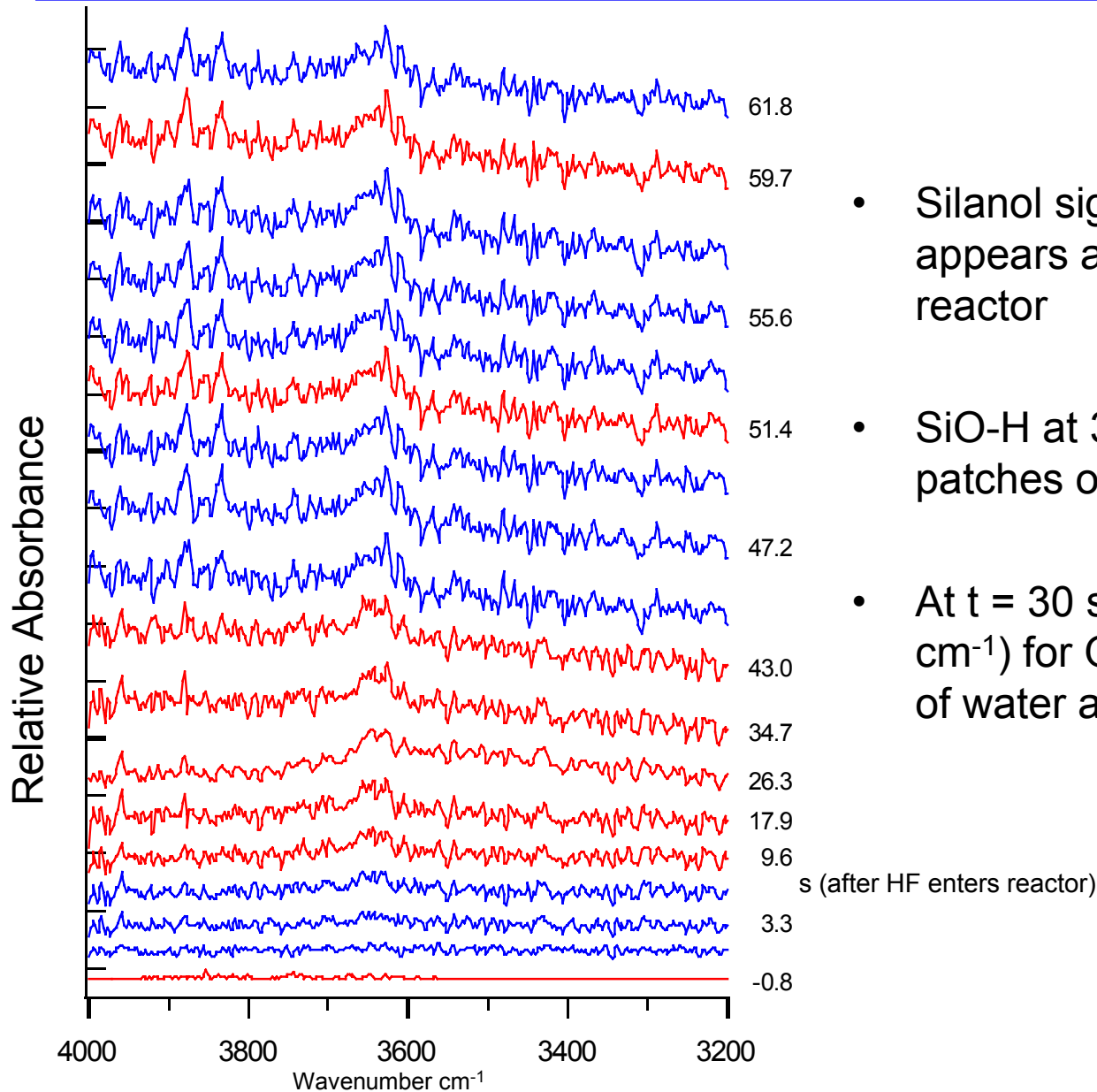
- Etching Induction Time
- SiF₄ appears before SiO₂ etches and decreases before etching is complete
 - Gas phase SiF₄ (1026 cm⁻¹) observed after 7 s
 - SiO₂ (1080 cm⁻¹) starts to etch after 14 s

Mid Frequency IR Vibrations



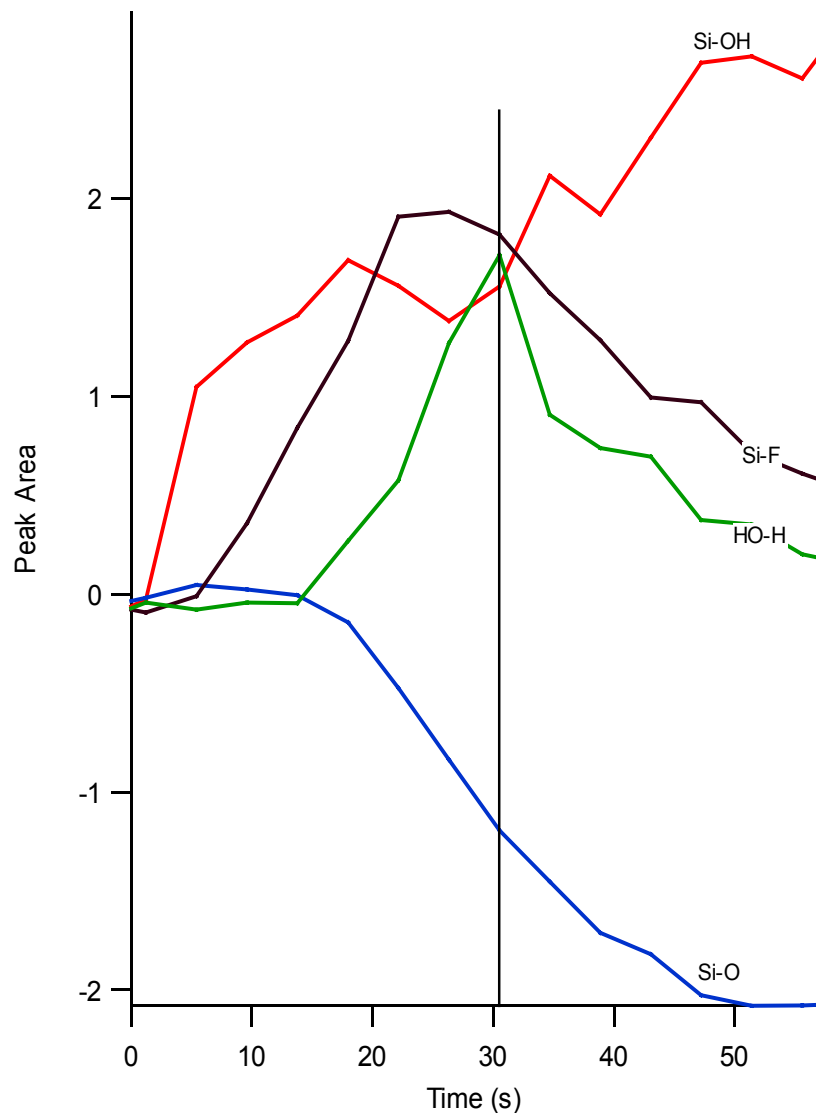
- Molecular water (1620 cm^{-1}) produced by reaction after 14 seconds.
 - Same time as SiO_2 starts to etch
- At $t \sim 30\text{ s}$ the 1620 cm^{-1} feature reaches its maximum

High Frequency IR Vibrations



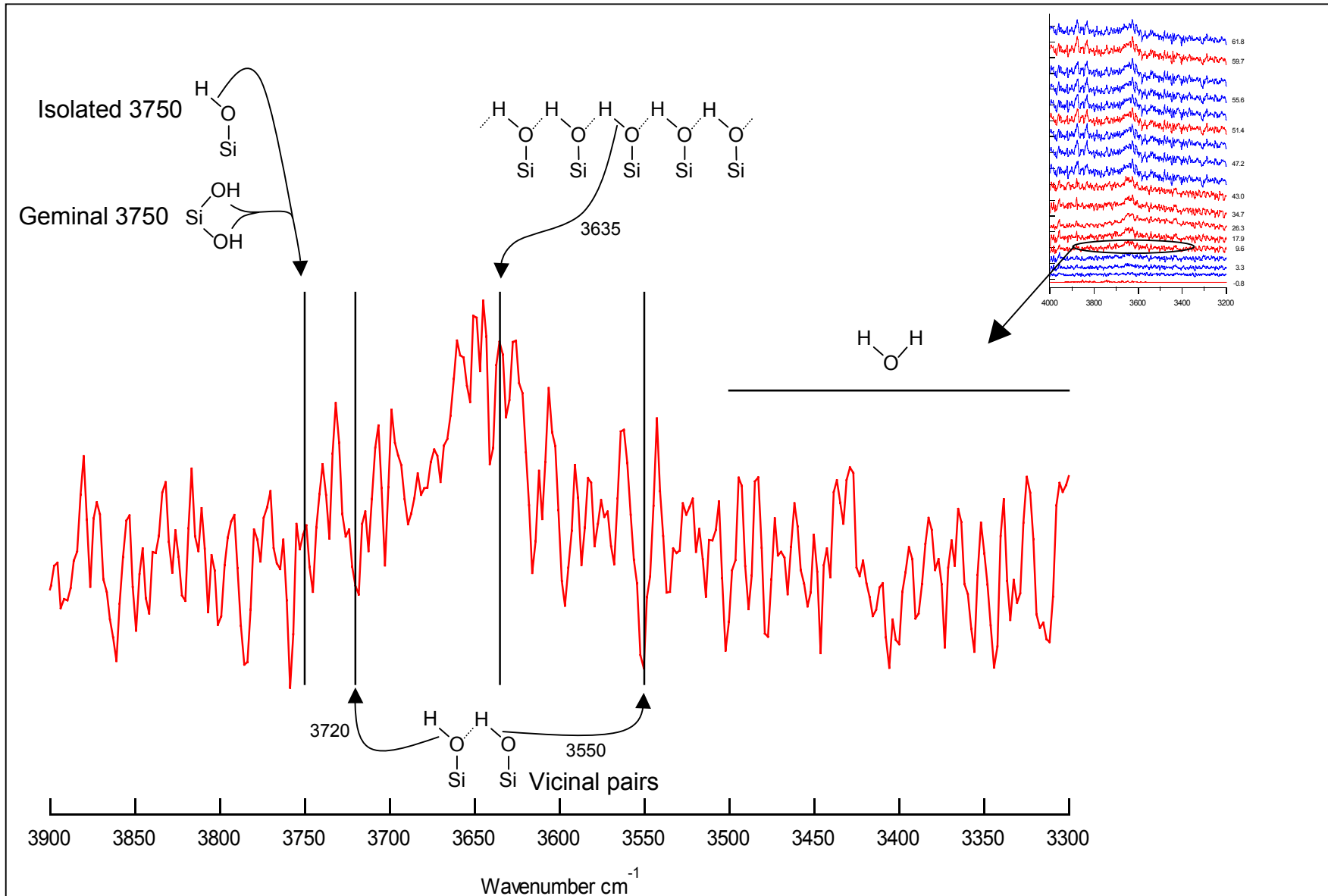
- Silanol signature (3650 cm^{-1}) appears as soon as HF enters reactor
- SiO-H at 3650 cm^{-1} indicative of patches of H-bonded silanols
- At $t = 30 \text{ s}$, envelope (at 3500 cm^{-1}) for O-H stretching modes of water apparent

Integrated FTIR Peak Areas



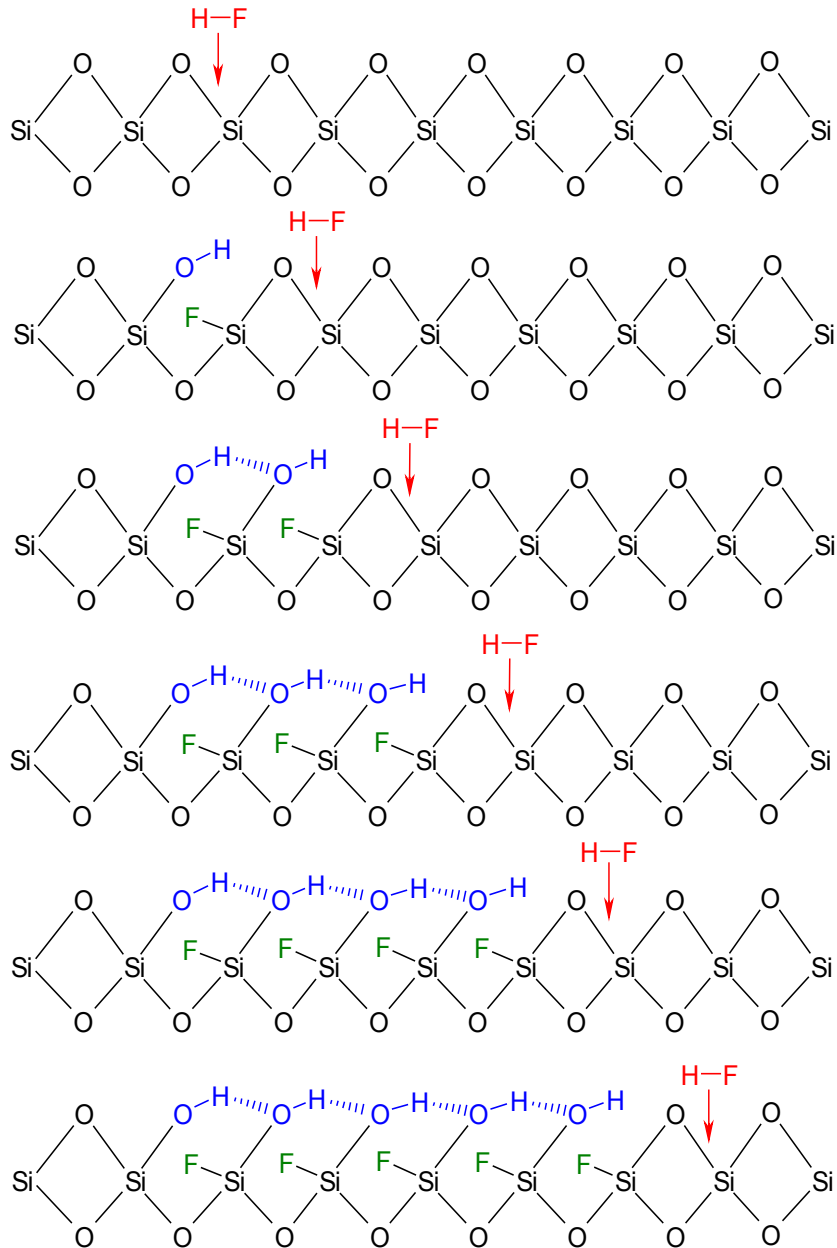
- SiO-H appears when HF enters reactor
- The SiF₄ signature appears 7 s after HF is introduced but before SiO₂ etching is apparent
- SiO₂ etching starts 14 s after HF enters reactor
- Max. in SiO₂ etching rate and surface HOH correspond
 - Occurs at ~ 1/2 initial oxide film thickness

Silanol FTIR Peak Identification



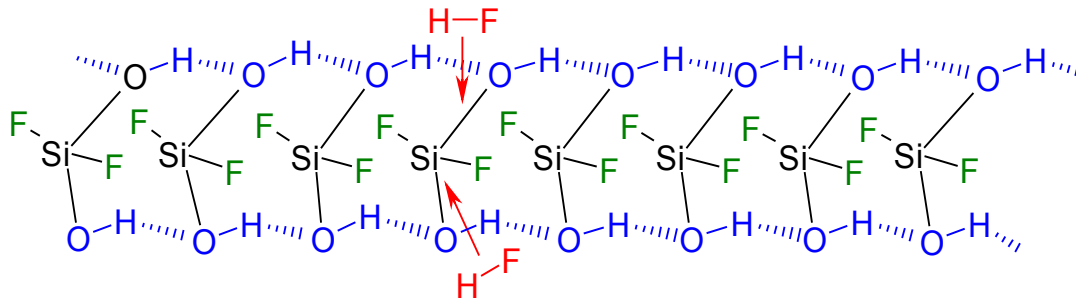
H. E. Bergna 1990

Proposed HF/H₂O Initiation Mechanism

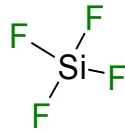


- HF attacks Si-O bond
 - Random
 - Activated
- SiO-H and Si-F produced
 - Si-F in FTIR?
- Succeeding HF molecules unzip oxide lattice
- Groups of H-bonded SiO-H produced during induction t

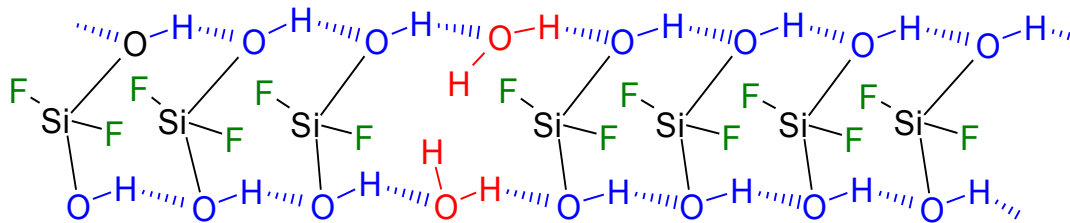
Proposed HF/vapor Bulk Etching Mechanism



- Hydroxylated Si-O surface
- HF attacks remaining Si-O bonds



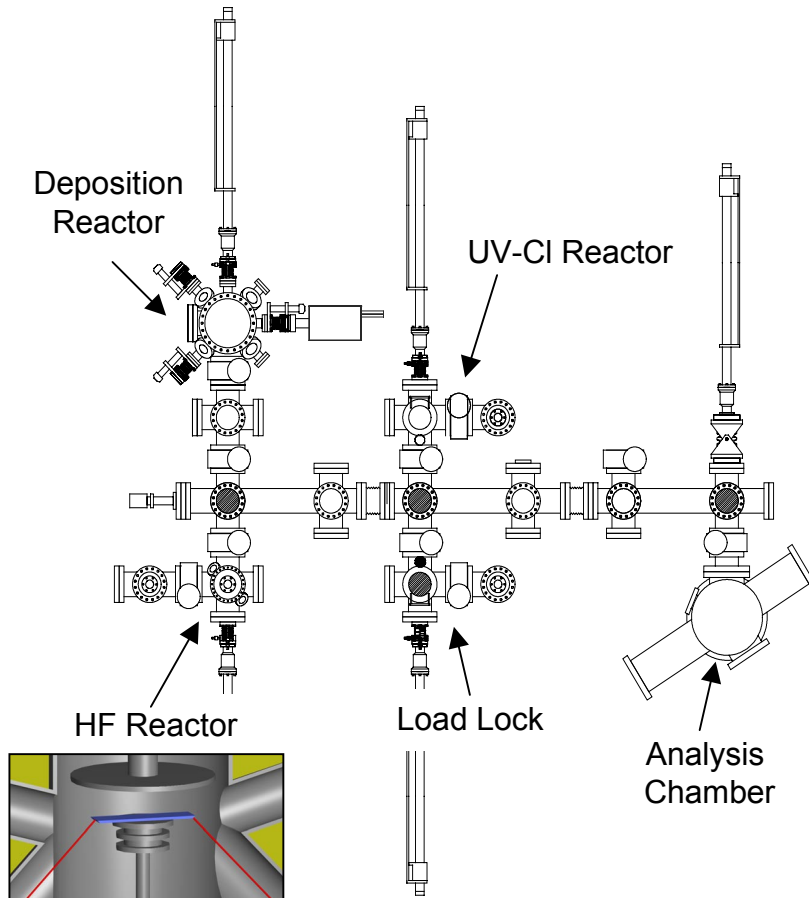
- SiF_4 produced
- Molecular water production and Si-O etching coincide



Summary

- Mixed acid residue formed on BPSG after AHF etching
- UV-Cl₂ process effective in removing residue
 - High temperature perpendicular UV exposure
 - Unrinsable residue
 - Low temperature parallel UV exposure
 - Eliminates water rinse
 - Process optimization/time
- HF/vapor Etching Mechanism
 - H-bonded silanols produced during induction time
 - Molecular water produced by HF attack of Si-OH bonds in F_xSi-OH groups
 - Liberate Si and etch film
 - Etching rate decreases before oxide film is stripped

Future Plans



- Determine details of the initial stages of HF/vapor etching
- Process doped oxide films in cluster apparatus
 - Develop UV-Cl₂ removal process
- Clean/prepare surfaces for Cu barrier and high-κ film deposition
- Develop metal removal mechanism with UV-Cl₂

Acknowledgements

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