
Selective Surface Preparation and Templated Atomic Layer Film Deposition

Rong Chen¹, Hyoungsub Kim³, Junsic Hong²
Paul C. McIntyre³, Stacey F. Bent²

1. Department of Chemistry
 2. Department of Chemical Engineering
 3. Department of Materials Science and Engineering
- Stanford University

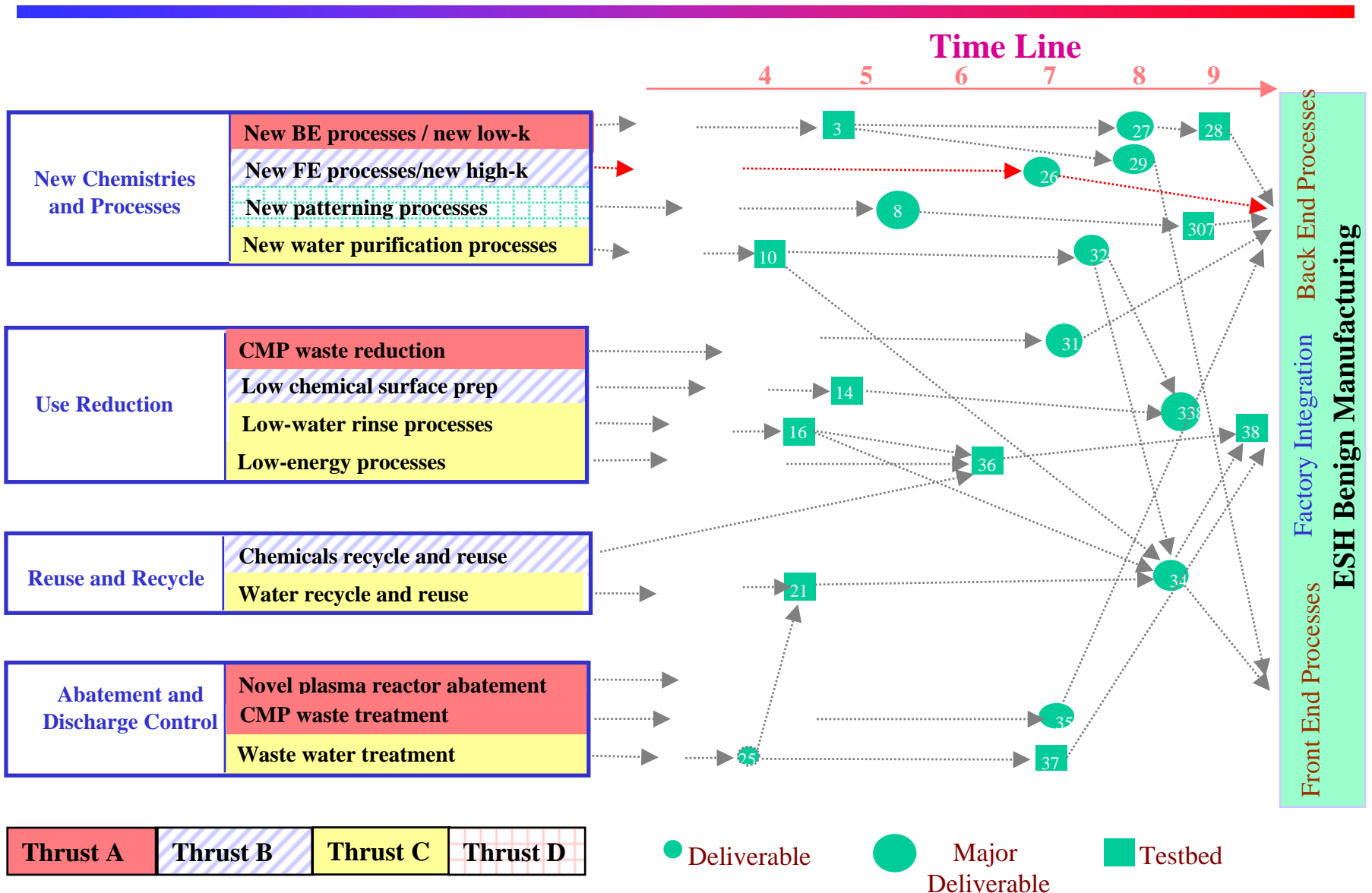
2003-12-11 Tele-seminar

NSF/SRC Engineering Research Center for Environmentally Benign Semiconductor Manufacturing

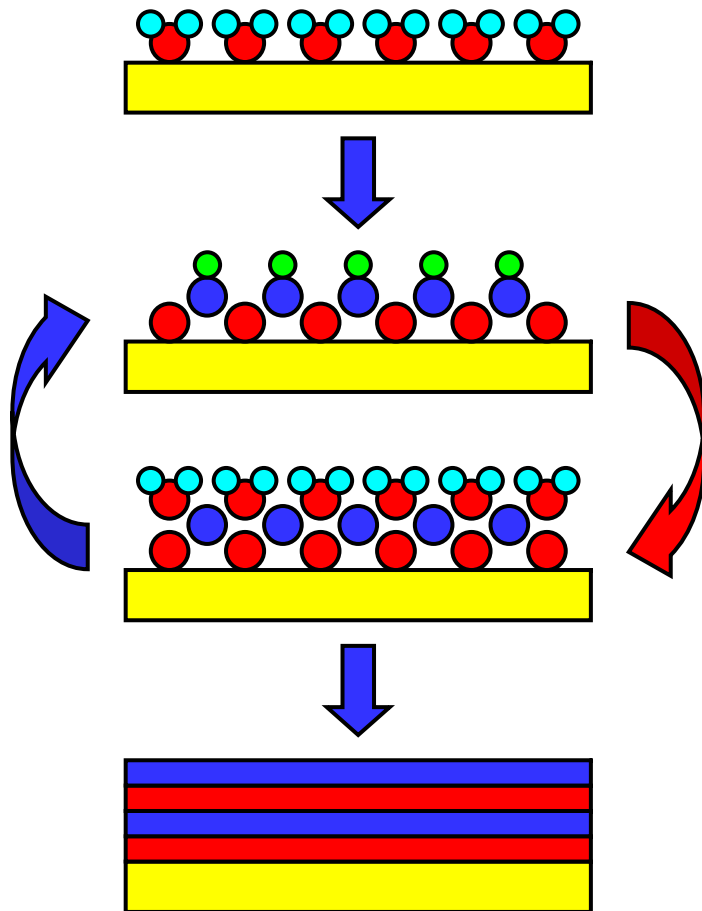


Stanford University
Department of Chemical Engineering
- <http://bentgroup.stanford.edu> -

Strategic Plan (Task B-2)

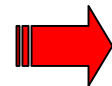


Surface Functional Groups Control ALD



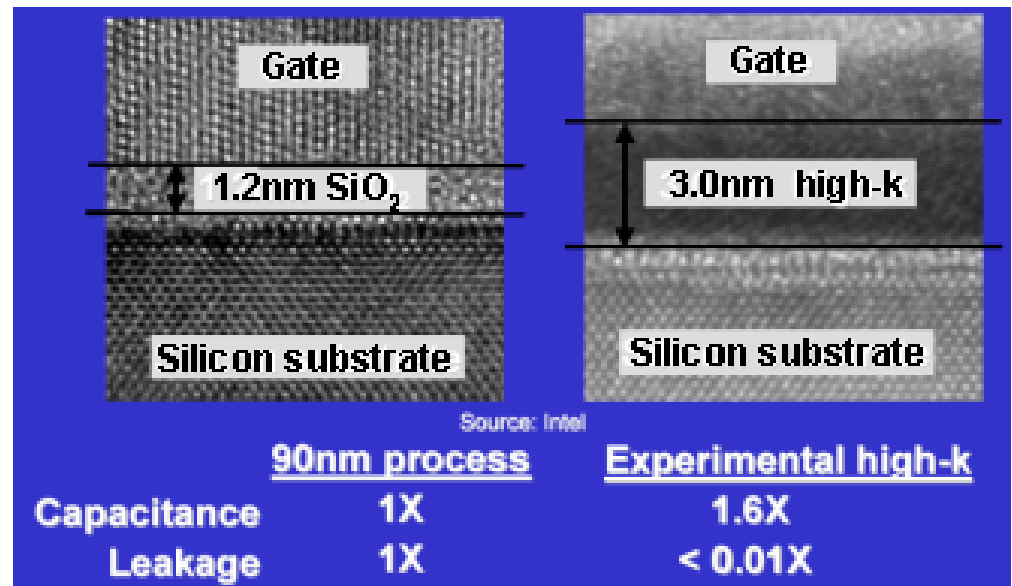
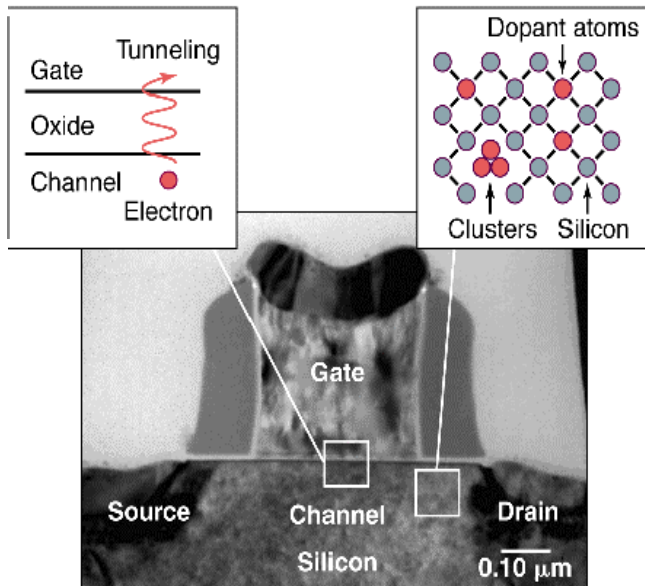
ALD process based upon chemical reaction between the precursors and the film surface.

The reactions depend on the specific reactive functional groups present at the surface.



Manipulate these surface groups to control the ALD process

The Need for High-κ Dielectric Materials



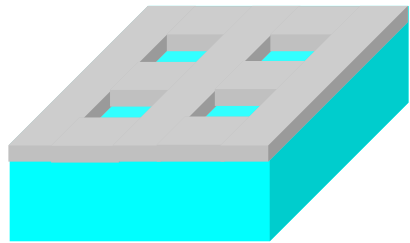
P.A. Packan, Science, 1999

Source: Intel

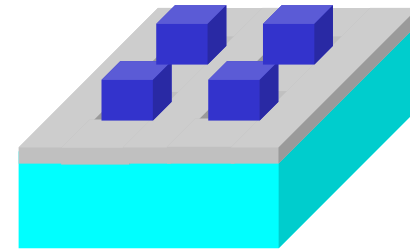
$$\text{Capacitance} \propto \frac{\kappa}{t}$$

- The scaling of metal-oxide-semiconductor (MOS) devices to sub-nanometer feature sizes requires **thin gate insulators**.
- **Leakage current** caused by **electron tunneling** increases exponentially with decreasing dielectrics thickness.
- Using high-κ materials allows deposition of thick films with an effective thickness equivalent to thin SiO₂ films.
- HfO₂ and ZrO₂ are promising candidates for future gate dielectrics

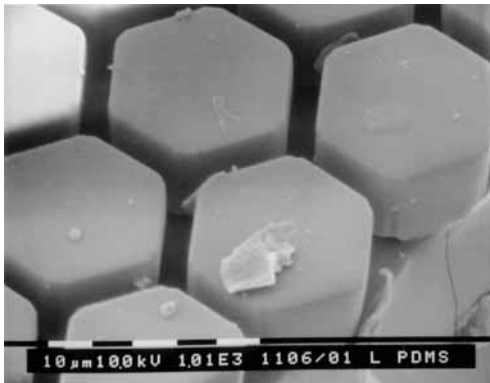
Generate 3-D Pattern from 2-D Template



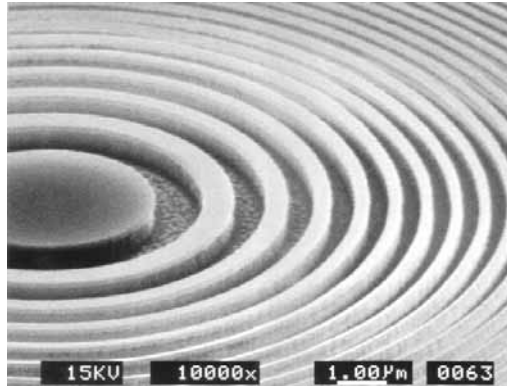
Area-Selective ALD



Microcontact
printing

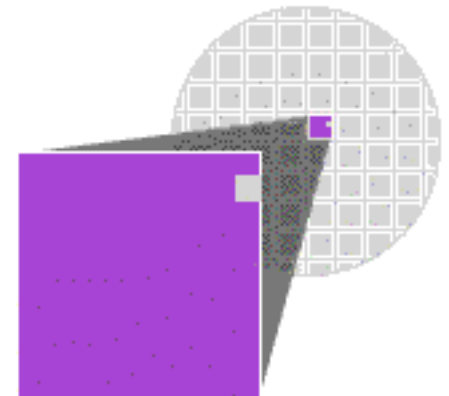


E-beam lithography
(direct writing)



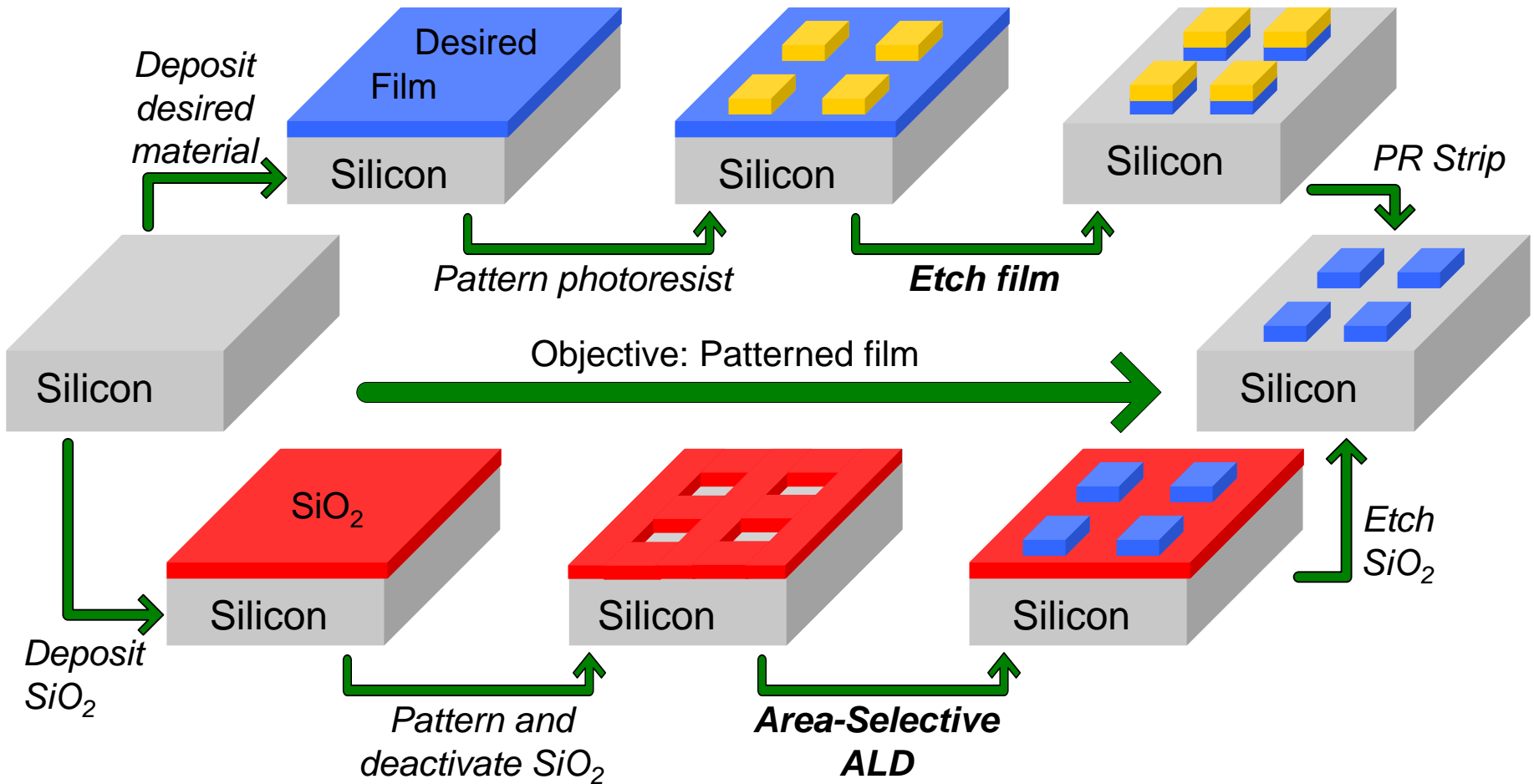
Source: Paul Scherrer Institut

Photolithography
patterned SiO₂/Si



Source: Intel

Additive vs. Subtractive Patterning

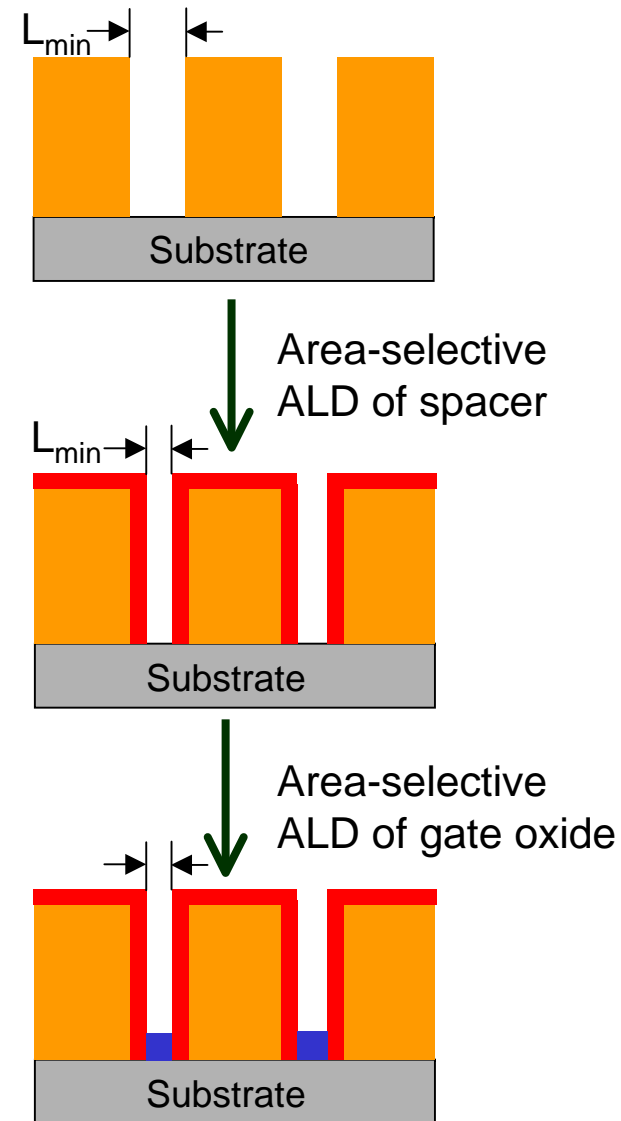


Advantages of additive patterning:

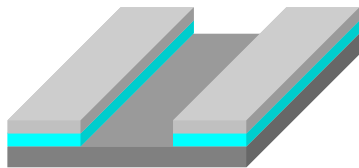
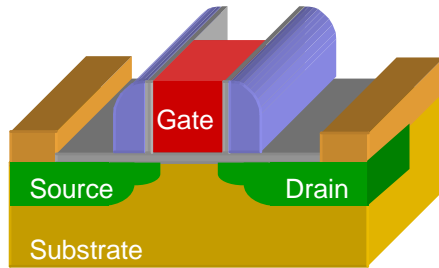
- Avoid difficult etch step
- Deposit material only where desired → cheaper, no residual contamination

Possible Applications of Area-Selective ALD

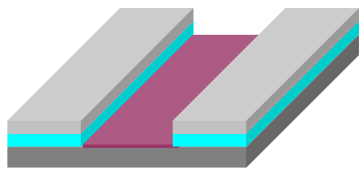
- **Self-aligned gate stack**
 - Deposit high- κ dielectric and metal gate by ALD.
 - Avoid high- κ etch
 - Avoid possible contamination
 - Proposed process sequence is self-aligned with respect to gate dielectric and metal.
- **Feature size reduction**
 - Selective deposition over mask
 - Precise dimensional control
- **Deposition of expensive materials**
- **Patterning of materials difficult to etch**



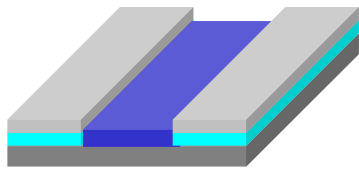
Process Flow for Area-Selective ALD



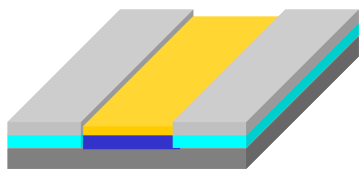
patterned and protected surface



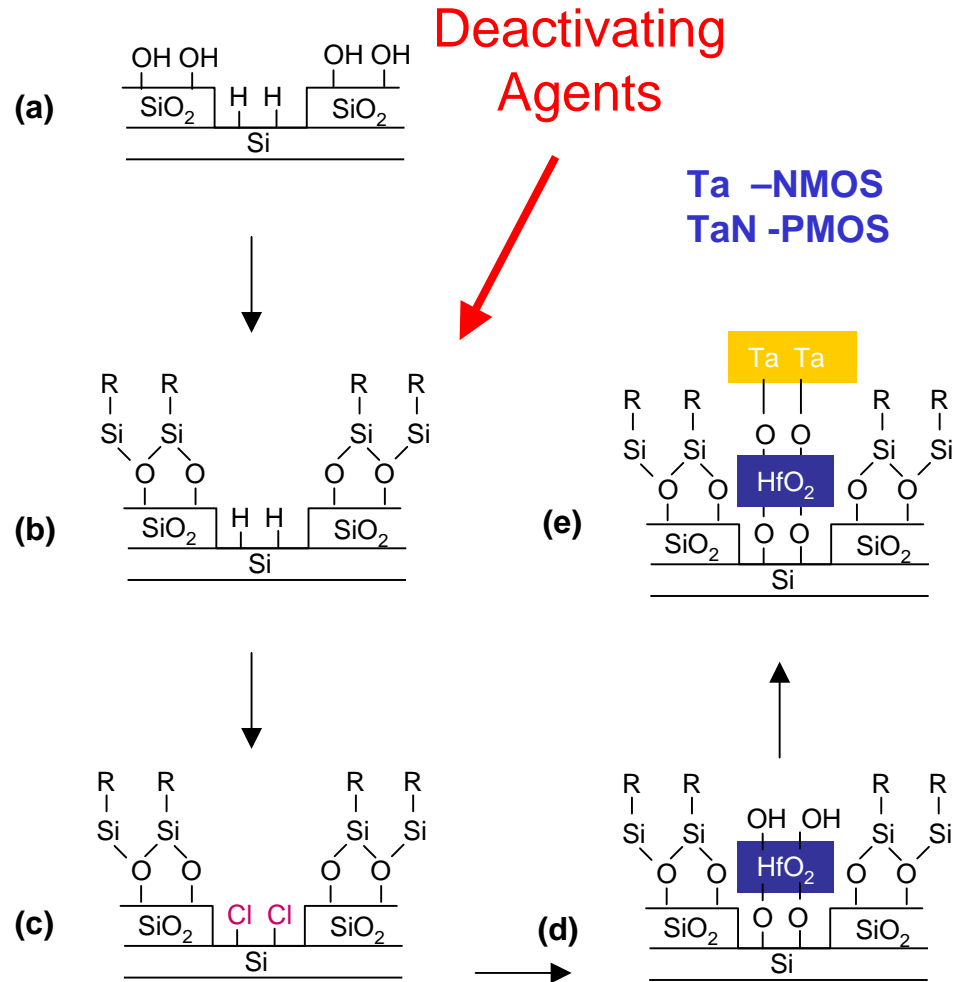
surface activation



high- κ material ALD deposition



metal gate ALD deposition



Requirements for Area Selective ALD

- **Fundamental**

- *Deactivate* surfaces to prevent growth
- *Activate* surfaces where growth is desired

- **Process Integration**

- *Selective deposition* **or** *patterning* of activating/deactivating agents
 - Compatible with front-end processes

- *Etching* of deactivating agents

- No residual contamination
- No damage to substrate/deposited films

- Activation agents must provide excellent interface

- No contamination, trapped charge, etc.

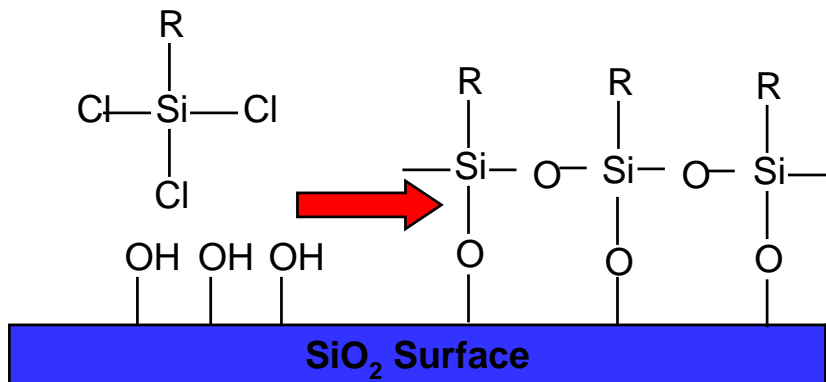
- Prevent incorporation of deactivating agents into growing film along interfaces (esp. sidewalls)

Deactivation of SiO₂

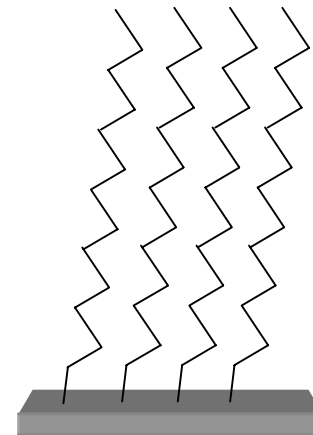
Good deactivating agents should have:

- High reactivity with surface Si-OH group
- Selectivity toward Si-OH over Si-H
- Stability at ALD temperature
- No reaction or competition with ALD precursors
- Ease of integration into process

- Self-assembled monolayers (SAMs) have been investigated for a few decades.
- We are exploring the use of a series of different length silylating organic molecules as the deactivating agents through both solution and gas phase delivery.

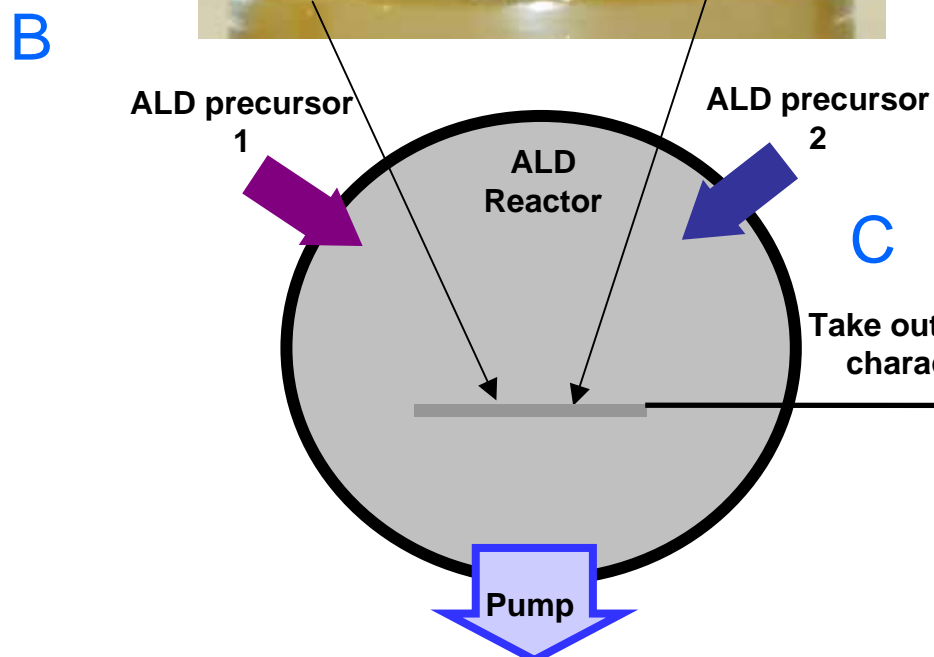
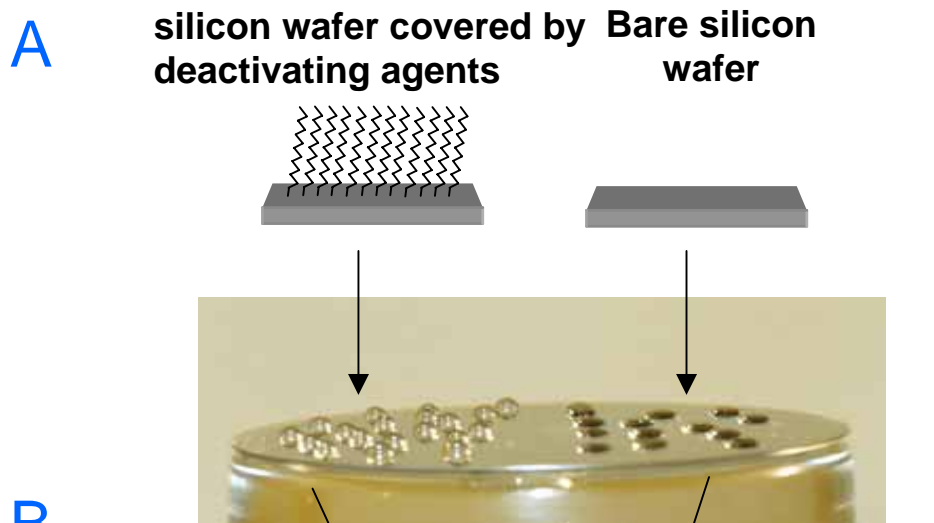


silylating reactions



self-assembled monolayer (SAMs)

Combined Deactivating Agents & ALD



- A.** Deactivating agents preparation and analysis;
- B.** ALD growth of ZrO_2 & HfO_2 ;
- C.** Sample characterization after deposition.

C

Take out samples for characterization

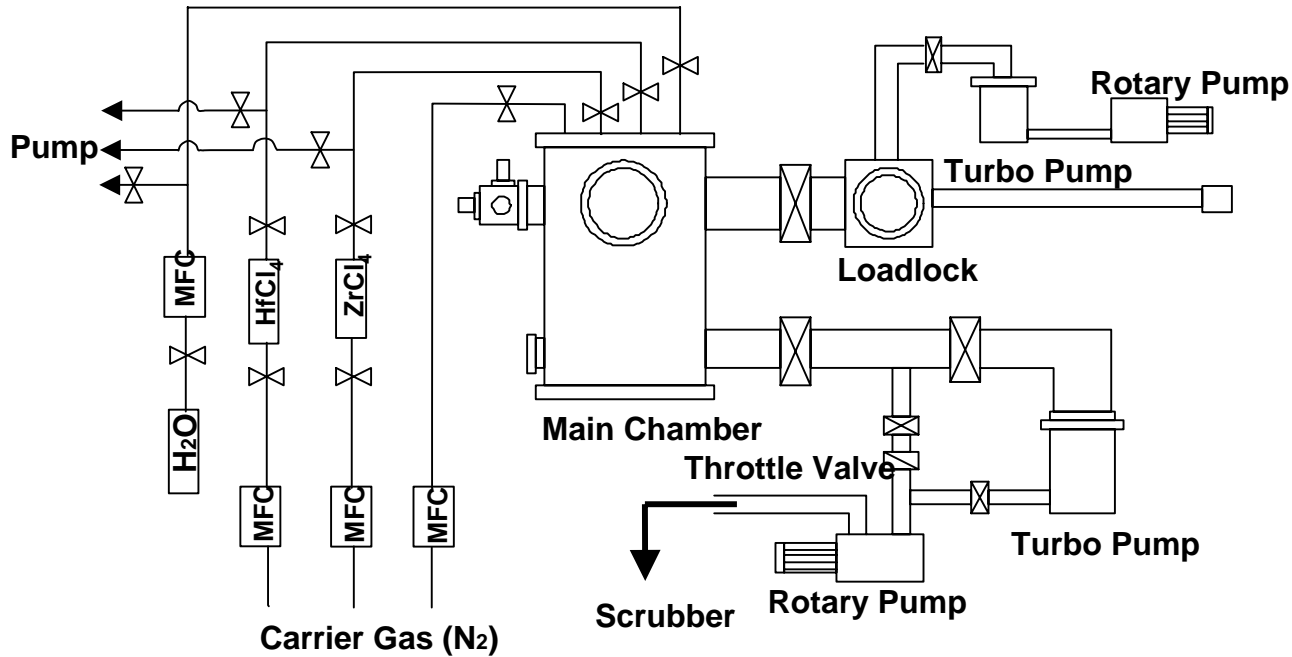
XPS;

Ellipsometry;

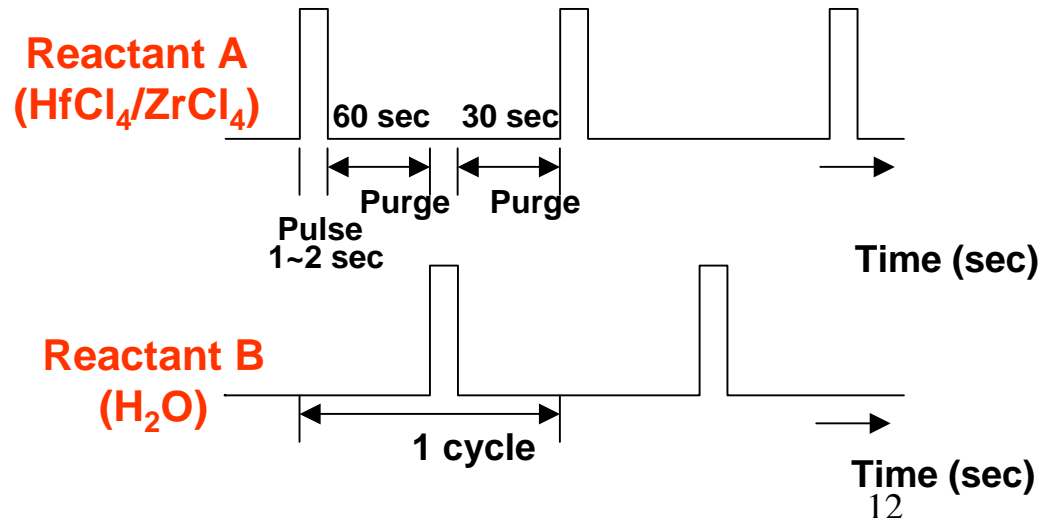
TEM

AFM

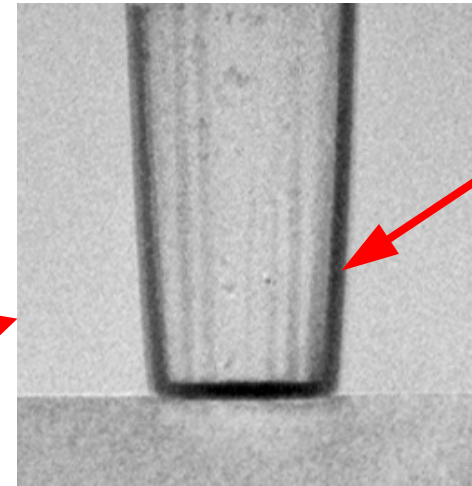
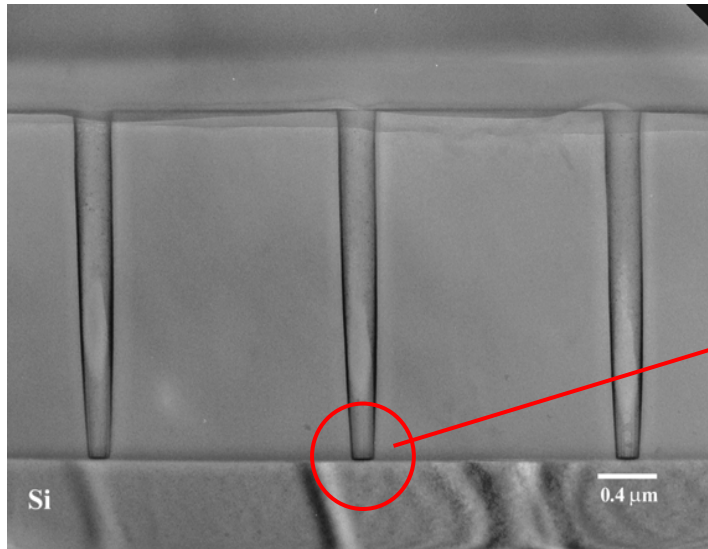
Schematic Diagram of ALD System



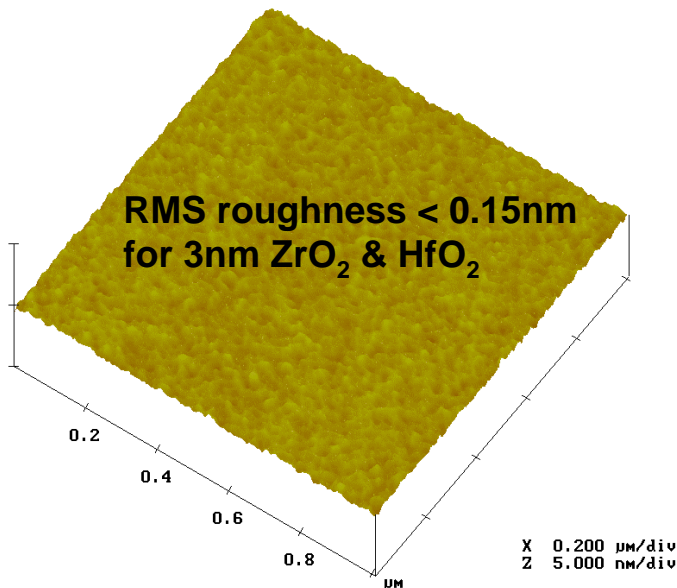
- Base pressure = $5 \cdot 10^{-8}$ Torr
- Process temperature : 300°C
- Process pressure : 0.5 Torr
- Source temperature :
 H₂O (liquid) = 20°C
 HfCl₄ /ZrCl₄(solid) = 150°C



Conformality and Surface Roughness of ALD Films



HfO₂

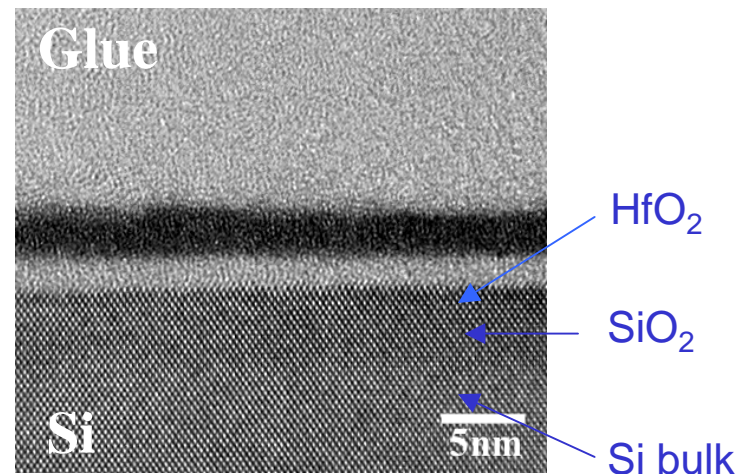
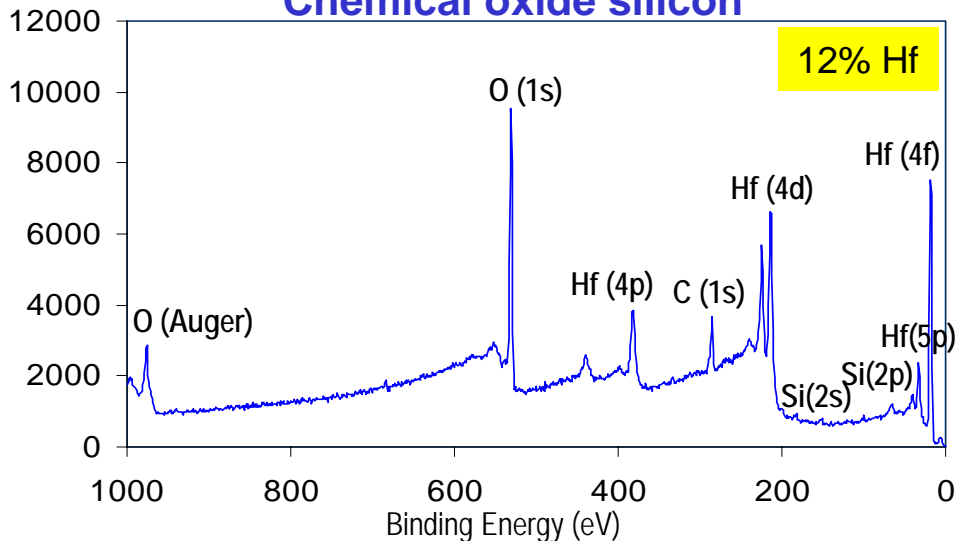


ALD deposition of metal-oxide films

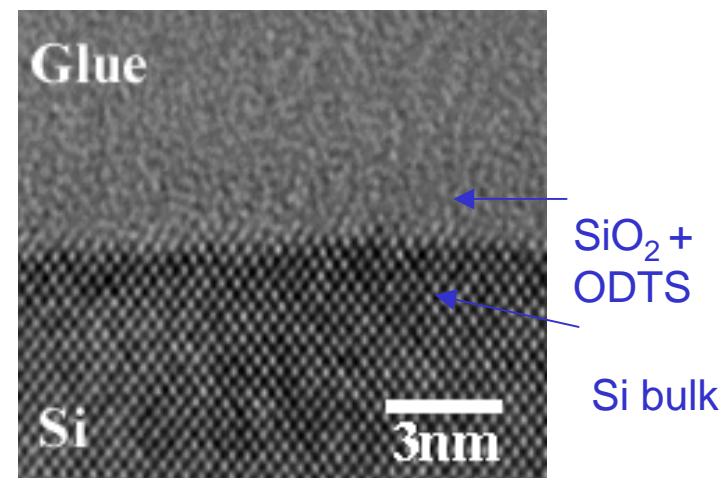
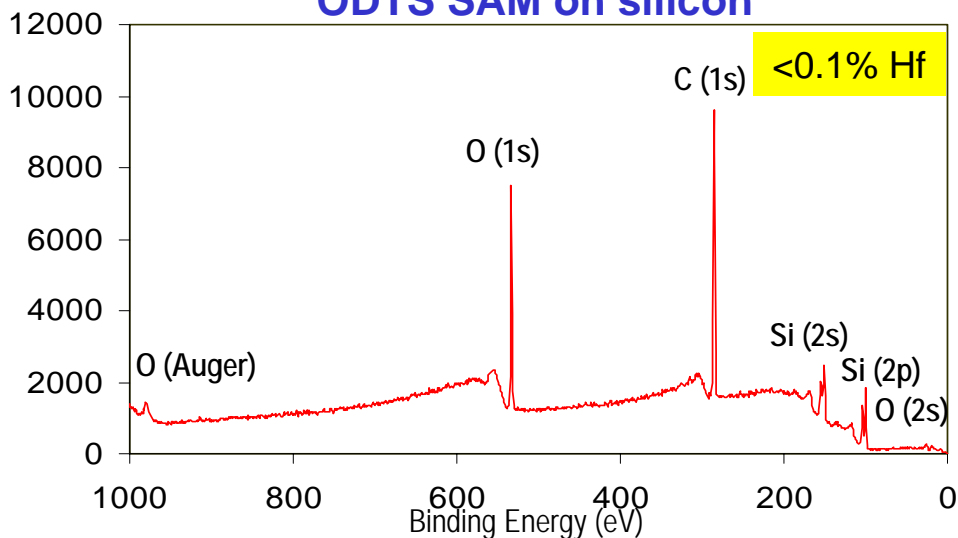
- Excellent step coverage (~100%) on complicated geometric structures
- Smooth and uniform deposition

ALD Inhibition by Octadecyltrichlorosilane (ODTS) SAM

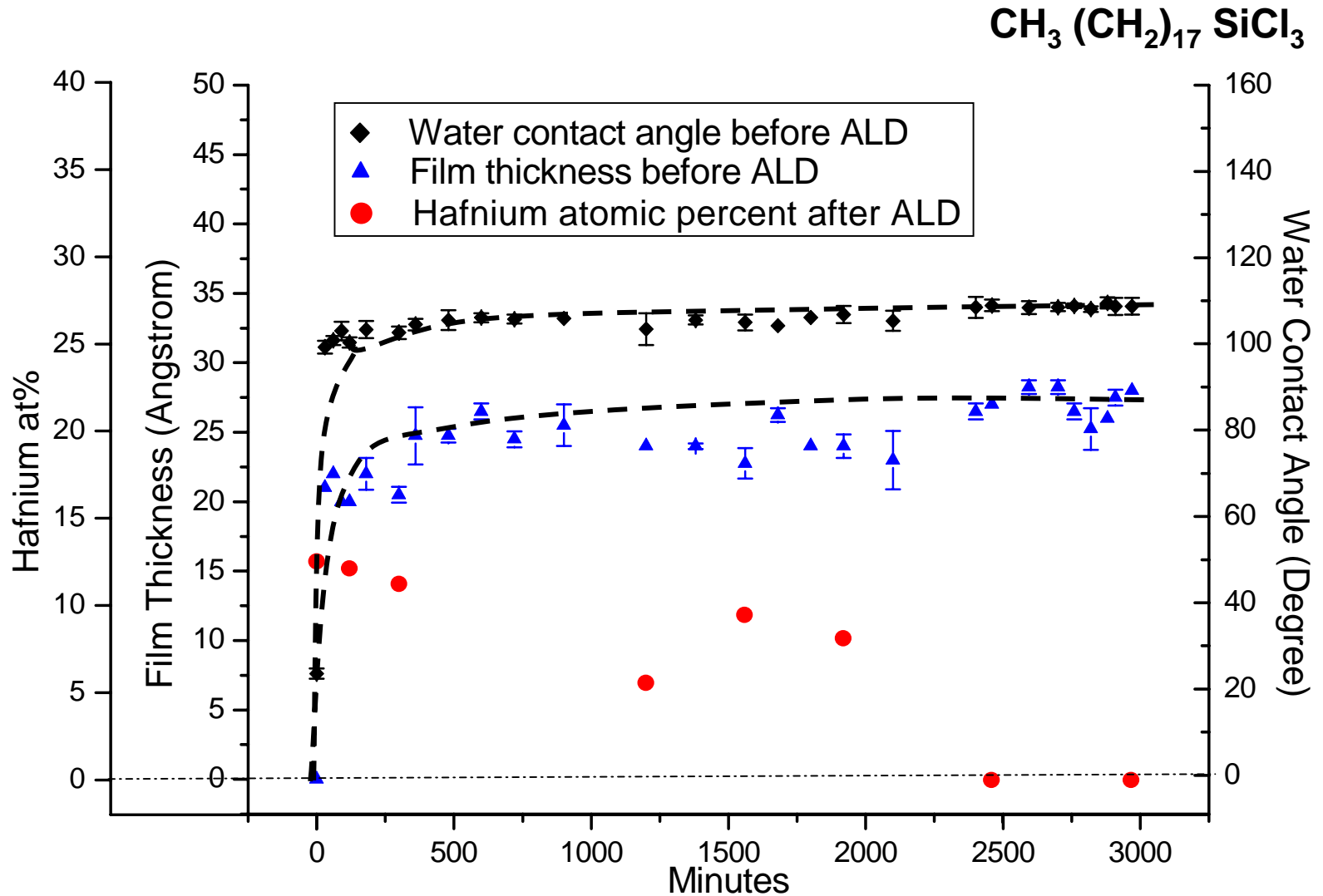
Chemical oxide silicon



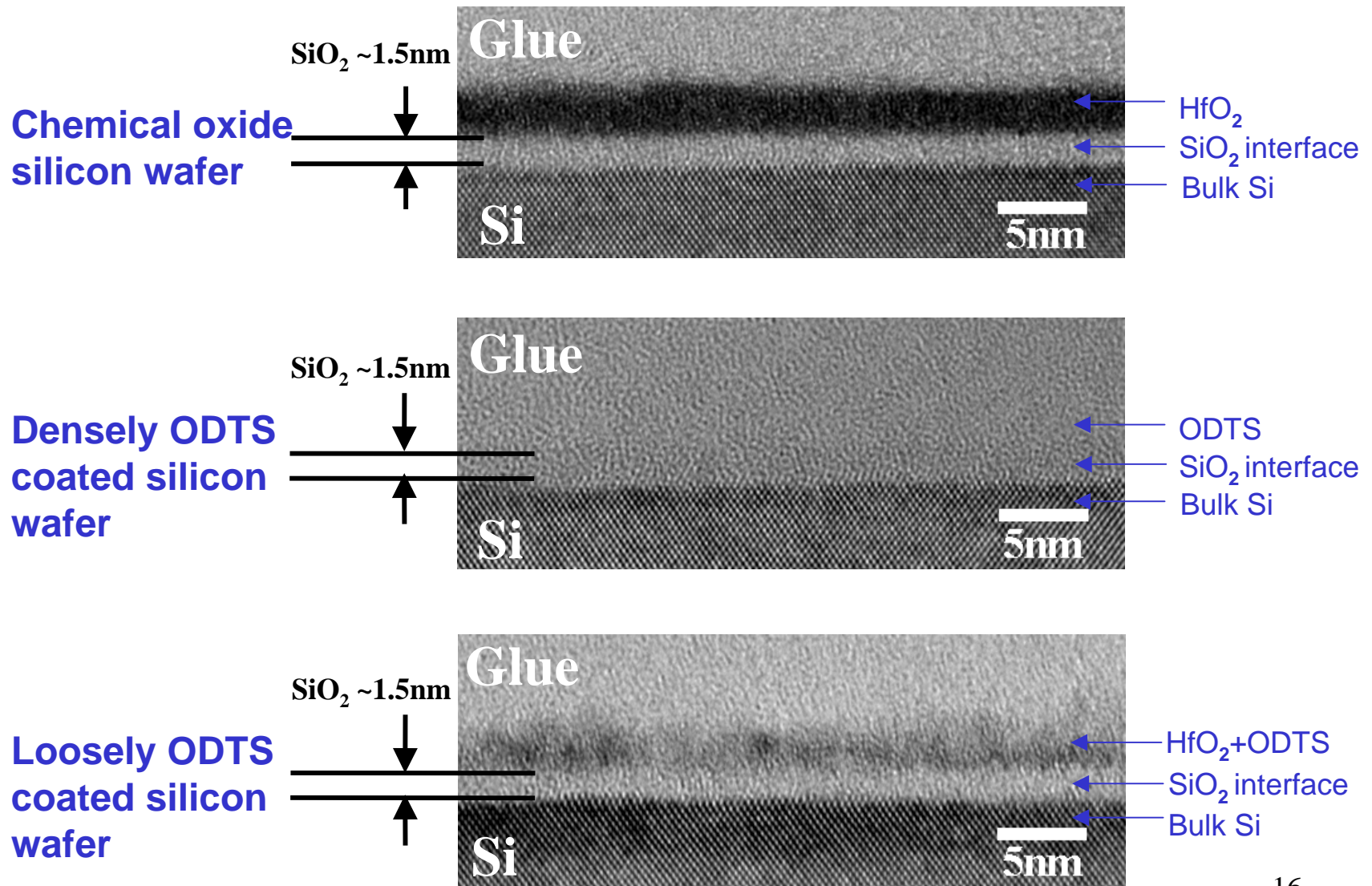
ODTS SAM on silicon



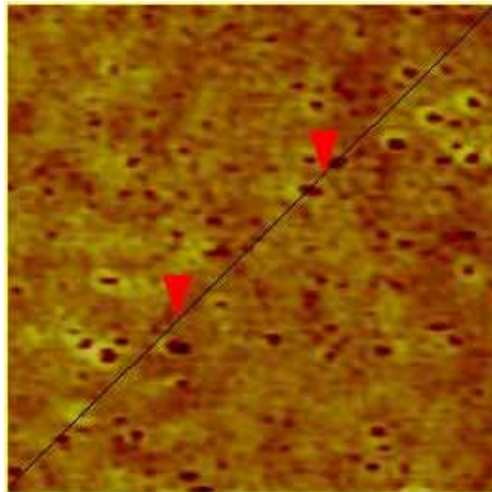
Silylation Time Dependence for ODTS



Interface Analysis by Cross-sectional TEM

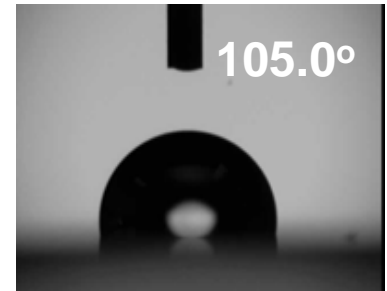
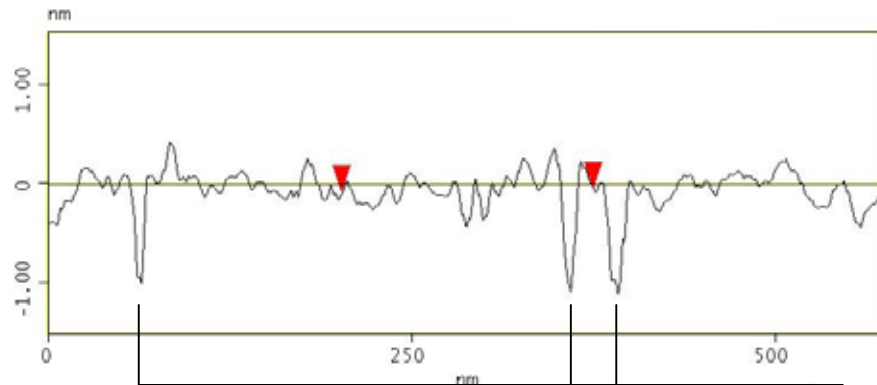


ODTS Films Analysis before ALD Process

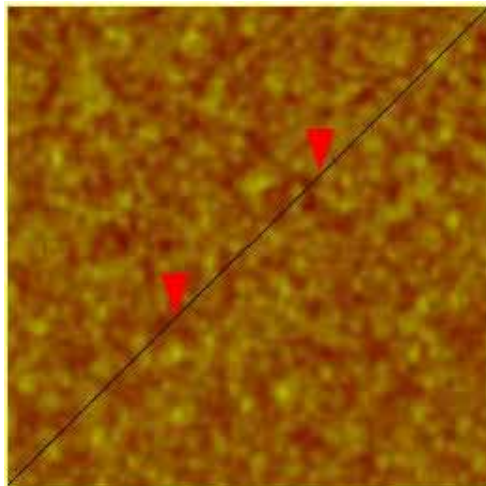


**Loosely packed
ODTS film**

WCA: 105.0°
RMS: 0.20nm

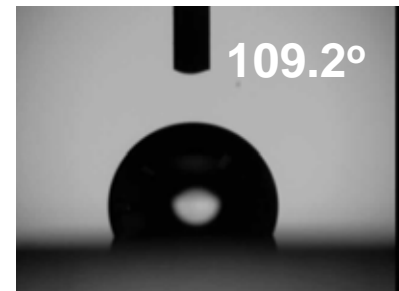
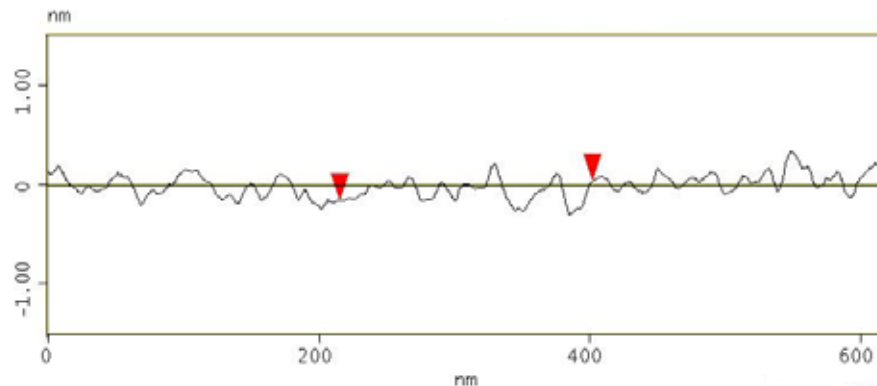


**Some pinholes
on SAMs**



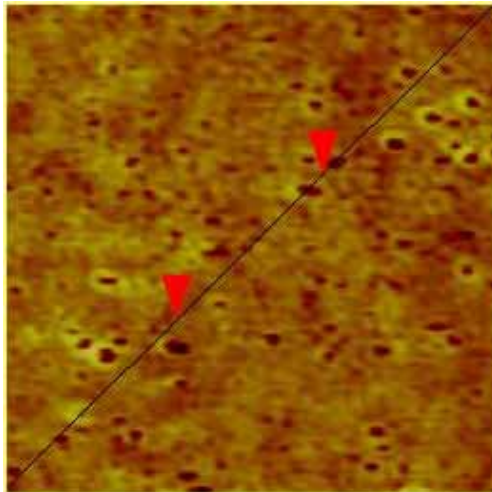
**Densely packed
ODTS film**

WCA: 109.2°
RMS: 0.10nm

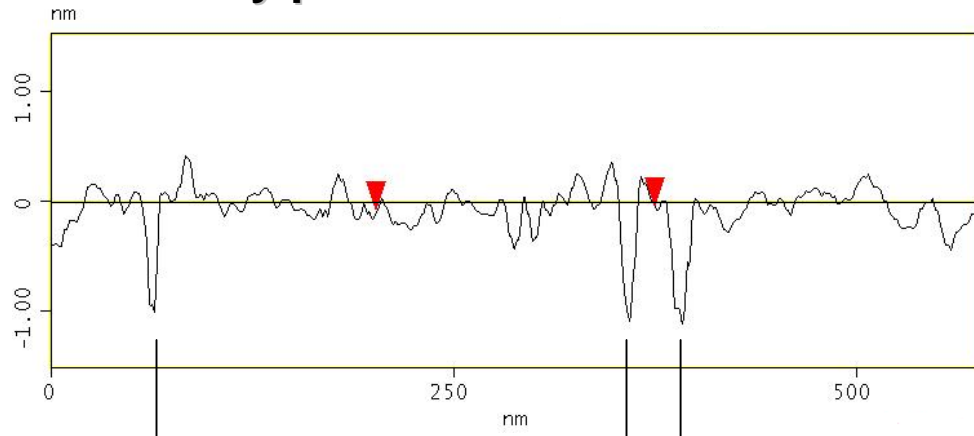


**Smooth and
Uniform SAMs**

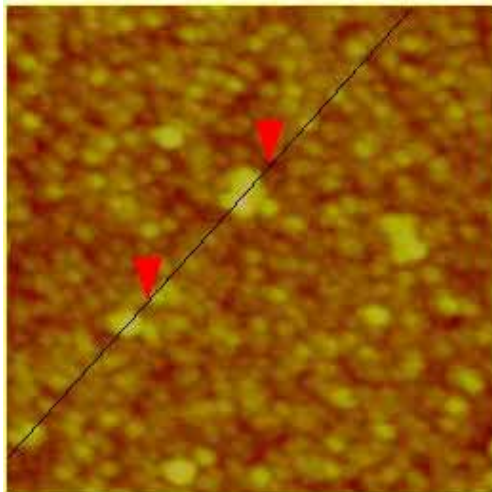
AFM Analysis of ODTS before & after ALD



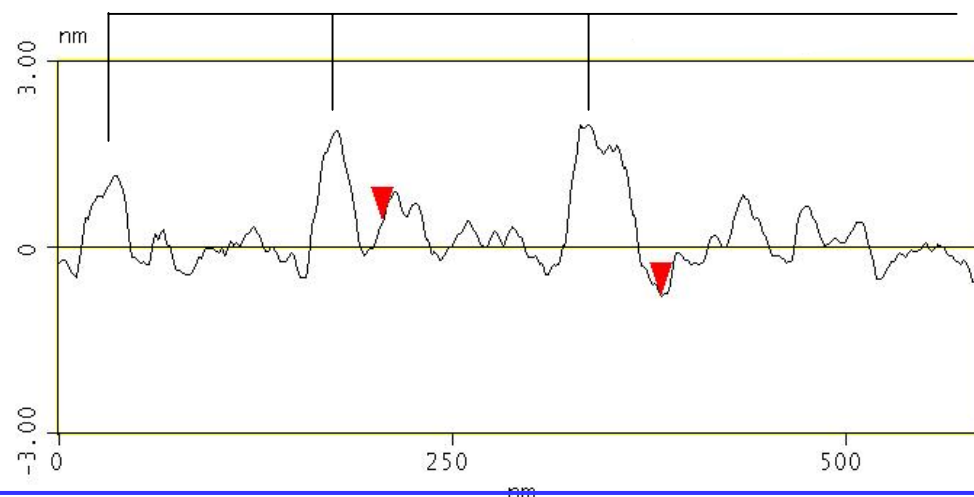
Loosely packed ODTS film **before** ALD



pinholes

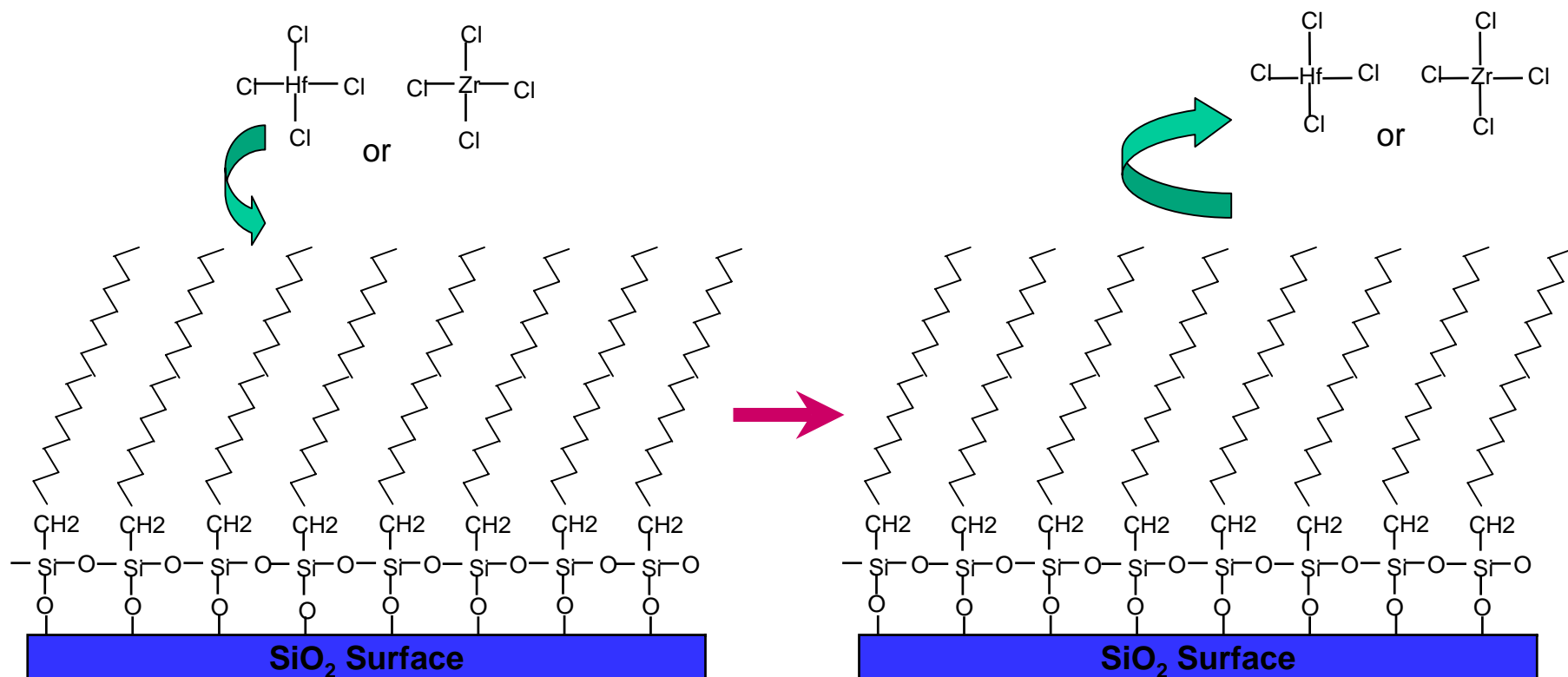


Loosely packed ODTS film **after** ALD



spikes

Mechanism of ALD inhibition on densely-packed ODTS

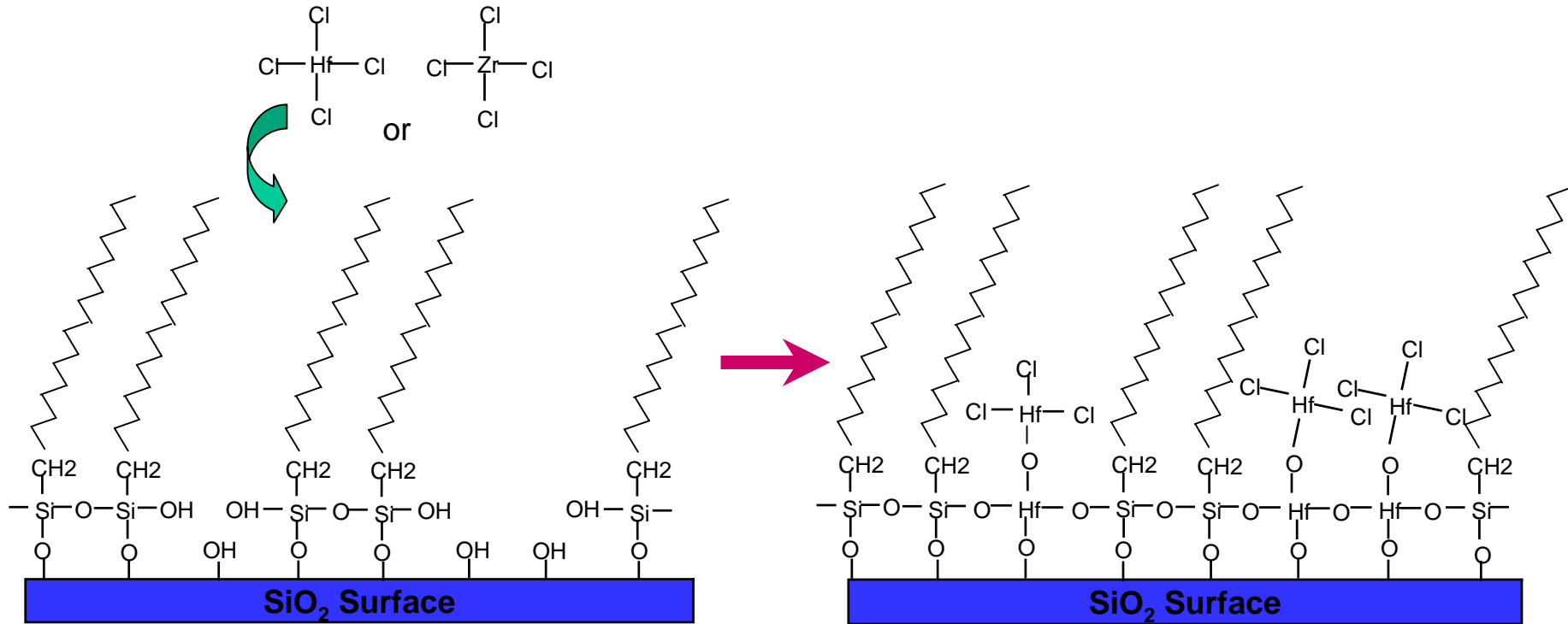


Densely packed ODTS

The organic SAM blocks the underlying oxide from the ALD precursors

Octadecyltrichlorosilane (ODTS), $\text{CH}_3(\text{CH}_2)_{17}\text{SiCl}_3$

Mechanism of ALD growth on loosely-packed ODTs



Loosely packed ODTs

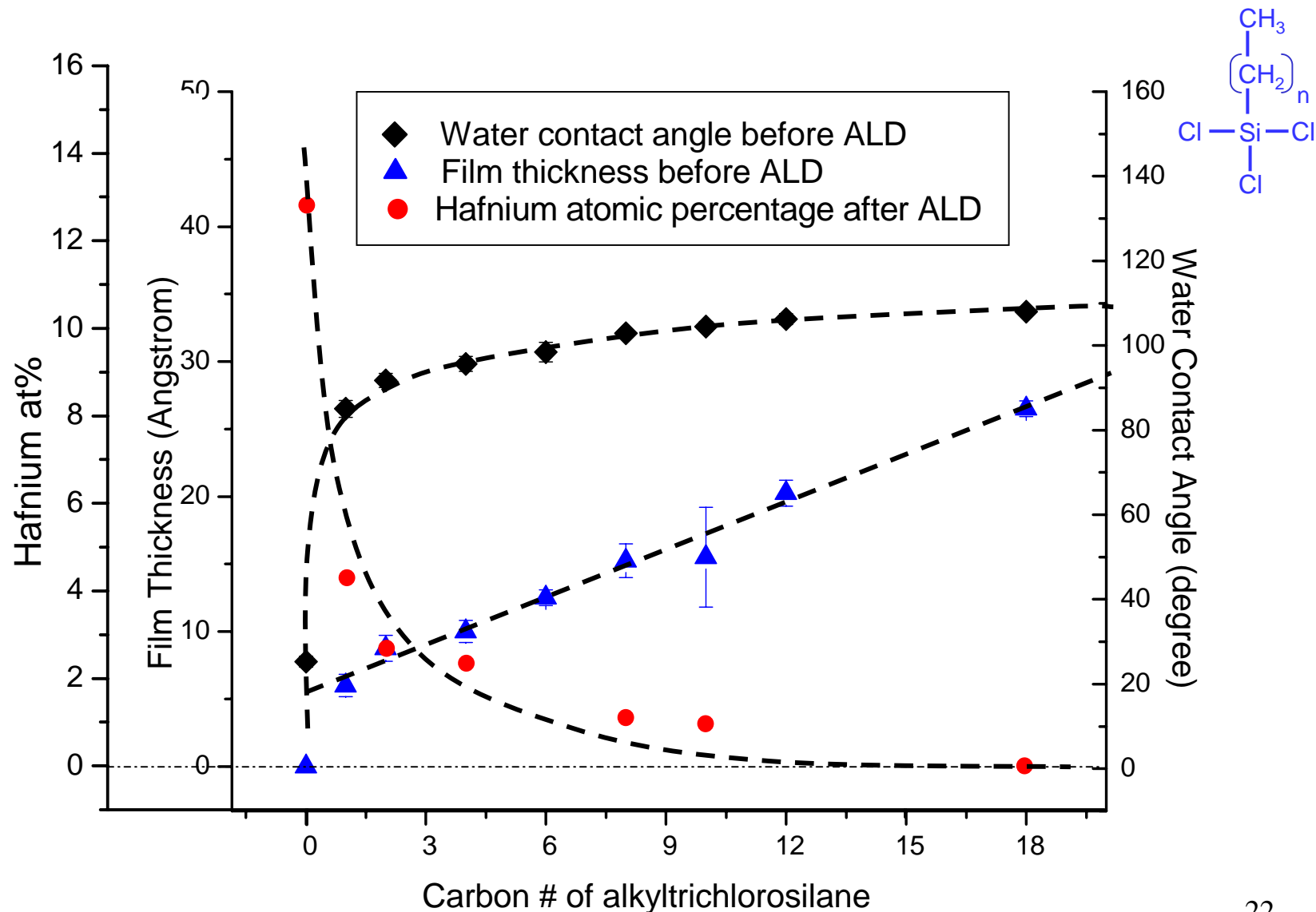
Pinholes in the SAM leave native oxide hydroxyl groups accessible to ALD precursors

ALD precursors react at these surface sites and growth nucleates from those sites

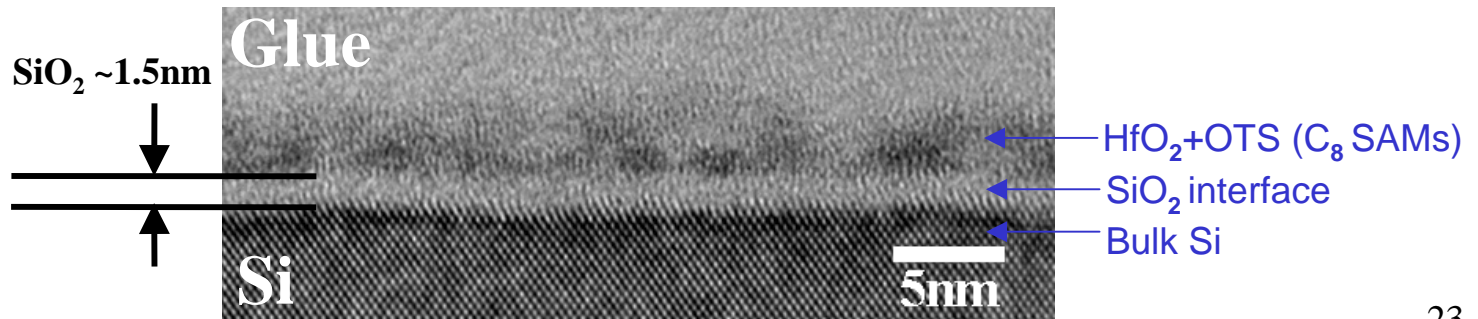
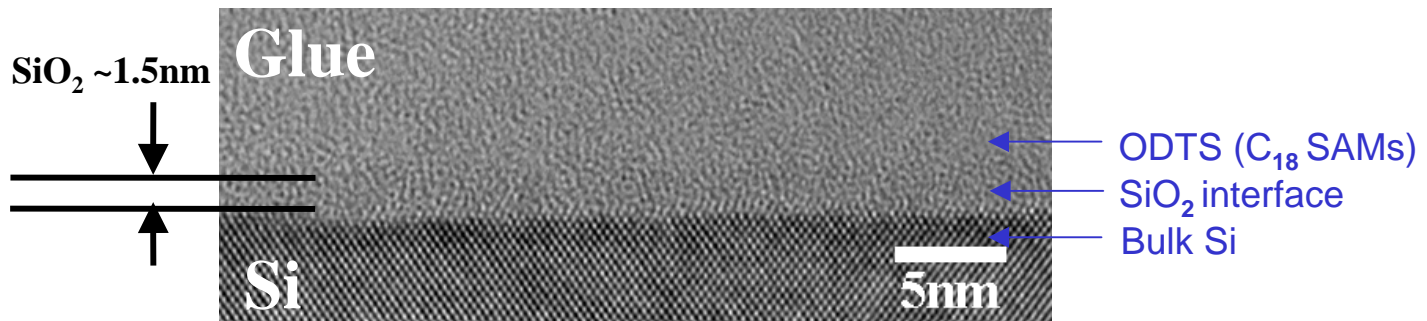
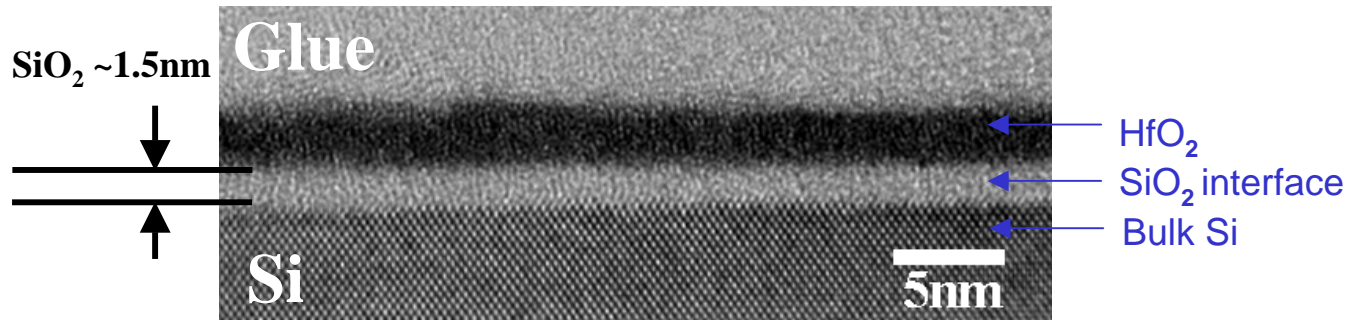
Octadecyltrichlorosilane (ODTS), $\text{CH}_3(\text{CH}_2)_{17}\text{SiCl}_3$

Let's examine other deactivating agents,
such as shorter chain alkylhalosilanes.

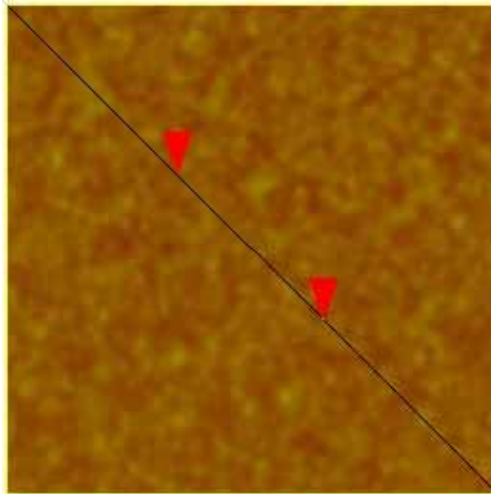
Alkyltrichlorosilane Chain Length Dependence



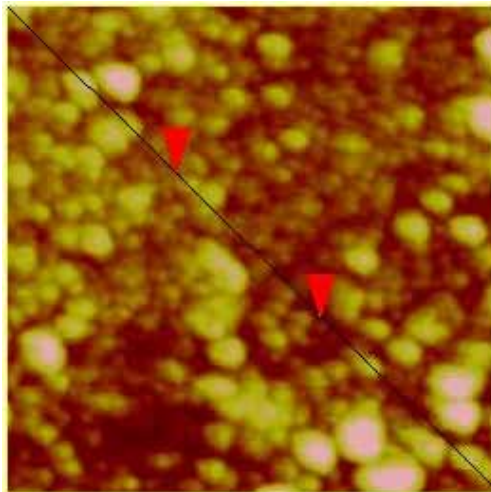
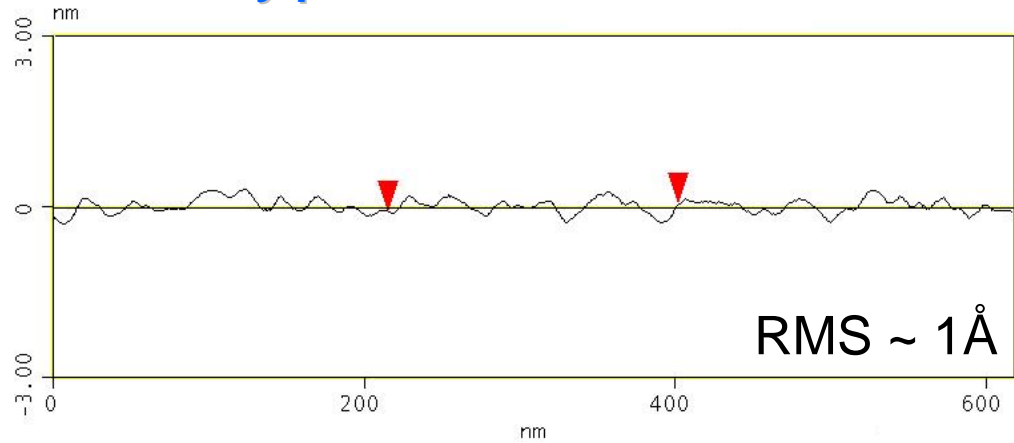
Cross-sectional TEM



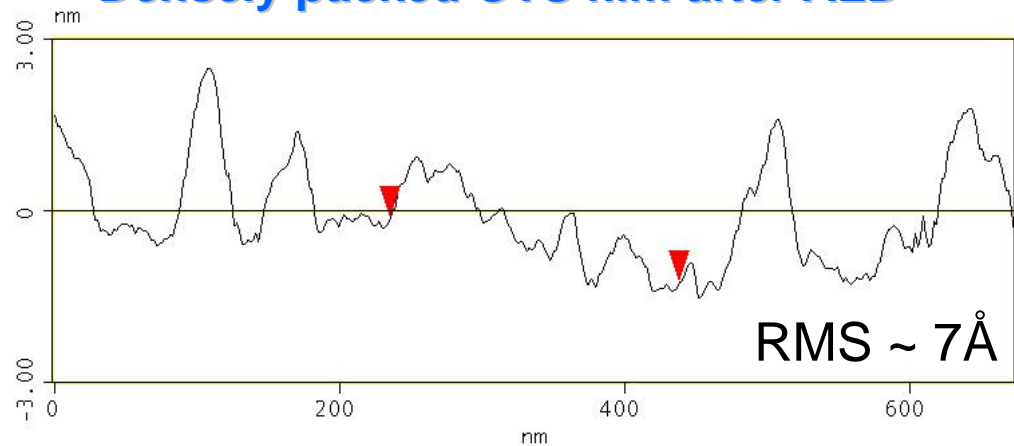
AFM Analysis of Octyltrichlorosilane before & after ALD



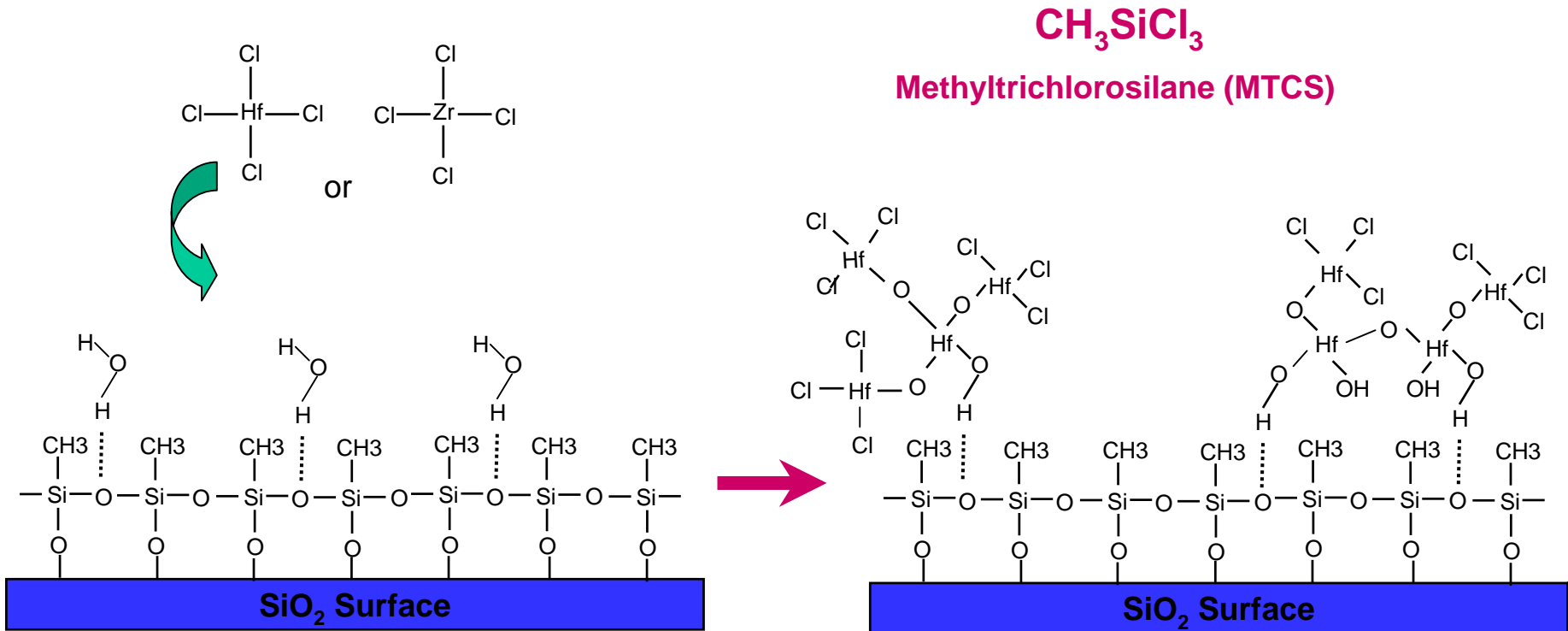
Densely packed OTS film before ALD



Densely packed OTS film after ALD



Mechanism of ALD growth on shorter chain SAM

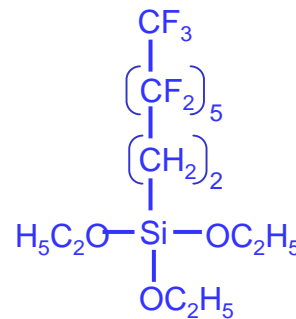
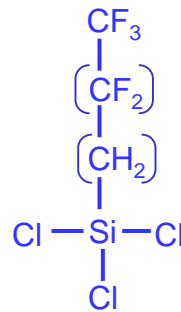
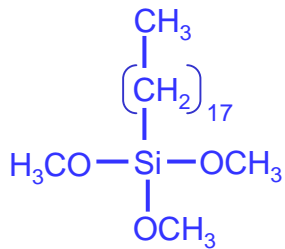
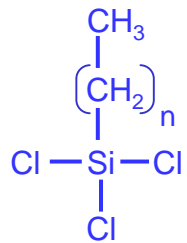


Densely packed MTCS
oxygen atoms provide hydrophilic
regions for ALD growth

ALD growth nucleates at exposed
oxygen atoms, but some precursor
attachment may be unstable

Deactivating Agents Studied

1. Chain lengths, reacting groups, and chain monomers

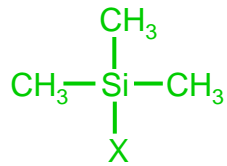


Alkyltrichlorosilane
n=0,1,3,5,7,9,11,17

2. Number of halide substituents

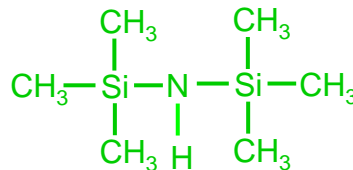


3. Reactive groups

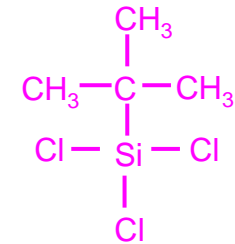
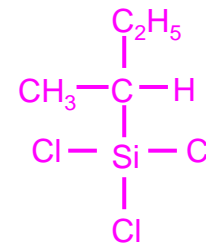
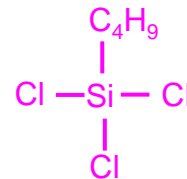
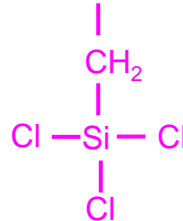
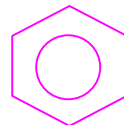


X=Cl, Br, I

Trimethylhalosilane



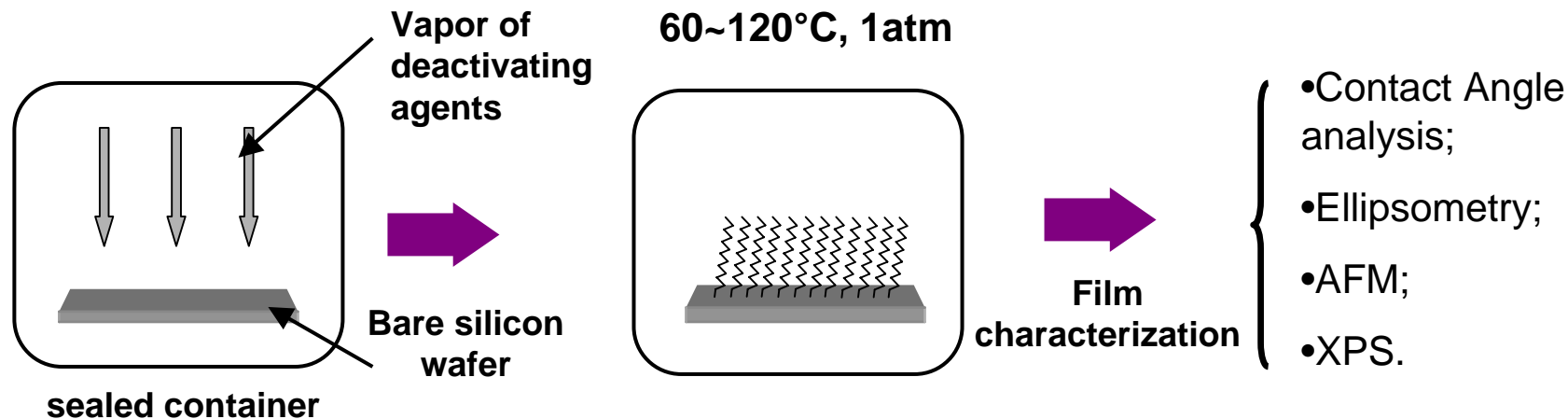
HMDS



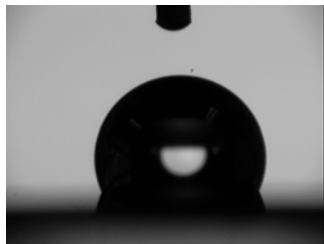
4. Type of carbon chain

Solution based delivery
Vs.
Gas phase delivery

Gas Phase Delivery of Deactivating Agents



Water Contact Angle



organosilane film
 $\theta = 110.5^\circ$
Hydrophobic

Ellipsometry monolayer thickness of organosilane

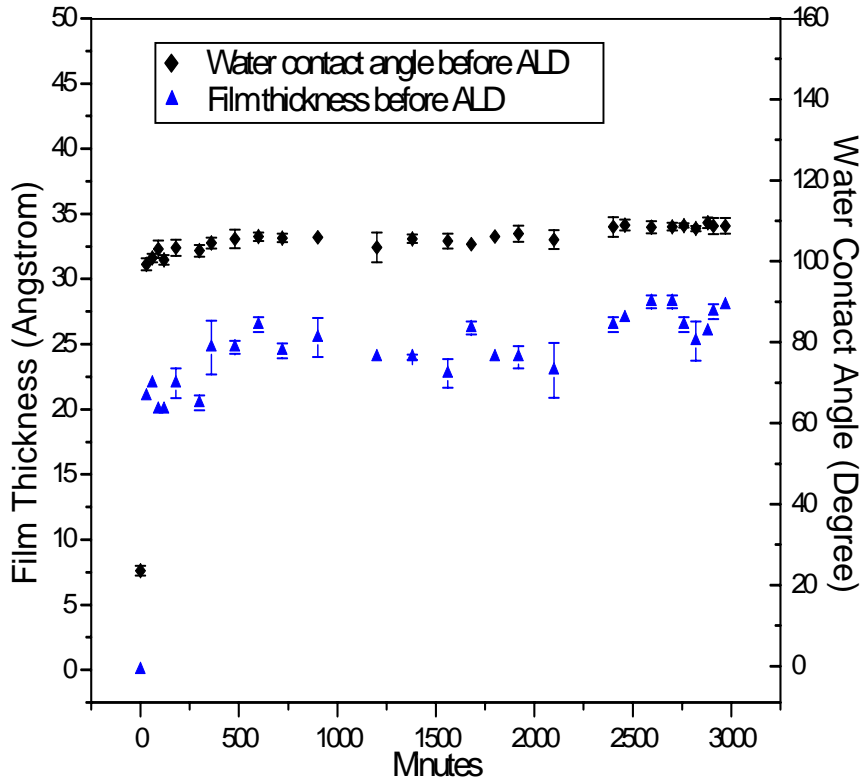
AFM analysis shows deactivating agents films formed by gas phase delivery are uniform and smooth.

Conclusion:

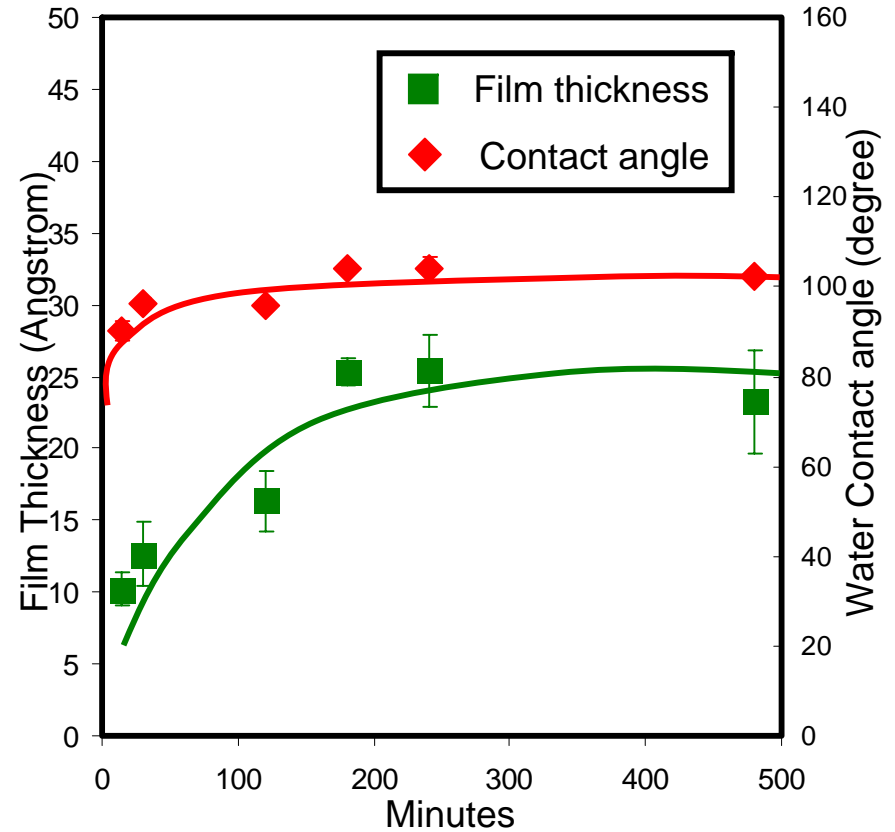
Both solution phase and gas phase delivery of deactivating agents can achieve uniform, high quality films, which can be used for direct patterning

Solution based Reaction vs. Gas Delivery Reaction

Solution Delivery Reaction



Gas Delivery Reaction

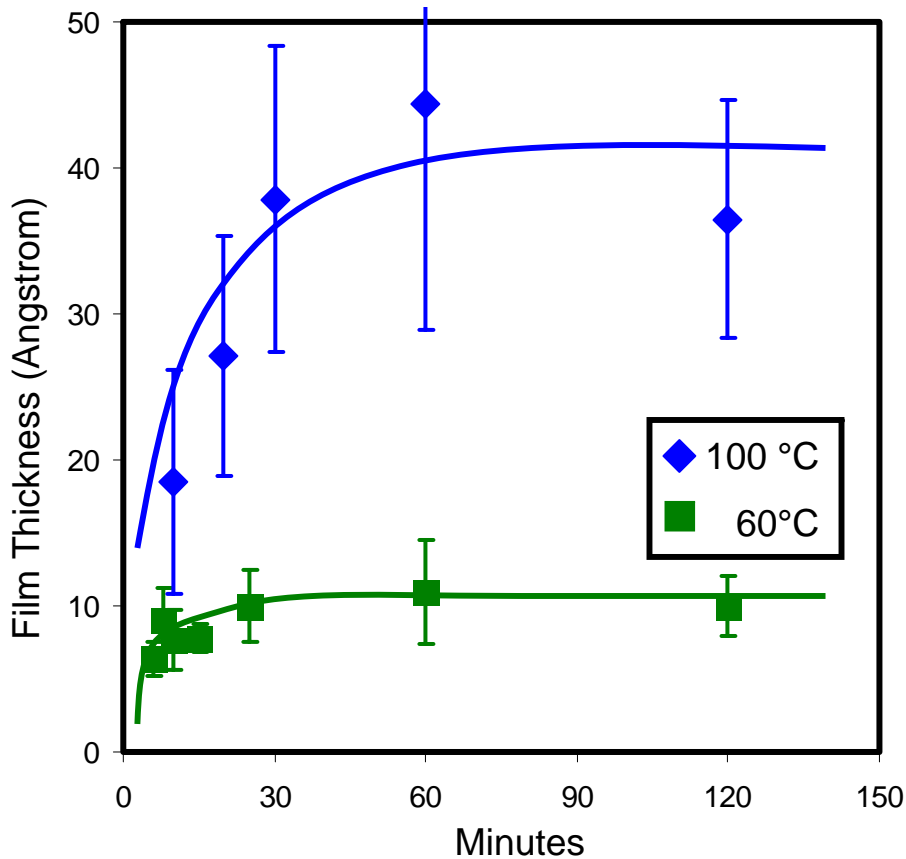


● Gas delivery reaction is much faster than solution based reaction

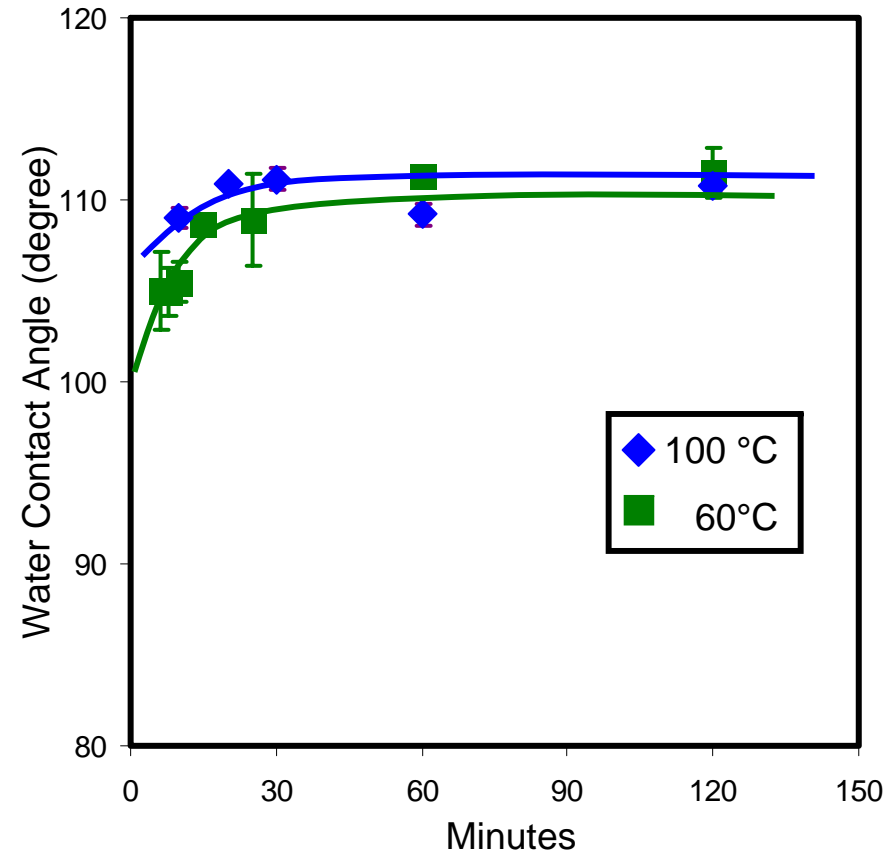
Surface Passivation as a function of CVD Temperature

- Tridecafluoro-1,1,2,2-tetrahydrooctyl trichlorosilane (FOTS)

Ellipsometry Measurement

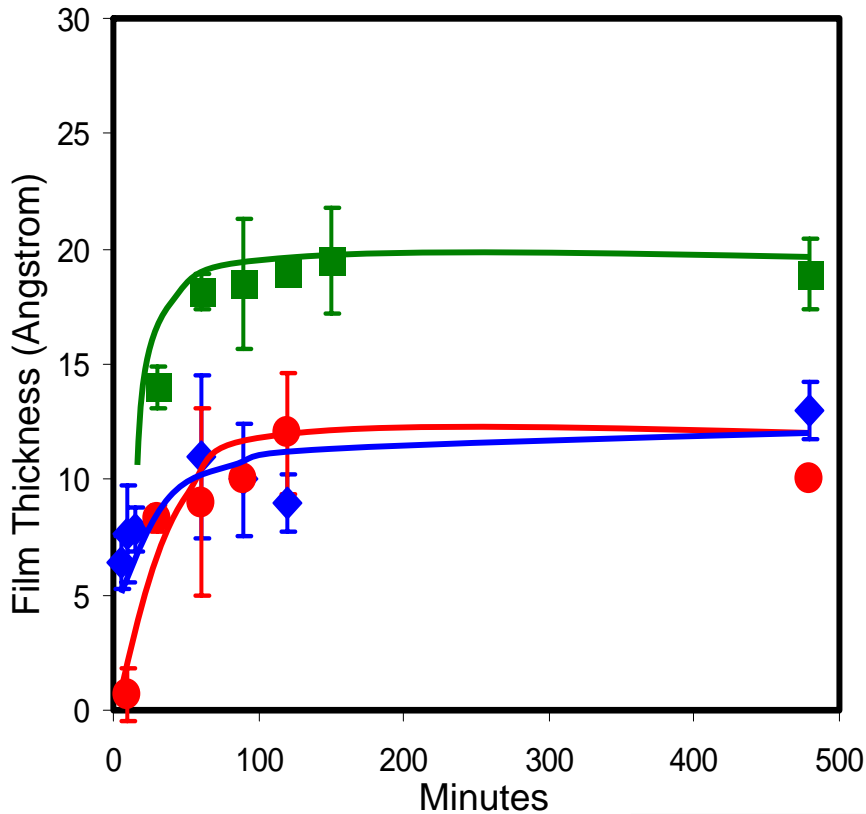


Contact Angle Measurement

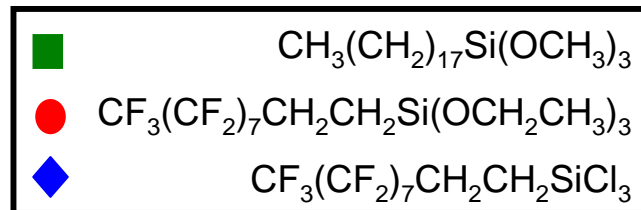
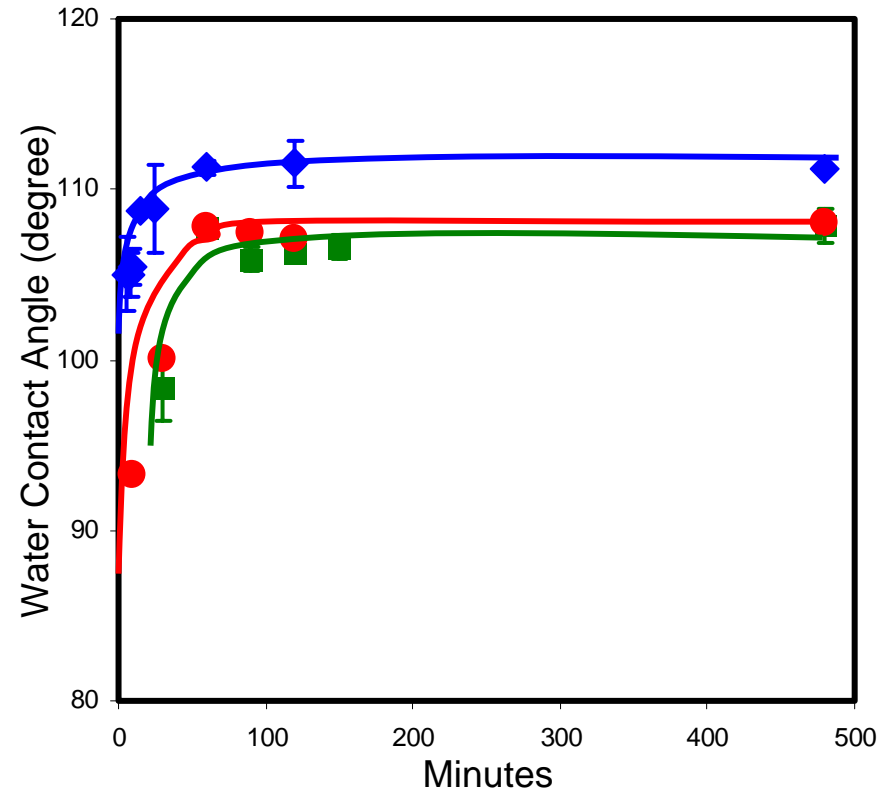


Surface Passivation of different molecules

Ellipsometry Measurement



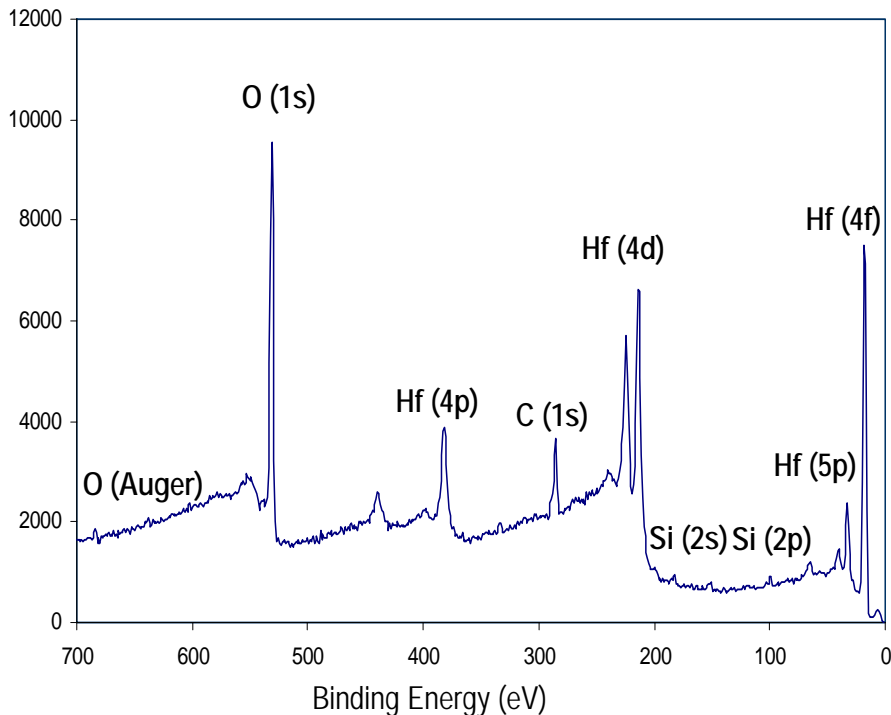
Contact Angle Measurement



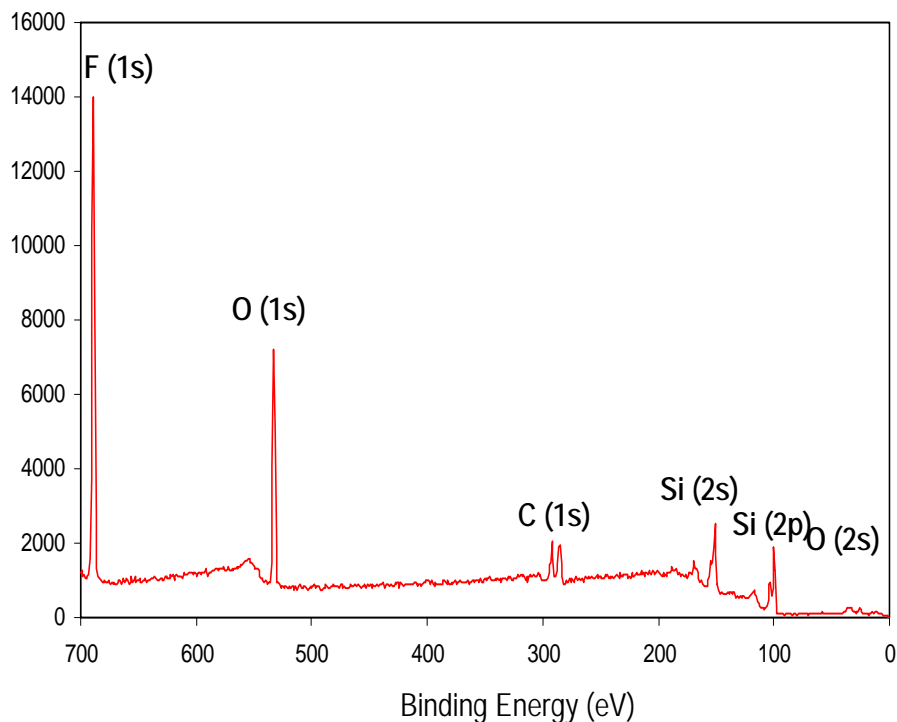
ALD Results of Gas delivered SAMs

- Tridecafluoro-1,1,2,2-tetrahydrooctyl trichlorosilane (FOTS)

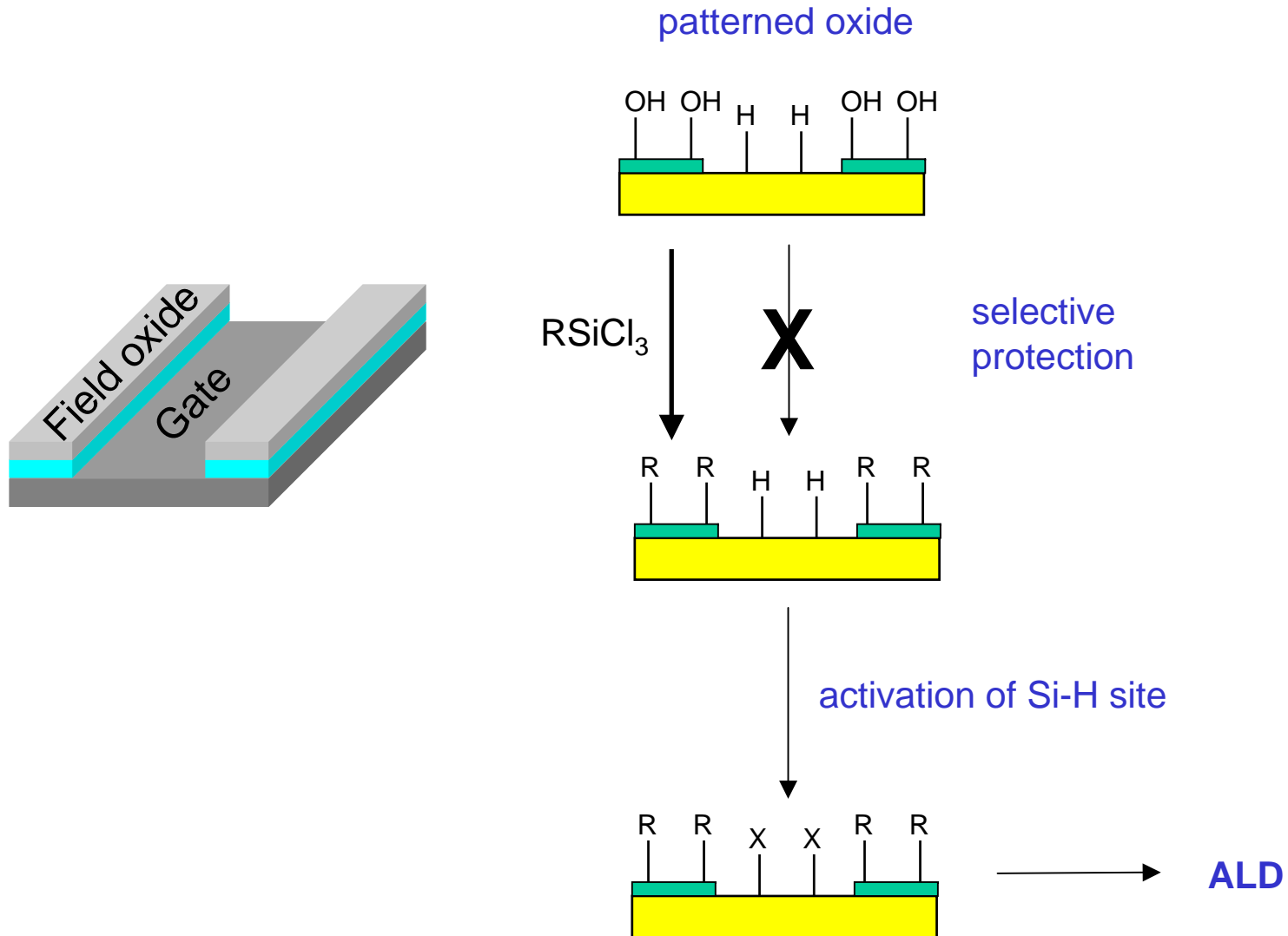
Chemical oxide silicon wafer



Silicon wafer coated with FOTS



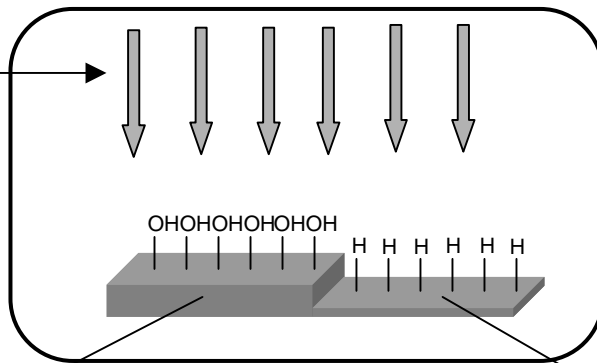
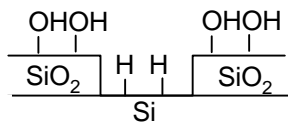
Can We Achieve the Necessary Selectivity?



Selectivity for Si-OH vs Si-H under Gas Phase Delivery

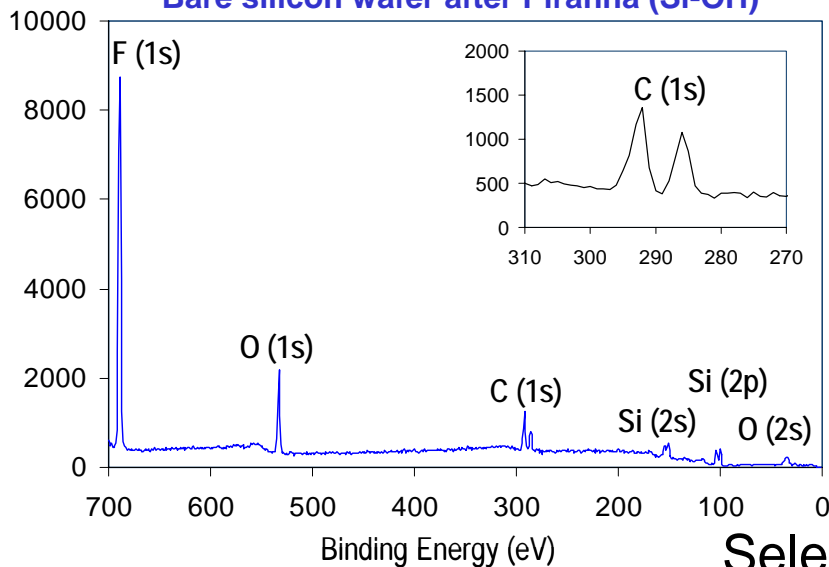
Vapor of deactivating agents

Tridecafluoro-(1,1,2,2 tetrahydrooctyl)-trichlorosilane (FOTS)

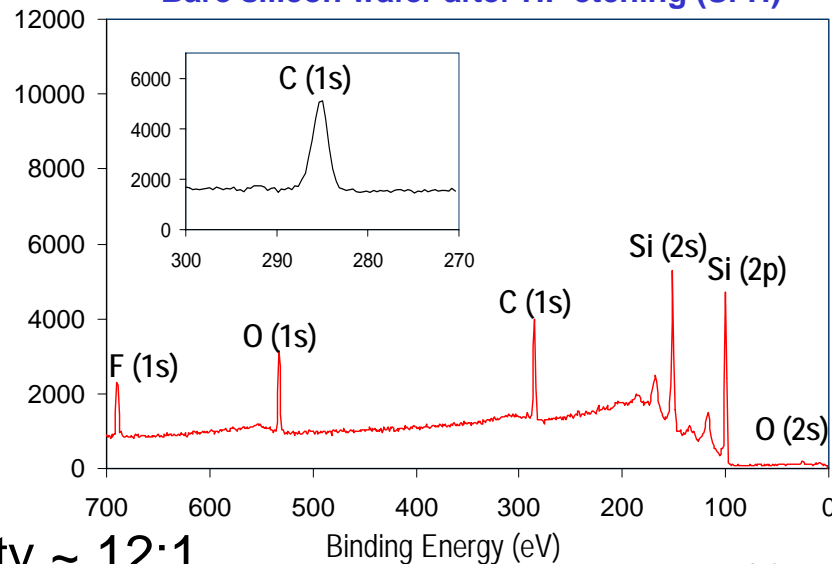


XPS Analysis

Bare silicon wafer after Piranha (Si-OH)



Bare silicon wafer after HF etching (Si-H)

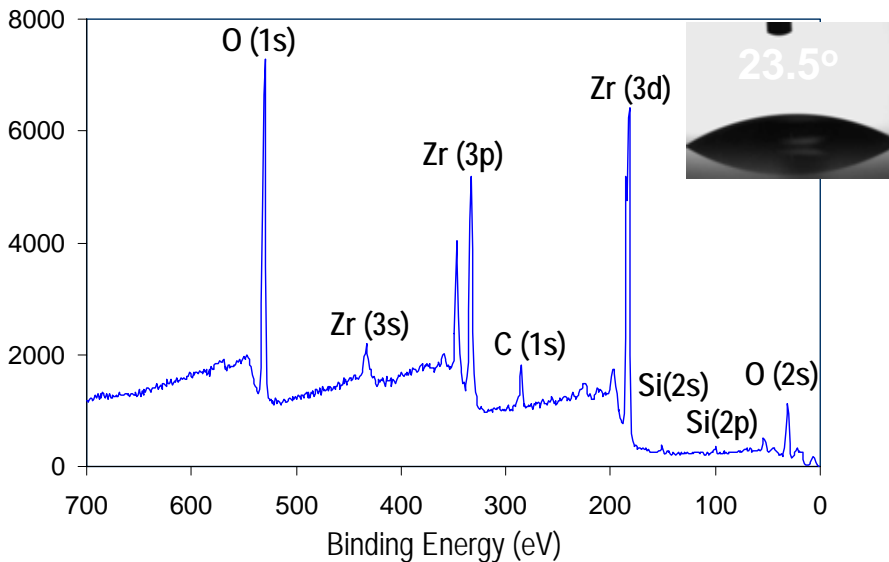


Selectivity ~ 12:1

Area Selective ALD for ZrO_2

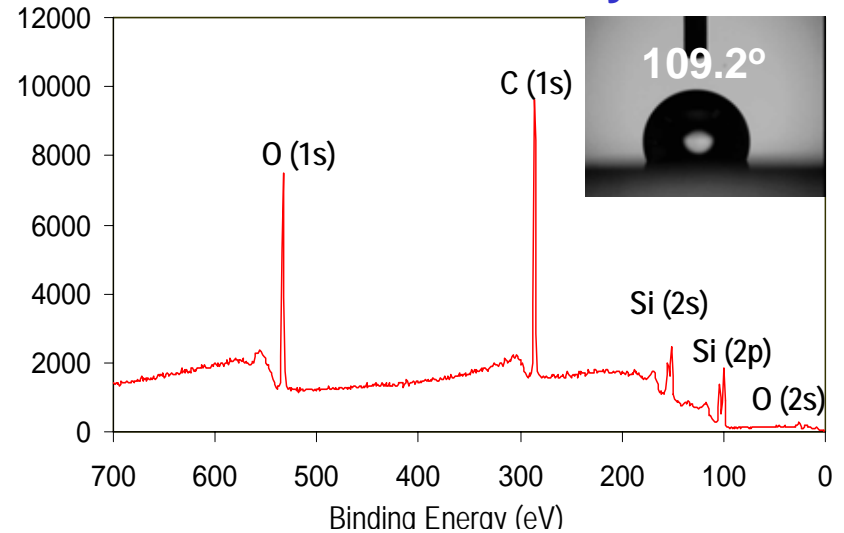
XPS Spectra after ZrO_2 ALD
on various surfaces:

Bare silicon wafer

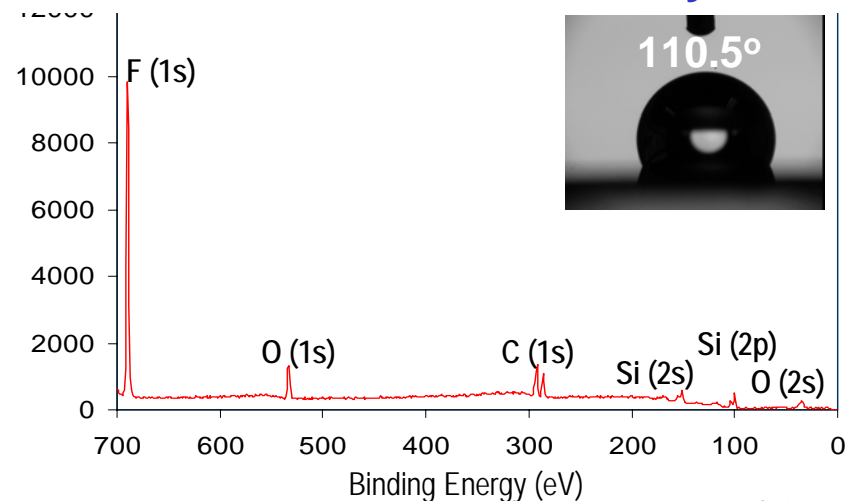


→ ZrO_2 ALD can also be effectively blocked by the deactivating agents for HfO_2 ALD.

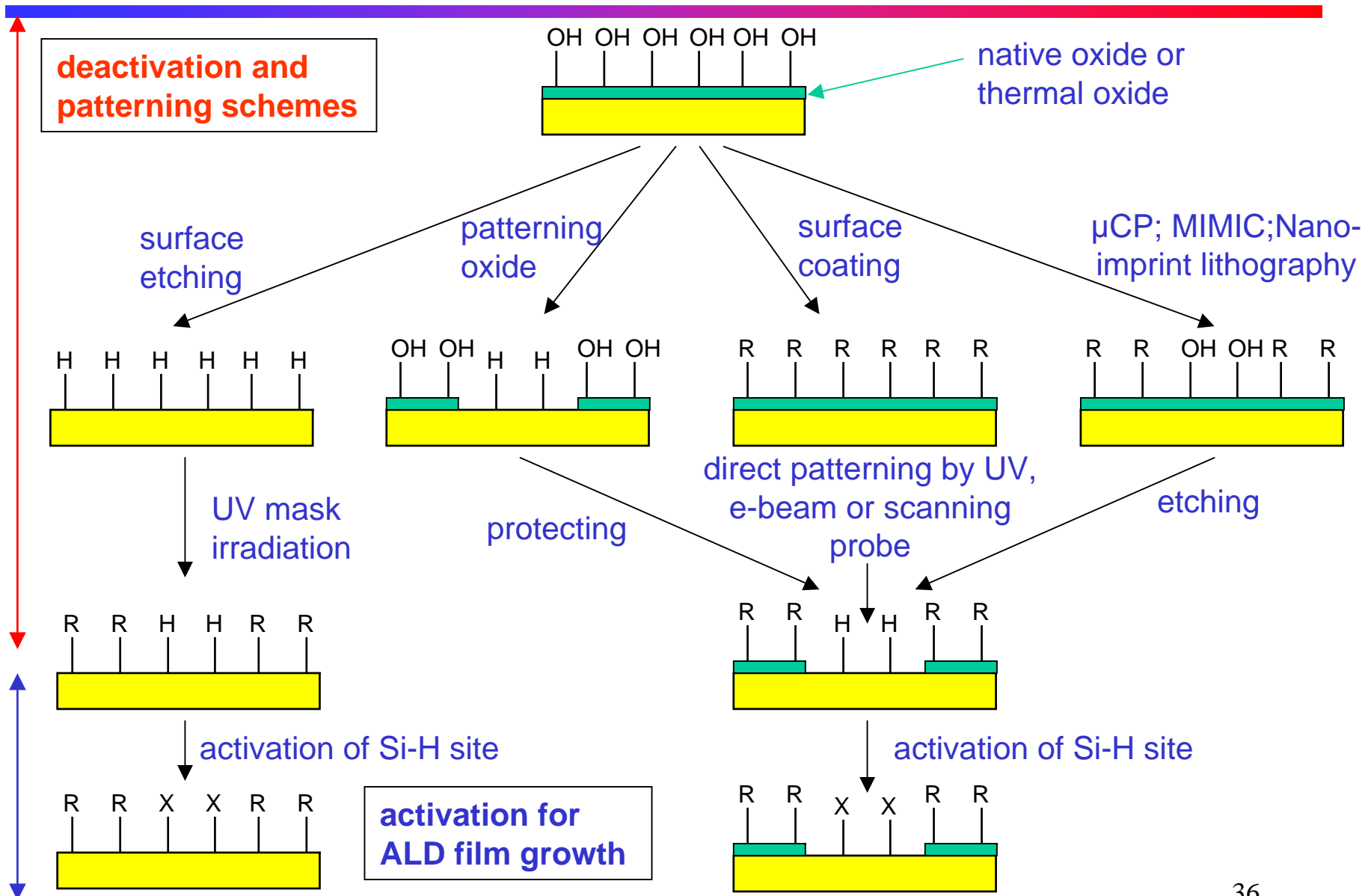
Si wafer coated with alkylsilane



Si wafer coated with fluoro-alkylsilane



Surface Modification for Selective ALD



Conclusions

- A variety of deactivating agents have been investigated
- Organosilanes are effective deactivating agents toward HfO₂ and ZrO₂ ALD
- Longer chain alkylhalosilanes which form more hydrophobic films can provide better deactivation toward ALD
- The blocking mechanisms have been investigated as a guidance for future experiment
- Both solution and gas phase delivery are promising methods for achieving high quality, dense SAMs, which can be used as a monolayer resist for area selective ALD
- Selectivity of Si-OH over Si-H is satisfactory for achieving area-selectivity

Future Work

- Deactivating strategies on Germanium substrate
- Activation strategies
- Patterning and etch methods for deactivating agents
- Extension of area-selective ALD to other chemistries
- Integration of area-selective ALD into CMOS process flow

Acknowledgments

Prof. Chris Chidsey
Prof. Krishna Saraswat
Prof. Charles Musgrave

Funding

- NSF/SRC Engineering Research Center for Environmentally Benign Semiconductor Manufacturing
- Stanford Center for Integrated Systems (CIS)
- Initiative for Nanoscale Materials and Processes (INMP)

Facilities



*Laboratory of Advanced
Materials
Stanford University*



*Stanford Nano-
Fabrication*



*Center on Polymer
Interfaces and
Macromolecular Assemblies*