Chemical Vapor Deposition of Organosilicon Composite
 Films for Porous Low-k
 Dielectrics

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# ••• Goals

- Create a Porous, Low- κ Film by CVD
  - Rigid Organosilicon Matrix
  - Thermally Labile
    Porogen
  - Deposition by Pulsed
    Plasma Enhanced CVD

	Composition SiO <sub>2</sub> Si:O:C:H (Organosilicate Glass – OSG)	Fully den: 4.0 2.7–3.0	se $\kappa$ Air $\kappa = 1.0$		
<u>% Porosity</u> <u>κ</u>					
		0	2.7		
		20	2.3		
		50	1.75		
		90	1.15		

# Technology Requirements - к

#### Interconnect Technology Requirements - Near Term

Year of Production	2001	2002	2003	2004	2005	2006	2007
Interlevel metal insulator effective k-value	3.0-3.6	3.0-3.6	3.0-3.6	2.6-3.1	2.6-3.1	2.6-3.1	2.3-2.7
Interlevel metal insulator bulk k-value	<2.7	<2.7	<2.7	<2.4	<2.4	<2.4	<2.1

#### Interconnect Technology Requirements - Long Term

Year of Production	2010	2013	2016
Interlevel metal insulator effective k-value	2.1	1.9	1.8
Interlevel metal insulator bulk k-value	<1.9	<1.7	<1.6

Source: ITRS Roadmap for Semiconductors: 2001 Update - Interconnect - http://public.itrs.net

# ••• Industry Outlook

• The widely used ILD material for  $0.13\mu m$  and older technologies are PECVD SiO<sub>2</sub> and SiOF

Materials/ Technology	0.13μm or 0.09μm	0.07µm	0.05µm	
Organic	SiLκ <sup>™</sup> , Flare <sup>™</sup> , Paralyne-F(N), αFC, PAE,etc.	Porous SiLĸ <sup>™</sup> , Porous Flare <sup>™</sup> , OXD, etc	Partial Air Gap, Complete Air Gap	
Organosilicates	Carbon Doped Oxide, SOG, etc	Porous CVD CDO, Porous SOD, CDO, etc.	Partial Air Gap, Complete Air Gap	
Range of ĸ	2.8 to 3.0	1.9 to 2.6	1.0 T to 1.5	

Dr. Eb Andideh, Intel Corporation (2003, MIT hosted ERC teleconference)

### Solventless Low-κ Dielectrics

 Manufacturing Metrics: Lowering the dielectric constant of current ILDs leads to fewest levels of interconnect → economic and environmental "win-win"



### CVD

- Thin, Conformal Coatings
- Solventless
- Ease of Integration
- Mechanical Properties
- Unclear Strategy for porous materials

#### Spin-On

 Extendibility to Future Generations by adding pores



Goolo/	Usa	Jsage Reduction		Emission Reduction			
Possibilities	Energy	Water	Chemicals	PFCx	VOCs	HAPs	Hazardous Wastes
Pulsed Plasma Enhanced CVD for k < 2.2	Data not available	NA	Order of magnitude greater consumption for spin-on (Hendricks)	Higher emissions for CVD due to chamber cleans	Great reduction vs Spin- on	Some reduction in acid vapors	Spin-on requires solids waste disposal



**OSG Mechanical Properties** CH<sub>3</sub> CH<sub>3</sub> O - Si - CH3-0-Si--0-SICH<sub>3</sub> CH<sub>3</sub> CH<sub>2</sub>

'M' Group Chain Terminating

'D' Group Chain Propagating

'T' Group Chain Crosslinking

### Condensation Reactions



## Monitoring Condensation





### ••• OSG Mechanical Properties



### OSG Mechanical Properties



#### Increasing the $H_2O_2$ flowrate ....

- Amount of 'T' groups relative to 'D' groups increases
- Both Modulus and Hardness increase
- Annealing increases percentage of T groups and Modulus

# ••• Composite Materials

#### Polystyrene Beads

- Bead Diameters: 15nm (std = 3), 96nm (std = 9)
- Health=0, Flammability=1, Reactivity=0
- 1% Styrene in Air: Health=1, Flammability=0, Reactivity=1

#### Cyclodextrin

• 1.54nm diameter



Spin On: Jin-Heong Yim, et al. 2003



Polystyrene Beads







### Reactor Configuration



### Reactor Configuration



#### **Ultrasonic Atomization**

- Uses low ultrasonic vibrational energy for atomization
- Dispenses microliters/min
- Pressureless atomization
  - Can handle up to 30% solids

#### Atomized Delivery of Water and Porogen



## ••• Porogen Detection

- Dextran
  - Molecular Weight: 3000 g/mol <sup>H</sup>

НÒ

 $\cap$ 

ĠН

Texas Red<sup>®</sup> labeling





### Dextran & OSG FTIR Analysis







# Conclusions for Porogens

- Porogen Delivery
  - Ultrasonic atomization used to deposit porogen under vacuum conditions
  - Dextran co-deposited with OSG matrix
  - Removal of dextran through annealing
- Pulsed Plasma Enhanced CVD process for depositing OSG films
  - OSG thin films were deposited using D<sub>4</sub>
  - $H_2O_2$  used as oxidant to promote –OH bonding
  - Condensation reaction improves hardness
  - Film properties linked to structure through FTIR



### Either increasing fragile porous low k materials must be integrated

or

A robust sacrificial layer must be integrated which can form air in the final step (sacrificial layer = 100% porogen)

> • Havemann and Jeng (TI), US Patent 5461003, 1995.

• Anand et al., IEEE, 1997.

Gleason Group Research by Tom Casserly and Kelvin Chan

### ••• Extreme Properties

- Air has the lowest possible dielectric constant of 1.0
  - reduced RC delay
  - lower power consumption
  - lower cross-talk noise
- Air has the lowest possible refractive index of 1.0
  - high index contrast in optical devices
    - (e.g. thin-film optical filters)

#### Future Microprocessor Interconnect Technology Requirements\*



\*Source: International Technology Roadmap for Semiconductors 2001

### Sacrificial Layers in Air Gap Formation

- Rate of Decomposition
- Pressure in Air Gap Cavity
- Diffusivity of Decomposition Products in Overlying Dielectric

#### **Temperature**



#### **Closed Cavity Air Gap**

# ••• Single-Level Structure





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