The background of the slide is a dark, circular image of a plasma reactor. It features a glowing blue plasma discharge within a cylindrical chamber. In the center, there is a horizontal substrate holder. The overall appearance is that of a high-tech laboratory equipment used for chemical vapor deposition.

Effect of Substrate Temperature for Plasma-Enhanced Chemical Vapor Deposition of Poly(methylmethacrylate) as a Sacrificial Material for Air Gap Fabrication

NSF/SRC ERB for Environmentally Benign Semiconductor Manufacturing

Tom Casserly, SRC/Novellus Fellow, Department of Chemical Engineering, MIT

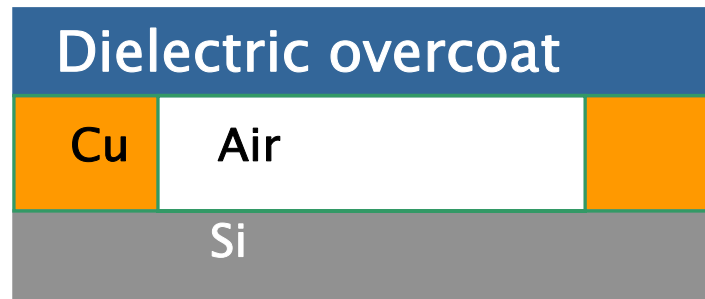
January 20, 2005

Thrust A2: Solventless Low-k Dielectrics

- **Why investigate air gaps?**
 - Dense solutions are approaching the limit for OSG materials
 - Integration of porous materials is an integration nightmare
 - Leap forward in performance increase as opposed to incremental improvements that face difficult challenges
- **Semiconductor International – January 1, 2005**
Senior Editor, Laura Peters
 - *Argues for air gaps as a low-k alternative needed at this time*
 - *“Porous dielectrics have too many problems — at least as many as their dense counterparts — and the overall benefit they deliver, represented as the effective k value, may be small given the integration, yield and reliability challenges they pose and the costs required to surmount them.”*

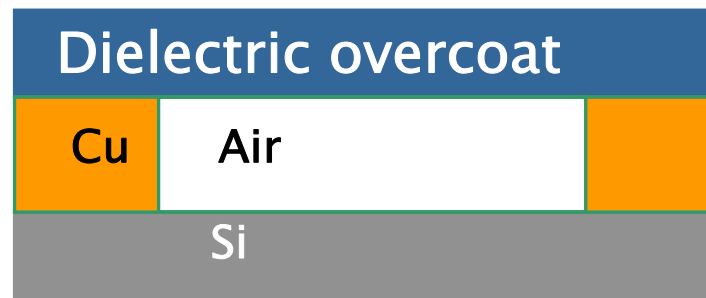
Why Air Gaps: Low- κ Solution

- **Air has lowest dielectric constant**
 - Faster Integrated Circuits
 - Decrease power consumption
 - Reduce cross-talk
 - Leads to fewer levels of interconnect
- **Air has lowest refractive index**
 - Use in optical filters/reflectors/refractors
- **Closed cavity air gap**
 - Additional applications in microfluidic devices



Anatomy of an Air Gap

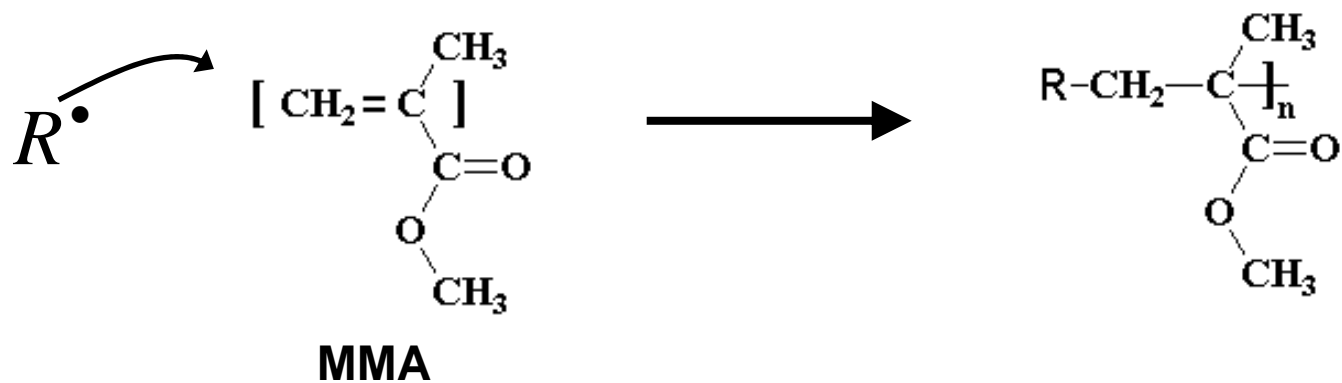
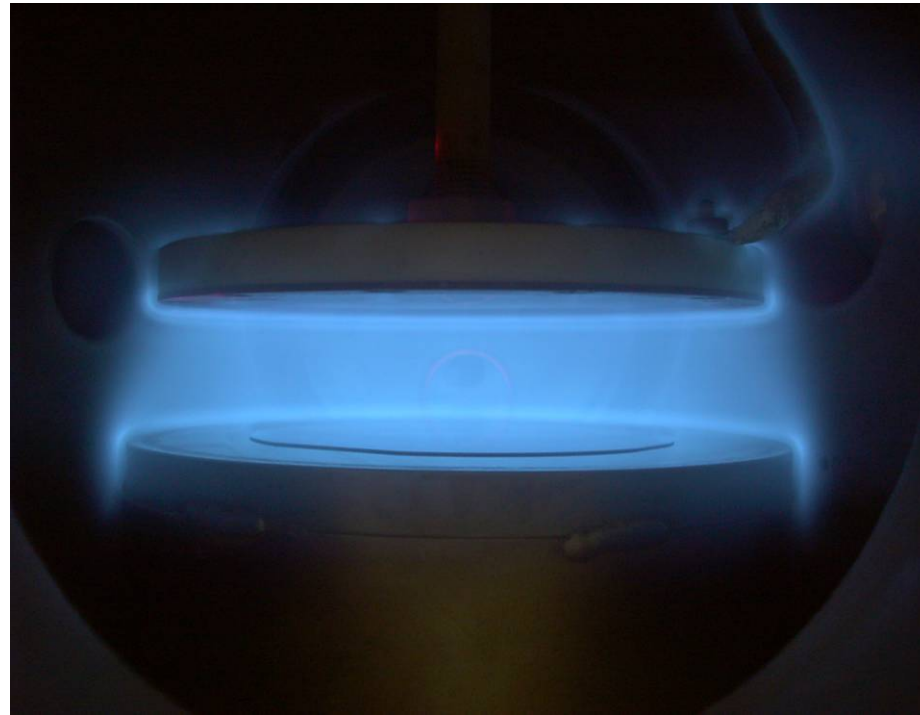
- **Overlying Dielectric**
 - Organosilicate Glass (OSG)
 - Remains as bridge layer
- **Metal Lines and Barrier Layers**
- **Sacrificial Material**
 - Poly(methylmethacrylate) (PMMA)
 - Cleanly removed to form air gap



Criteria for Sacrificial Materials

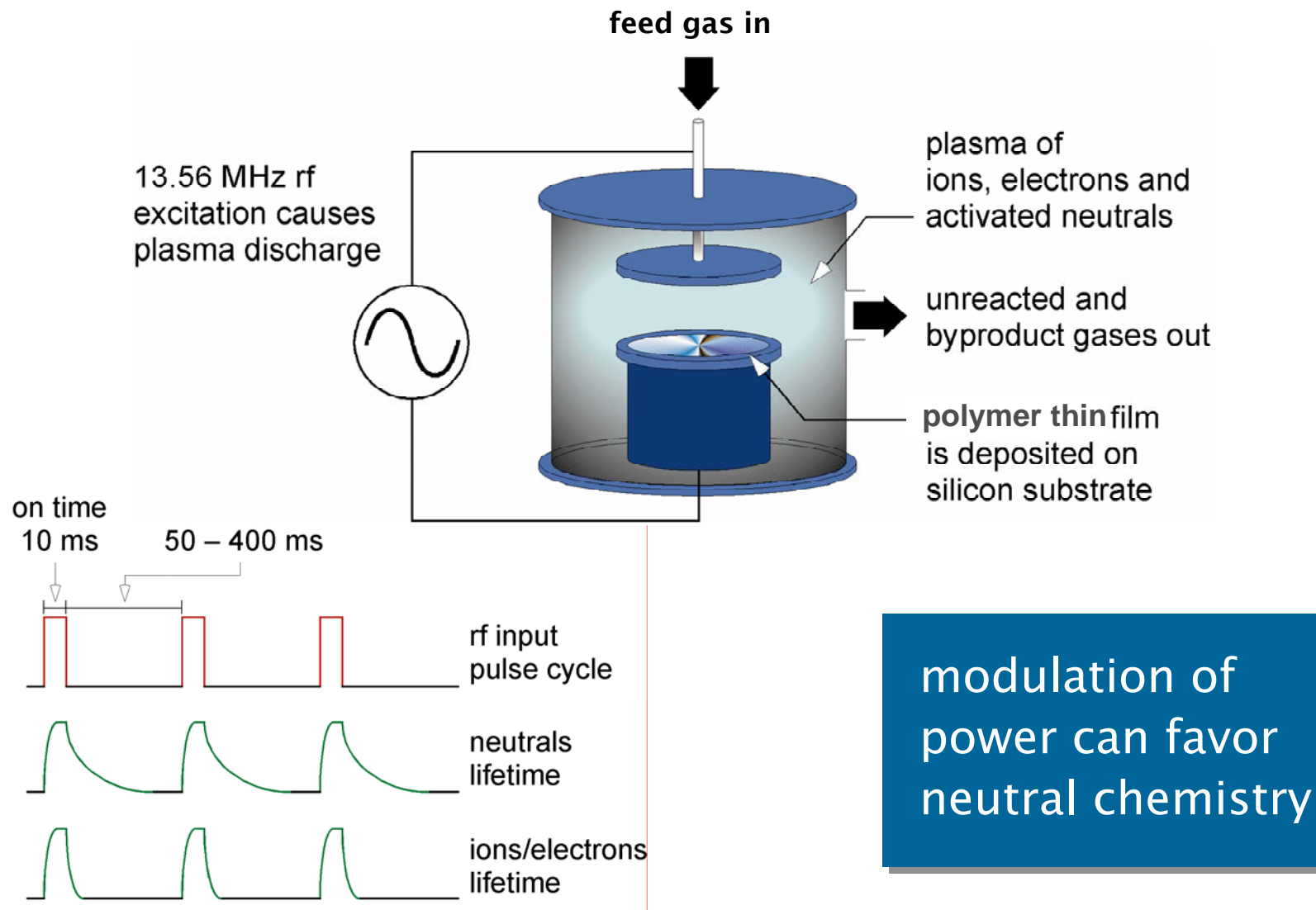
- **Ease of synthesis**
- **Decomposition in the absence of oxygen**
- **Desirable temperature range for decomposition**
- **Short decomposition time (small degradation fragments)**
- **Negligible residue**
- **Does not affect ILD and metal during deposition and removal**
- **Ability to be patterned**
- **Good mechanical properties & adhesion**
- **Survives aqueous and solution rinses**

PECVD of Poly(methyl methacrylate) (PMMA)



Free radical polymerization of MMA gas to form PMMA solid

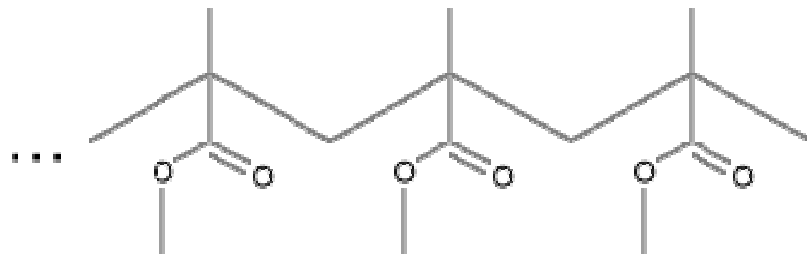
Pulsed Plasma-Enhanced CVD (PPECVD)



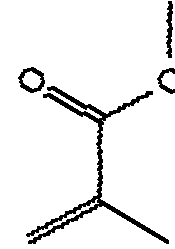
Process variables

- Substrate T: 50 to 130°C
- Power conditions: 10W CW v. 100W 10/90 Pulsed
- Pressure: 0.5 to 3 Torr

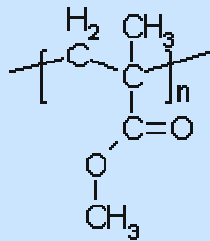
Poly(methyl methacrylate) (PMMA)



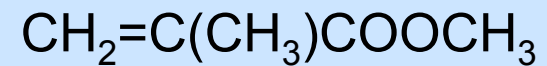
CLEAN decomposition



PMMA



Methyl Methacrylate



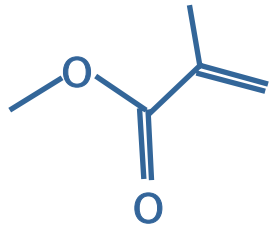
Bulk PMMA is insoluble in water and IPA at room temperature.

As a bulk polymer, PMMA is known for its outstanding mechanical properties.

Applications

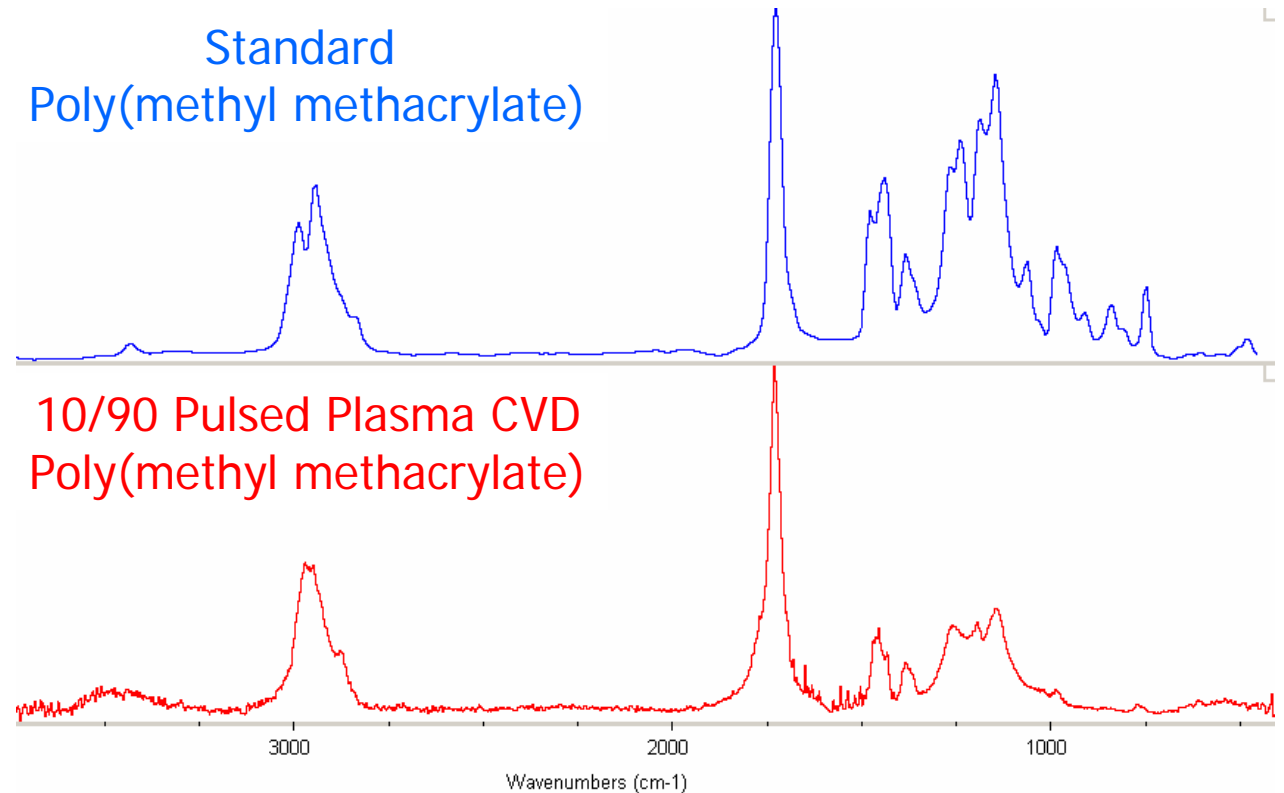
- Used in orthopedic surgery
 - Fix prosthetic components
 - Bone graft template
 - Femoral window plug in total hip replacement
- Used in various plastics, floor polish, glues, and packaging

PMMA Deposition and Characterization

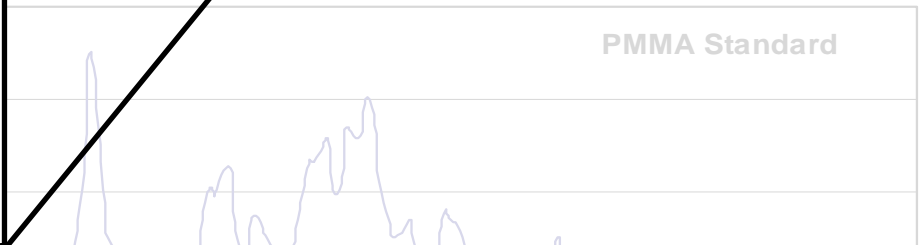
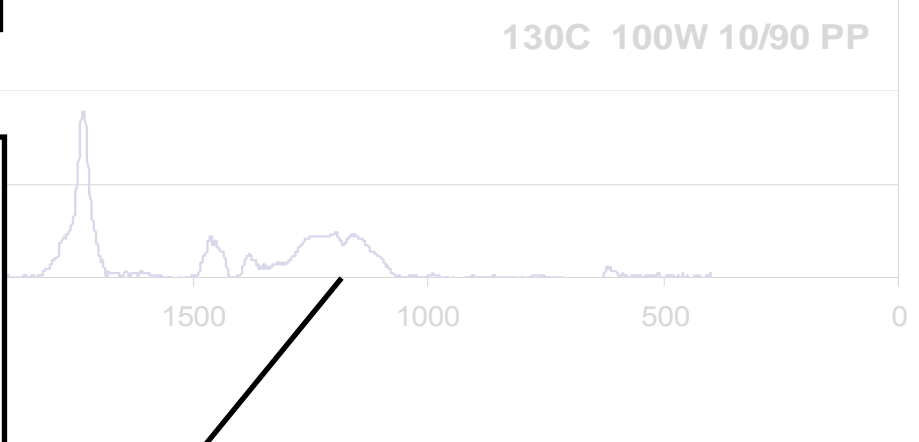
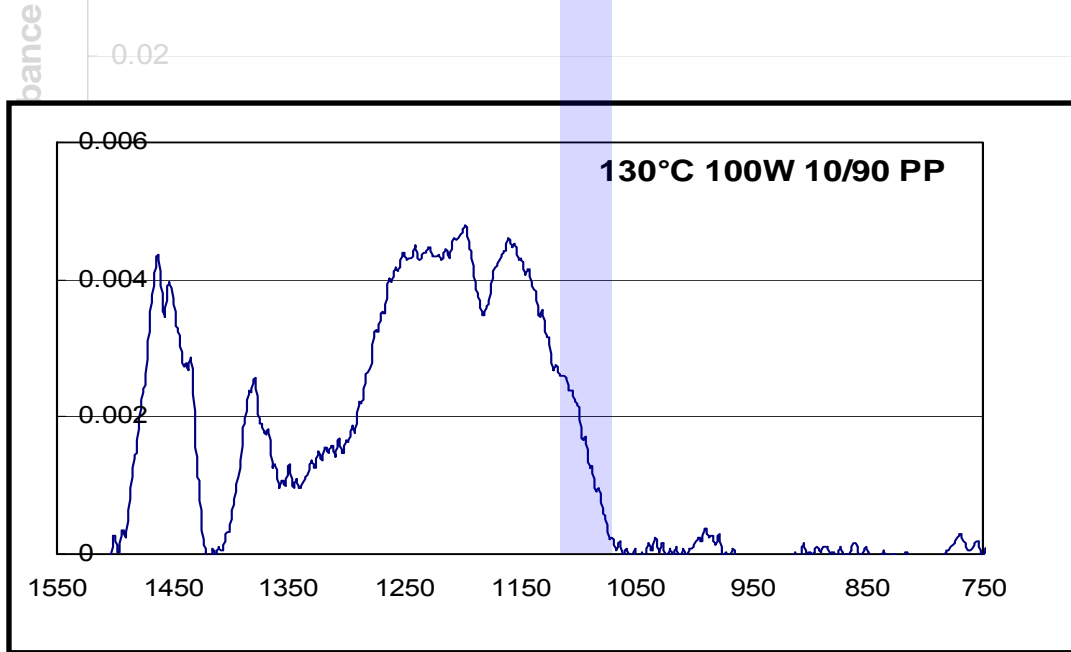
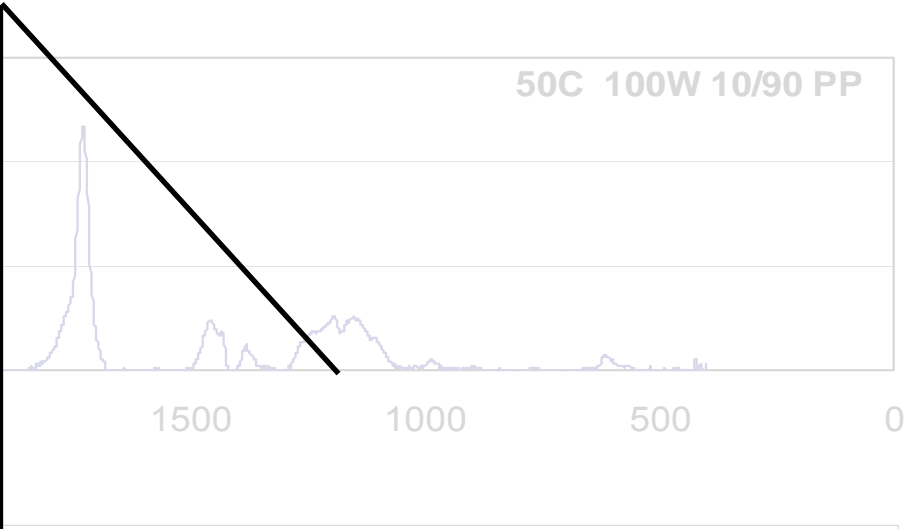
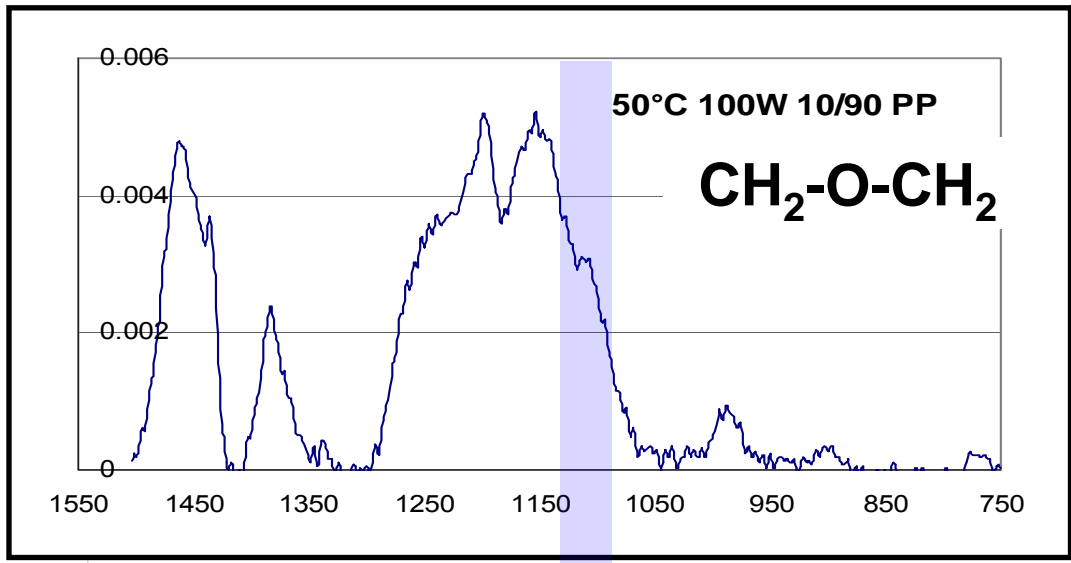


Methyl Methacrylate

- Polymerizes via vinyl bond to Poly(methyl methacrylate) (PMMA)
- Low duty cycle PPECVD retains much of PMMA functionality
- Thermally decomposes in the absence of oxygen to monomer form below 350°C

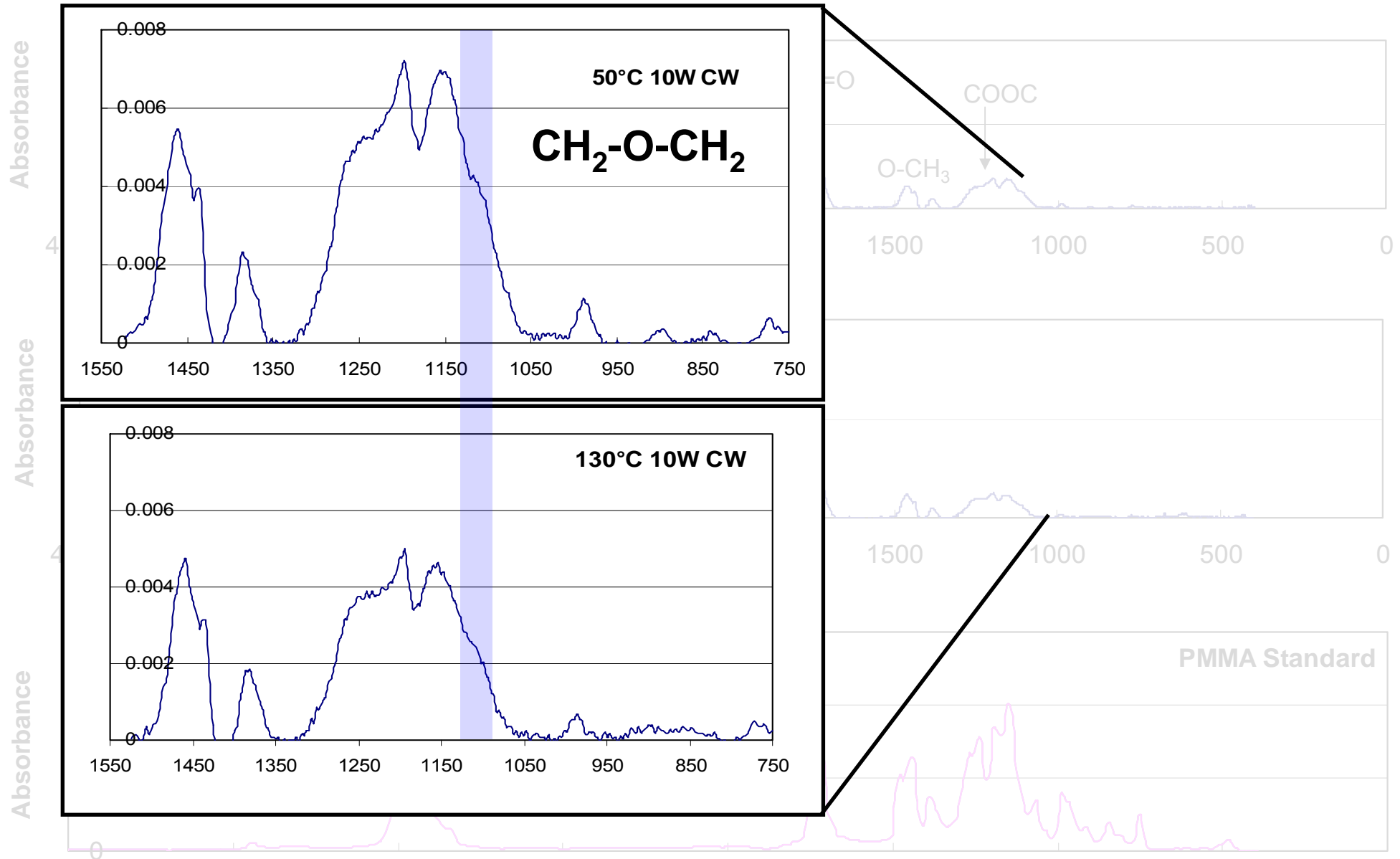


FTIR of Pulsed PECVD PMMA – 100W 10/90



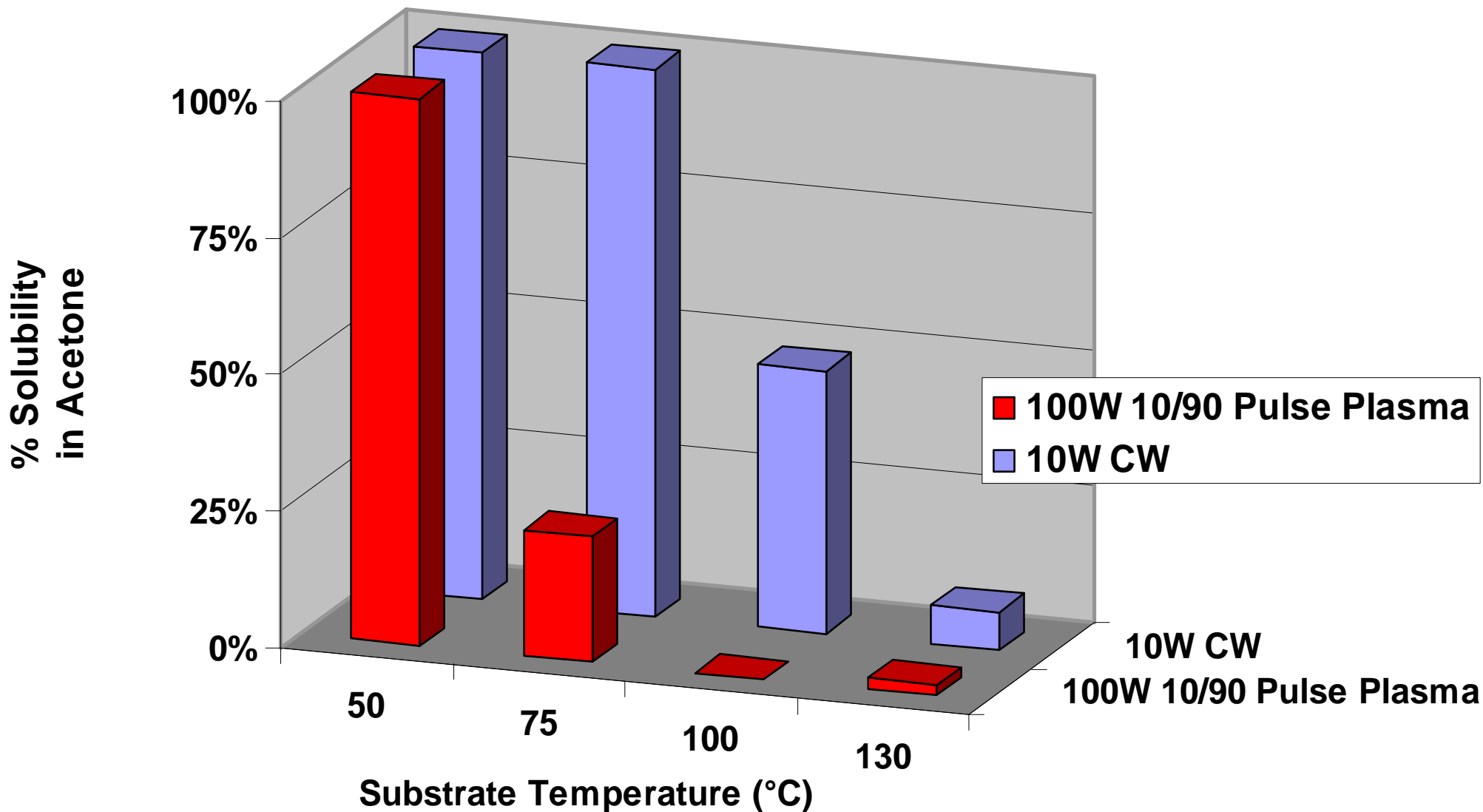
Increased substrate temperature decreases oxygen incorporation into polymer backbone
Improves the thermal stability requiring a higher temperature for onset of decomposition

FTIR of PECVD PMMA – 10W Continuous Plasma



Increased substrate temperature decreases oxygen incorporation into polymer backbone
Improves the thermal stability requiring a higher temperature for onset of decomposition

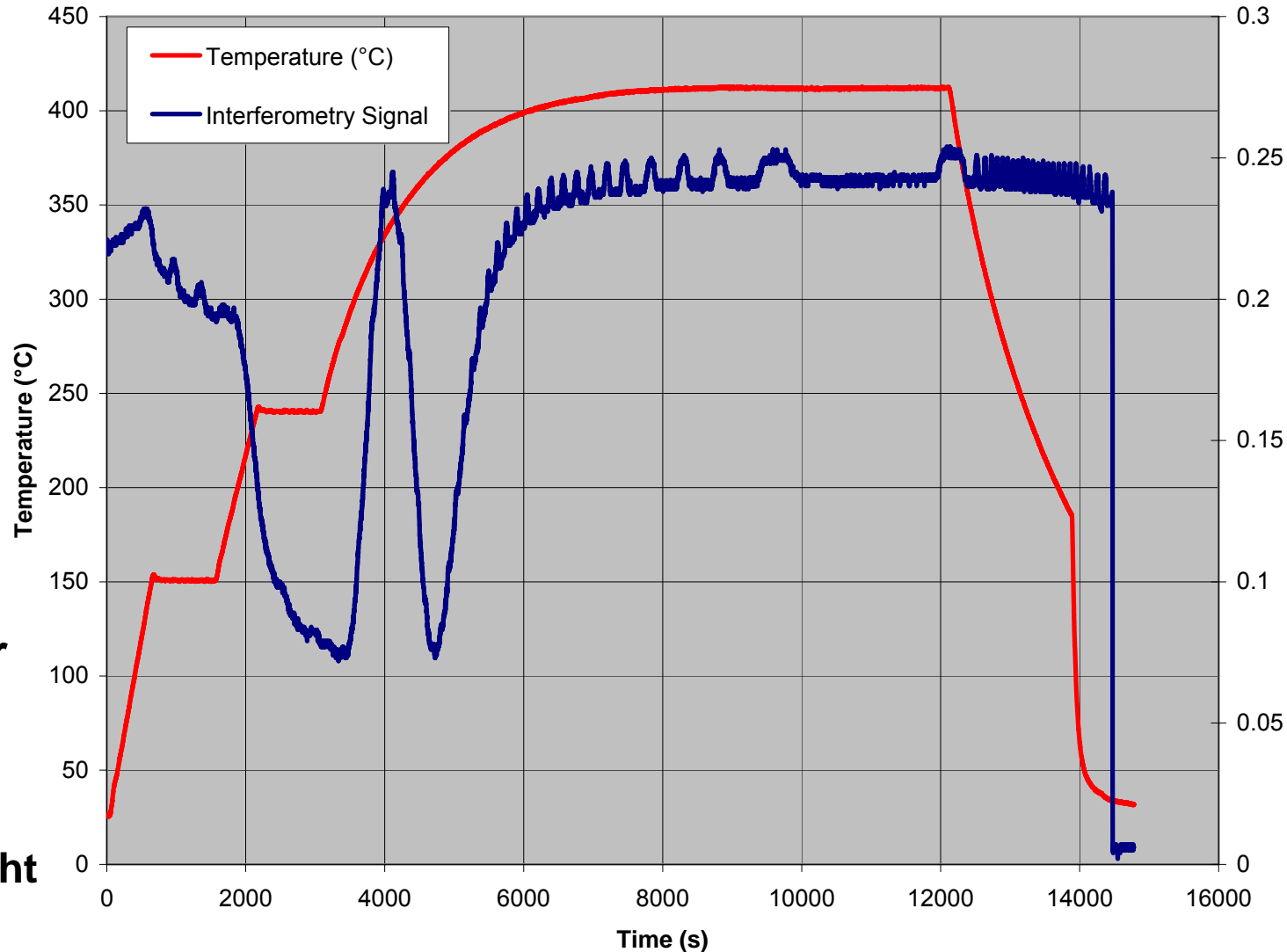
Solubility in Acetone v. Substrate Temperature



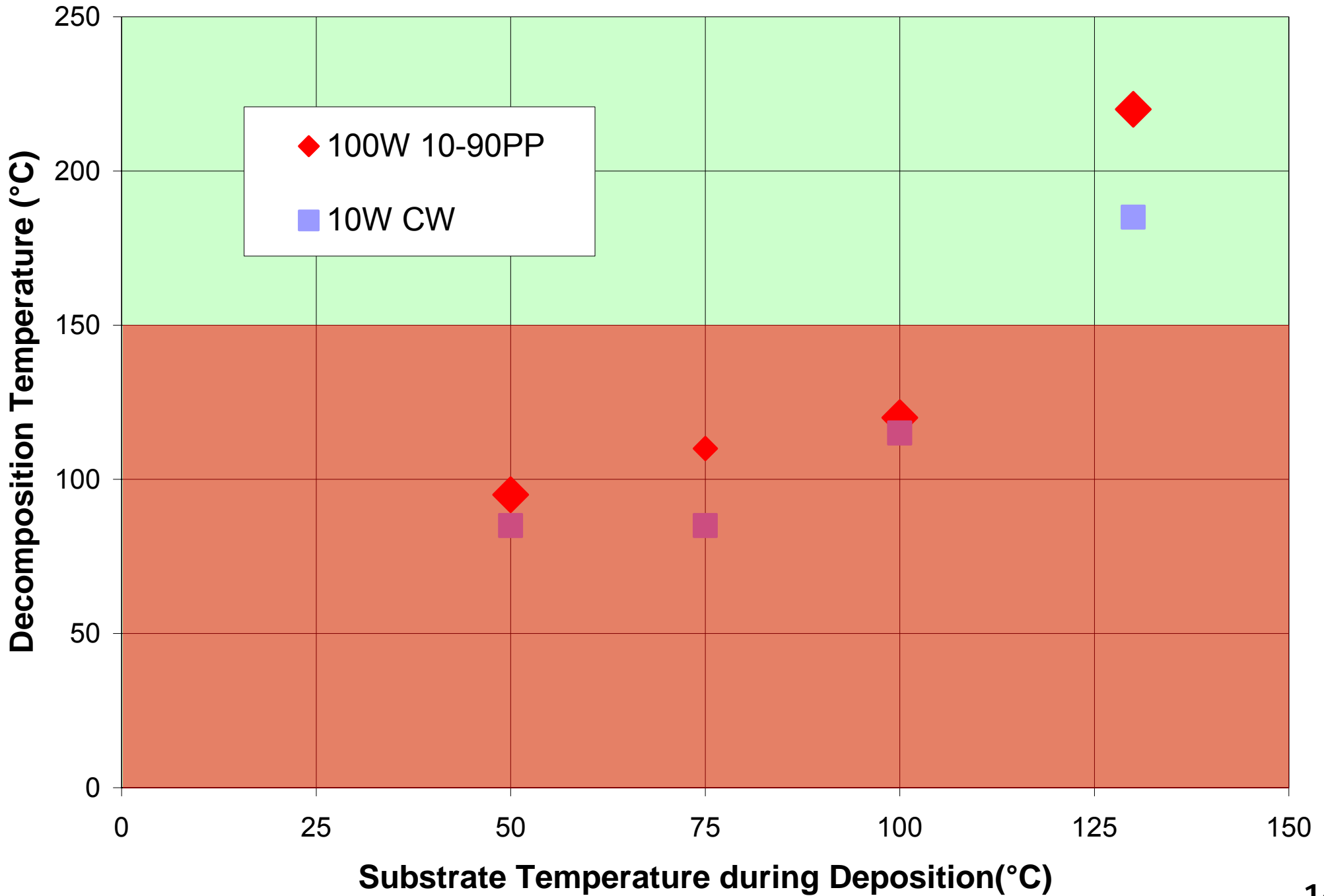
Interferometry for Thermal Stability (ITS)

- Nitrogen purged annealing chamber
- Ramp to 150°C
- Hold for 30 min
- Ramp to 240°C
- Hold for 30 min
- Ramp to 410°C
- Hold for 1 hour
- Cool

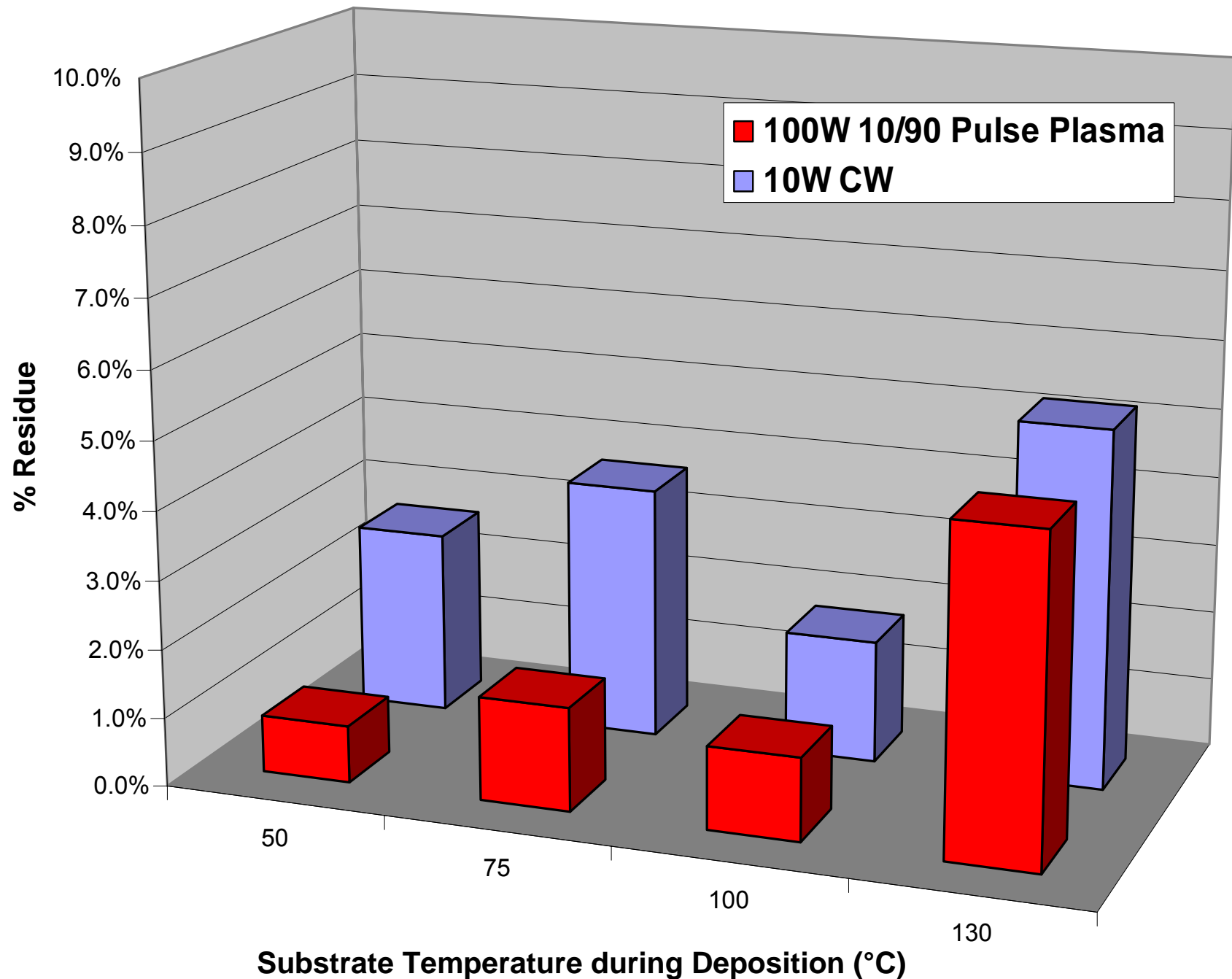
- Example ITS scan for 10W CW sample deposited at 100°C
 - Film swelling
 - Low molecular weight and weak bond breaking
 - Macro decomposition



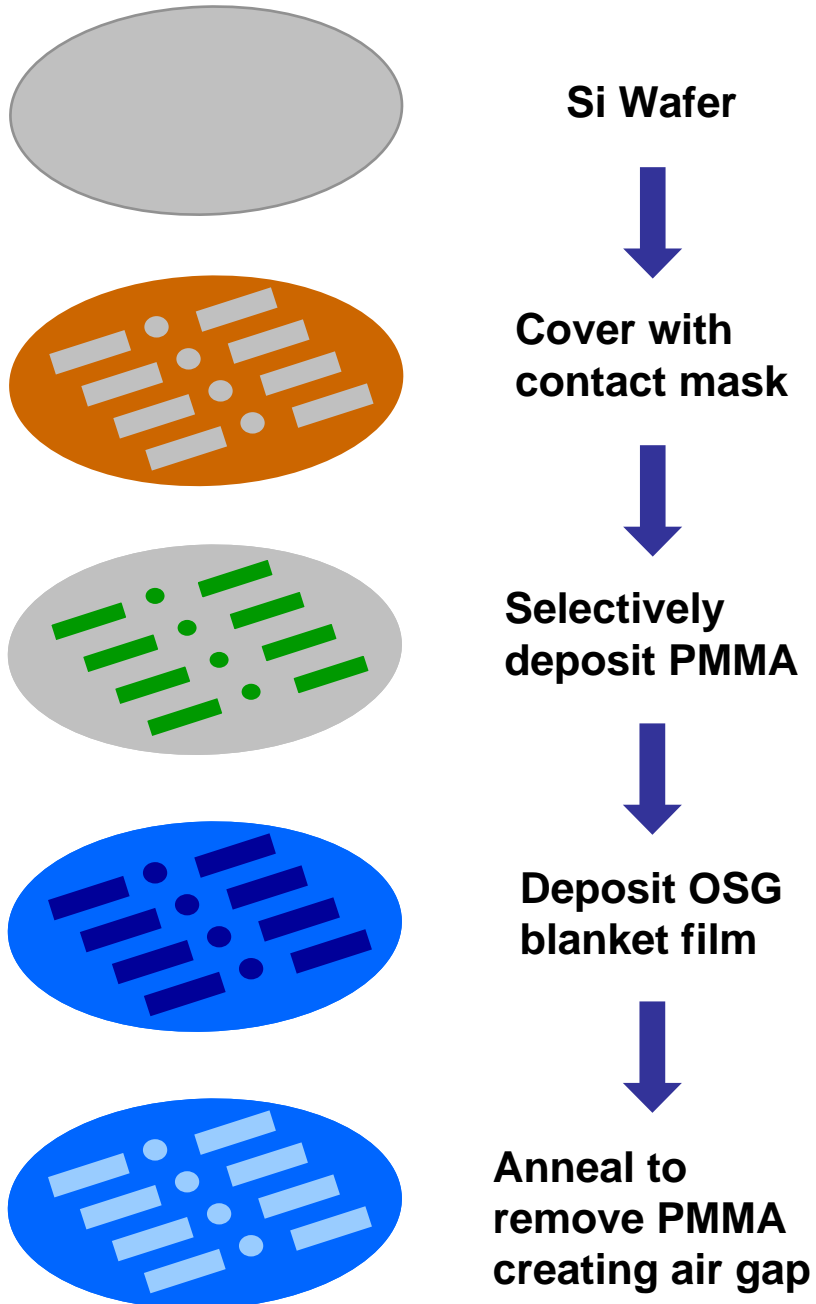
Onset of Thermal Decomposition v. Substrate Temperature



Residue after ITS anneal ramp to 410°C

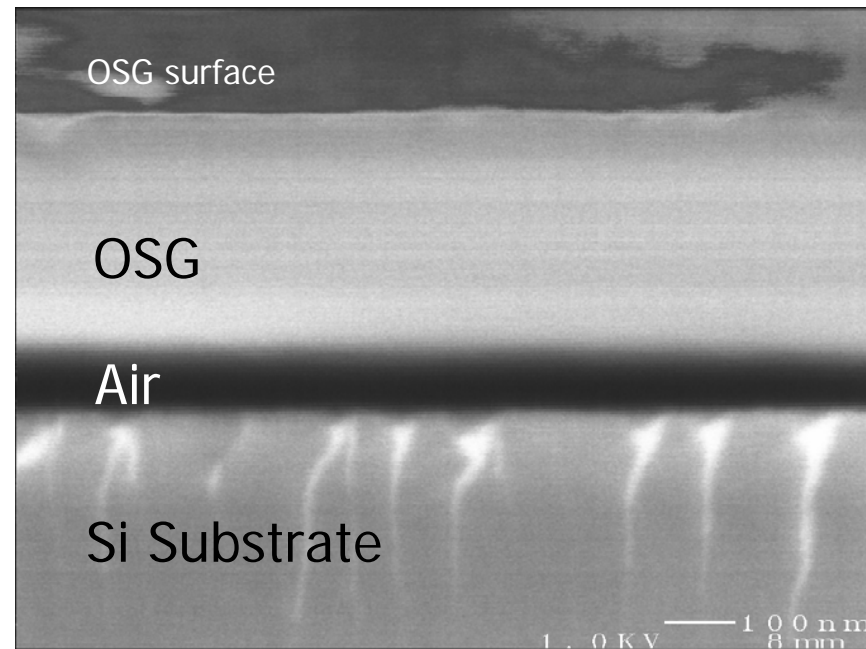


Fabrication of Enclosed Void utilizing PMMA



■ Proof of Concept

- Large scale features created by depositing sacrificial layer using a contact mask



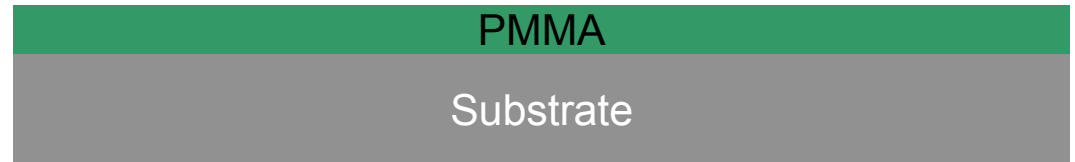
SEM of Air Gap from PMMA

~4000Å OSG deposited over ~1000Å PMMA
Anneal at 350°C creates ~1000Å Air Gap

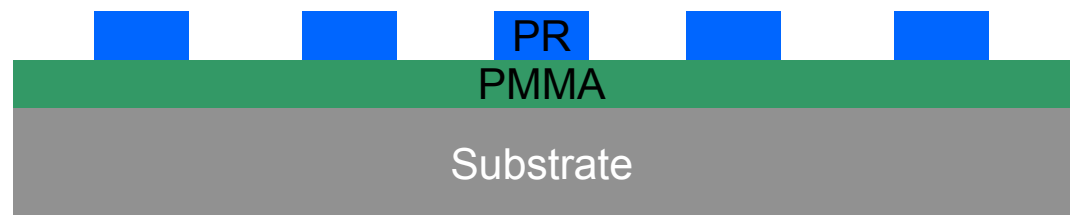
Microscale Air-Gap Fabrication

No Hard Mask Required

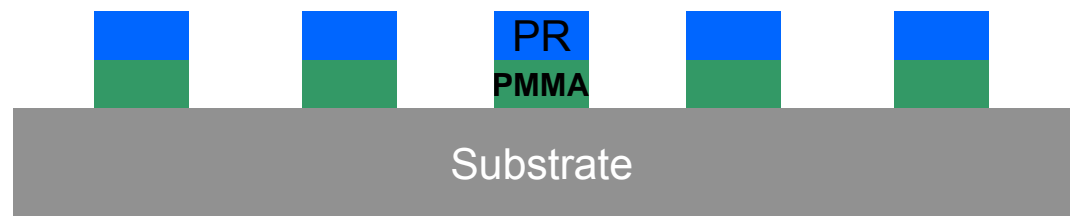
PMMA Deposition



Photolithography



O₂ Plasma Etching of PMMA



Photoresist Acetone Strip
SiO₂ Deposition



PMMA Decomposition



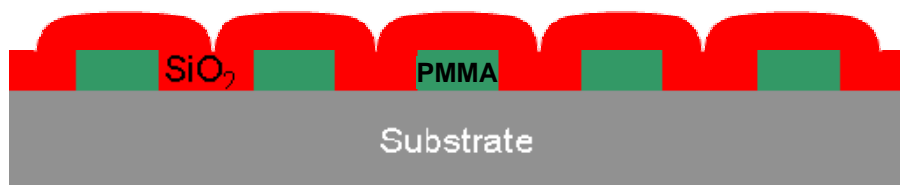
Etch Selectivity PMMA/PR ≥ 2

Fabrication of Enclosed Void

Air-Gap Microchannels



Width: 1 – 10 μm



Anneal ↓



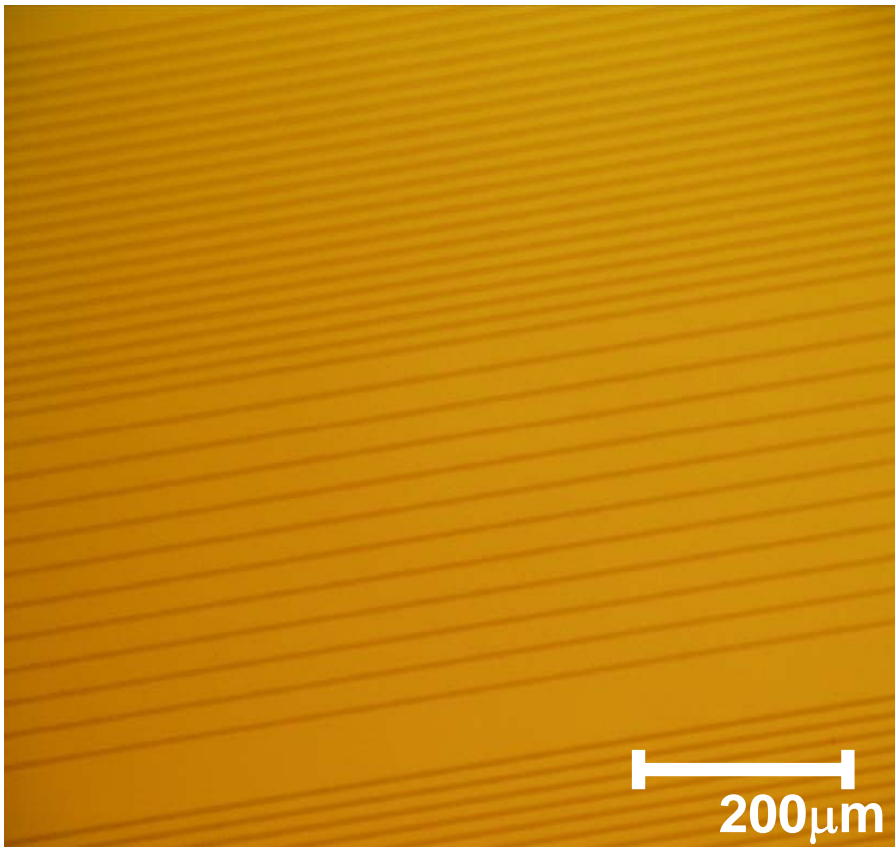
Decomposition Profile

1. Quick ramp to 150°C
2. Hold for 10 min
3. 150 – 250°C: 5°C/min
4. Hold for 30 min
5. 250 – 300°C: 1°C/min
6. Hold for 30 min

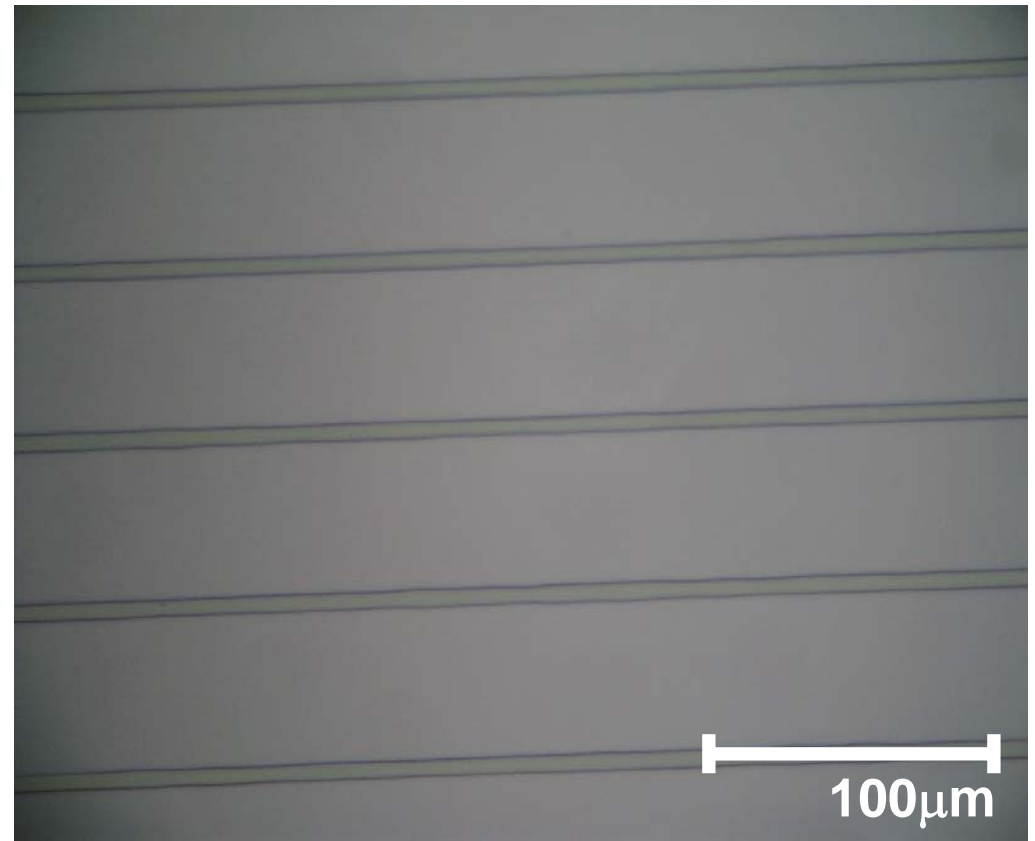
Total Time: 140 min

Ambient: 1 atm (Air or N₂)

Top Down Optical Micrographs



200X – Spin Coated Resist over PMMA

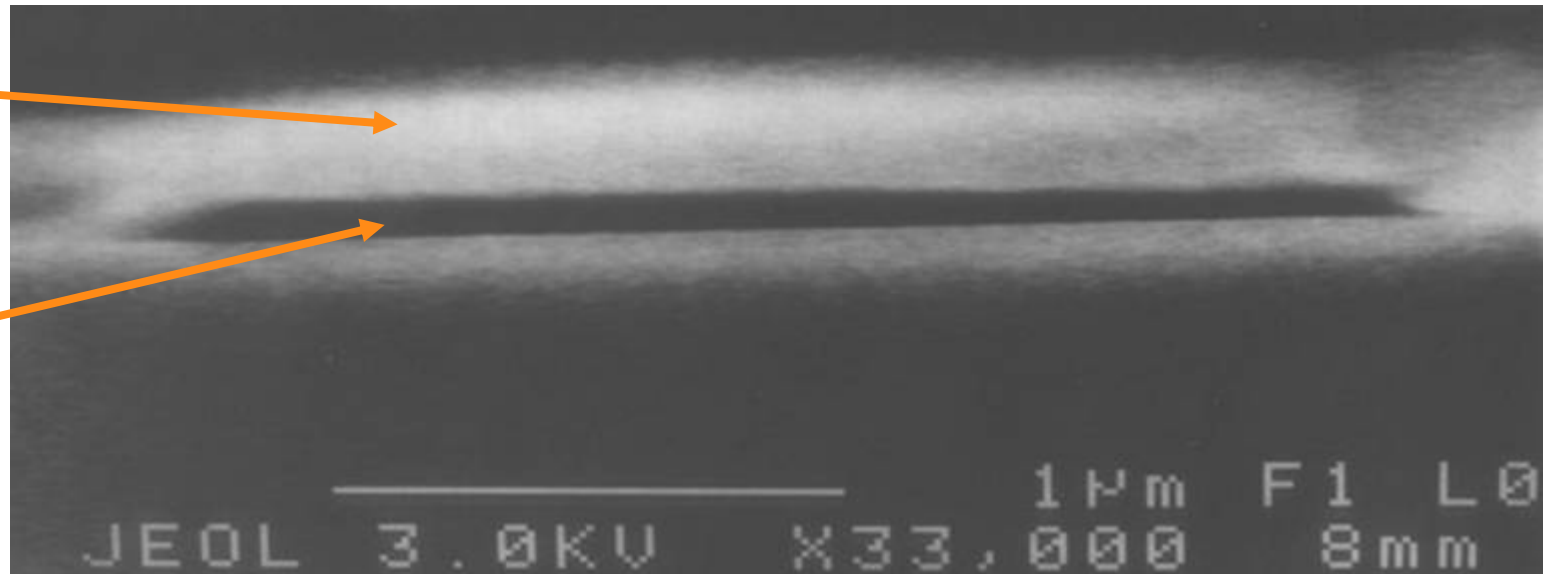


500X – OSG over patterned PMMA

Cross-Sectional SEM

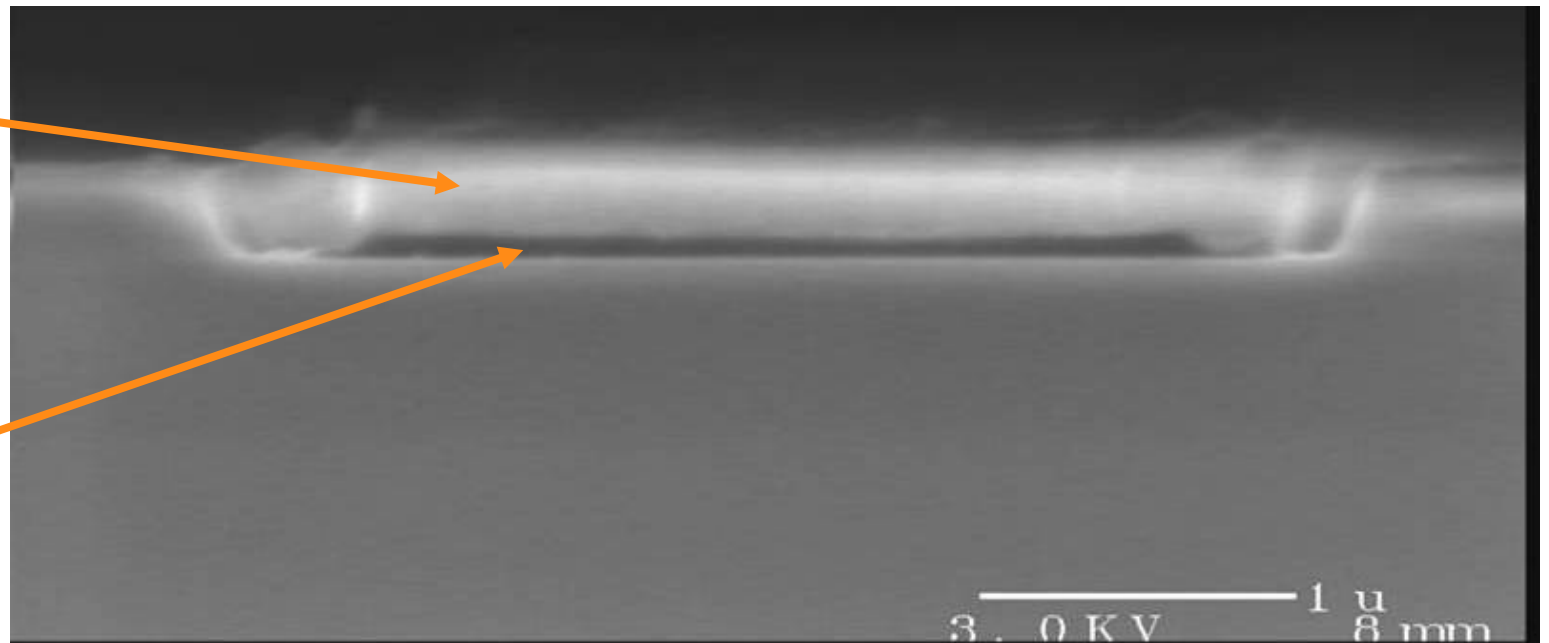
3000 Å SiO₂

1000 Å Void



3000 Å SiO₂

1000 Å Void



PMMA – Summary and Key Findings

- Low power PECVD is a viable method for depositing PMMA
 - FT-IR confirms structural similarity to bulk PMMA
- PECVD PMMA has the desired properties
 - Decomposes at 230 to 350 °C
 - Negligible residue
- Closed-cavity air-gap microstructures successfully fabricated using conventional photolithographic and deposition tools
- Solubility of PMMA in acetone and isopropyl alcohol is inversely proportional to both the deposition power and substrate temperature
- Films deposited at lower temperatures contain $\text{CH}_2\text{-O-CH}_2$ linkages in the polymer backbone decreasing its thermal stability
- Onset of thermal decomposition increases with increased deposition temperature

Plan for Future Work

- **Collaborate with Thrust D**
- **Create Air-Gaps at the nanometer level**
 - **Low-power CW PMMA (complete removal)**
 - **Direct pattern at Scanning Electron Beam Lithography (SEBL) Facility**
 - **Cut wafer with Focused Ion Beam (FIB) and image with SEM**

Acknowledgements

- **The Gleason Group at MIT**
- **Elizabeth Sutton, UROP**
- **SRC/Novellus for Fellowship Funding**
- **NSF/SRC ERC for EBSM**