

EHS Assessment of Chelators and Biocides Utilized in Semiconductor Manufacturing

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Objectives

 Review trends in the use of biocides and chelators in wet cleans & CMP.



- **II)** Literature review:
 - EHS characteristics of main biocides and chelators
 - Impact on:
 - Biological wastewater treatment processes.
 - Species used in effluent ecotoxicology monitoring.





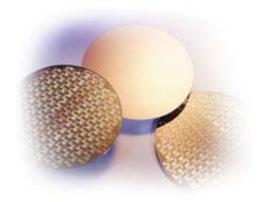




Method of Approach

Inventory of biocides and chelators in wet cleans and CMP:

Consultation with industry, suppliers, experts



EHS assessment / fate in biological treatment systems:

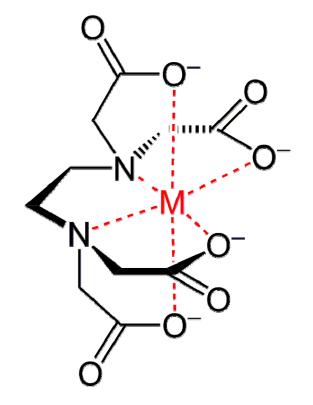
Review of literature and public databases







Chelators



Chemical structure of EDTA-metal chelate.





CMP

Used to complex Cu ions.

Back-end-of-line cleaning

- Used for stripping photoresists, and
- Removing organic/inorganic residues from substrates subjected to gas phase etching, post-etching and CMP.

Front-end-of-line cleaning

Generally do not utilize chelators



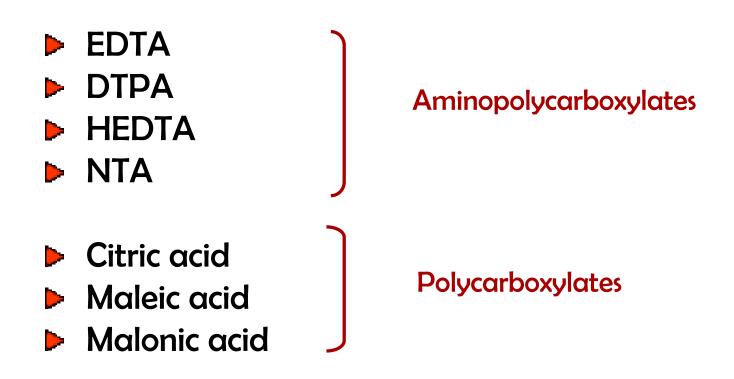






Survey of Chelator Use by ISMI Member Companies

Organic chelators listed in the survey conducted by ISMI among member companies:







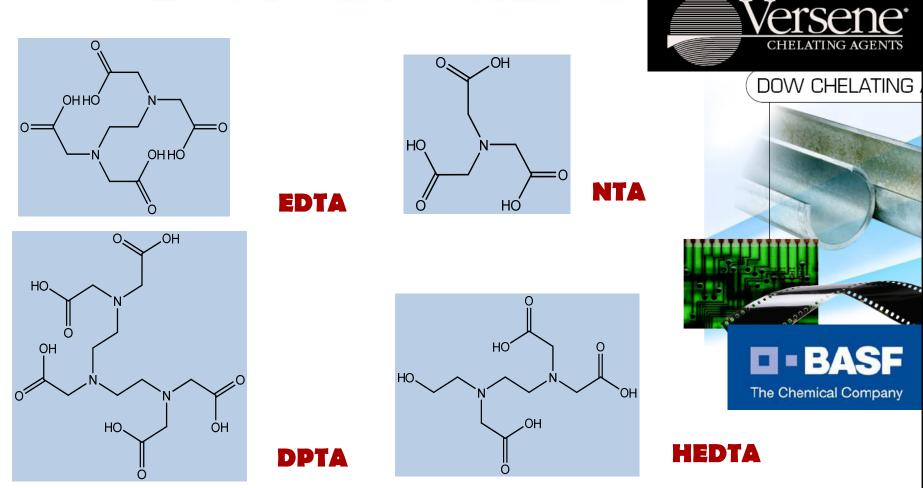
- Amino-polycarboxylates (eg. EDTA, active components in Versene / Trilon[®])
- Polycarboxylates (eg. citric acid) & simple organic acids (eg. lactate)
- <u>Amino acids</u> and derivatives (eg. glycine)
- Organic amines (eg. ethylenediamine)
- (Phosphonates)







Aminopolycarboxylate chelators



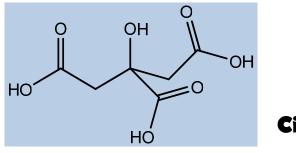
EDTA (Ethylene-diaminetetraacetic acid); DTPA (Diethylenetriaminepentaacetic acid); NTA (Nitrilotriacetic acid) HEDTA (N-(hydroxyethyl)-ethylenediaminetriacetic acid)



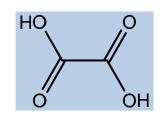


Other chelators: Polycarboxylates, Glycine derivatives

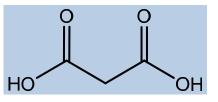
Polycarboxalate chelators



Citric acid

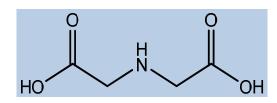


Oxalic acid

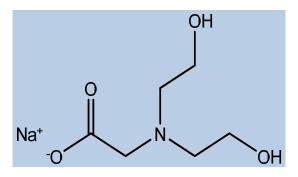


Malonic acid

Glycine-based chelators



Glycine (iminodiacetic acid)



Dihydroxyethylglycine





Compounds are	Compounds are solid at room temperature, low vapor pressure			Poison 😣
Exposure: Potential for dermal and inhalation exposure			Corrosive	
Hazards:	Reactivity:	No		Flammable 💋
	Ignitability:	No		Evenla si va
	Corrosion:	Yes		Explosive
	Health:	Yes		Oxidizer 🔶

EDTA- Approved food additive

Acceptable daily intake of Calcium-disodium EDTA (WHO) = 2.5 mg/kg+ body weight

Health hazard Acute / chronic toxicity Oxalic acid NTA (possible carcinogen) / DTPA (suspected teratogen)





Aminopolycarboxylates: Pollution ranges of waste water effluents

	EDTA ($\mu g/L$)	NTA ($\mu g/L$)	DTPA ($\mu g/L$)		
Effluents of municipal sewage treatment plants					
Typical pollution	10-250	1-15	1-30		
High pollution	1000	200	300		
Effluents of industrial sewage treatment plants					
Typical pollution	100-20,000	100 - 2000	50-5000		
High pollution	400,000	5000	20,000		

(Schmidt et al. Environ Pollut, 2004, 131, 107-124)





Chelators: Fate in POWTs

Chelators	Biodegradation	Sorption to sludge	Removal	Microbial toxicity
Aminopolycarboxylates	EDTA – Low Others- Moderate	No/Low	EDTA –Low Others- Moderate	Yes (high conc)
Polycarboxylates/ Amino acid derivatives	Yes	No	High	No





Aminopolycarboxylates chelators:

Widspread environmental contaminants



- EDTA, NTA, and DPTA are among the most abundant pollutants detected in surface water in USA and Europe.
- Typical concentrations of EDTA in European rivers: 0-100 µg/L.

(Schmidt et al. Environ Pollut, 2004, 131, 107-124) (Reemtsma. Environ. Sci. Technol. 2006, 40, 5451-5458)



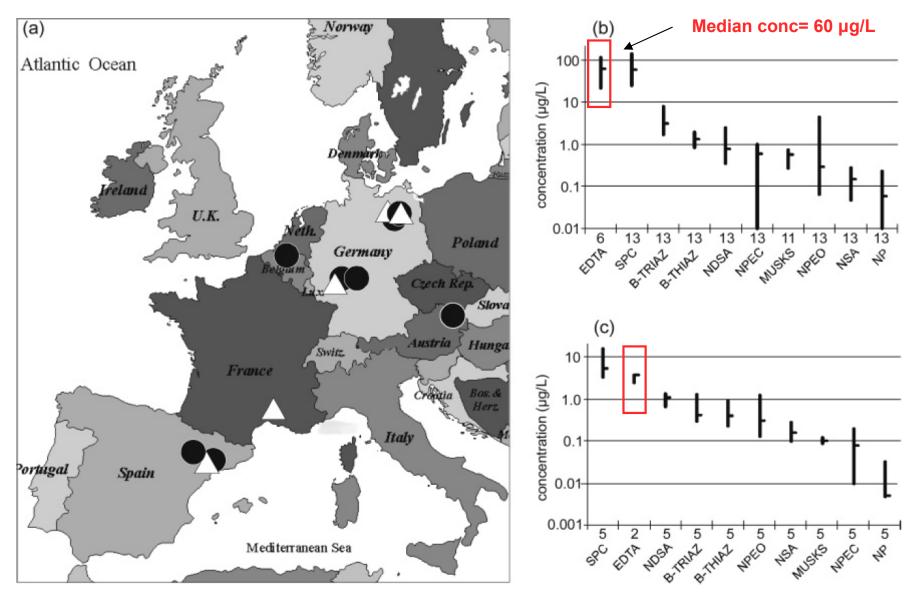


FIGURE 1. (a) Sampling locations for WWTP effluents (black dots) and surface water monitoring (white triangles). Median, 25 and 75 percentile concentrations of polar pollutant classes determined in the monitoring campaign (in μ g/L) (b) of 13 effluent samples of eight WWTP in four European countries and (c) in five surface waters of different catchments in Europe. *n* denotes the numbers of samples analyzed.

Acceptable daily intake of Calcium-disodium EDTA (WHO) = 2.5 mg/kg+ body weight.





Microbial and Aquatic Toxicity:

Aminopolycarboxylates

Microbial Toxicity

- No negative effects on biological treatment processes in POWTs
- Microbial toxicity could be a problem in the biological treatment of industrial wastewaters containing high chelator levels.

Aquatic Toxicity

- Moderate-low acute toxicity: LC50 ≈ hundreds of mg/L
 - e.g. NTA:

LC50 (water fleas):

LC50 (fathead minnows):

600-900 mg/L

: 127 mg/L

Environmental concs. of these chelators are unlikely to cause aquatic toxicity







Microbial & Aquatic Toxicity: Aminopolycarboxylates

- <u>No/low intrinsic toxic</u>, toxicity due to their ability to decrease the bioavalability of essential metals.
- Toxicity depends on chelator speciation (pH, metal levels, hardness)
- May cause mobilization of toxic metals in the environment.

Precipitated metal salt + EDTA \rightarrow Metal-EDTA (soluble)

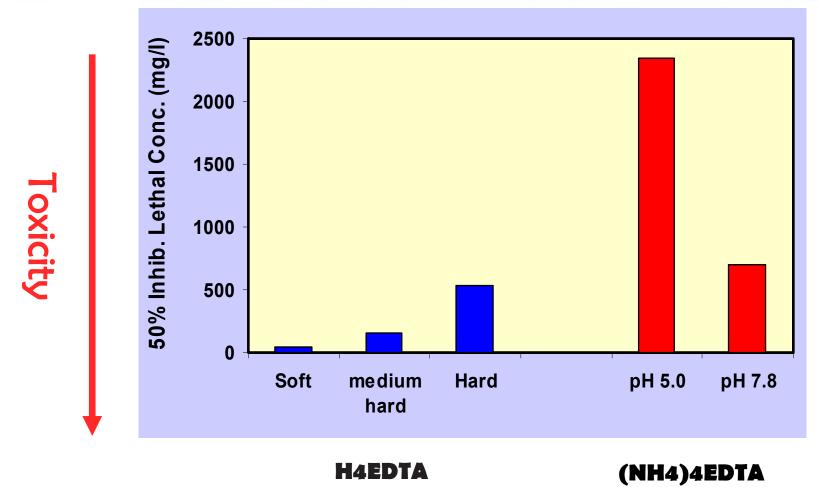
Metal-EDTA + Fe³⁺ \rightarrow Fe-EDTA + Metal⁺ⁿ (Very stable) Log K = 10²⁵







Water hardness, pH, Cation conc. Affect Aminopolycarboxylate Toxicity



Acute toxicity (96 h-LC50) of EDTA species towards bluegill sunfish (Lepomis macrochirus).

Batchelder et al. Bull. Environ. Contam. Toxicol. 1980, 24, 543



Biocides in Semiconductor Manufacturing

CMP

Prevent growth of bacteria, mold, slime and fungi.

Wet clean processes and plating operations

• Not often used. Extreme pH prevents microbial growth

(Cooling towers and ultra pure water)

Used to prevent microbial fouling







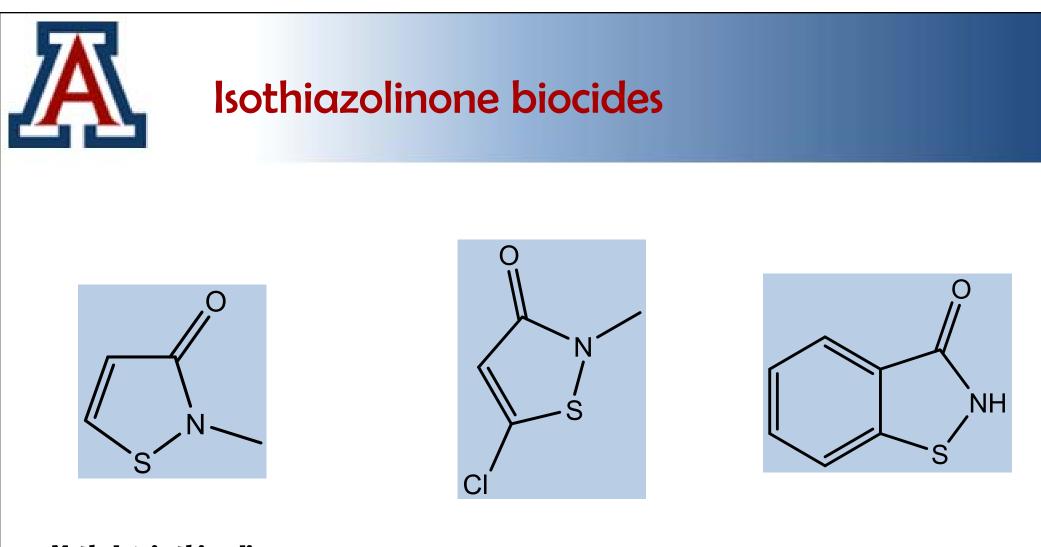


ROHM: HAAS 🛽

Isothiazolinone derivatives (active ingredients in Kathon®)

- **Benzotriazoles** (e.g. benzotriazole, tolylbenzotriazole)
- (Quaternary ammonium salts)





2-Methyl-4-isothiazolin-3-one

5-Cl-2-methyl-4-isothiazolin-3-one

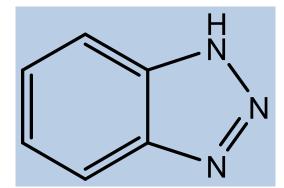
1,2-Benzi\$othiazolin-3-one

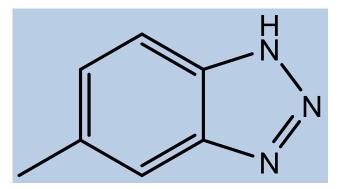
Dichloro-2-n-octyl-4-isothiazolin-3-one





Benzotriazole derivatives





Benzotriazole

(BTA)

Tolyl- benzotriazole

4- or 5-methyl-1H-benzotriazole





Biocides: EHS Characteristics

Compounds are solid at room temperature, low vapor pressure

Exposure: Potential for dermal and inhalation exposure

Hazards	Reactivity:	No
	Ignitability:	No
	Corrosion:	Yes
	Health:	Yes

Health hazard (Acute toxicity) Isothiazoliones / Benzotriazoles





Biocides: Fate in POWTs

Chelators	Biodegradation	Sorption to sludge	Removal	Microbial toxicity
l\$othiazolione\$	Yes ?? Benzothiazolones – No	Low	No data	Yes
Benzotriazole ;	No	Low	Low	No (only at high conc)





Benzotriazole (BTA) and tolylbenzotriazole (TT)

often detected in treated effluents and surface water in USA and Europe

BTA & TT detected in effluents of German POWTs at concs. from below 10 to 100 µg/L (Voutsa et al. 2006).

TT detected in 32% of U.S. streams sampled at conc. > 0.1 μ g/L. Max. and median values were 2.40 and 0.40 μ g/L (Kolpin et al. 2002).





Tolyl-benzotriazole in U.S. Surface Waters

(95 organic contaminants monitored in 139 streams across 30 states)

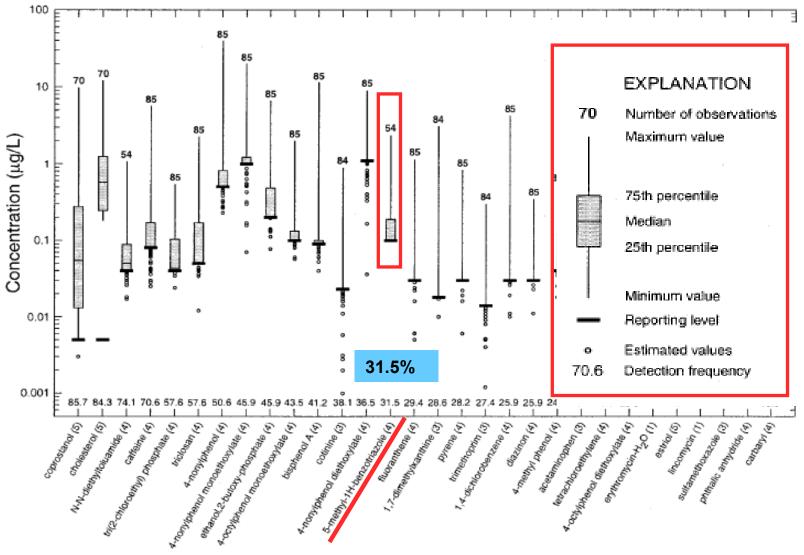


FIGURE 2. Measured concentrations for the 30 most frequently detected organic wastewater contaminants. Boxplots show concentration distribution truncated at the reporting level. Estimated values below the reporting level are shown. Estimated maximum values for coprostanol and cholesterol obtained from Method 5 (Table 1) are not shown. The analytical method number is provided (in parentheses) at the end of each compound name. An explanation of a boxplot is provided in Figure 3.

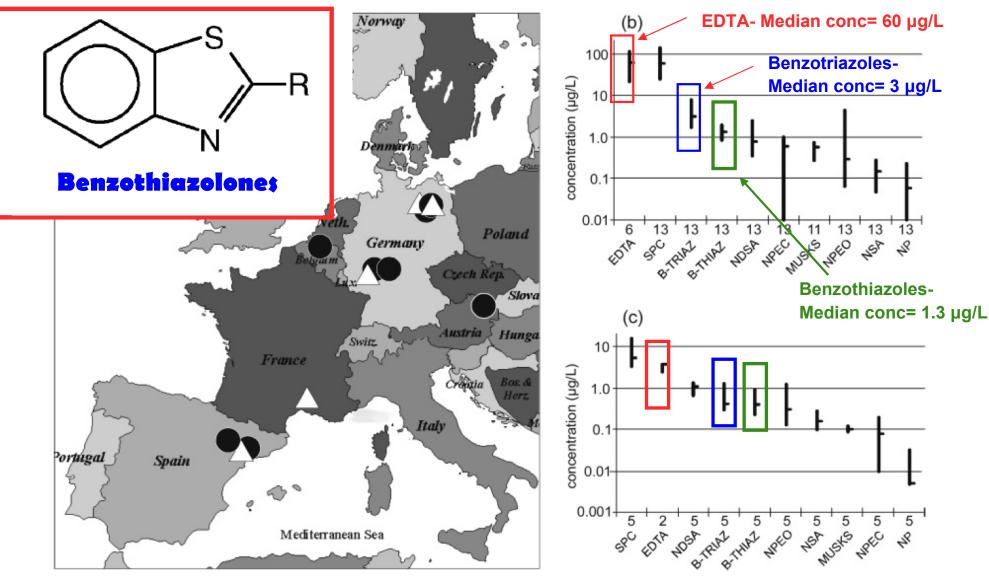
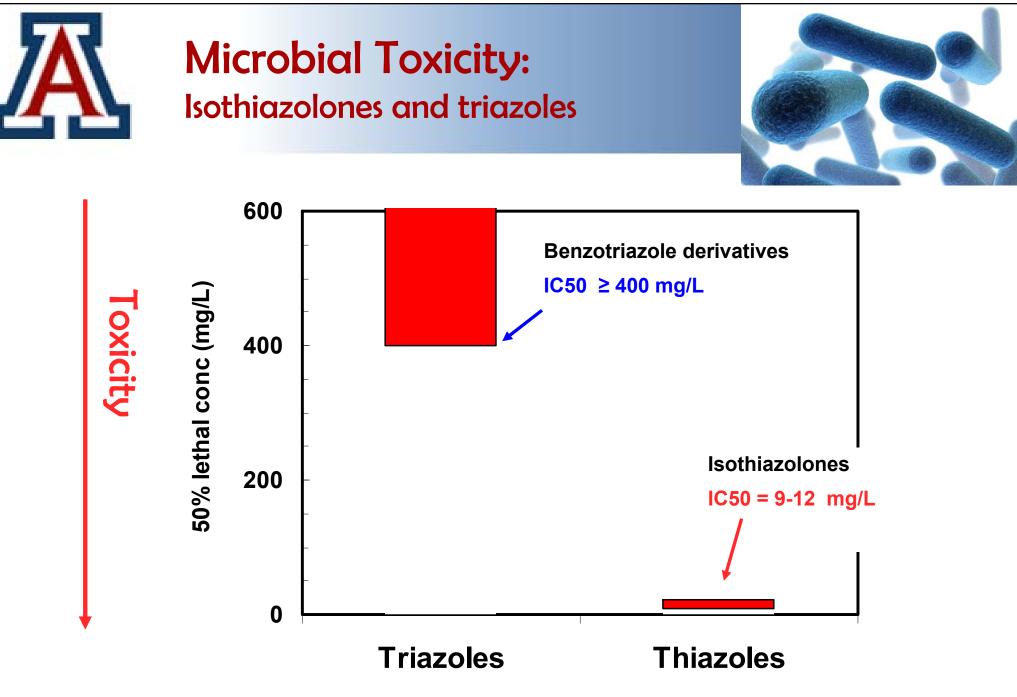


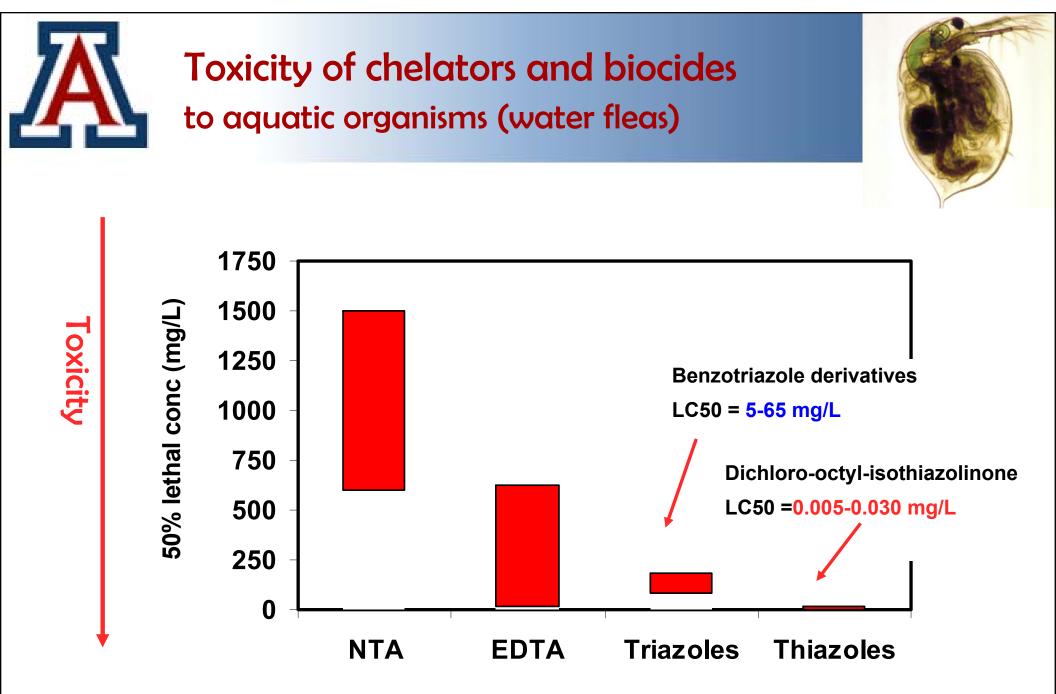
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Reemtsma et al. Environ. Sci. Technol. 2006, 40, 5451-5458



= BTA is a potent inhibitor of nitrification











Compounds	ESH	Aquatic Toxicity	Persistence
Aminopolycarboxylate chelators (EDTA, NTA, etc)	YES*	Moderate/ Low	YES (EDTA)
Benzotriazole derivatives	NO (slight hazard)	LOW	YES
Isothiazoliones	YES	HIGH	YES (Benzothiazoles)

* NTA, DPTA





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