



Radical Oxidation of Ge for Interface Gate Dielectric GeO_2 Formation in MOS gate stack

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Introduction: Ge MOS technology



- Ge as a performance booster

- High electron/hole mobility
- High process compatibility
- Low temperature process
- Possible V_{dd} scaling

	Si	Ge
Electron μ (cm ² /Vs)	1600	3900
Hole μ (cm ² /Vs)	430	1900
Band gap (eV, 300K)	1.12	0.66
Dielectric constant	11.9	16
Melting point (°C)	1415	937

- Key issues: Interface property of Ge MOS gate stack

- GeO₂ is regarded as a promising interface gate dielectrics*
- Since GeO₂ decomposition/GeO evaporation temperature is very low (430°C), low temperature oxidation is needed with high density of oxidant source

*D. Kuzum, IEDM2007, T. Takahashi, IEDM2007, Y. Nakakita, IEDM2008

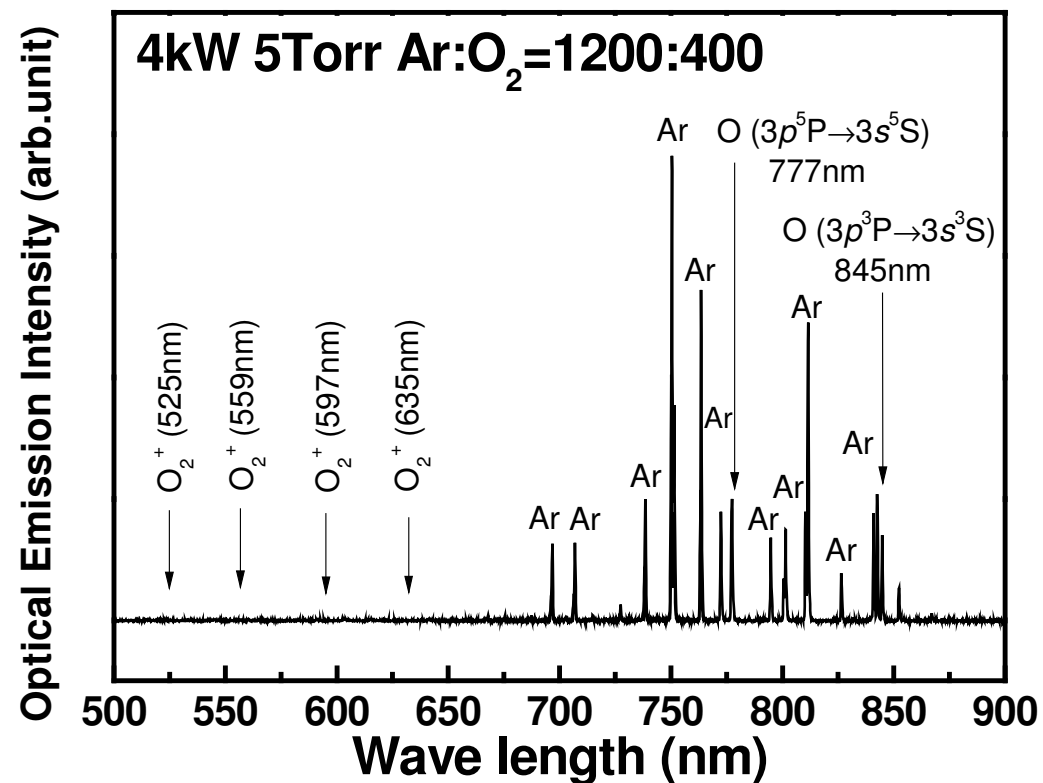
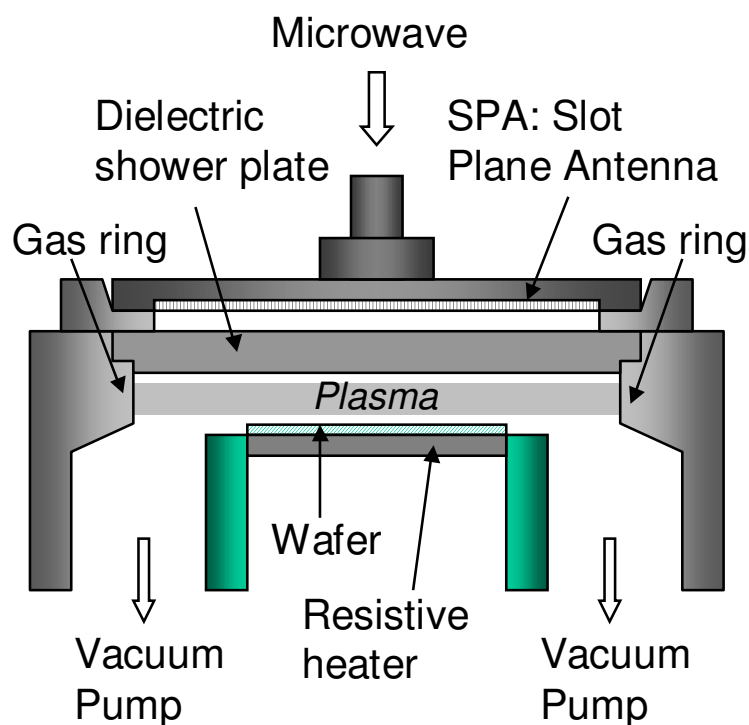
Slot-Plane-Antenna (SPA) system*



- High density/highly reactive atomic oxygen radicals (O^*) generation
- Enables low temperature oxidation
 - Below 400°C** for Si and Ge

*T. Ohmi et al., Proc of IEEE 2001

T. Sugawara et al., JJAP 2005



Objectives



- Study oxidation kinetics of radical oxidation
 - Comparing to thermal oxidation
- Examine the electrical property of interface gate dielectric GeO_2
 - CV characteristics of MOS capacitor
 - Thermal stability
 - Band alignment



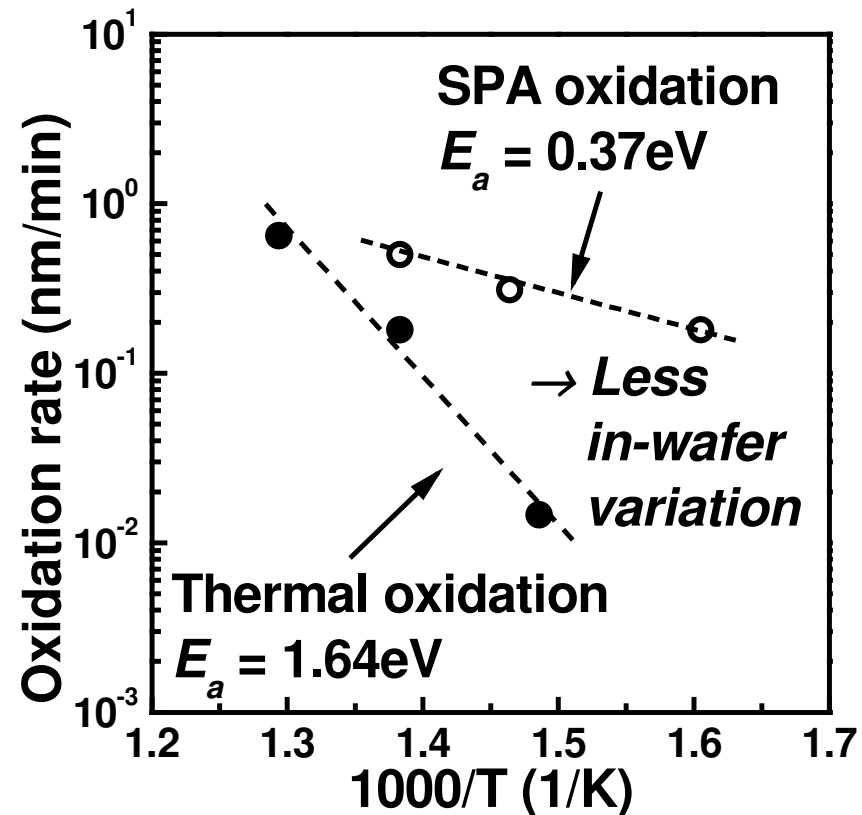
Experiments

- Sample preparation
 - (100) and (111) Ge surface was cleaned by PRS100 organic remover and by HCl/HF
 - Surface was oxidized by Slot-Plane-Antennal (SPA) radical oxidation system
 - Thermal oxidation was also done as a reference
- Kinetics study
 - Thickness was measured by ellipsometry and XPS
- Electrical property
 - 5nm ALD Al_2O_3 was deposited on GeO_2/Ge
 - Sputtered Al metal pad
 - 400°C FGA anneal
 - XPS was used to identify surface chemical property
 - Synchrotron radiation photoemission spectroscopy (SRPES) was used for band offset measurement

Activation energy of Ge oxidation



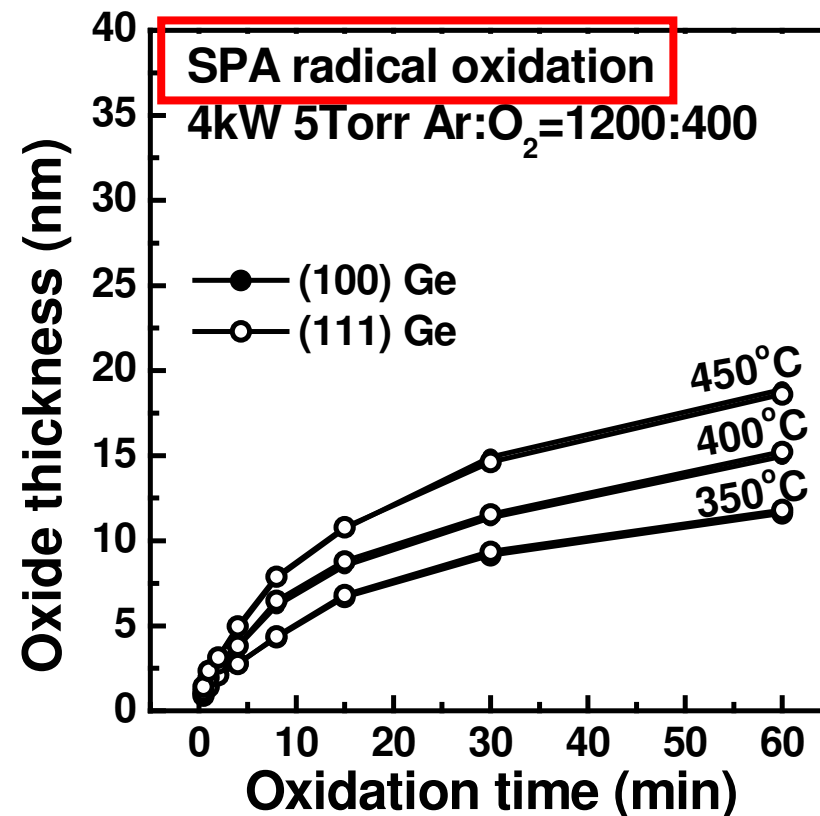
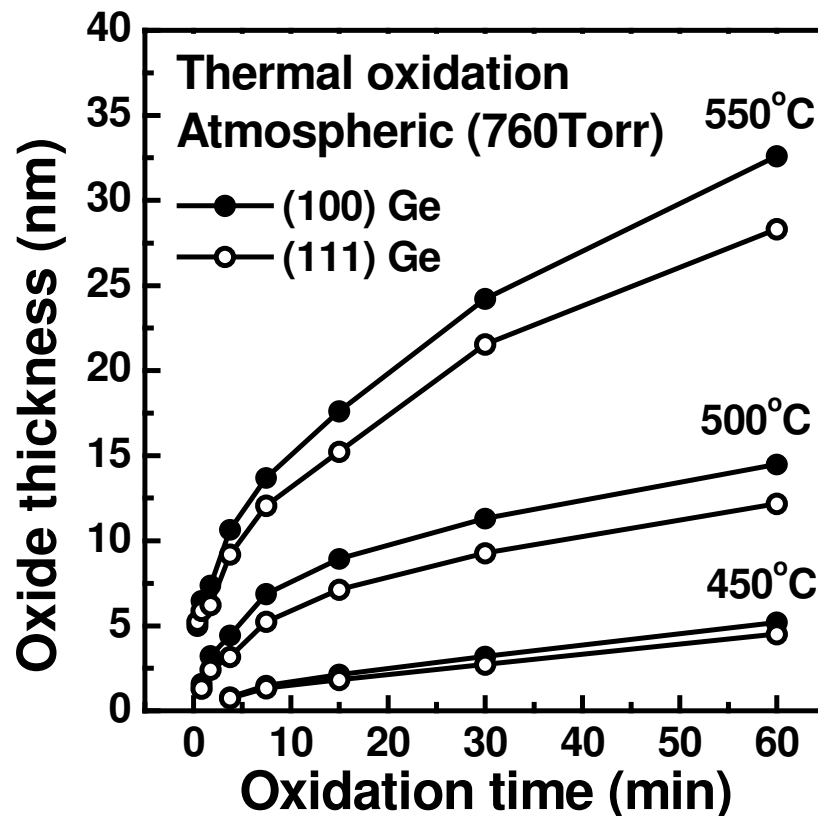
- Temperature dependence
 - Radical oxidation has lower activation energy (0.37eV) than thermal oxidation (1.64eV)
 - Due to high reactivity of radicals with small radius





Oxidation rate (1)

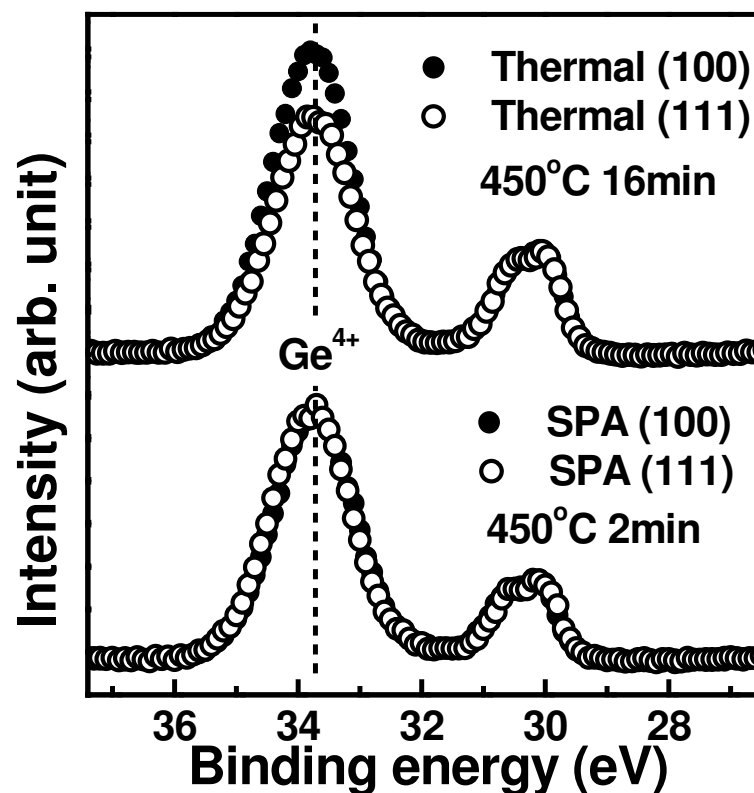
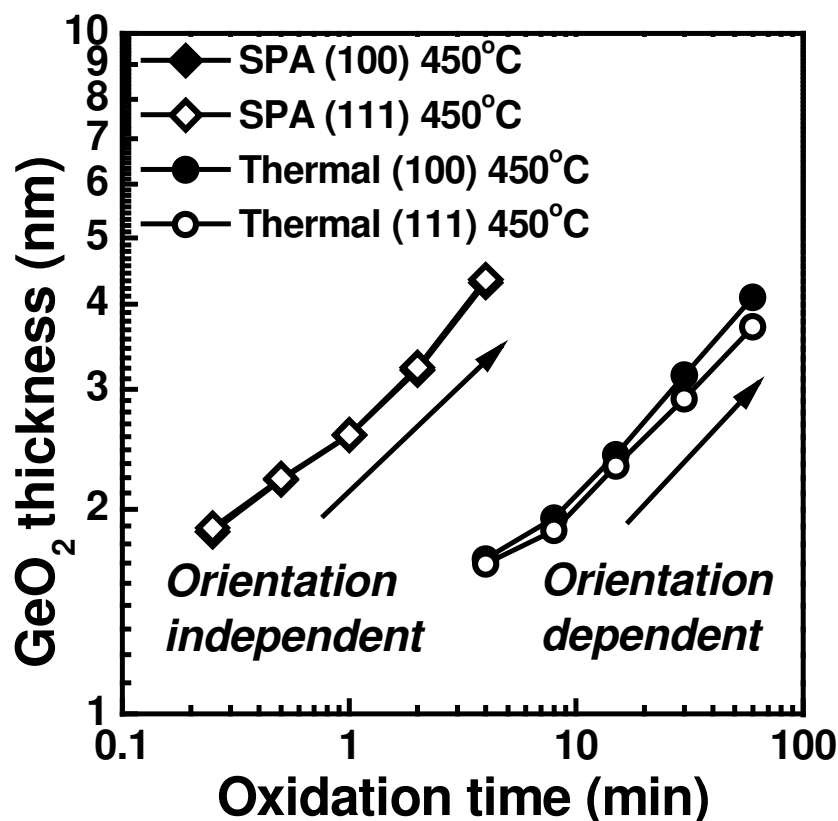
- Radical vs thermal oxidation (ellipsometry)
 - No orientation dependence was observed in radical oxidation between (100) and (111) Ge
 - Faster oxidation on (100) Ge, opposite to Si



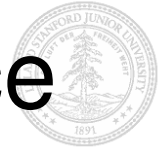


Oxidation rate (2)

- Radical vs thermal oxidation (XPS)
 - No orientation dependence was observed in radical oxidation between (100) and (111) Ge
 - Faster oxidation on (100) Ge, opposite to Si

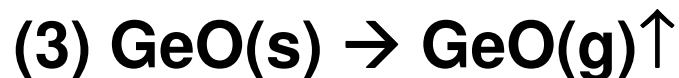
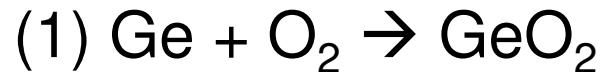


Mechanism of orientation independence



- Thermal oxidation

- Decomposition/evaporation limited oxidation process because of low decomposition temperature

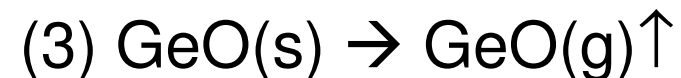


- In Si, SiO₂ decomposition/SiO evaporation rate is higher on (111) because of higher stress in SiO₂ on (111)*
- By the analogy from Si, gross oxidation rate is higher on (100)

*J. R. Engstrom, Surf. Sci 1991

- SPA radical oxidation

- Oxide formation is dominant because of highly reactive oxygen radicals



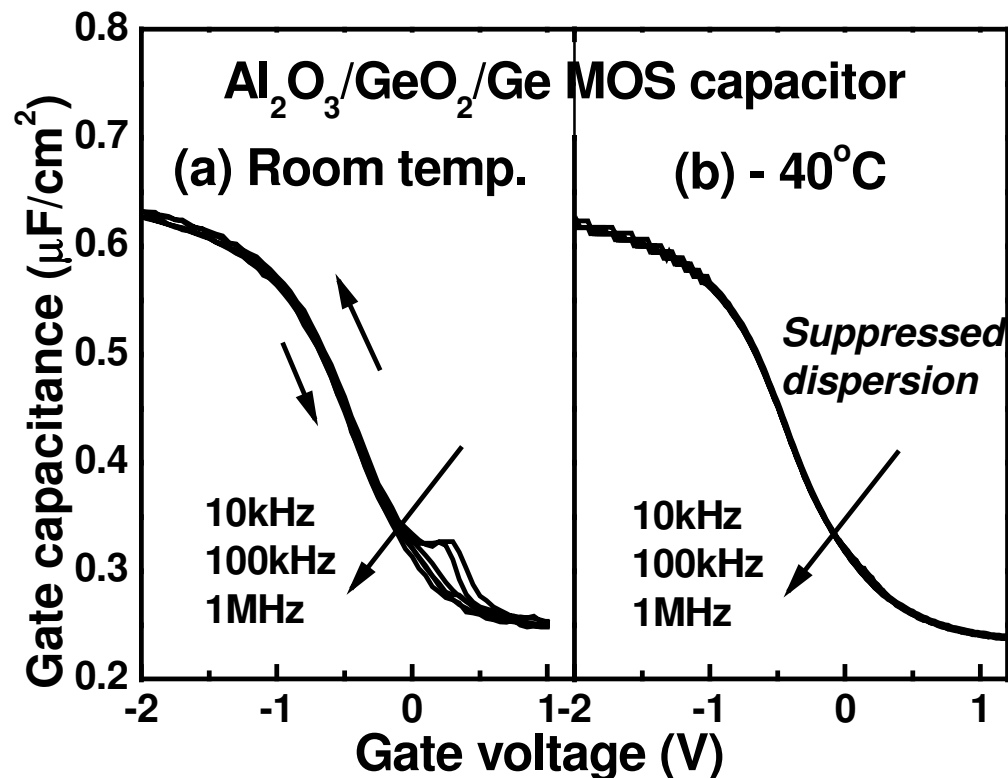
- Lower oxidation process can also suppress decomposition/evaporation
- Highly reactive oxygen radicals with small radius can penetrate on any orientated surface nearly at the same rate*

*T. Ohmi, Proc. IEEE 2001

Electrical property of GeO₂/Ge interface (1)



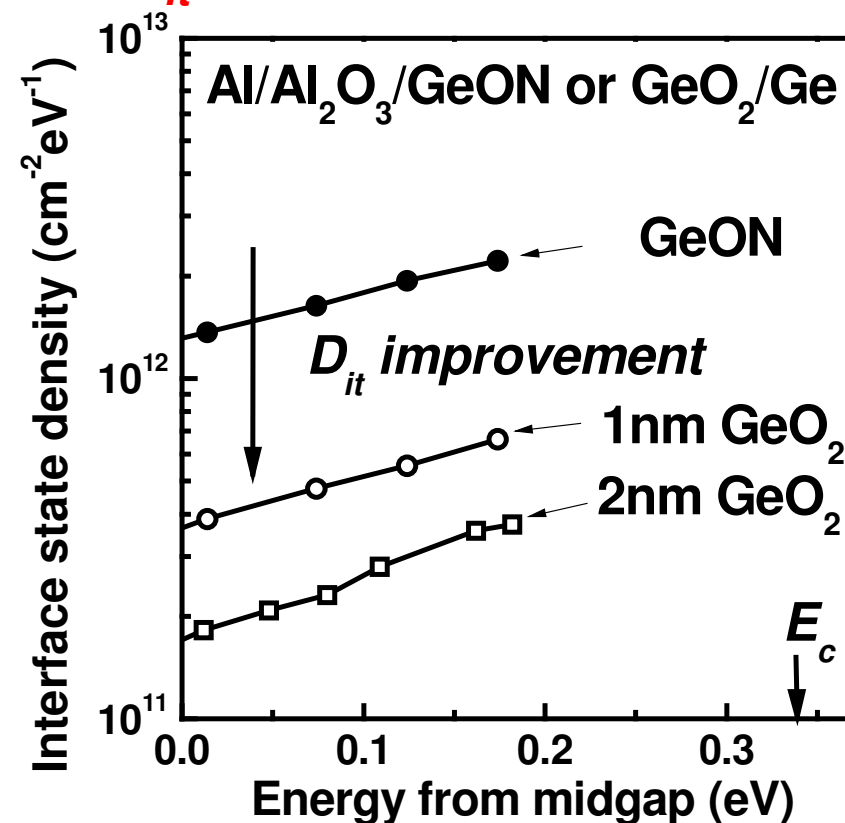
- Al/Al₂O₃/GeO₂/Ge Ge MOS capacitor:
 - 350°C ALD Al₂O₃ deposition + 400°C FGA anneal
 - Very small hysteresis and frequency dispersion
 - Low temperature measurement suppresses frequency dispersion due to minority carrier response



Interface state density (D_{it}) of GeO_2/Ge



- Comparison between GeON and GeO_2
 - D_{it} was measured by conductance method
 - Significant improvement from GeON
 - Achieved $D_{it} = 1.4 \times 10^{11} \text{ cm}^{-2} \text{ eV}^{-1}$ at midgap

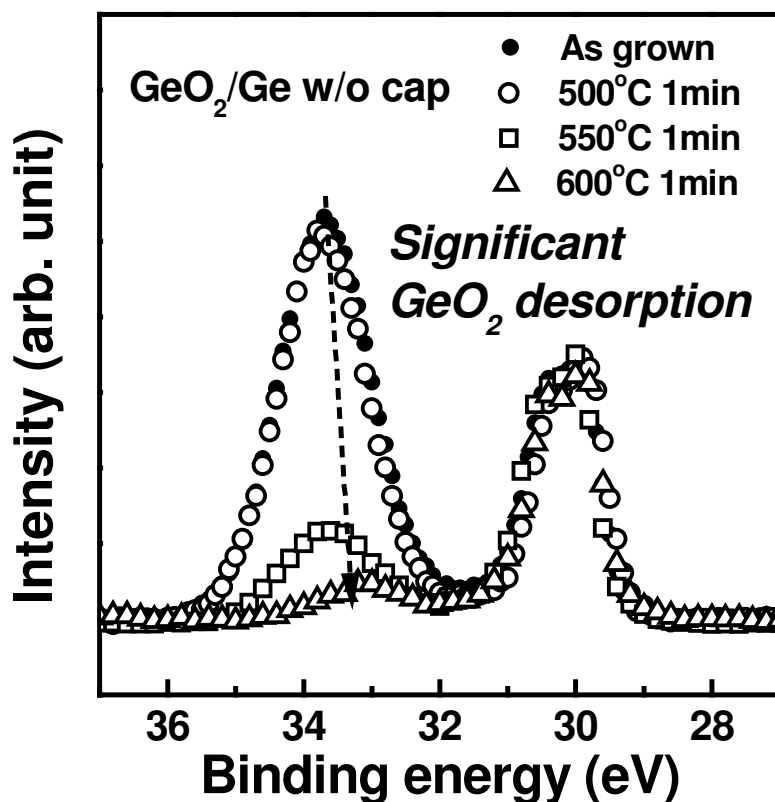




Thermal stability

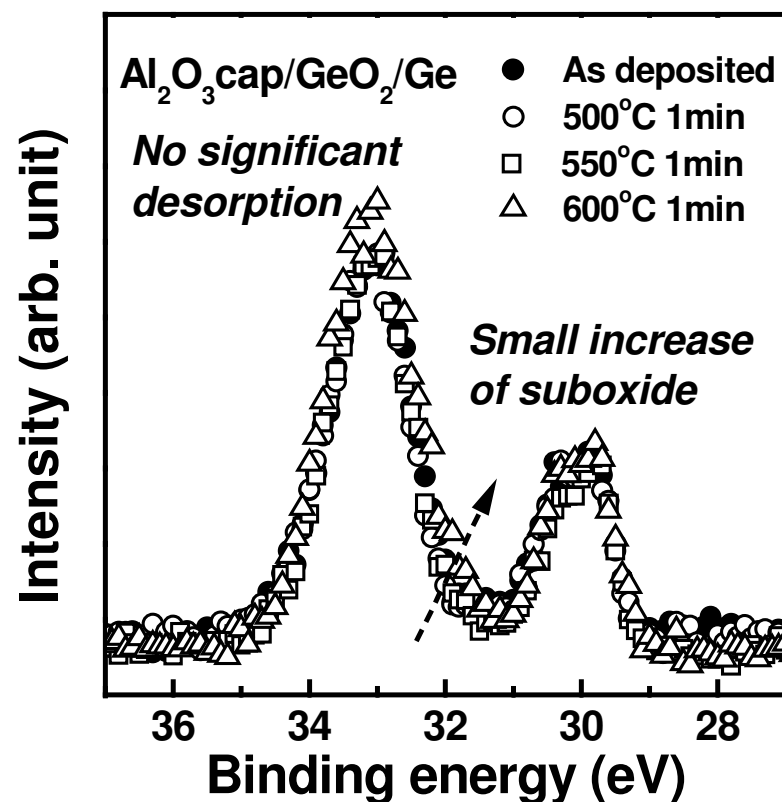
- GeO_2/Ge

- Significant GeO_2 decomposition and GeO out-diffusion



- $\text{Al}_2\text{O}_3/\text{GeO}_2/\text{Ge}$

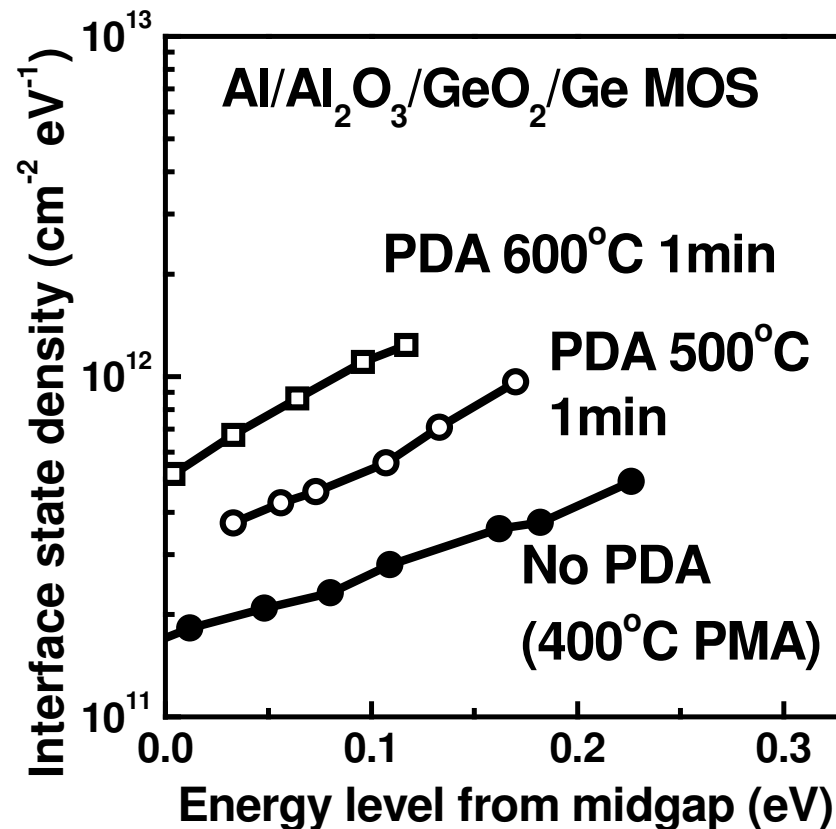
- Al_2O_3 works as an out-diffusion barrier and maintained GeO_2
- Suboxide formation



Thermal stability of electrical property



- D_{it} change after PDA
 - D_{it} can still be kept on the order of $10^{11} \text{cm}^{-2} \text{eV}^{-1}$
 - However, as low thermal budget as possible is required to avoid degradation due to suboxide formation

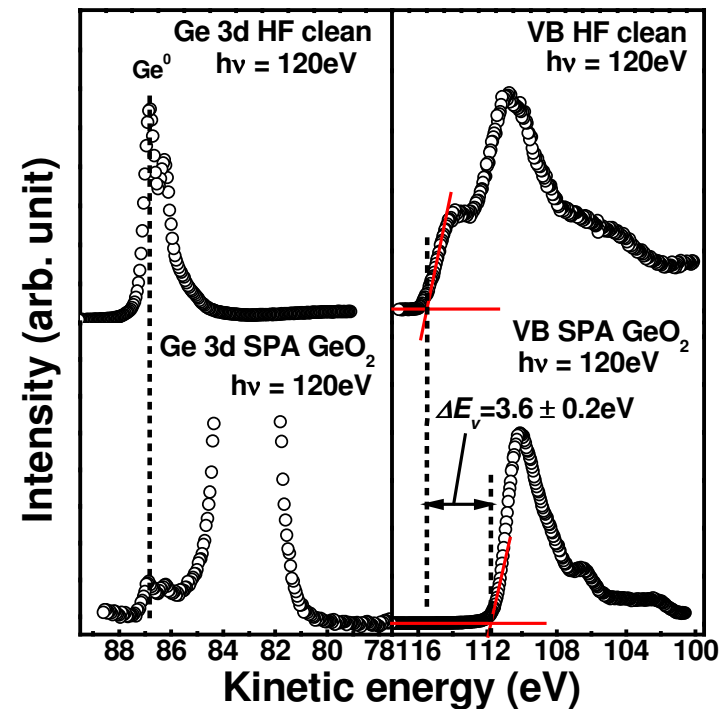
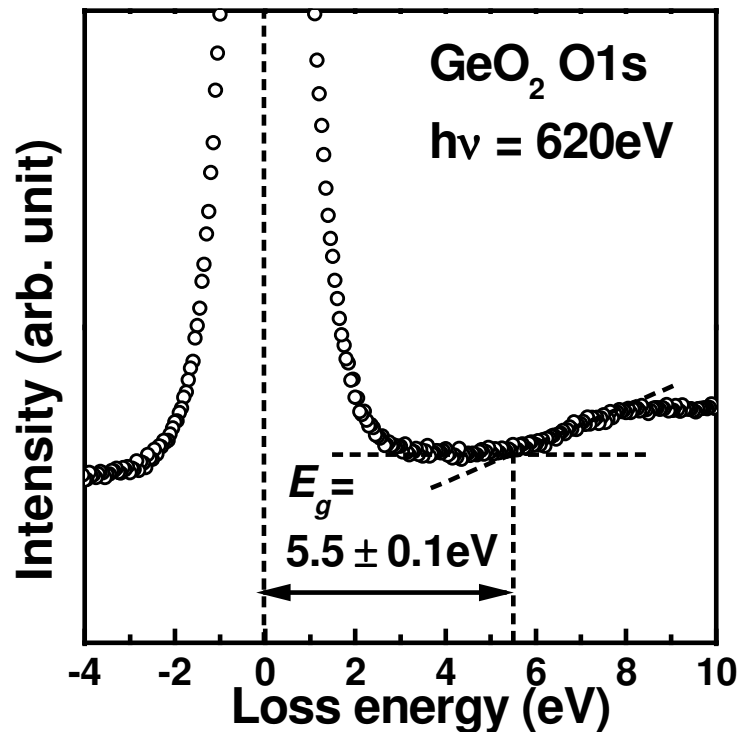


Band diagram by SRPES



- Build band diagram by SRPES
 - $E_{g\text{GeO}_2} = 5.5\text{eV}$
 - $E_c = 1.2 \pm 0.3\text{eV}$, $E_v = 3.6 \pm 0.2\text{eV}$, which satisfies criterion* of band offset $> 1\text{eV}$ to suppress leakage

*J. D. Wilk, JAP2001



Summaries



- Orientation-independent oxidation was observed by SPA radical oxidation
 - Highly reactive oxygen radicals with small radius and low activation energy
 - Useful for uniform oxide formation in multi-gate FinFETs or nano-wire FETs
- Obtained excellent CV characteristics and achieved very low D_{it} ($1.4 \times 10^{11} \text{cm}^{-2} \text{eV}^{-1}$)
- Build band diagram
 - Band offset ($E_c = 1.2 \pm 0.3 \text{eV}$, $E_v = 3.6 \pm 0.2 \text{eV}$) satisfies requirement of leakage suppression



Future works

- Gaurav Thareja will take over this project
- Scalability of GeO_2 interfacial layer
- MOSFET performance with GeO_2 interfacial layer grown by SPA radical system
- Application to multi-gate FET or nano-wire FETs