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**2013 Environmental Nanotechnology
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Detection of carbon nanotubes in environmental and biological matrices using programmed thermal analysis



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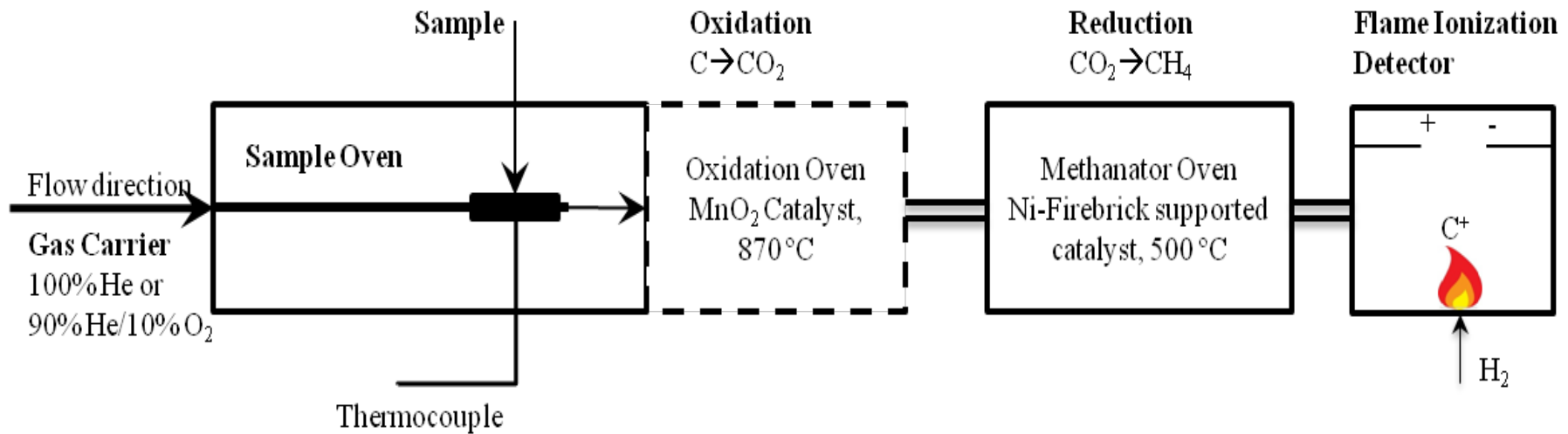
Carbon Nanotubes



- Optimization of analysis scheme
 - Many techniques exist
 - Here we optimize a thermal analysis method
- Objective is to remove organic carbon and “soot” signals in thermal analysis method
- Application to CNT analysis in rat lungs
 - Objective is to demonstrate spike recoveries in real lung tissue
- Future work – apply extraction and analysis in eco-toxicity testing and CNT release from products (e.g., thermal packaging)



Elemental Carbon/Organic Carbon Analyzer



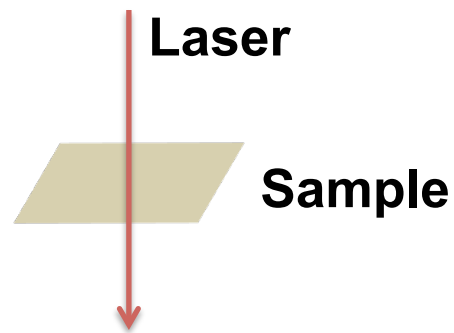
**Sunset Laboratories
Lab OC-EC Aerosol
Analyzer**



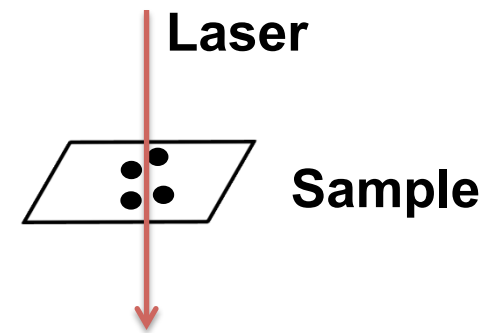
Manual OC-EC Cutoff Required



**Homogeneously
loaded sample
(e.g., air)**



**Heterogeneously
loaded sample
(e.g., CNTs)**



Reasons using the laser transmission to separate OC from EC automatically for CNTs does not work:

- As oxidation occurs the CNTs are removed and the laser may or may not still be hitting the “sample.”
 - This is even more pronounced when the CNTs are embedded in a matrix such as lung tissue.
- CNTs are generally too black for the laser to pass through, which results in a poor baseline for transmission readings.
- Typically, automatic correction will mistaken CNTs as OC or OC as EC.



Not All CNTs are Equivalent – So we examined 15 different CNTs

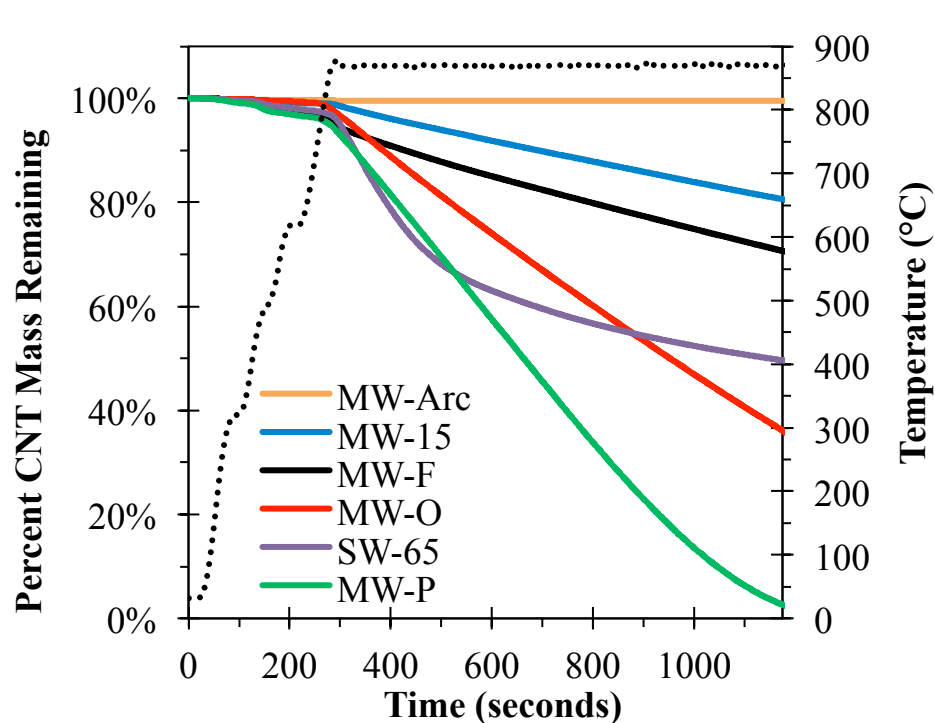
Table 1. Properties of CNTs Used in This Study

CNT ID	CNT type	state	purity ^a	metal content ^b	outer diameter (nm)	inner diameter (nm)	length (μm)
MW-O	MWCNT	raw	>95%	<6%	20–30	5–10	10–30
MW-P	MWCNT	purified	>98%	<2%	20–30	5–10	10–30
MW-F	MWCNT	functionalized	>99.9%	<0.01%	20–30	5–10	10–30
MW-15	MWCNT	raw	>95%	<5%	7–15	3–16	0.5–200
MW-20	MWCNT	raw	>95%	<5%	10–20	5–10	0.5–200
MW-30	MWCNT	raw	>95%	<5%	10–30	5–10	0.5–500
MW-100	MWCNT	raw	>95%	<5%	60–100	5–10	0.5–500
MW-OH	MWCNT	functionalized	>95%	<1.5%	8–15	3–5	10–50
MW-COOH	MWCNT	functionalized	>95%	<1.5%	8–15	3–5	10–50
MW15G ^c	MWCNT	annealed	>97%	<1%	7–15	3–6	0.5–100
MW-Mitsui	MWCNT	raw	>98%	<1%	20–70	NA	NA
MW-arc	MWCNT ^d	raw	<50%	0%	5–10 ^e	NA	NA
SW	SWCNT	raw	<50%	<10%	1.1	NA	0.5–100
SW-65	SWCNT	purified	<75%	<10%	0.8	NA	0.45–2

^aCNT content reported by manufacturer. MW-P and MW-F calculated assuming no amorphous carbon remaining. ^bMetal content reported by manufacturer except for MW-F and MW-P determined using energy-dispersive X-ray spectroscopy and MW-15G using thermogravimetric analysis. ^cMW-15 annealed at ~2000 °C in UHP He. ^dSynthesized using arc method; all others are CVD. ^eObtained from TEM images; all others reported by manufacturer.



Thermal Properties of CNTs

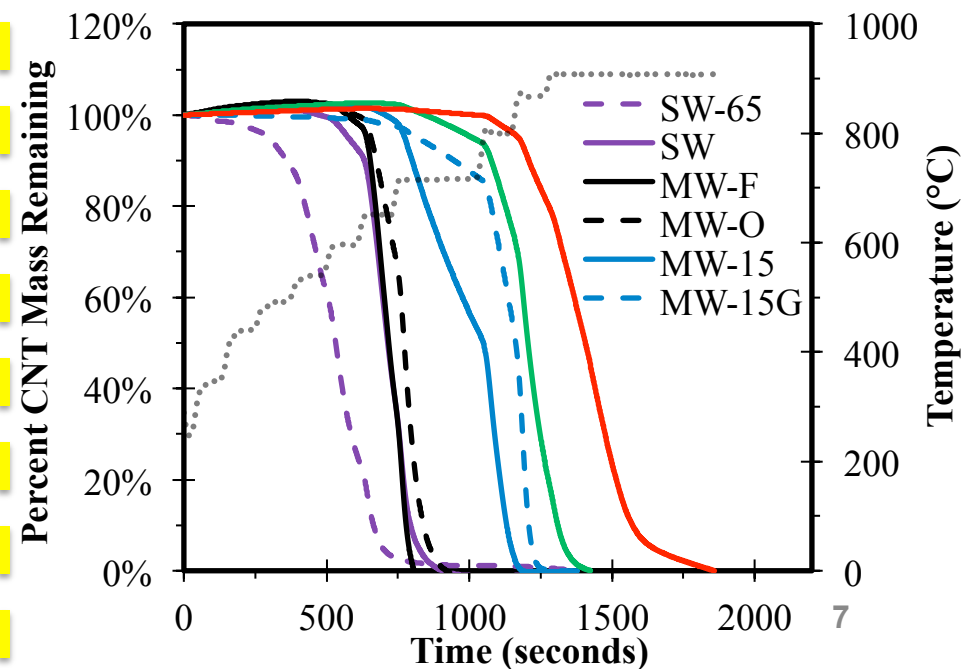


**Oxidizing conditions
(90% He/ 10% O₂)**

**Conclusion: Not all
CNTs “burn” at the
same temperature**

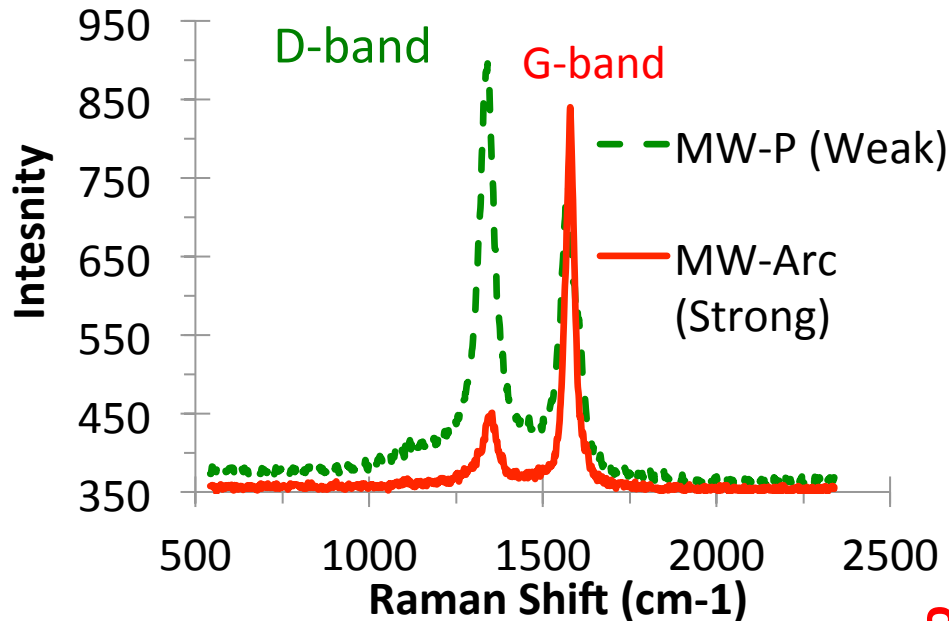
Inert conditions (100% He)

**Conclusion: Surface
oxygen groups allow
some CNTs to burn in
inert gas environment**



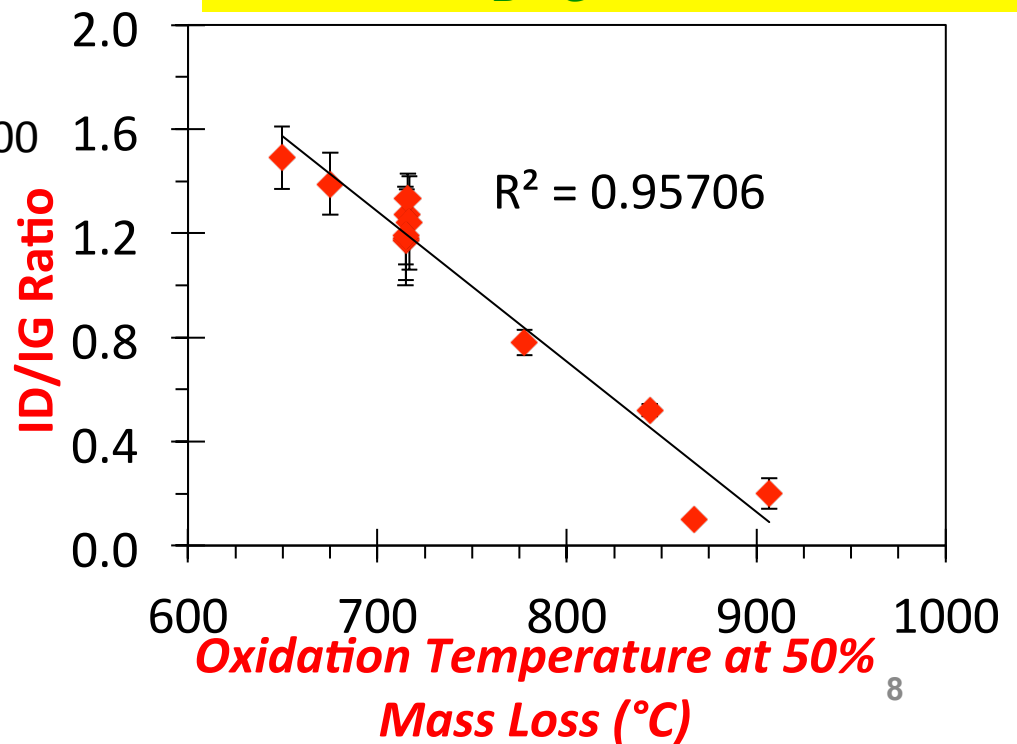


Oxidation “Capability” Can be Predicted from Raman Spectra



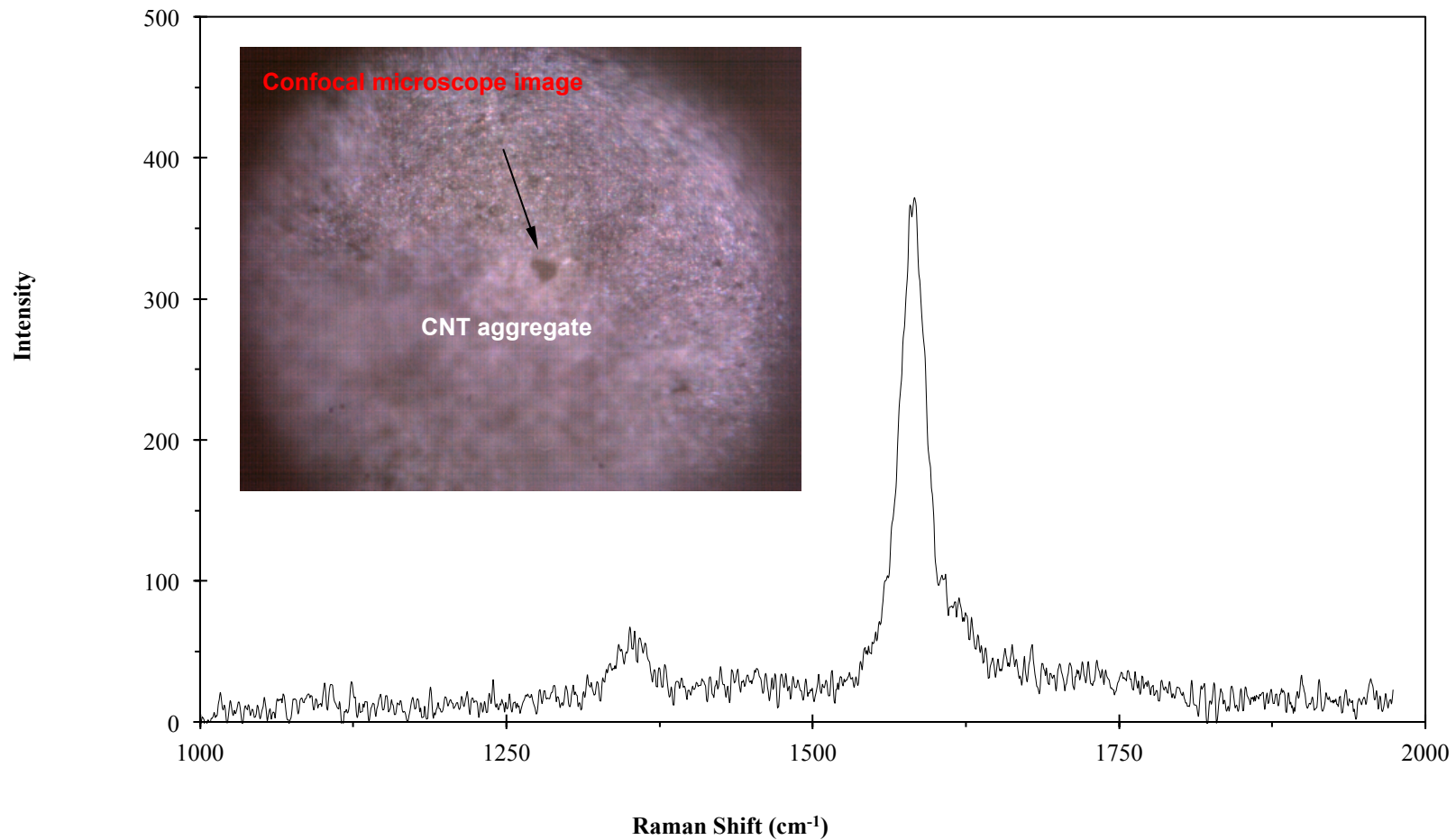
Raman spectra of two CNTs

Correlation between oxidation temperature at 50% mass loss and the I_D/I_G ratio



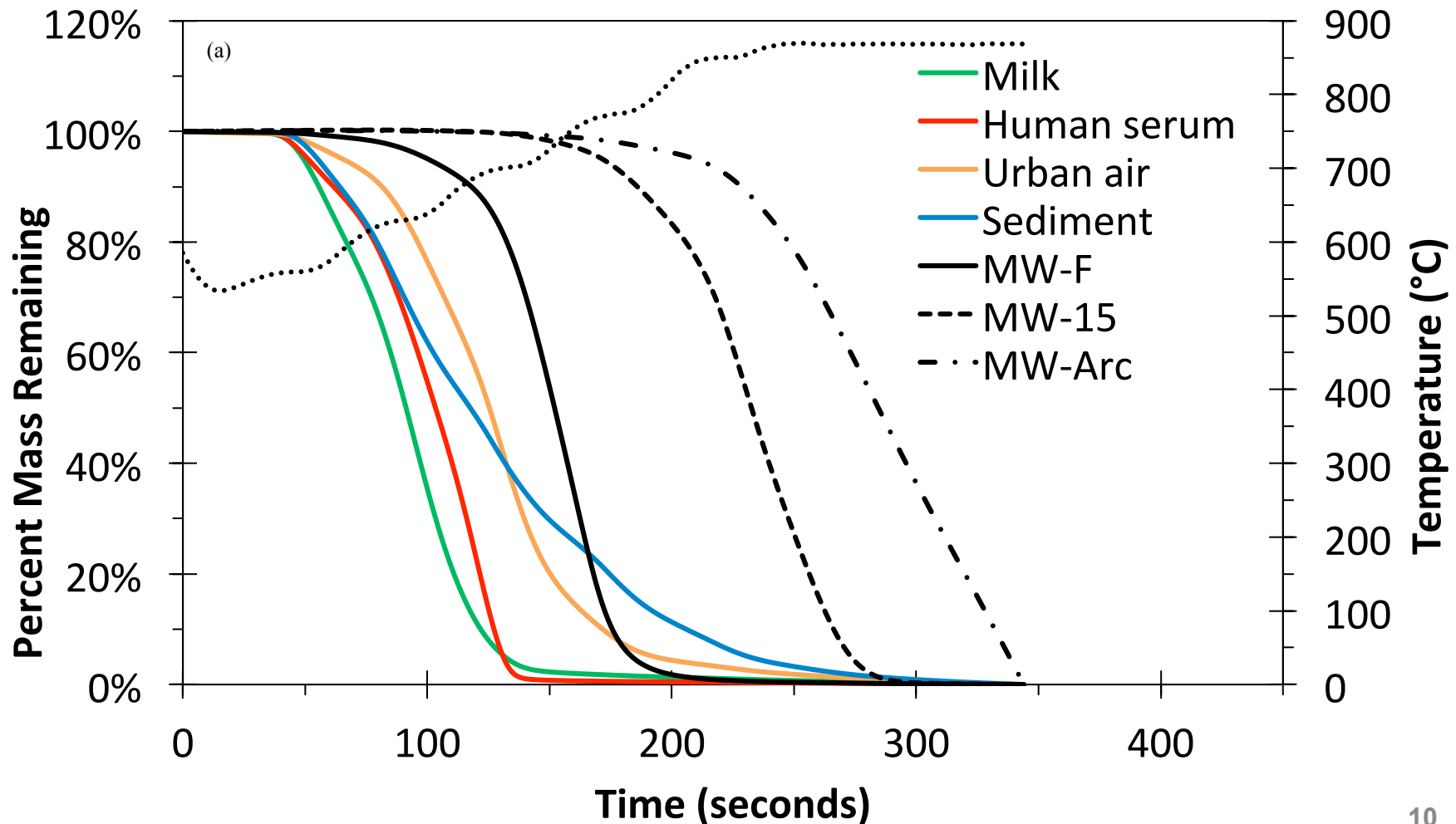


Raman spectrum of CNTs extracted from Cyanobacteria



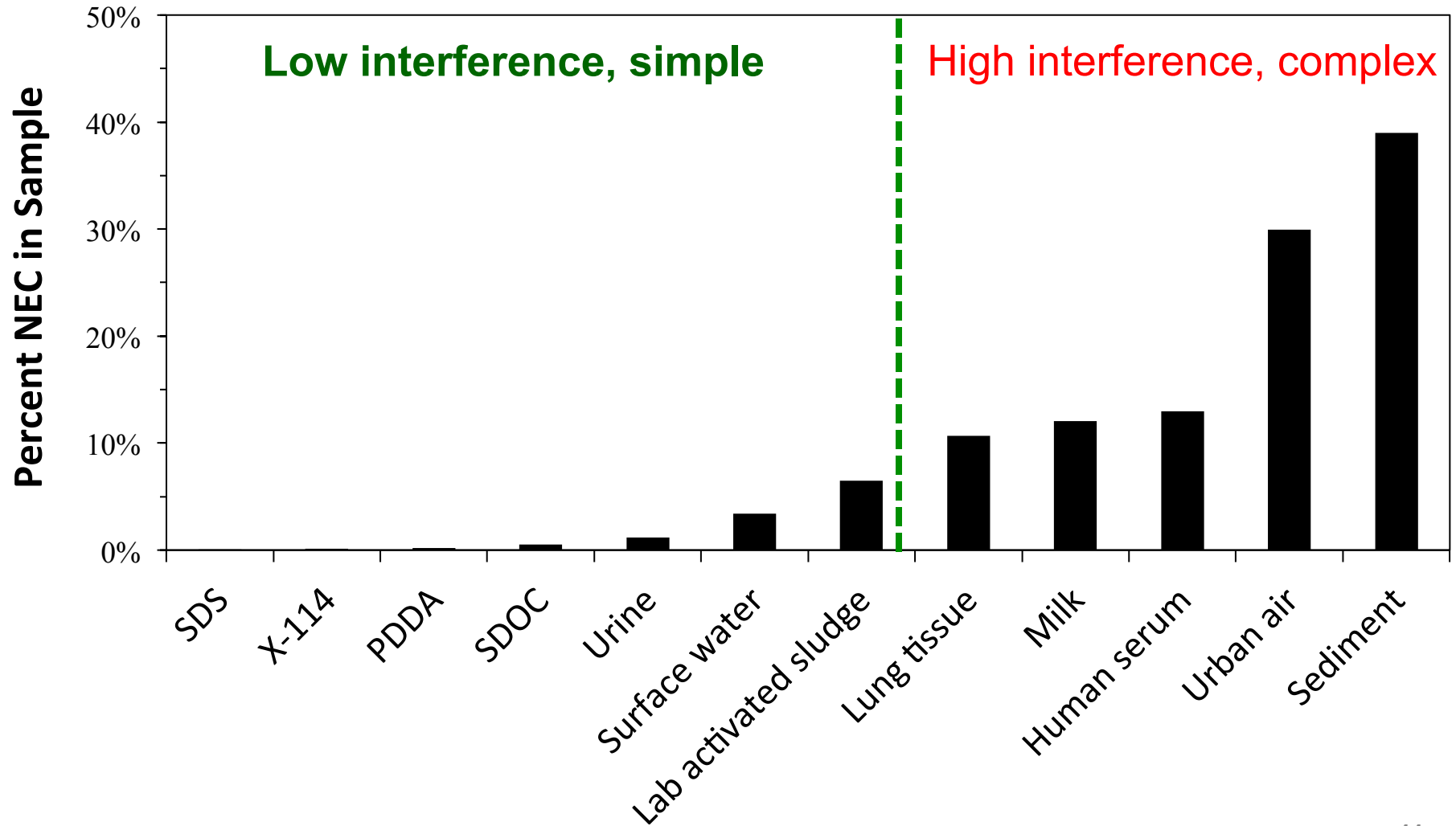


Organic materials produce non-elemental carbon (NEC) that causes interference where CNTs respond



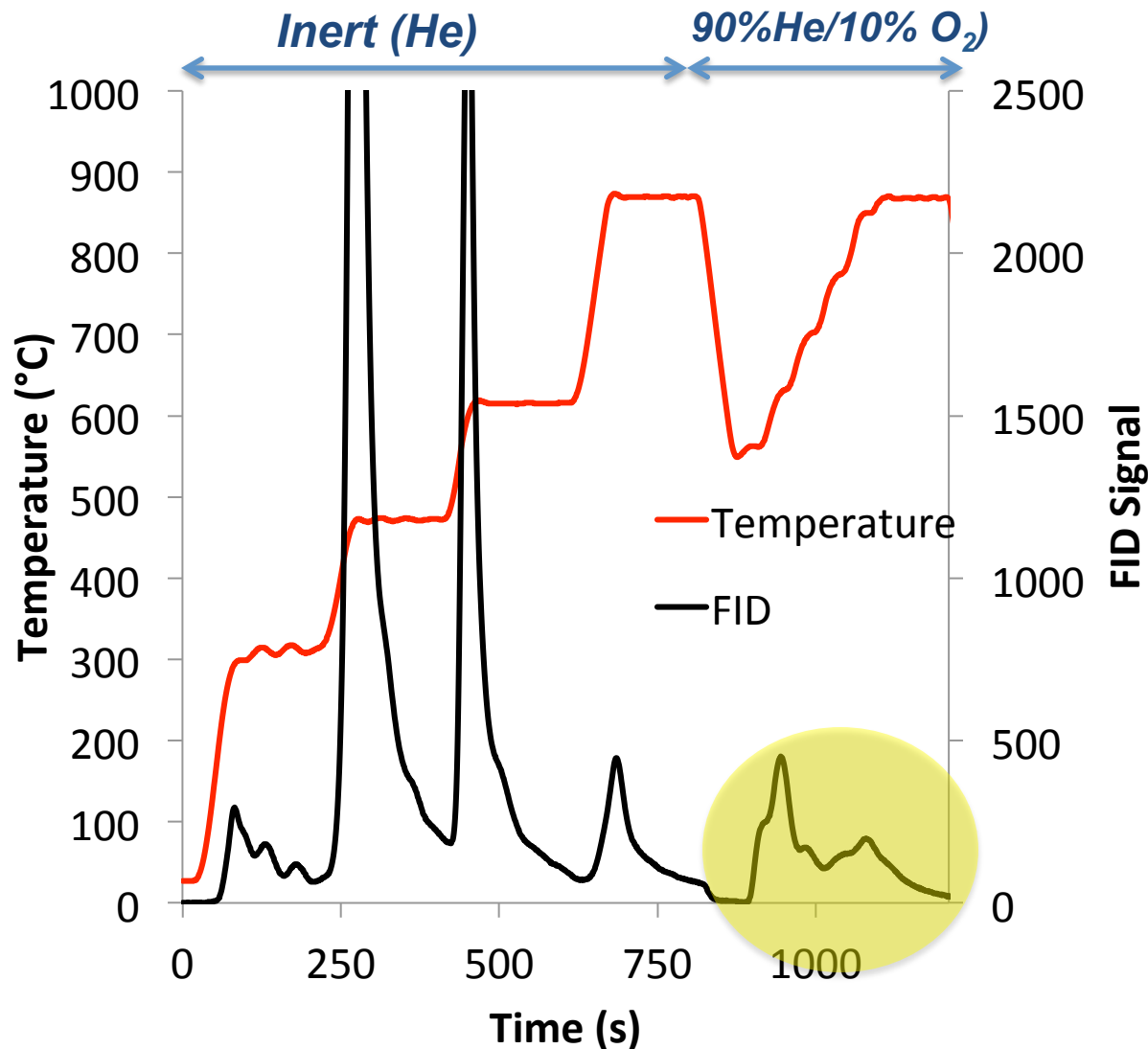


Thermal Behavior of Various Environmental and Biological Matrices





Tissue Sample Example: Lung Tissue



Even a small amount of lung tissue (a few mg) cause significant charring (EC) response



Challenge with Thermograms

Organic material in tissue and environmental samples “char” and produce *responses* which could be interpreted as CNTs

Critical: Determine temperature dependency of various biological and environmental matrices to produce non-elemental carbon (NEC) as a function of the total carbon in the sample



Challenge: How to digest away organic matrix without affecting CNT properties

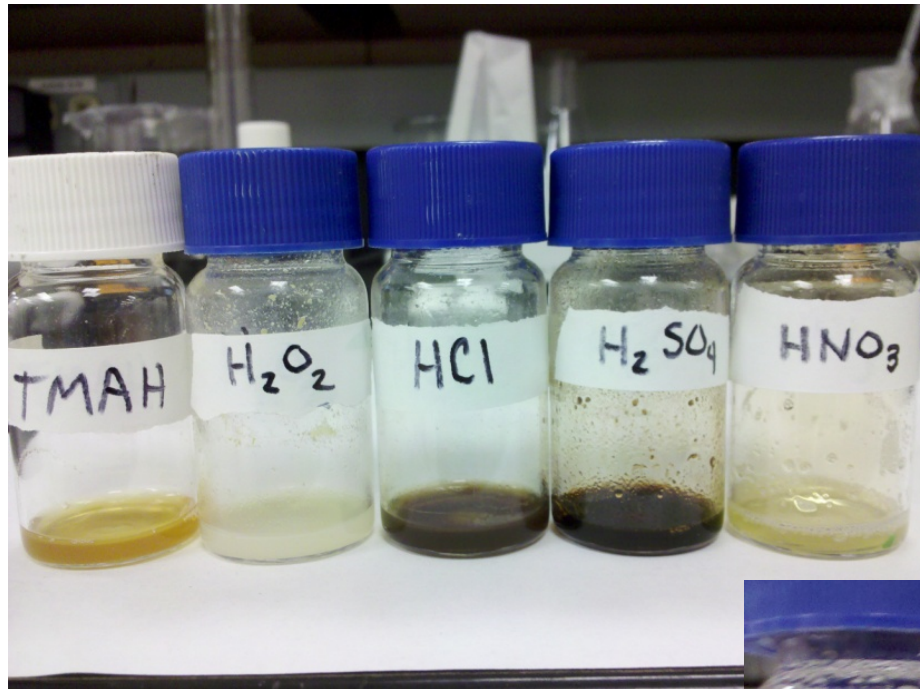
Approach

We investigated a variety of
digestates using rat lung tissue

***Experiments used Mitsui CNTs (CVD method)
(Strong CNT classification)***

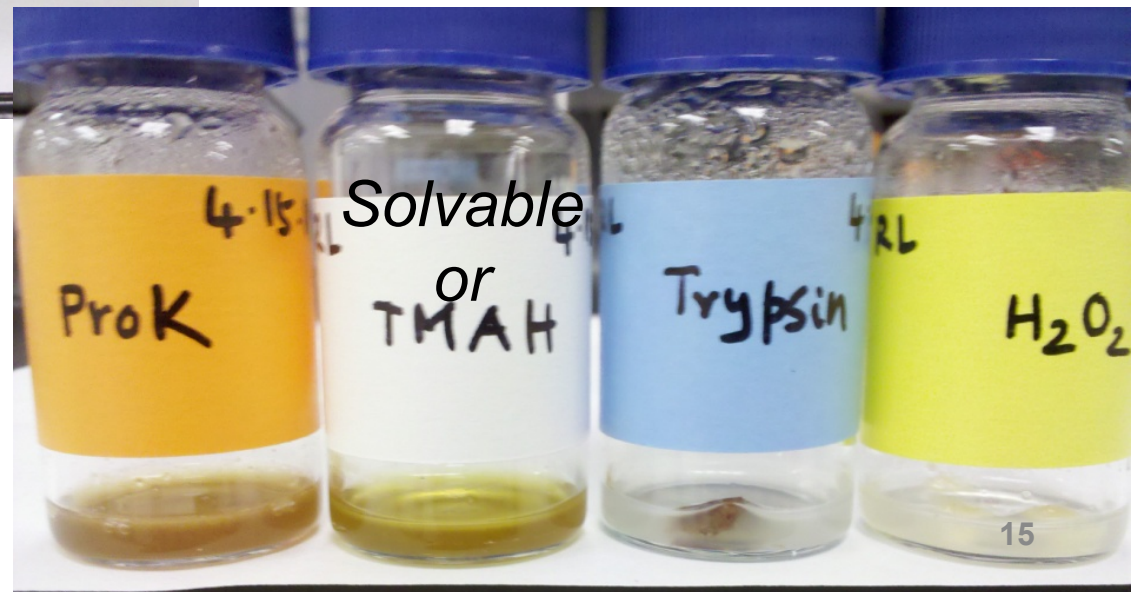


Comparison of Different Chemical Digestion Methods



**Color of rat lungs
after chemical
digestion**

**Black color results in
highest NEC response**





CNT Extraction from Rat Lung Tissue

Matrix	<u>Acids</u>		<u>Acid/Oxidants</u>		<u>Oxidants^b</u>		<u>Hydrolysis</u>		<u>Enzymatic</u>
	HCl	HF	HNO ₃ ^a	H ₂ SO ₄	H ₂ O ₂	NaOCl	TMAH ^c	Solvable ^d	Pro K ^e
Rat lung			✓	✓	✓	✓	✓	✓	✓
CNT	✓	✓	✓				✓	✓	✓

Check mark:

- ✓ **When rat lung tissue was successfully dissolved from solid to liquid** (for solid-liquid separation of CNTs)
- ✓ **When CNTs were not mineralized**; Subsequent TGA analysis showed many acids also lead to surface oxidation of CNTs which can change their thermal profile

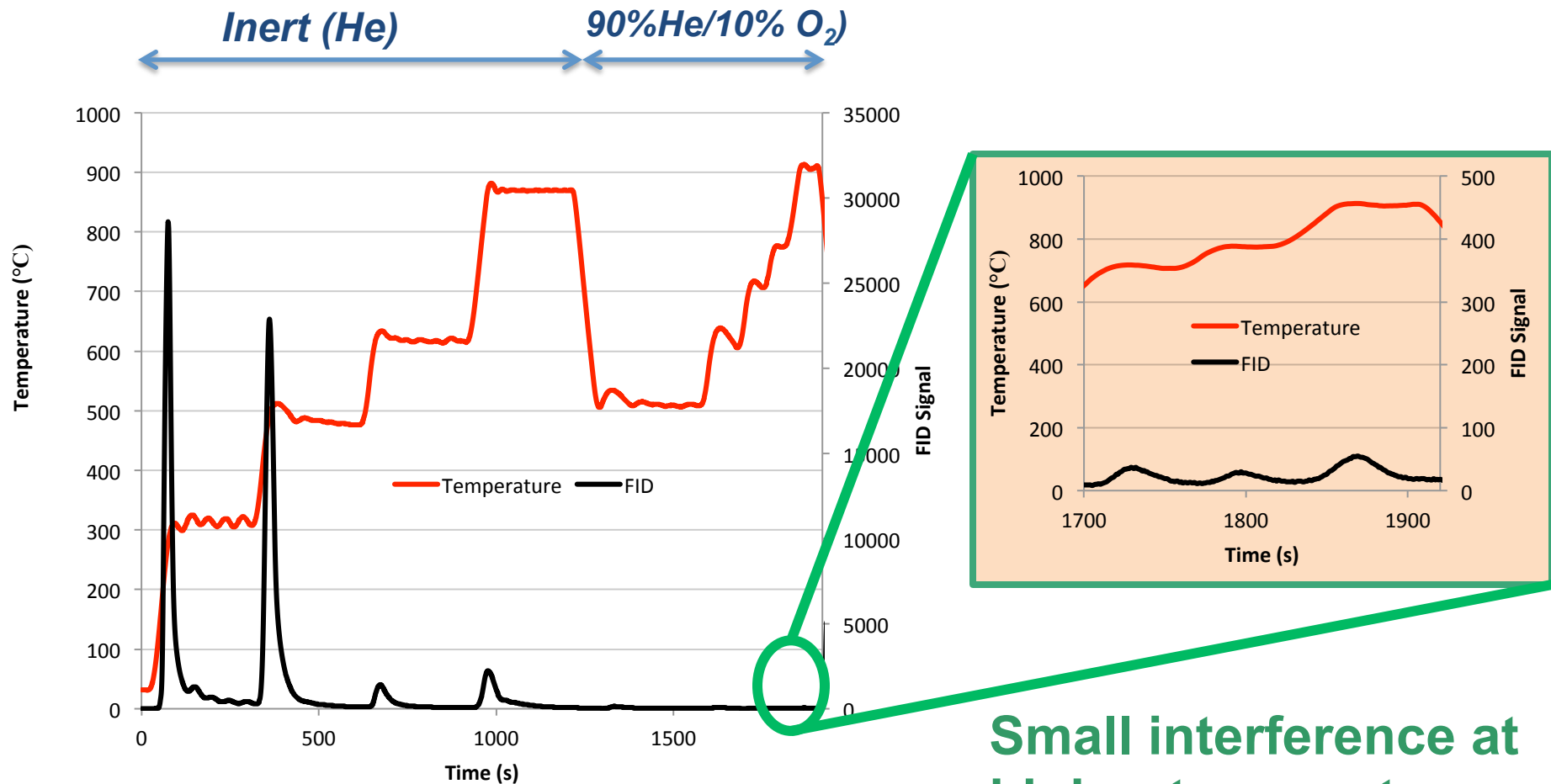


Highlights of previous slide (Lessons Learned)

- Acids and oxidants are messy, hazardous to the analytical instrument, and some mineralize CNTs.
- Not all hydroxides are equal: TMAH and Solvable worked best, simple salt forms were not ideal (e.g., NaOH, KOH).
- A two-step digestion process involving Solvable (or TMAH) and Pro K is necessary to remove high-temperature stable carbon.
- Filtration is not an option due to sticking; centrifugation should be done at a minimum of 25,000 Gs for 20 min.

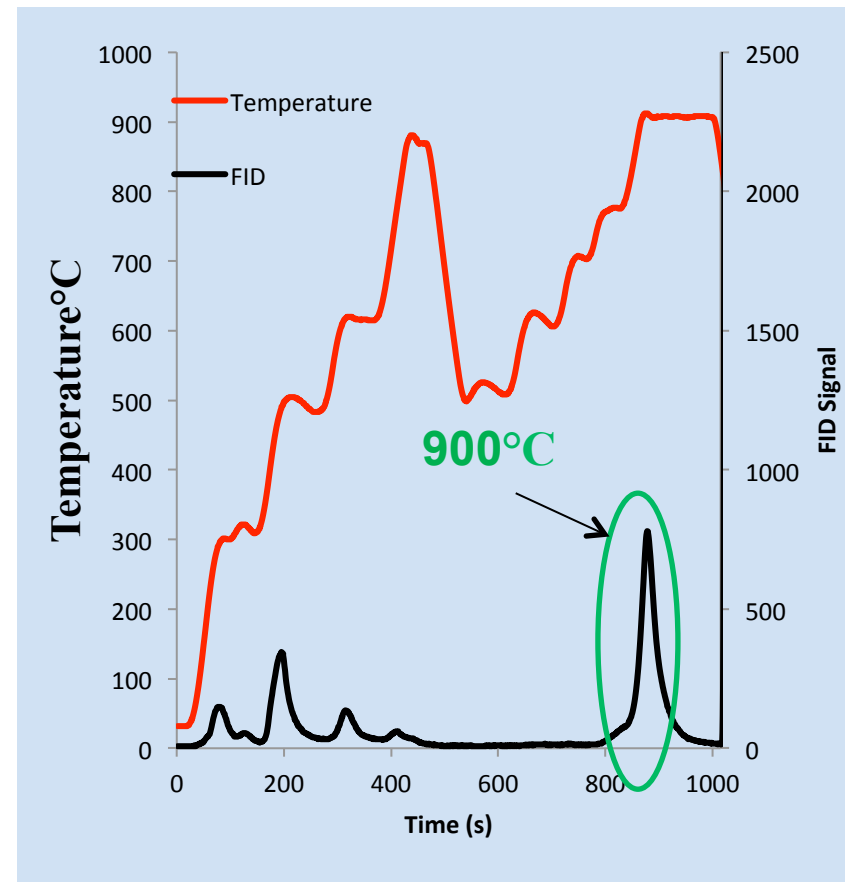
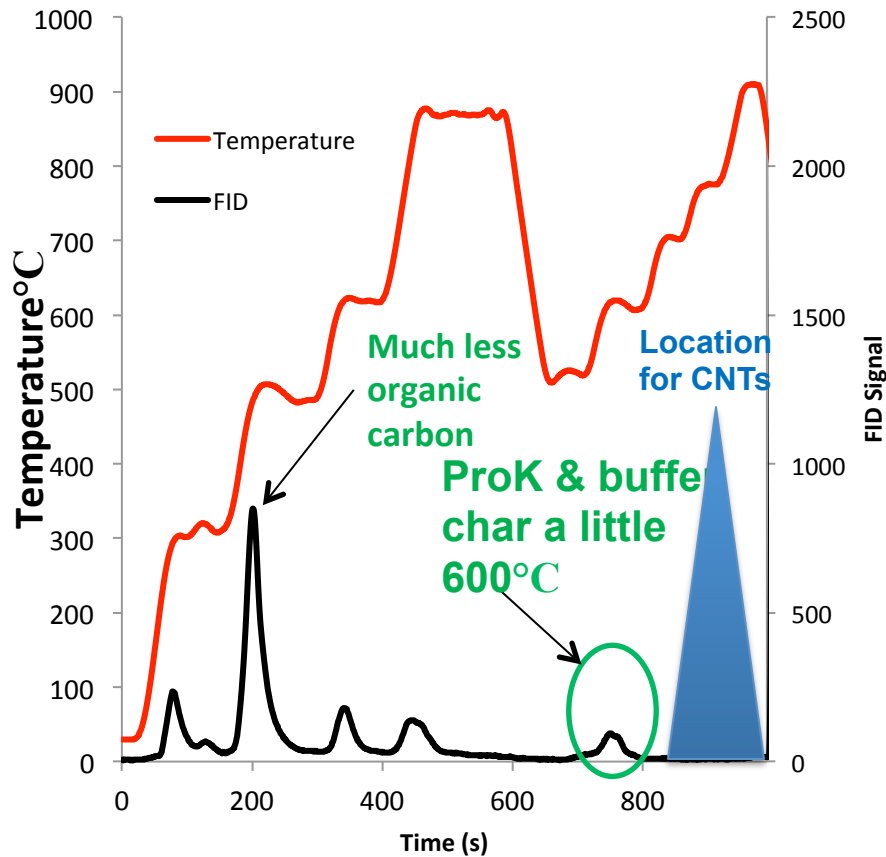


Thermograms of Whole Rat Lungs digested using Solvable (alone)



Small interference at higher temperatures where CNTs respond

Rat Lung Digested with Solvable + ProK reduces interferences more than Solvable Alone



Solvable and Pro K reduce charred OC & EC; EC does not interfere with CNTs

CNT Control – treated with Solvable and Pro K¹⁹

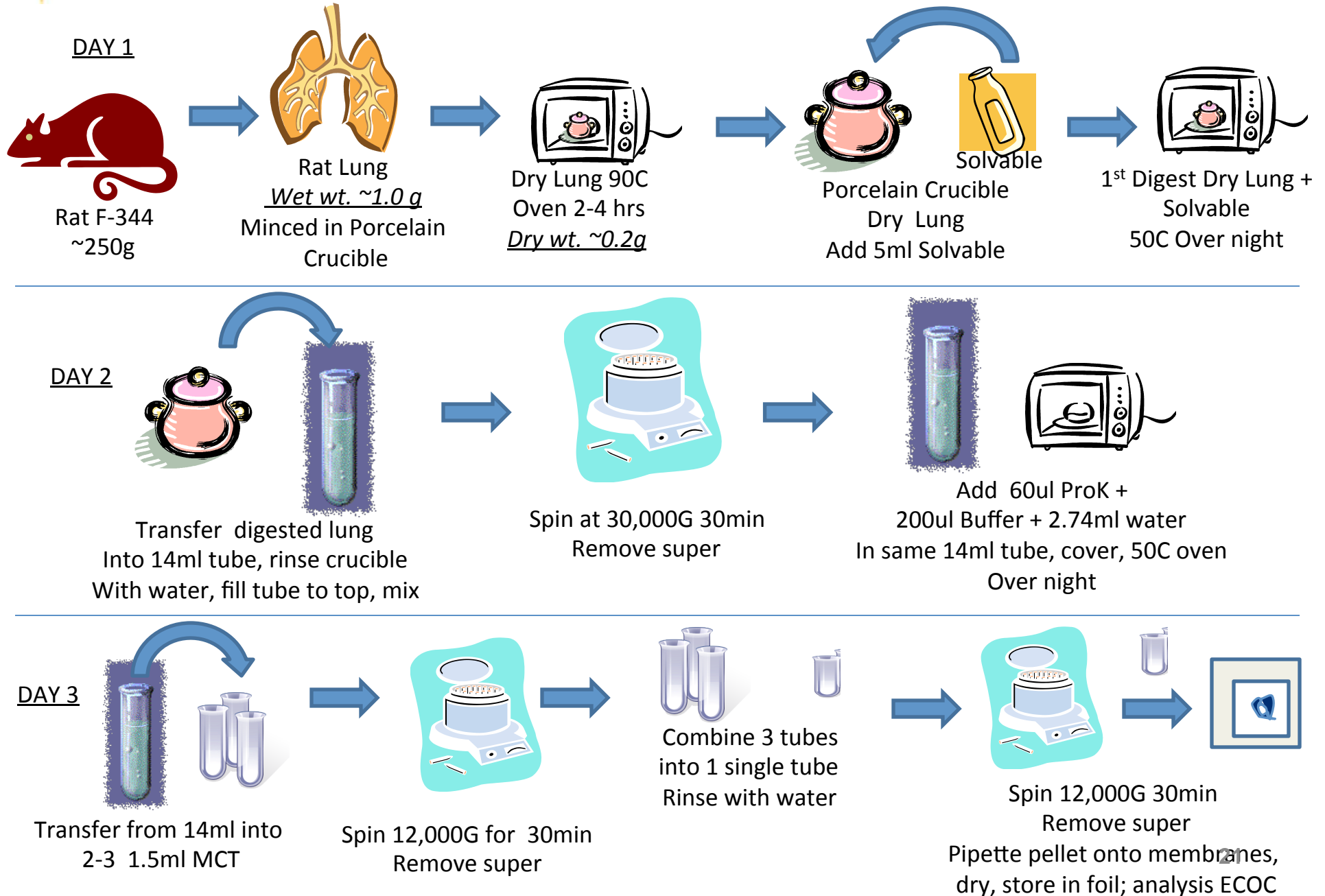


Separation of CNTs from digested lung tissue

- **Need: a method to separate the CNTs from the digested lung liquid (5-15 mL)**
- **Centrifugation or filtration ?**
- Since we load samples onto quartz fiber filters (QFF) for TGA analysis we tried this first. Unfortunately, some sample passed through the filter because it was dispersed.
- Inducing aggregation of CNTs with salt allowed GFF to capture CNTs but they also stuck to the syringe and the filter housing.
- Centrifugation: Speeds below 25,000 Gs and times less than ~20 minutes did not produce good pellets, and some CNTs were left in suspension in the centrate. **Washing with 1 M KCl in place of water enhanced the pellet stability.**

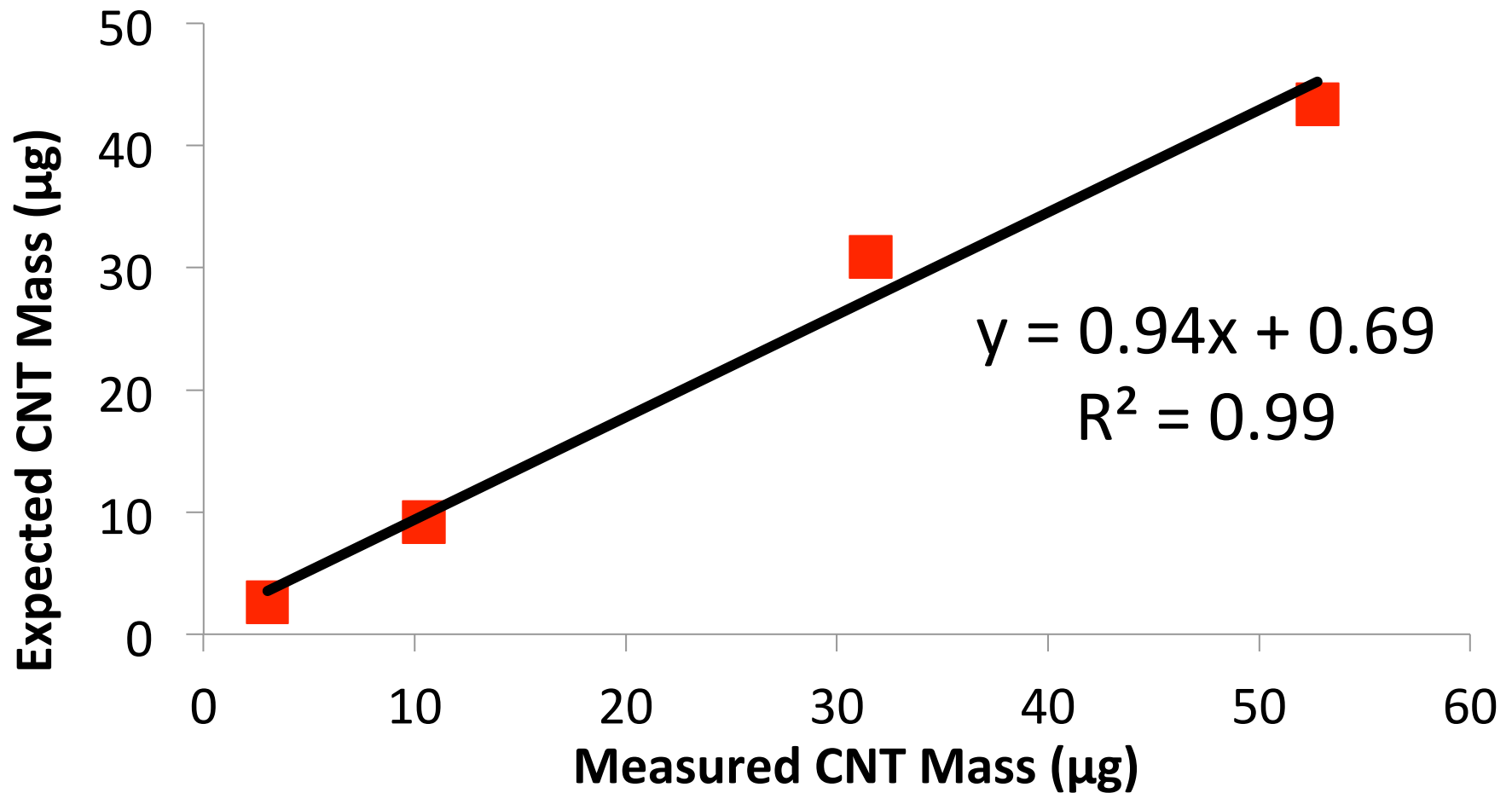


Optimized CNT Extraction from Rat Lungs





CNT Mass – Spike Recovery on Controls: CNTs treated with Solvable + Pro-K)





CNT Mass Recovery with Instilled Rat Lungs

- Rat lungs instilled with 3 μg of CNTs
- Digested using solvable + Pro-K and separated using previous protocol
- Residual material analyzed using programmable thermal analysis
- Controls (no lungs) for 3 μg of CNTs
 - Recovery: 3.30 $\mu\text{g} \pm 0.09 \mu\text{g}$
- Instilled rat lungs for 3 μg of CNTs
 - Recovery: 2.80 $\mu\text{g} \pm 0.44 \mu\text{g}$
 - Average recovery: 85%
 - Recovery was consistent
 - Losses attributed to extraction processes – CNTs lost during centrifugation & washing



CNT Summary

- Two distinct thermal classes of CNTs
 - Weak as described by an I_D/I_G ratio less than 0.75
 - Includes most MWCNTs and SWCNTs
 - Strong as described by an I_D/I_G ratio greater than 0.75
 - Includes MWCNTs with low defect density
- Hydrolysis and enzymatic digestion methods are most appropriate for CNTs and rat lung tissue



Future Work

- A robust CNT analytical method was developed
- Semi-conductor industry uses CNTs
 - Thermal packaging
 - Electrical conducting properties
- We plan to evaluate:
 - Workplace exposure (air) to CNTs
 - Release during mechanical wear of CNTs from thermal packaging



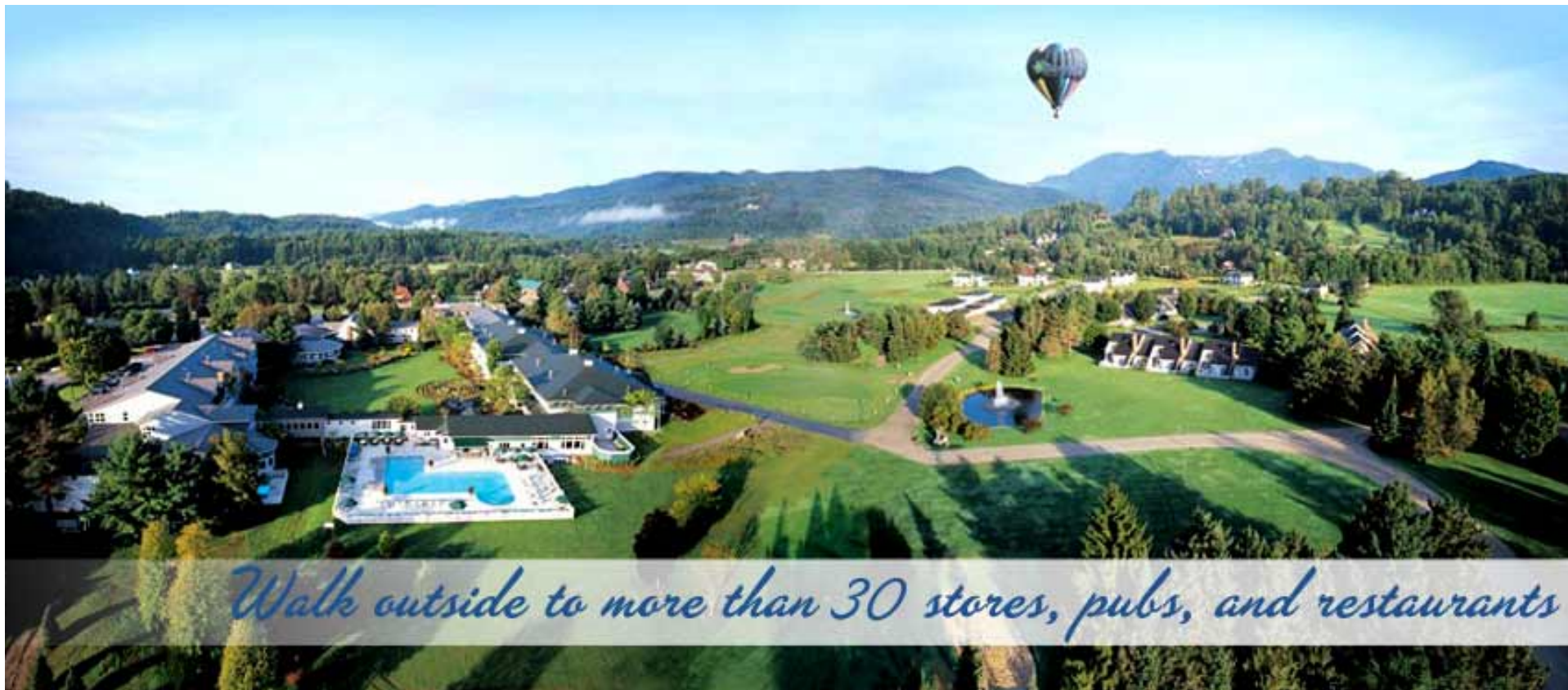
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