

SRC Engineering Research Center for Environmentally Benign Semiconductor Manufacturing

Ceria Nanoparticles: Environmental Impacts on Particle Properties and Potential Effects on Biological Systems

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S. Seal

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Teleseminar Aug 8, 2013

EMSL is located at PNNL





Nanomaterials can help solve many important problems





http://www.nisenet.org/catalog/programs/nanotech_consumer_products ,



Tiny traits cause big headaches Nanotech medicines held up by lack of particle characterization. Daniel Cressey

"Characterization is the biggest challenge to this field," Simon Holland, GlaxoSmithKline

"Everybody accepts that as an academic community we haven't been characterizing enough," Kenneth Dawson, University College Dublin

http://www.nature.com/news/2010/100914/full/467264b.html

Nanoparticles designed for medical application





- "The problem with determining atomic structure at the nanoscale"
 - S. J. L. Billinge and I. Levin, Science (2007).
- "Common pitfalls in nanotechnology...,"
 - R. M. Crist, et al. Integr. Biol. (2013).
- "The characterization bottleneck,"
 - E. K. Richman and J. E. Hutchison, ACS Nano (2009).
- "Discriminating the states of matter in metallic nanoparticle transformations: What are we missing?"
 - J. M. Pettibone, et al. ACS Nano (2013)
- "Tiny traits cause big headaches...".
 - D. Cressey, Nature (2010)



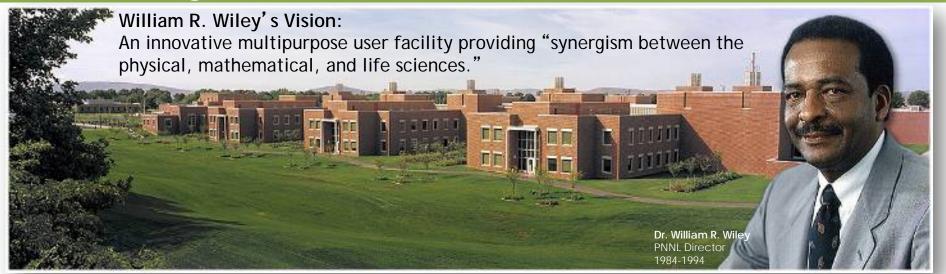
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What is the problem?



Environmental Molecular Sciences Laboratory

EMSL



EMSL, a national scientific user facility at the Pacific Northwest National Laboratory, provides *integrated experimental and computational resources* for discovery and technological innovation in the environmental molecular sciences to support the needs of DOE and the nation. Focus on the environment and energy

What are the issues? What we have learned from many studies



- Focused beneficial and potential negative impacts of three nanoparticles
 - Iron metal-core oxide-shell particles to remove environmental contaminants in ground water
 - Ceria nanoparticles which have a wide variety of uses
 - **Silver** nanoparticles in wide use in consumer products
- Common characteristics that complicate understanding and characterization
- Ceria as an example
 - **Conflicting** Biological impacts
 - Environmentally induced changes in chemical state
 - Dynamic, not static

7

Synthesis and delivery challenges

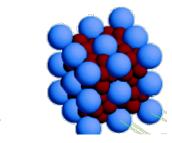




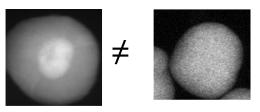
Characteristics that complicate understanding and characterization



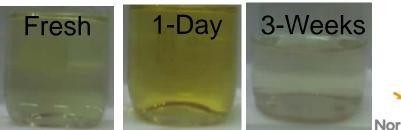
Most atoms are near a surface or interface



Nanoparticles are not created equal



Particles often change as a function of time Particles are dynamic – not static



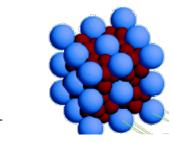




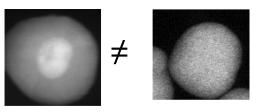
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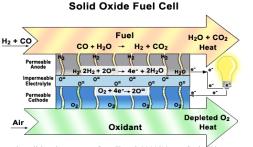


An example: Cerium oxide (Ceria, CeO₂) – contradictory behaviors



Solid Oxide Fuel Cells

(Ce 3+ and Ce 4+)



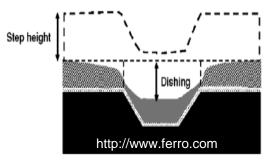
http://ciencia.nasa.gov/headlines/y2003/18mar_fuelcell.htm

Catalysis (Ce3+ and Ce4+)



Microelectronics (CMP)

(Ce3+)



Particles

Films or powder pellets Particles Bio-medical Applications

(Ce3+ and Ce4+) Particles UV-protection Cancer treatment



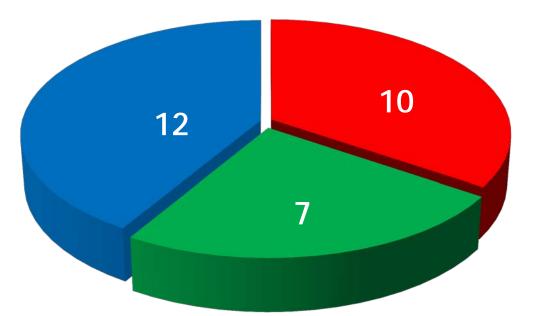


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Contradictions in biological impact: Cerium oxide nanoparticles – Antioxidative or oxidative? EMSL

Oxidative Neutral Antioxidative



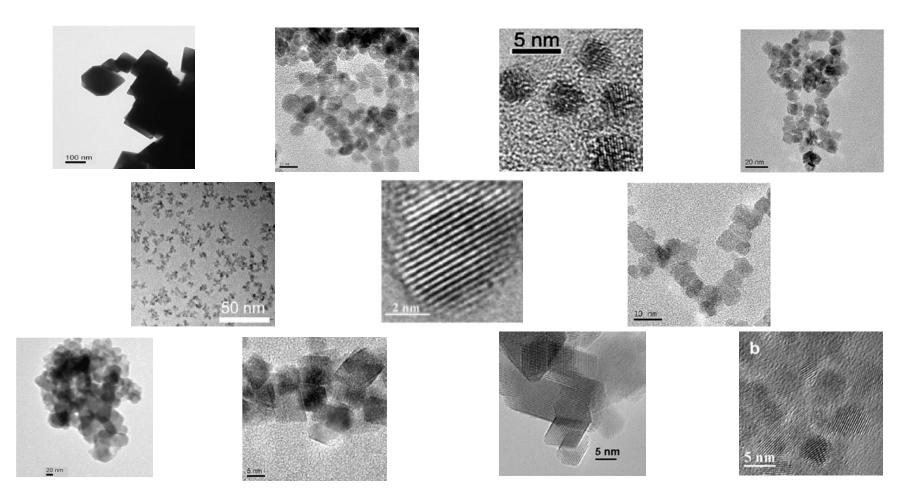
Scattered observations in literature on biological effects of cerium oxide nanoparticles (Dec 2010)





Cerium oxide nanoparticles – What is being tested





Huge variation in size, shape, crystalline state and agglomeration

Karakoti et al Surface and Interface Analysis - 2012





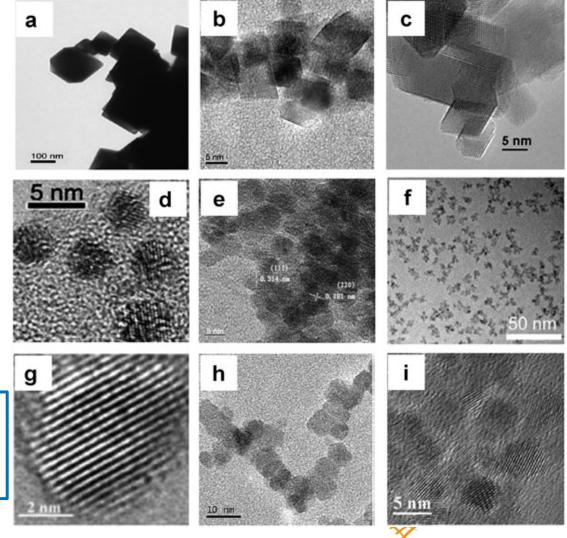
Ceria synthesis involved different tempertures



Heated or calcined at high temperature (>300°C)

Heated (<100°C) in solution (with or without surfactants)

Room temperature synthesis (with or without surfactants)



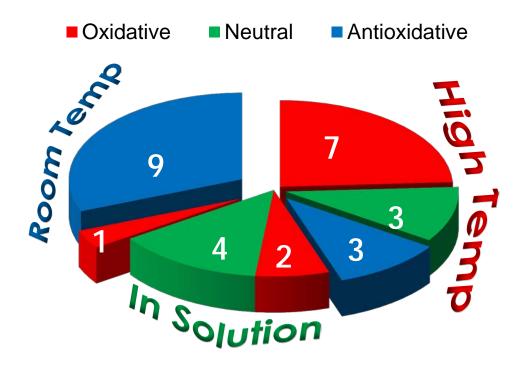
¹³Karakoti et al Surface and Interface Analysis - 2012

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Cerium oxide nanoparticles – Biological impacts based on synthesis





- Nanoparticles are not created equal
- Lack of control over properties of engineered nanomaterials may lead to varying biological responses
- Only one of the possible reasons not the only possible reason

Karakoti et al Surface and Interface Analysis - 2012





Take home messages



- Synthesis methods, processing conditions and history of nanomaterials have an important impacts on their properties
- Nanomaterials are like chameleons
 - Nanomaterials are not static entity and often change with conditions
 - Impacts shelf life and product stability
 - Important for understanding behavior in the environment and biological impacts







Inconsistency chemical and physical properties



- Contradictory reports of biological impacts of ceria nanoparticles.
 - Many good anti-oxidant effects
 - Some toxic behaviors reported
- Inconsistent report of physical and chemical Properties
 - Data reported measuring quantum confinement
 - Seen in different size ranges (or not seen) by different groups

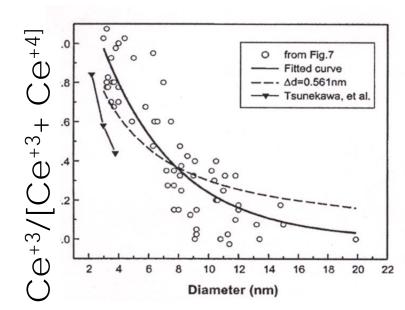


Inconsistency chemical and physical properties



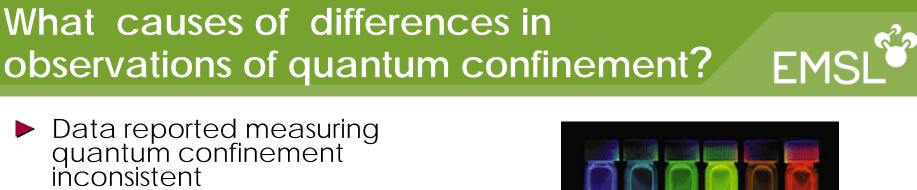
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- Switching of chemical state important. Can that explain property variations?

Size and chemical state

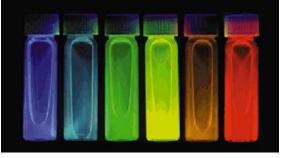


Wu et al. Phys. Rev B 69, 125415, 2004



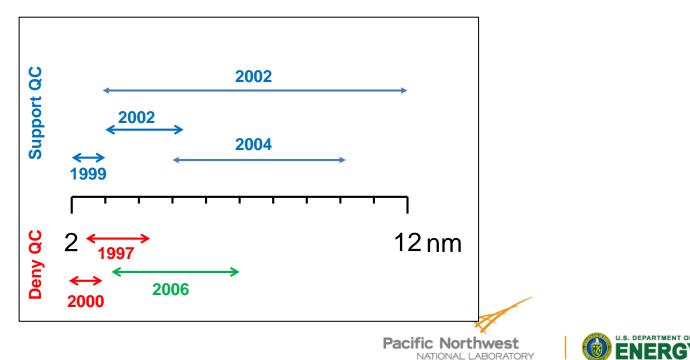


 Seen in different size ranges (or not seen) by different groups



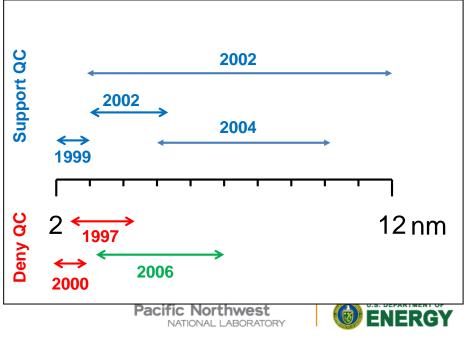
2 nm

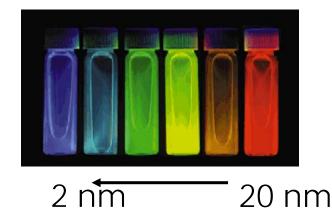
20 nm



What causes of differences in observations of quantum confinement?

- Data reported measuring quantum confinement inconsistent
 - Seen in different size ranges (or not seen) by different groups
- Quantum confinement identified by measurements of band gap.
- Can we reproduce measurement of changes in band gap as function of size
- Might the BG differences observed be due to chemical state effects rather than quantum confinement?



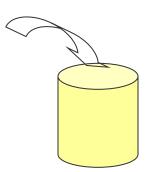


EMS

Formation of Ceria Nanoparticles to study of size dependent oxidation state



 $Ce^{3+} + OH^{-} + \frac{1}{2} H_2O_2 \rightarrow Ce(OH)_2^{2+}$ $Ce(OH)_2^{2+} + 2 OH^{-} \rightarrow Ce(OH)_4 \rightarrow CeO_2.2H_2O$



Particles form quickly when peroxide added salt solution

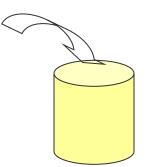




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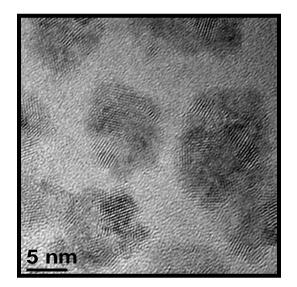


 $\begin{array}{l} \operatorname{Ce}^{3+} + \operatorname{OH}^{-} + \frac{1}{2} \operatorname{H}_{2}\operatorname{O}_{2} \xrightarrow{} \operatorname{Ce}(\operatorname{OH})_{2}^{2+} \\ \operatorname{Ce}(\operatorname{OH})_{2}^{2+} + 2 \operatorname{OH}^{-} \xrightarrow{} \operatorname{Ce}(\operatorname{OH})_{4} \xrightarrow{} \operatorname{CeO}_{2}.2\operatorname{H}_{2}\operatorname{O} \end{array}$



Particles form quickly when peroxide added salt solution

TEM of particles harvested within an hour show 3-5 nm particles in 15-20 nm agglomerates. Particles appear the same to TEM analysis for all conditions to follow



15-20nm agglomeratesNo specific morphology

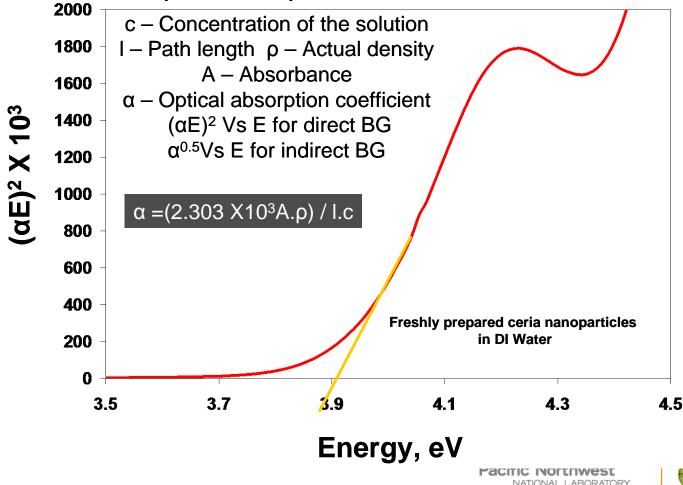
PACITIC NORTINWEST NATIONAL LABORATORY



Apparent band gap often determined by optical adsorption



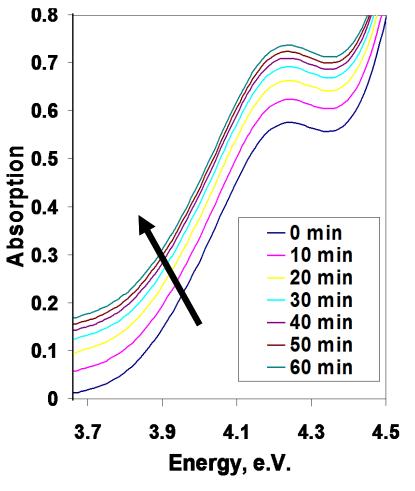
Band-gap value can be obtained from the energy intercept of the plot between αE^2 and E







Absorption Data Collected Soon After Nucleation Appears Consistent with Quantum Confinement

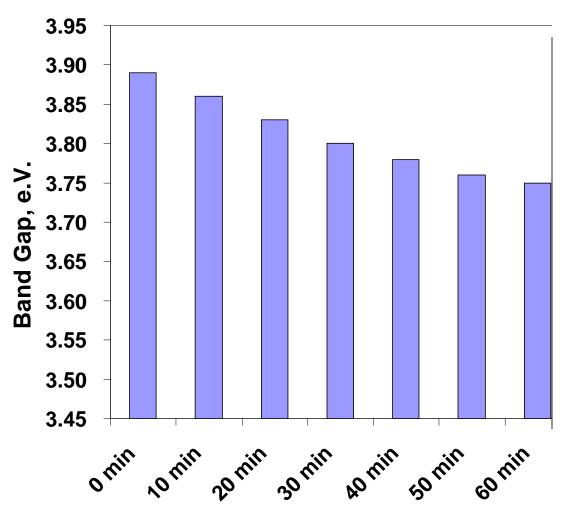






Band gap appears to shrink as particles grow Bohr radius ~7.0nm



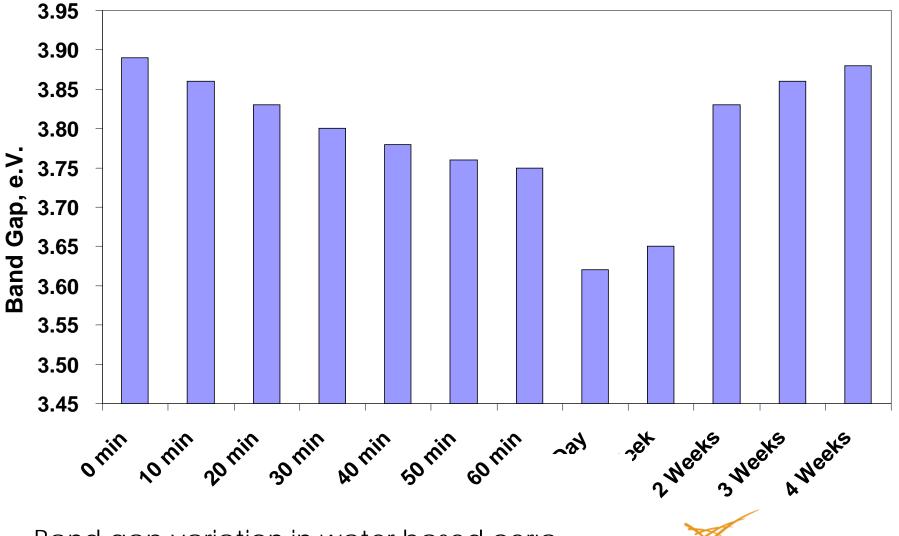


Band gap variation in water based ceria nanoparticles as a function of time





Band gap appears to shrink as particles grow Bohr radius ~7.0nm



Band gap variation in water based ceria nanoparticles as a function of time

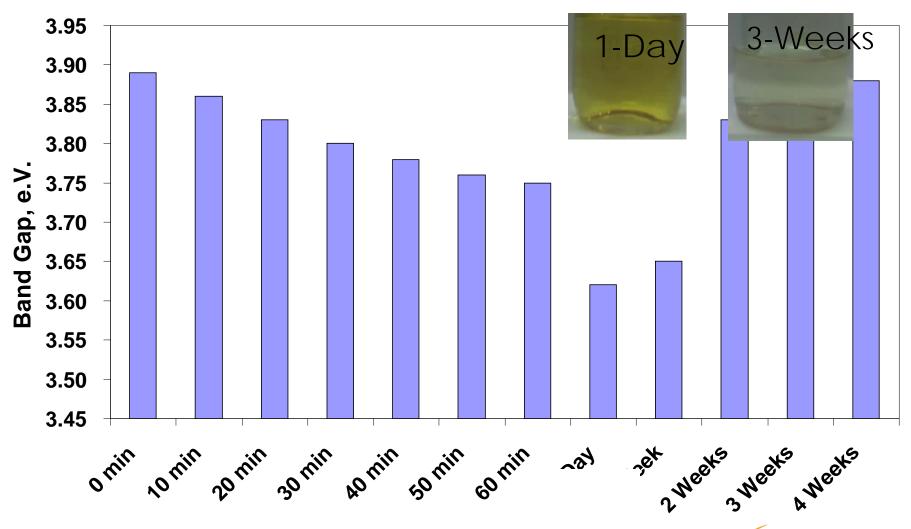
Pacific Northwest



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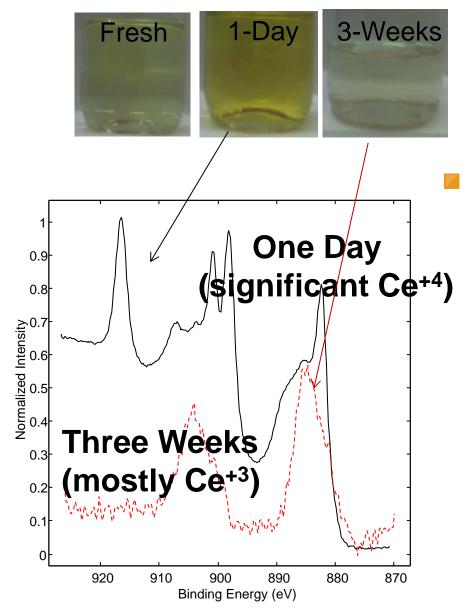


Band gap variation in water based ceria nanoparticles as a function of time

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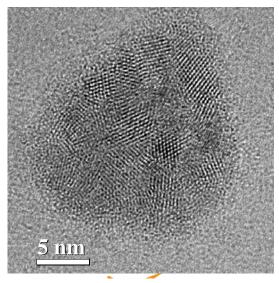


Ceria particles can change chemical state: Time and Environment factors



TEM data at 1 day and 21 days shows 10-20 nm agglomerates made up of 3 to 5 nm particles for most conditions

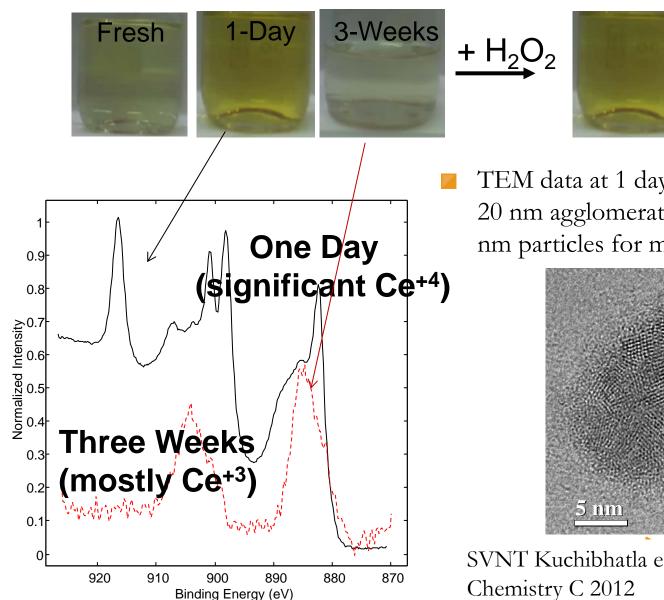
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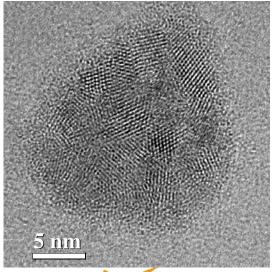
SVNT Kuchibhatla et al. Journal of Physical Chemistry C 2012

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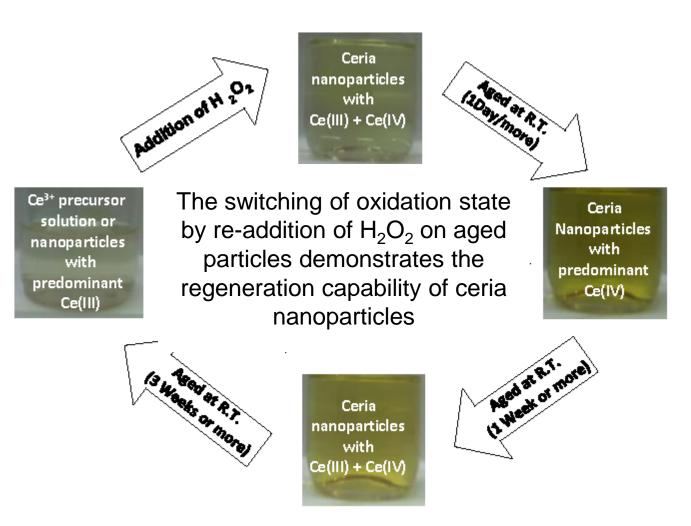
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SVNT Kuchibhatla et al. Journal of Physical Chemistry C 2012

Oxidation reduction cycle for one type of ceria nanoparticle



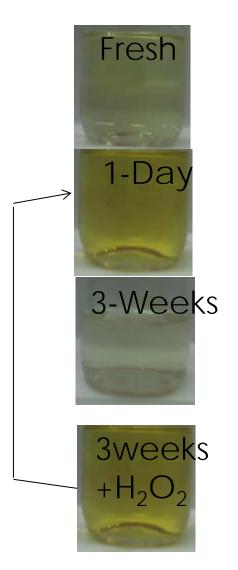


SVNT Kuchibhatla et al. Journal of Physical Chemistry C 2012



What is happening to the particles in solution?







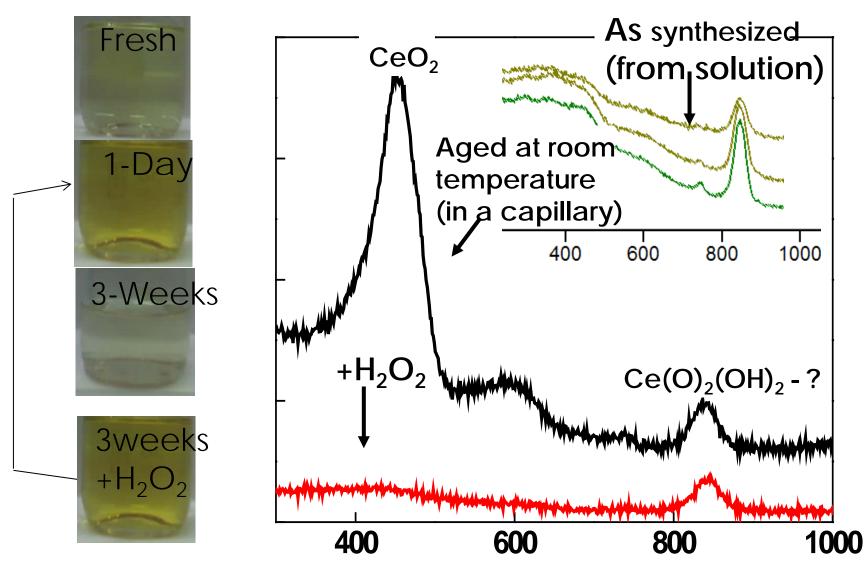


What is happening to the particles in solution?

31



Add Optical Method - In situ Raman



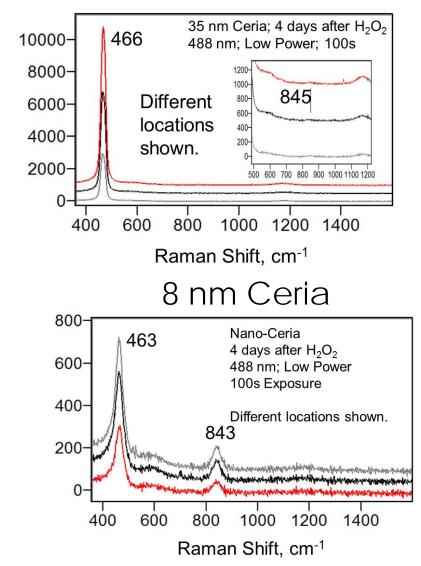
Raman Data: Ceria particles of different sizes



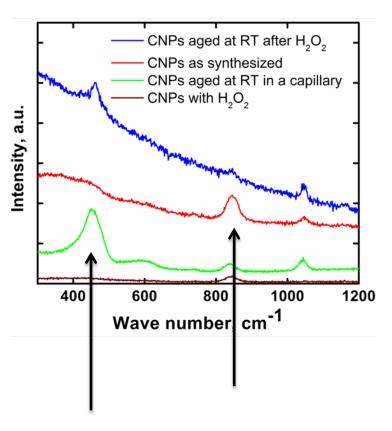
RTMENT OF

RGY

35 nm ceria



3 nm ceria



ceria Oxyhydroxide

Ceria nanoparticles can do the unexpected ! – environment again



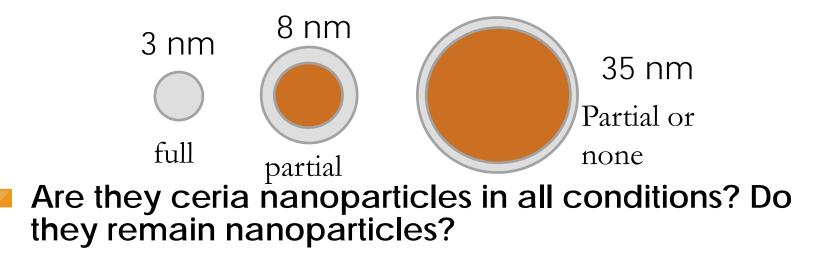
- Particles can change oxidation state in solution depending on the oxidizing potential of the solution
- TEM shows the particles to be similar in either chemical state (may be due to vacuum and/or probe damage)
- Dynamics light scattering sees particles in solution in either condition
- However XRD and laser Raman indicate that the small particles can fundamental switch structures between an oxyhydroxide (when Ce⁺⁴) and a defected ceria structure (when Ce⁺³)



Ceria nanoparticles can do the unexpected ! – environment again



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- Small particles switch completely; larger partially or not at all



What are the issues? What we have learned from many studies



- Focused beneficial and potential negative impacts of three nanoparticles
 - Iron metal-core oxide-shell particles to remove environmental contaminants in ground water
 - Ceria nanoparticles which have a wide variety of uses
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- Common characteristics that complicate understanding and characterization

Ceria as an example

- Conflicting Biological impacts
- Environmentally induced changes in chemical state
- Dynamic, not static
- Synthesis and delivery challenges Pacific

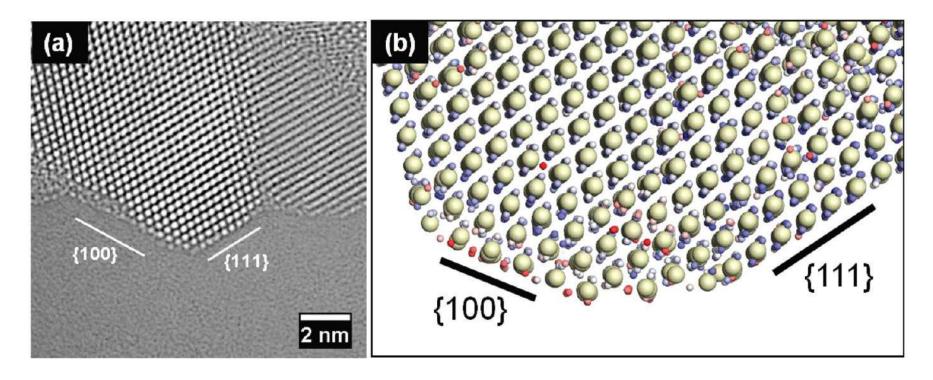




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Proudly Operated by Battelle Since 1965

We think of surfaces as static





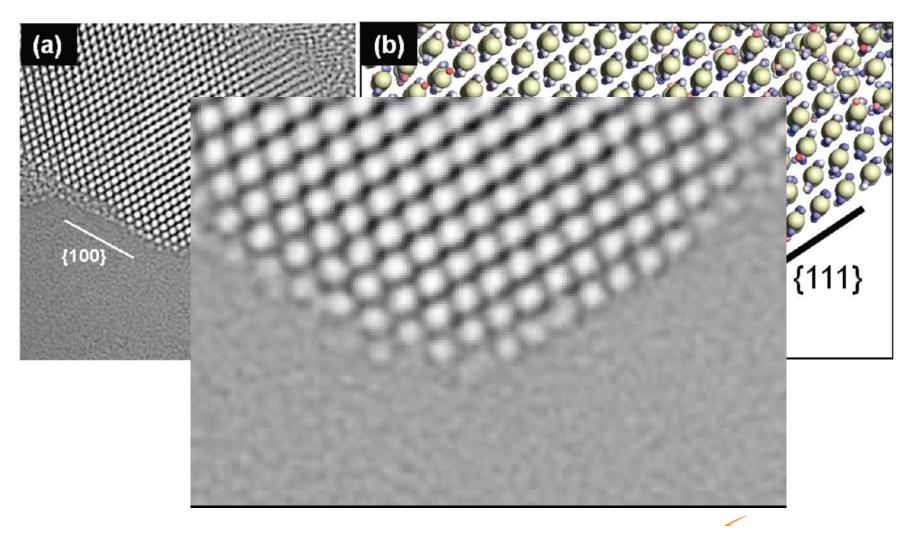
Electron microscopy observations of ceria nanoparticle

BHATTA ET AL. ACS Nano VOL. 6 ' NO. 1 ' 421–430 ' 2012 www.acsnano.org





We think of surfaces and objects as static: They are not! Think dynamics, motion and change EMSL



BHATTA ET AL. ACS Nano VOL. 6 ' NO. 1 ' 421–430 ' 2012 <u>www.acsnano.org</u> Can find movie on website as supplemental material



Challenges in Delivering Particles for Biological Studies



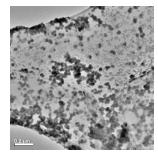
Precursors – Cerium nitrate hexahydrate (99% and 99.999% Sigma Aldrich) and hydrogen peroxide (30% w/v) (Sigma Aldrich)

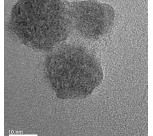
- Aged at room temperature in lab conditions
- TEM and UV-Visible spectrophotometry as a function of aging
- ➢Oxidation state changes from 4+ to 3+ over time

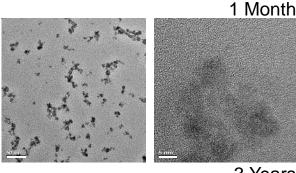
Time	Zeta potential (mV)	Size by DLS (primary) (nm)	рН
Fresh	35 ± 7	15-25	2.9
1 month	35 ± 7	30-45	3.5
2 months	35± 7	30-45	3.6
6 months	35±7	70-100	3.8

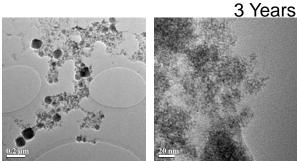
Summary of various samples

Particles relatively stable ~ year









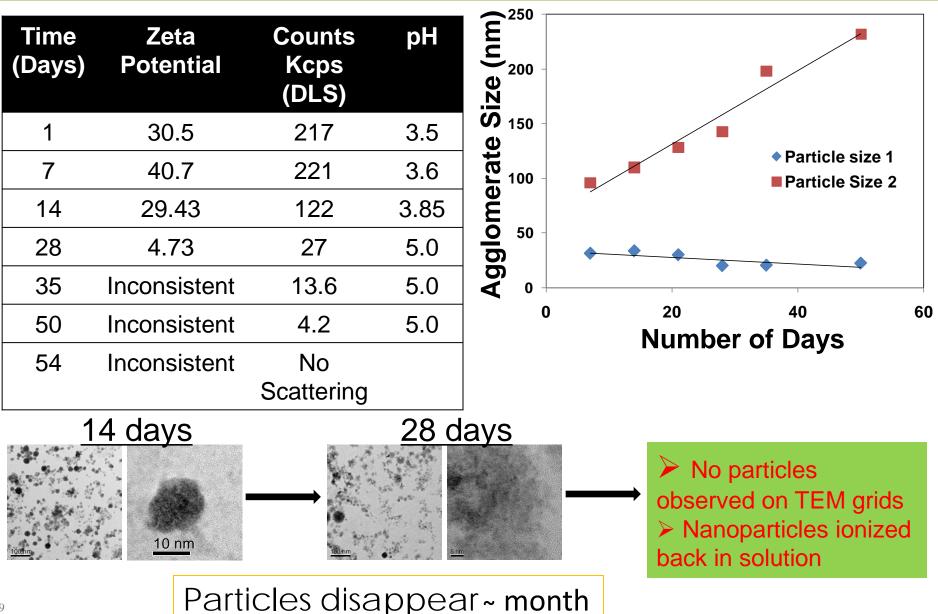
Samples sent back for TEM analysis by collaborator Prof. James McGinnis





Room Temperature Synthesis of Nanoceria – Repeated at EMSL

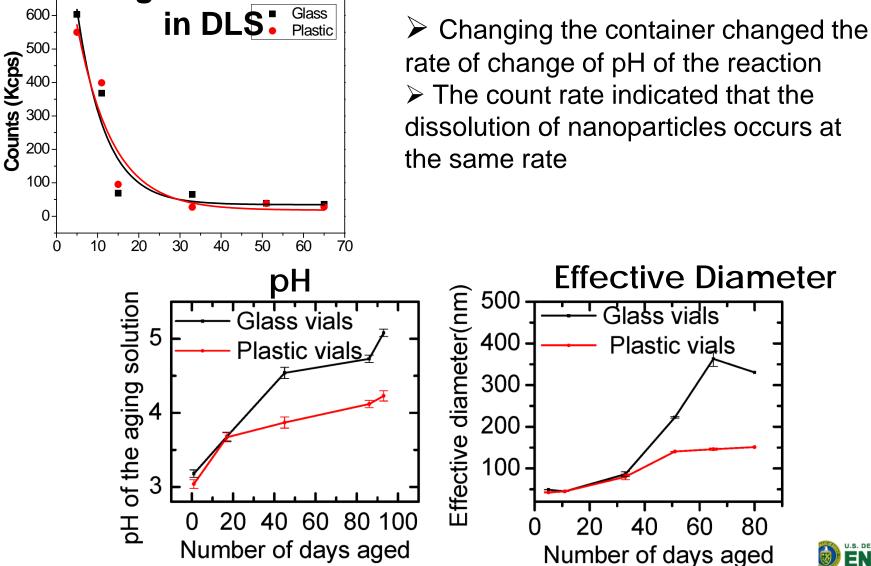




What is different – Glass and Plastic **Containers, Water, Chemical Batch**



Change in count rate



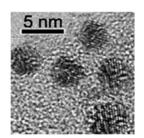


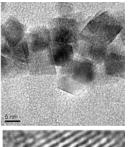
80

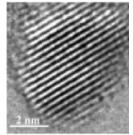
What have we learned about ceria nanoparticles?



- Biological outcomes can depend on synthesis method and processing history
- Chemical state of ceria nanoparticles is dependent on particle size, time and the environment
- XRD and laser Raman indicate that small particles are not always ceria in solution but can switch between an oxyhydroxide (when Ce⁺⁴) and ceria structure (when Ce⁺³)
- Only partial or no changes occurring for larger particles
- The same people using the same process can get different results when "minor" variable are changed





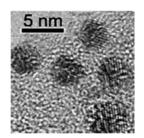


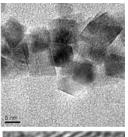


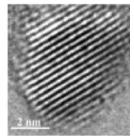
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GY

Time and Environment Matter - Synthesis differences and **time/environmental changes** can impact biological and environmental studies - need to be recorded and reported

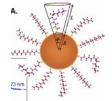
There are more general lessons

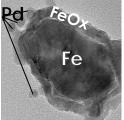
- EMSL
- Nano "object" characterization is more important, more challenging, more interesting than expected
- Synthesis differences and time/environmental changes can impact technological, biological and environmental studies, shelf life, product stability

Challenges –

- Are the particles really what we think they are?
 - How much and what types of characterization is needed?
 - Is contamination present on the surface?
- History makes a difference?
 - How have they been made and processed?
 - What is the shelf life? How long can they be stored?
 - How fast do they change in the working environment?
- Particles are dynamic! How quickly do they transform/change in the environment of interest?







What we have learned?

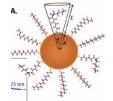


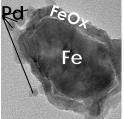
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- Particles are dynamic! How quickly do they transform/change in the environment of interest?
- Changes are an opportunity as well as a challenge







Acknowledgements



- Research was performed at the W.R. Wiley Environmental Molecular Sciences Laboratory (EMSL), a national scientific user facility at the Pacific Northwest National Laboratory (PNNL) sponsored by the USDOE-BER (www.EMSL.PNNL.gov)
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 - University of Central Florida, Sudipta Seal
- National and International Participants from ISO Committee TC201 and ASTM Committee E42